

Appendix B

Quality Assurance Project Plan Addendum

Supplemental Soil Characterization Work Plan

Appendix B

Quality Assurance Project Plan Addendum

Pines Area of Investigation

AOC II

Docket No. V-W-'04-C-784

Revision 3

May 2015

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QAPP Approvals

**Quality Assurance Project Plan Addendum
Supplemental Soil Characterization Work Plan
Pines Area of Investigation
Revision 3**

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Acronyms

ALS	The ALS Group
AOC II	Administrative Order on Consent, 2004; Docket No. V-W-'04-C-784
bgs	below ground surface
°C	degrees Celsius
CAS	Chemical Abstracts Service
CCB	Coal Combustion By-product
CCBK	Continuing Calibration Blank
COC	Constituent of Concern
DOE	Department of Energy
DQL	Data Quality Level
EM	Electron Multiplier
EML	Environmental Measurements Laboratory
FS	Feasibility Study
FWHM	Full Width at Half Maximum
GEL	General Engineering Laboratory
HASL	Health and Safety Laboratory
IC	Inductively Coupled
ICBK	Initial Calibration Blank
ICP-AES	Inductively Coupled Plasma-Atomic Emission Spectroscopy
ICP-MS	Inductively Coupled Plasma-Mass Spectrometry
IDL	Instrument Detection Limit
keV	kiloelectron volt
LCS	Laboratory Control Sample
LOI	Loss on Ignition
MDC	Minimum Detectable Concentration
MDL	Method Detection Limit
mg/kg	milligram per kilogram
mL	milliliter
MRL	Method Reporting Limit

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MS/MSD	Matrix Spike/Matrix Spike Duplicate
NA	Not Applicable
NIPSCO	Northern Indiana Public Service Company
NIST	National Institute of Standards and Technology
NORM	Naturally Occurring Radioactive Material
ORP	Oxidation-Reduction Potential
oz	ounce
pCi/g	Picocuries per gram
PLM	polarized light microscopy
PRG	Preliminary Remediation Goal
QA	Quality Assurance
QAPP	Quality Assurance Project Plan
QC	Quality Control
%R	Percent Recovery
RI	Remedial Investigation
RI/FS	Remedial Investigation and Feasibility Study
RJ Lee	RJ Lee Group
RL	Reporting Limit
RPD	Relative Percent Difference
RSD	Relative Standard Deviation
RSL	Regional Screening Level
SOP	Standard Operating Procedure
SOW	Statement of Work
SR	Sample Result
SSC	Supplemental Soil Characterization
TOC	Total Organic Carbon
ug/L	micrograms per liter
USEPA	United States Environmental Protection Agency
XRD	X-Ray Diffraction
XRF	X-Ray Fluorescence

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Standard Chemical Abbreviations

Ac	Actinium
Al	Aluminum
Am	Americium
As	Arsenic
Ba	Barium
Be	Beryllium
Bi-	Bismuth
Co	Cobalt
Cs	Cesium
Cr	Chromium
Fe	Iron
K	Potassium
Pa	Protactinium
Pb	Lead
Ra	Radium
Th	Thorium
Tl	Thallium
U	Uranium
V	Vanadium

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Disclaimer

This document is a draft document prepared under a federal administrative order on consent. This document has not undergone formal review by U.S. Environmental Protection Agency (USEPA), however, this document has incorporated comments provided by USEPA on the previous draft version of the report (see Appendix D of the Supplemental Soil Characterization (SSC) Work Plan). The opinions, findings, and conclusions expressed are those of the author and not necessarily those of USEPA.

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A.0 Project Management

A.1 Introduction

In April 2004, the United States Environmental Protection Agency (USEPA) and the Respondents (Brown Inc., Ddalt Corp., Bulk Transport Corp., and Northern Indiana Public Service Company (NIPSCO)) signed an Administrative Order on Consent (AOC II) (Docket No. V-W-'04-C-784) to conduct a Remedial Investigation and Feasibility Study (RI/FS) at the Pines Area of Investigation, located in the environs of the Town of Pines, Indiana, as set forth in Exhibit I to II (AOC II, 2004). AOC II (Section VII. 22) and its attachment, the Statement of Work (SOW) (Task 8), require the Respondents to conduct a Feasibility Study (FS) as part of the RI/FS process. As documented in the FS (AECOM, 2014a), additional soil investigations will be conducted. A Supplemental Soil Characterization (SSC) Work Plan (AECOM, 2014b) provides the details for conducting this work.

This Quality Assurance Project Plan (QAPP) Addendum incorporates the RI/FS QAPP (ENSR [now AECOM], 2005, as revised in 2008) by reference and was prepared to reflect the scope of work described in the SSC Work Plan, and is provided as Appendix B of the SSC Work Plan.

A.2 Project Schedule

The project schedule for the initial nine properties sampled (Appendix G of the SSC Work Plan) is presented in Section 6.0 of the SSC Work Plan.

The schedule for sampling the remaining properties (Appendix H of the SSC Work Plan) will consist of the same components as outlined in Section 6.0 of the SSC Work Plan. Access agreements to conduct work on private properties have been submitted to property owners and the majority of the agreements have been signed and received. The Respondents will continue to work cooperatively with USEPA to obtain the remaining access agreements. Field work is tentatively scheduled to commence May 18, 2015 and is expected to be completed within approximately four weeks; however, delays in receiving any of the remaining access agreements may extend this schedule. Once the field activities are completed, compilation of field notes, laboratory analysis and other report preparation tasks continue for another 10 weeks. Note that expedited laboratory turn-around times (5 business days following sample receipt) have been requested and data maps/tables for each property will be provided to USEPA within one week following receipt of the validated data package for each property.

A.3 Distribution List

The QAPP Addendum, and any subsequent revisions, will be distributed to the personnel shown on the Distribution List that immediately follows the approval page.

A.4 Project/Task Organization

The lines of authority and communication specific to the Quality Assurance (QA) program for this additional soil investigation are presented in Figure A-1.

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A.5 Problem Definition and Background

Background and the objectives specific to the additional soil investigation are provided in Sections 1.1 and 1.2, respectively, of the SSC Work Plan. Additional background information is provided in Appendix H of the SSC Work Plan.

A.6 Project/Task Description

To meet the objectives defined in the SSC Work Plan, three data collection tasks will be conducted:

1. Gamma count rate and gamma dose rate surveys.
2. Coal combustion by-product (CCB) visual inspection confirmation sampling.
3. Soil sampling of selected properties for specific constituents; constituents are specific to the individual sampling tasks, as outlined in Appendix G and Appendix H of the SSC Work Plan.

The number of field and quality control (QC) samples that will be collected for each analytical parameter is presented in Table A-1. A summary of analytical parameters by medium is presented in Table A-2. Target compounds and analytical parameters for all matrices are presented with their respective laboratory detection limits and data quality levels (DQLs) in Tables A-3 (Metals) and A-4 (primary radionuclide target analytes), and in Attachment B (gamma spectroscopy library).

All data generated from field activities or from the analytical program will be reviewed through a tiered review process and validated prior to reporting. All of the laboratory data will be validated as described in Section D.0.

A.7 Quality Objectives and Criteria for Measurement Data

The objectives for precision, accuracy, and sensitivity are provided in Table A-5.

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B.0 Measurement/Data Acquisition

B.1 Sampling Process Design

The rationale for the sample design is provided in Section 3.0 of the SSC Work Plan for the work identified in Appendix G of the SSC Work Plan.

The rationale for the sample design for the work identified in Appendix H of the SSC Work Plan is provided in that Appendix.

B.2 Sampling Methods Requirements

B.2.1 Field Measurements

Field measurements associated with the work described in Appendix G will include a gamma walk-over survey and a gamma dose rate survey. These procedures are described in Sections 3.2.1 and 3.2.2 of the SSC Work Plan. Standard operating procedures (SOPs) are included in Appendix C of the SSC Work Plan.

Field measurements associated with the work described in Appendix H consist of screening selected samples with a field-portable X-ray fluorescence analyzer (FP-XRF), the Operating Procedure for which is provided in Appendix H, Attachment 2.

B.2.2 Sampling Procedures

The SOPs that will be utilized for CCB visual inspection, confirmation sampling, and private property soil sampling (Sections 3.3 and 3.4 of the SSC Work Plan, respectively) are provided in Appendix C of the SSC Work Plan.

An updated procedure for conducting visual inspections associated with the work described in Appendix H is provided in Appendix H, Attachment 2.

B.3 Sample Handling and Custody

B.3.1 Sample Containers, Preservation, and Holding Times

A summary of sample container, preservation, and holding time requirements is presented in Table B-1.

B.3.2 Sample Labeling

For the work described in Appendix G, labeling of sample containers is described in Section 4.2.2.2 of the SSC Work Plan and Section 8 of Appendix G. For the work described in Appendix H, labeling of containers is described in Section 7 of Appendix H.

B.4 Analytical Methods

Analyses will be performed by laboratories that have been utilized for previous investigations at the Pines Area of Investigation and approved by USEPA. Chemical analyses of soil samples for metals for the scope in Appendix G (including hexavalent chromium), except total uranium will be performed by The ALS Group ([ALS]; formerly

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Columbia Analytical Services, Inc.) in Rochester, NY. Gamma spectroscopy analyses for the radioisotopes and inductively coupled plasma – mass spectroscopy (ICP-MS) for total uranium (calculated from U-235 and U-238) will be performed by General Engineering Laboratories, LLC (GEL) in Charleston, SC. Particulate matter analyses will be performed by RJ Lee Group (RJ Lee) in Monroeville, PA.

Chemical analyses of soil samples for metals for the Appendix H sampling will be performed by ALS in Kelso, WA.

B.4.1 Laboratory Analytical Procedures

B.4.1.1 Metals

A list of the ALS and GEL SOPs for metals analyses is provided in Table B-2. The SOPs are included in Attachment A.

B.4.1.2 Particulate Matter

The particulate matter analysis by RJ Lee will consist of polarized light microscopy (PLM), X-ray diffraction (XRD), X-ray fluorescence (XRF), and loss on ignition (LOI). The PLM analysis will be used to identify crystalline and non-crystalline glassy and organic phases in the sample and to identify particles with fly ash or bottom ash morphologies. The PLM analyses will be performed in general accordance to EPA 600/R-93/116, *Test Method for the Determination of Asbestos in Bulk Building Materials*. The SOPs for the particulate matter analysis are not included in Attachment A because RJ Lee considers them to be proprietary. The SOPs are available for review at the RJ Lee facility. A summary of the procedures are presented below.

The sample will be spread on a grid on one or more glass slides, and 400 random points on that grid are observed. Particles will be identified by the microscopist as either fly ash or bottom ash and these particles will be counted. If one particle of fly ash is counted in any of those 400 grid squares, the sample will be reported as containing 0.25% fly ash. If a single fly ash article is observed in the light microscopy field, but it is not contained in any of the 400 grid squares randomly selected for counting, a result of "trace" will be reported, which quantitatively is <0.25% fly ash. The same method will apply to the bottom ash results.

Particle Counting Interpretations Observations on a 400-point Grid

Result	Observations on a 400-point Grid
ND	No particles observed within or outside of the grid
<0.25%	No particles observed within the grid, but particle(s) observed upon review of the entire slide
0.25%	1 particle observed within the grid
0.5%	2 particles observed
0.75%	3
1%	4
1.5%	6
2%	8

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The XRD analysis will be used to identify the crystalline phases present in the samples and will be performed in general accordance to ASTM D934, *Standard Practices for Identification of Crystalline Compounds in Water-Formed Deposits by X-ray Diffraction*. The LOI analysis (based on ASTM D7348, *Standard Test Methods for Loss on Ignition (LOI) of Solid Combustion Residues*) will be used to determine the weight of percent organic and volatile constituents in the sample and to prepare the samples for XRF analysis by removing the carbonaceous material, carbon dioxide, and water content. The XRF analysis will be used to determine the elemental composition of the residual material in the sample. XRF will be performed in general accordance to ASTM D4326, *Standard Method for Analysis of Coal and Coal Ash by X-ray Fluorescence*.

B.4.1.3 Radionuclides

A list of the GEL SOPs is provided in Table B-2. The SOPs are included in Attachment A.

For the radionuclides by gamma spectroscopy, GEL will dry and prep samples according to their SOPs and seal samples in a 100 cc “tuna can” sample container for a minimum of 28-days prior to analyses. The samples will be counted using the more sensitive detector (N-type) and will be counted for a sufficient amount of time, to a maximum count time of 1000 minutes, in order to meet the minimum detectable concentration (MDC) requirements in Table A-4. The laboratory gamma spectroscopy library includes those isotopes that are used to directly or indirectly quantify the primary radionuclide target analytes. These isotopes are also listed in Table A-4 as primary target analytes with MDCs. Due to the parent/progeny relationship with the isotopes in the gamma spectroscopy library and Table A-4, MDCs are not established for all isotopes on the library list. MDCs in Table A-4 establish the detection limits for the primary radionuclide target analytes. The radionuclides that will be reported for the soil samples will be based on the gamma spectroscopy library that is provided in Attachment B.

The following limits on sensitivity parameters will be applied for gamma spectroscopy analysis:

- Energy Tolerance: 1.5 kiloelectron volts (keV)
- Peak Sensitivity: 3.0
- Abundance Limit: 75%

GEL will conduct a thorough review of 'Unidentified Peaks' and note such peaks with comments in the batch narrative. Any nuclides identified in the 'Unidentified Peaks' report that are not already being reported may be added to the library following approval by the AECOM project manager. For each radionuclide in the gamma library, GEL will report the concentration along with its sample specific uncertainty for each sample as obtained during analysis even if the concentration is less than the MDC or is negative (Section 16.6 in MARLAP, 2004).

USEPA plans to collect split samples for independent analysis. The specific procedures for sample selection, collection, and shipment will be provided by USEPA and will be accommodated in the field.

B.4.2 List of Project Target Constituents and Detection Limits

A listing of project target constituents and reporting limits (RLs) or MDCs for each analyte group listed in Table B-2 can be found in Tables A-3 (metals) and A-4 (primary radionuclide target analytes) and in Attachment B (gamma spectroscopy library).

B.5 Quality Control

Table B-3 summarizes the QC for the analytical methods. Field QC will include field duplicates, equipment blanks, and matrix spike/matrix spike duplicate (MS/MSD) samples, as appropriate for the analytical method. Table A-5

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presents the limits for each field QC sample. Procedures and frequencies will be consistent with those described in the RI/FS QAPP.

B.6 Instrument/Equipment Testing, Inspection, and Maintenance

B.6.1 Field Instrument Maintenance

The maintenance schedule and trouble-shooting procedures for field instrument are indicated in Table B-4.

B.7 Laboratory Instrument Preventative Maintenance

Table B-5 provides the frequency with which components of key analytical instruments will be serviced. Table B-6 provides a summary of the monitoring of laboratory equipment.

B.8 Instrument/Equipment Calibration and Frequency

B.8.1 Field Instruments

Calibration of field instruments will be performed according to the manufacturer's instructions and the SOPs included in Appendix C of the SSC Work Plan. A summary of calibration procedures and frequencies is provided as Table B-7.

B.8.2 Analytical Instrumentation

The SOP for each analysis performed in the laboratory describes the calibration procedures, their frequency, acceptance criteria, and the conditions that will require recalibration. This information is summarized in Table B-8. The SOPs are included as Attachment A.

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C.0 Project Assessment/Oversight

No audits of field activities or the laboratory analyses associated with the investigation to be conducted under the SSC Work Plan are planned.

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D.0 Data Validation and Usability

All chemical and radiochemical data generated as part of the Appendix G scope of work will be subjected to limited validation. Ten percent of the data associated with the Appendix H scope of work will be subjected to full data validation; the remaining chemical data will be subjected to limited validation. Validation of the laboratory data associated with the Appendix H scope of work will be performed by Environmental Standards, Inc. The elements of full and limited validated are described in the RI/FS QAPP (ENSR, 2008). Consistent with the RI/FS QAPP, all metals and radionuclide data packages will be submitted by the laboratories as Level IV data deliverables in the event that a more thorough data validation is required at a later date.

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AECOM. 2014a. Feasibility Study. Pines Area of Investigation. July 2014.

AECOM. 2014b. Supplemental Soil Characterization Work Plan. Pines Area of Investigation. November 2014.

ENSR. 2005. Remedial Investigation/Feasibility Study Work Plan, Volumes 1-7. September 16, 2005.

ENSR. 2008. Remedial Investigation/Feasibility Study Work Plan, Volume 3, Quality Assurance Project Plan.
Revised March 31, 2008.

MARLAP. 2004. Multi-Agency Radiological Laboratory Analytical Protocols Manual. July 2004.

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Table A-1 Sample Summary

Matrix	Number of Locations (Samples)	Field Parameters	Analytical Parameters	Equipment Blanks ¹	Field Duplicates ²	MS/MSDs ^{2,3}
Soil (Quadrant Sampling Associated with Appendix G of the SSC Work Plan)	<ul style="list-style-type: none"> 9 properties 4 quadrants per property⁴ (up to 7 quadrants if special uses) 1 composite (from 5 locations) per quadrant 3 depths⁵: 0 - 6 inches below ground surface (bgs), 6 -18 inches bgs, and 1.5 - 5 feet bgs. Total samples: 72 - 189	None	Aluminum Arsenic Chromium (total) Hexavalent Chromium Cobalt Iron Thallium Total uranium Vanadium Radionuclides ⁶	9	9	9 pairs
Soil (Quadrant Sampling Associated with Appendix H of the SSC Work Plan)	37 Identified Properties <ul style="list-style-type: none"> 4 quadrants per property⁴ (up to 7 quadrants if special uses) 1 composite (from 5 locations) per quadrant 2 to 3 depths: <ul style="list-style-type: none"> 0 - 6 inches bgs, 6 -18 inches bgs, and 18-36 inches bgs⁷. Replicate (triplicate) sampling at 15% of Quadrants 	None	Arsenic Lead Thallium	37-74	37	37
	32 Additional Properties <ul style="list-style-type: none"> 4 quadrants per property⁴ (up to 7 quadrants if special uses) 1 composite per property, if no CCBs are observed, plus 1 composite per property if CCBs are observed along the road 1 depth: 0 - 6 inches bgs Total samples: 420 – 1,075	None	Arsenic Lead Thallium	2-4	4	4
				39-78	41	41
Soil/CCB Mixture (CCB VI Verification Sampling)	15	None	Particulate matter (PLM, XRF, XRD, LOI)	NA	0	NA

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Matrix	Number of Locations (Samples)	Field Parameters	Analytical Parameters	Equipment Blanks ¹	Field Duplicates ²	MS/MSDs ^{2,3}
<p>¹ Equipment rinsate blanks will be collected at a minimum frequency of one per property per non-disposable or non-dedicated piece of equipment used.</p> <p>² For Appendix G scope: Collected at a minimum frequency of one per 10 samples submitted for analysis for the CCB/soil mixture samples, and a frequency of one per property sampled for the soil samples. For Appendix H scope: Collected at a minimum frequency of one per Identified Property and at a minimum frequency of one per 20 samples collected from Additional Properties.</p> <p>³ Collected at a frequency of one per property sampled, except not applicable to radionuclides.</p> <p>⁴ Up to 7 quadrants may be established based on the property size and to address the three specific property uses defined in the SSC Work Plan: gardens, children's play areas (based on the presence of swing sets and/or other outdoor play equipment), and unpaved driveways where CCBs are observed.</p> <p>⁵ Samples collected at the 1.5 - 5 foot horizon will be submitted for analysis only if CCBs are visually observed in at least 1 of the 5 samples to be composited from a single quadrant.</p> <p>⁶ Refer to Table A-4 and Attachment B (gamma spectroscopy library) for the specific list of radionuclides.</p> <p>⁷ Samples collected from the 18-36 inch horizon will be submitted for analysis if all or a portion of the 18-to-36-inch interval contains no visual evidence of CCBs in at least 1 of the five samples in a quadrant; samples with no visually-observed CCBs will be composited from a single quadrant.</p> <p>bgs – below ground surface CCB – Coal Combustion By-product LOI – Loss on Ignition NA – Not Applicable MS/MSD – Matrix Spike/Matrix Spike Duplicate PLM – Polarized Light Microscopy SSC – Supplemental Soil Characterization XRF – X-ray Fluorescence XRD –X-ray Diffraction</p>						

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Table A-2 Laboratory Parameters by Sample Medium

Parameter	Soil ⁴ (Quadrant Sampling – Appendix G)	Soil (Quadrant Sampling – Appendix H)	Soil/CCB Mixtures ⁴ (CCB VI Verification Sampling)
Metals ¹	X	X	
Hexavalent chromium	X		
Percent moisture	X		
Radionuclides ²	X		
Particulate matter ³			X
¹ Refer to Table A-3 for the specific list of analytes. ² Refer to Table A-4 and Attachment B (gamma spectroscopy library) for the specific list of radionuclides. ³ PLM, XRF, XRD, LOI ⁴ Note that particulate matter analyses will be conducted on samples collected from locations where CCBs were previously identified to be present. CCBs may also be present in the soil samples identified for collection in Table A-1, however, those samples will be collected without biasing the sample locations based on the presence or absence of CCBs. CCB – Coal Combustion By-product LOI – Loss on Ignition PLM – Polarized Light Microscopy XRD – X-Ray Diffraction XRF – X-Ray Fluorescence			

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Table A-3a Target Analytes, Reporting Limits, and Data Quality Levels for Metals (Appendix G)

Analytes	CAS Number	MDL ¹ or IDL ¹ (mg/kg)	RL ¹ (mg/kg)	DQL – USEPA May 2014 Residential RSL (mg/kg) ^{2,3}	
Metals by ICP-AES					
Aluminum	7429-90-5	2.8	10	7700	nc
Cobalt	7440-48-4	0.053	5	2.3	nc
Iron	7439-89-6	5.23	10	5500	nc
Metals by ICP-MS					
Arsenic	7440-38-2	0.003	0.10	0.67	c
Chromium (total)	7440-47-3	0.00438	0.20	12000	nc
Thallium	7440-28-0	0.00239	0.10	0.078	nc
Uranium (total)	7440-61-1	see below		23	nc
U-235	see total uranium	0.002	0.014	see total uranium	
U-238	see total uranium	0.0132	0.040	see total uranium	
Vanadium	7440-62-2	0.0164	0.20	39	nc
Hexavalent Chromium					
Chromium (hexavalent)	18540-29-9	0.057	0.40	0.3	c

¹ Laboratory RLs, MDLs (GEL), and IDLs (ALS) are updated periodically and are dependent on aliquot weight and percent moisture of the samples. MDLs and IDLs are updated periodically; the current MDLs and IDLs at the time of analyses will be used.

² Nondetects for ICP-AES and ICP-MS analyses will be reported at the MDLs or IDLs to ensure achievement of the DQL.

³ Regional Screening Levels for Chemical Contaminants at Superfund Sites. May 2014. <http://www.epa.gov/region09/superfund/prg/index.html>. Values for residential soil. If RSL is based on a noncancer endpoint (nc), the RSL is adjusted to a hazard quotient of 0.1 by multiplying the RSL by 0.1. The risk level is 1E-6 if the RSL is based on a cancer endpoint (c).

CAS – Chemical Abstracts Service

DQL – Data Quality Level

ICP-AES – Inductively Coupled Plasma-Atomic Emission Spectroscopy

ICP-MS – Inductively Coupled Plasma-Mass Spectrometry

IDL – Instrument Detection Limit

MDL – Method Detection Limit

mg/kg – milligrams per kilogram

RSL – Regional Screening Level

RL – Reporting Limit

USEPA – United States Environmental Protection Agency

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Table A-3b Target Analytes, Reporting Limits, and Data Quality Levels for Metals (Appendix H)

Analytes	CAS Number	MDL ¹ or IDL ¹ (mg/kg)	RL ¹ (mg/kg)	DQL – USEPA May 2014 Residential RSL (mg/kg) ^{2,3}
Metals by ICP-AES				
Arsenic ⁴	7440-38-2	2	4	0.67 c
Lead ⁴	7439-92-1	0.7	2	N/A
Thallium ⁴	7440-28-0	0.8	2	0.078 nc
Metals by ICP-MS				
Arsenic	7440-38-2	0.2	0.5	0.67 c
Lead	7439-92-1	0.02	0.05	N/A
Thallium	7440-28-0	0.002	0.02	0.078 nc
¹ Laboratory RLs, MDLs and IDLs are updated periodically and are dependent on aliquot weight and percent moisture of the samples. MDLs and IDLs are updated periodically; the current MDLs and IDLs at the time of analyses will be used. ² Nondetects for ICP-AES and ICP-MS analyses will be reported at the MDLs or IDLs to ensure achievement of the DQL. ³ Regional Screening Levels for Chemical Contaminants at Superfund Sites. May 2014. http://www.epa.gov/region09/superfund/prg/index.html . Values for residential soil. If RSL is based on a noncancer endpoint (nc), the RSL is adjusted to a hazard quotient of 0.1 by multiplying the RSL by 0.1. The risk level is 1E-6 if the RSL is based on a cancer endpoint (c). ⁴ This analyte will be initially analyzed by method SW-846 6020; however, the analyte may need to be reanalyzed by SW-846 6010 if the analyte concentration exceeds the calibration range of SW-846 6020. If possible, the associated sample(s) will be reanalyzed by SW-846 6020 at an appropriate dilution to bring the analyte concentration within the SW-846 6020 calibration range; however, if the dilution level is a source of failed QC results, then the associated sample(s) will be reanalyzed via SW-846 6010. CAS – Chemical Abstracts Service DQL – Data Quality Level ICP-AES – Inductively Coupled Plasma-Atomic Emission Spectroscopy ICP-MS – Inductively Coupled Plasma-Mass Spectrometry IDL – Instrument Detection Limit MDL – Method Detection Limit mg/kg – milligrams per kilogram N/A – not applicable; lead is being analyzed solely for use in evaluating the accuracy of field screening instrumentation (see Section 13 of Appendix H for further discussion). RSL – Regional Screening Level RL – Reporting Limit USEPA – United States Environmental Protection Agency				

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Table A-4 MDC and Data Quality Levels for Primary Radionuclide Target Analytes

Radionuclides ¹	MDC ² (pCi/g)	DQL – Residential Soil PRG (pCi/g) ³
Potassium-40	0.5	8.36E-02
Cobalt-60	0.1	3.20E-02
Barium-137m	0.1	1.60E+05
Cesium-137 ⁴	– ⁴	– ⁴
Thallium-208	0.1	2.05E+04
Lead-210	1.0	1.90E-02
Bismuth-211	– ⁵	2.69E+06
Lead-211	– ⁵	3.95E+01
Bismuth-212	0.5	1.70E+01
Lead-212	0.1	6.44E-01
Bismuth-214	0.1	6.43E+01
Lead-214	0.1	4.73E+01
Radon-219	– ⁵	7.04E+07
Francium-223	– ⁵	2.11E+00
Radium-223	– ⁵	6.65E-02
Thorium-227	– ⁵	3.35E-01
Actinium-228	0.2	8.58E+00
Protactinium-231	– ⁵	8.86E-02
Thorium-231	– ⁵	7.36E+00
Protactinium-234m	4.5	1.03E+07
Thorium-234	4.5	6.95E-01
Uranium-235	0.2	1.07E-01
Americium-241 ⁴	– ⁴	– ⁴
¹ The radionuclides listed in this table are based on the USEPA approved gamma spectroscopy library presented in Attachment B. ² MDCs are based upon sample volume, instrument background, detector efficiency, count time and other statistical factors, as well as specific isotopic values such as abundance and half-life. ³ USEPA Preliminary Remediation Goals for Radionuclides. Updated November 2014. As the download tables were not available as of 11/6/2014, the on-line calculator was used to derive PRGs for the residential scenario using all defaults from the calculator. http://epa-prgs.ornl.gov/radionuclides/download.html . DQLs for which Residential Soil PRGs are available for the isotope and its progeny. The DQL represents the lower of the Residential Soil PRGs for the isotope and its progeny. ⁴ Cs-137 and Am-241 will be reported for each sample, but the results will be used by the laboratory as QC checks on instrument calibration (i.e., efficiency and energy). These radionuclides are not naturally occurring radioactive material (NORM), but are in the environment due to world-wide nuclear activities. ⁵ The laboratory gamma spectroscopy library includes those isotopes that are used to directly or indirectly quantify the primary target radionuclides listed above. Due to the parent/progeny relationship with the isotopes in the gamma spectroscopy library, MDCs are not established for all isotopes on the library list. MDCs establish the detection limits for the primary radionuclide target analytes. pCi/g – picoCuries per gram		

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Radionuclides ¹	MDC ² (pCi/g)	DQL – Residential Soil PRG (pCi/g) ³
DQL – Data Quality Level MDC – Minimum Detectable Concentration NORM – Naturally Occurring Radioactive Material PRG – Preliminary Remediation Goal QC – Quality Control		

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Table A-5a Quality Control Performance Criteria for Soil Samples (Appendix G)

Analytes or Radionuclides	Field and Lab Blanks	Field Duplicate %RPD ¹	LCS % R ²	Matrix Spike % R ²	Duplicate % RPD ²
Metals by 3050/6010C/6020A & 3060/7199					
Aluminum	<RL	30	47 – 152 ³	75-125	20
Arsenic	<RL	30	82.3 - 117 ³	75-125	20
Chromium (total)	<RL	30	81.8 - 118 ³	75-125	20
Chromium (hexavalent)	<RL	30	80 - 120 ³	85-115	20
Cobalt	<RL	30	83.2 -116 ³	75-125	20
Iron	<RL	30	50.6 -149 ³	75-125	20
Thallium	<RL	30	78.2 -120 ³	75-125	20
Uranium (total) (calculated from U-235 and U-238)					
Uranium-235	<RL	30	75-125	75-125	20
Uranium-238	<RL	30	75-125	75-125	20
Vanadium	<RL	30	73.5 -126 ³	75-125	20
Radionuclides by Gamma Spectroscopy					
Potassium-40	<MDC	30	75-125	NA	20
Cobalt-60	<MDC	30	75-125	NA	20
Barium-137m	<MDC	30	75-125	NA	20
Cesium-137 ⁴	<MDC	30	75-125	NA	20
Thallium-208	<MDC	30	75-125	NA	20
Lead-210	<MDC	30	75-125	NA	20
Bismuth-211	<MDC	30	75-125	NA	20
Lead-211	<MDC	30	75-125	NA	20
Bismuth-212	<MDC	30	75-125	NA	20
Lead-212	<MDC	30	75-125	NA	20
Bismuth-214	<MDC	30	75-125	NA	20
Lead-214	<MDC	30	75-125	NA	20
Radon-219	<MDC	30	75-125	NA	20
Francium-223	<MDC	30	75-125	NA	20
Radium-223	<MDC	30	75-125	NA	20
Thorium-227	<MDC	30	75-125	NA	20
Actinium-228	<MDC	30	75-125	NA	20
Protactinium-231	<MDC	30	75-125	NA	20
Thorium-231	<MDC	30	75-125	NA	20
Protactinium-234m	<MDC	30	75-125	NA	20

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Analytes or Radionuclides	Field and Lab Blanks	Field Duplicate %RPD ¹	LCS % R ²	Matrix Spike % R ²	Duplicate % RPD ²
Thorium-234	<MDC	30	75-125	NA	20
Uranium-235	<MDC	30	75-125	NA	20
Americium-241 ⁴	<MDC	30	75-125	NA	20
<p>¹ RPD criteria when results are less than 5x RL \leq 60%.</p> <p>² Control limits current at the time of analyses will be used.</p> <p>³ Represents the limits on the Certificate of Analysis supplied by the vendor or limit supplied by ALS Rochester.</p> <p>⁴ Cs-137 and Am-241 will be reported for each sample, but the results will be used by the laboratory as QC checks on instrument calibration (i.e., efficiency and energy). These radionuclides are not NORM, but are in the environment due to world-wide nuclear activities.</p> <p>⁵ The gamma spectroscopy analysis will quantify gamma-emitting radionuclides provided in Table A-4. Many of these radionuclides are decay progeny of parent radionuclides that are Pines Area of Investigation constituents of concern (COCs) and they will be used to directly quantify the activity of the parent COC. Parent COC concentrations are equivalent to or a constant factor of the progeny radionuclides given that the decay chains will be in secular equilibrium following a 28-day buildup time.</p> <p>LCS – Laboratory Control Sample MDC – Minimum Detectable Concentration (radionuclides only) NA – Not Applicable NORM – Naturally Occurring Radioactive Material QC – Quality Control RL – Reporting Limit % R - Percent Recovery RPD – Relative Percent Difference</p>					

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Table A-5b Quality Control Performance Criteria for Soil Samples (Appendix H)

Analytes	Field and Lab Blanks	Field Duplicate %RPD ¹	LCS % R ²	Matrix Spike % R ²	Duplicate % RPD ²
Metals by 3050/6010C/6020A					
Arsenic	<RL	30	82 - 117 ³ 78 - 122 ⁴	75-125	20
Lead	<RL	30	82-118 ³ 79-121 ⁴	75-125	20
Thallium	<RL	30	78-121 ³ 79-120 ⁴	75-125	20
¹ RPD criteria when results are less than 5x RL \leq 60%. ² Control limits current at the time of analyses will be used. ³ Represents the limits on the Certificate of Analysis supplied by the vendor or limit supplied by ALS Kelso for Method 6010C. ⁴ Represents the limits on the Certificate of Analysis supplied by the vendor or limit supplied by ALS Kelso for Method 6020A. LCS – Laboratory Control Sample RL – Reporting Limit % R - Percent Recovery RPD – Relative Percent Difference					

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Table B-1 Summary of Sample Container, Preservation, and Holding Time Requirements for Soil and CCB/Soil Mixture Samples

Parameter	Container ^{1,2}	Preservation	Holding Time ³
Metals (Appendix G) Al, As, Co, Cr, Fe, Ti, U (total), and V	Wide-mouth 500-mL glass or plastic jar ⁴	Cool, 0 - 6°C	180 days
Metals (Appendix H) As, Pb, Ti	Wide-mouth 500-mL glass or plastic jar ⁴	Cool, 0 - 6°C	180 days
Hexavalent chromium	Wide-mouth 500-mL glass or plastic jar ^{4,5}	Cool, 0 - 6°C	24 hours for ORP, pH, and ferrous iron 30 days to extraction, 7 days from extraction to analysis for hexavalent chromium
Radionuclides ⁶	Wide-mouth 500 mL glass or plastic jar ⁴	None required	None required
Particulate matter (PLM, XRD, XRF, LOI)	Wide-mouth 4-oz glass jar ⁴	None required	None established

¹ Additional volume will be collected for MS/MSD samples (metals only).

² Laboratory may provide alternate containers as long as the containers meet the requirements of the method and allow the collection of sufficient volume to perform the analyses.

³ Holding time begins from date and time of sample collection.

⁴ Glass containers will be placed in zipper-lock bags prior to shipping.

⁵ In association with the hexavalent chromium analyses, additional parameters will be analyzed. These parameters either confirm the reducing/oxidizing tendency of the samples or are indirect indicators of the reducing/oxidizing tendency, and include pH, oxidation reduction potential (ORP), Total Organic Carbon (TOC), sulfides, and ferrous iron. To ensure adequate volume, a separate container from other metals will be collected. These results will be used qualitatively.

⁶ Refer to Table A-4 and Attachment B (gamma spectroscopy library) for the list of radionuclides that will be reported for this program.

mL – milliliter

oz – ounce

°C – degrees Celsius

CCB – Coal Combustion By-product

LOI – Loss on Ignition

MS/MSD – Matrix Spike/Matrix Spike Duplicate

ORP – Oxidation Reduction Potential

PLM – Polarized Light Microscopy

TOC – Total Organic Carbon

XRD – X-Ray Diffraction

XRF – X-Ray Fluorescence

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Table B-2 Analytical Methodologies

Analyte Group ¹	Laboratory SOP Number ²	Equivalent Method Number
Metals (Appendix G)		
Metals by ICP-AES (Al, Co, Fe)	MET-3050B, Rev. 5 MET-200.7/6010C, Rev. 14	USEPA SW-846 Method 3050B USEPA SW-846 Method 6010C
Metals by ICP-MS (As, Cr, Tl, V)	MET-3050B, Rev. 5 MET-6020A/200.8, Rev. 2	USEPA SW-846 Method 3050B USEPA SW-846 Method 6020A
Uranium (Total) (calculated from U-235 and U-238)	GL-MA-E-009, Rev. 22 GL-MA-E-014, Rev. 25	USEPA SW-846 Method 3050B USEPA SW-846 Method 6020A
Hexavalent Chromium	GEN-3060, Rev. 3 GEN-7199, Rev. 6	USEPA SW-846 Method 3060A USEPA SW-846 Method 7199
Metals (Appendix H)		
Metals by ICP-AES (As, Pb, Tl) ³	MET-3050B, Rev. 14 MET-200.7/6010C, Rev. 25	USEPA SW-846 Method 3050B USEPA SW-846 Method 6010C
Metals by ICP-MS (As, Pb, Tl) ³	MET-3050B, Rev. 14 MET-6020A/200.8, Rev. 16	USEPA SW-846 Method 3050B USEPA SW-846 Method 6020A
Radionuclides		
Radionuclides by Gamma Spectroscopy	GL-RAD-A-013, Rev. 25 GL-RAD-A-021, Rev. 20 GL-RAD-I-001, Rev 19	DOE EML HASL 300
Particulate Matter		
PLM	See Section B.4.1.2	Modified, EPA 600/R-93/116, <i>Test Method for the Determination of Asbestos in Bulk Building Materials</i> .
XRD	See Section B.4.1.2	ASTM D934, <i>Standard Practices for Identification of Crystalline Compounds in Water-Formed Deposits by X-ray Diffraction</i>
XRF	See Section B.4.1.2	ASTM D4326, <i>Standard Method for Analysis of Coal and Coal Ash by X-ray Fluorescence</i>
LOI	See Section B.4.1.2	ASTM D7348, <i>Standard Test Methods for Loss on Ignition (LOI) of Solid Combustion Residues</i>
<p>¹ See Tables A-3 and A-4 and Attachment B (gamma spectroscopy library) for the analytes in each analyte group.</p> <p>² The version of the SOP that is current at the time of sample analysis will be utilized. Any modification to the approved SOP will require USEPA notification and concurrence.</p> <p>³ If the analyte(s) concentration(s) exceeds the calibration range of SW-846 6020, the associated sample(s) will be analyzed at an appropriate dilution in order to bring the analyte(s) concentration(s) within the 6020 calibration range; however, if the dilution level is a source of failed QC results then the associated sample(s) will be reanalyzed via SW-846 6010.</p> <p>DOE – Department of Energy EML – Environmental Measurements Laboratory HASL – Health and Safety Laboratory ICP-AES – Inductively Coupled Plasma-Atomic Emission Spectroscopy ICP-MS – Inductively Coupled Plasma-Mass Spectrometry LOI – Loss on Ignition NA – Not Available</p>		

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Analyte Group ¹	Laboratory SOP Number ²	Equivalent Method Number
PLM – Polarized Light Microscopy SOP – Standard Operating Procedure USEPA – United States Environmental Protection Agency XRF – X-ray Fluorescence XRD – X-ray Diffraction		

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Table B-3 Analytical Quality Control Checks

Parameter/ Method	QC Check	Frequencies ¹	Control Limits	Laboratory Corrective Actions
Metals 6010C	Reagent/prep/ ICBK blanks	One per preparation batch	No analytes above RL	Repreparation/reanalysis of entire prep batch
	MS samples	One per preparation batch	One per preparation batch	Analyze post-digestion spike
	Duplicate samples	One per preparation batch	Refer to Table A-5	Check analytical system, flag results
	LCS	One per preparation batch	Vendor limits (refer to Table A-5)	Repreparation/reanalysis of entire prep batch
	Dilution test	One per preparation batch	Within 10% of original sample results	Flag results
	Interference check	Beginning of each analytical run	20% of true values	Recalibrate and reanalyze any sample with interfering elements
Metals 6020A	Reagent/Prep/ ICBK/CCBK blanks	1 per analytical batch of 20 samples or less, CCBs every 10 samples in analytical run	No analytes above RL	Repreparation/reanalysis of entire prep batch
	MS Samples	1 per analytical batch of 20 samples or less	Refer to Table A-5	Analyze post digestion spike
	MS Duplicate Samples	1 per analytical batch of 20 samples or less	Refer to Table A-5	Check analytical system, flag results
	LCS	1 per analytical batch of 20 samples or less	Vendor limits (refer to Table A-5)	Repreparation/reanalysis of entire prep batch
	Dilution Test	1 per analytical batch of 20 samples or less	10% (results >4 x RL)	Flag results
	Interference check	Beginning of each analytical run	80-120% R	Recalibrate, reanalyze any sample with interfering elements
Hexavalent Chromium 7199	Reagent/prep blanks	One per analytical batch of 20 samples or less	Not detected above MRL	Repreparation/reanalysis of entire batch
	MS samples	One per analytical batch of 20 samples or less	Refer to Table A-5	Repreparation/reanalysis of entire batch
	Duplicate samples	One per analytical batch of 20 samples or less	Refer to Table A-5	Check analytical system, flag results
	LCS	One per analytical batch of 20 samples or less	Vendor limits (refer to Table A-5)	Repreparation/reanalysis of entire batch

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Parameter/ Method	QC Check	Frequencies ¹	Control Limits	Laboratory Corrective Actions
Radionuclides Gamma spectroscopy	Reagent/prep blanks	One per preparation batch	Not detected above MDC	Repreparation/reanalysis of entire batch
	Duplicate samples	One per preparation batch	RPD <20; RPD<100 if SR <5x MDC; RPD NA if SR <MDC	Check analytical system, flag results
	LCS	One per preparation batch	75-125% R	Repreparation/reanalysis of entire batch
Particulate Matter	Duplicate samples	One per preparation batch	RPD <20 or current laboratory limits	Check results, flag results

¹ Preparation Batch defined as maximum of 20 field samples of a similar matrix unless otherwise specified.

CCB – Coal Combustion By-product
 CCBK – Continuing Calibration Blank
 ICBK – Initial Calibration Blank
 LCS – Laboratory Control Sample
 MDC – Minimum Detectable Concentration
 MRL – Method Reporting Limit
 MS – Matrix Spike
 NA – Not Applicable
 %R – Percent Recovery
 QC – Quality Control
 RL – Reporting Limit
 RPD – Relative Percent Difference
 SR – Sample Result

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Table B-4 Maintenance Procedures and Schedule for Field Instruments

Instrument	Maintenance Procedures/Schedule	Spare Parts in Stock
Portable radiation detection instruments	Physical check of instrument/daily Check the battery and change/recharge if necessary/daily High voltage check/daily Response correlation check/beginning of the program	Batteries Source standard

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Table B-5 Maintenance Procedures and Schedule for Analytical Instruments

Instrument	Spare Parts	Activity	Frequency
ICP-AES (SW-846 Method 6010C)	Gases O-rings Tubing	Check gases	Daily
		Check argon tank pressure	Daily
		Check aspiration tubing	Daily
		Check vacuum pump gauge	Daily
		Check cooling water system	Daily
		Check nebulizer	Daily
		Check capillary tubing	Daily
		Check peristaltic pump tubing	Daily
		Check high voltage switch	Daily
		Check exhaust screens	Daily
		Check torch, glassware, aerosol injector tube, bonnet	Daily
		Clean plasma torch assembly	Monthly or as needed
		Clean nebulizer and drain chamber	Monthly or as needed
		Clean filters	Monthly or as needed
		Replace tubing	Monthly or as needed
		Check o-rings	Monthly or as needed
ICP-MS (SW-846 Method 6020A)	Gases O-rings Tubing	Clean nebulizer tip after use	As needed
		Replace peripump sample introduction tubing	As needed
		Change pump hoses on drain systems	As needed
		Check drain waste collection containers, and empty as necessary	As needed
		Check Neslab water level and add water if required	As needed
		Clean/replace interface cones	As needed
		Clean/replace nebulizer	As needed
		Clean/replace torch	As needed
		Check/replace water filter	As needed
		Change oil in interface rotary pump (or as needed).	Quarterly
		Clean ion lenses 4-6 months (or as needed).	Quarterly
		Clean air filters	6 months
		Change pump oil in backing rotary pump	12 months
Ion Chromatography Method (SW-846 Method 7199)		Evaluate/replace EM	
		Rinse IC pump and valves	Weekly
Gamma Spectrometer (DOE EML HASL 300)		Lubricate pump	Every 6 months
		Energy and FWHM calibration	Annual
		Efficiency calibration	Annual
		Instrument check	Daily
		Background	Weekly
		Liquid nitrogen fill	Weekly
		Software backups	Monthly
		Filter cleaning	Quarterly

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Instrument	Spare Parts	Activity	Frequency
DOE – Department of Energy EM – Electron Multiplier EML – Environmental Measurements Laboratory FWHM – Full Width at Half Maximum HASL – Health and Safety Laboratory IC – Ion Chromatograph ICP-AES – Inductively Coupled Plasma-Atomic Emission Spectroscopy ICP-MS – Inductively Coupled Plasma-Mass Spectrometry			

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Table B-6 Laboratory Equipment Monitoring

Equipment Type	Activity	Frequency
Ovens	Temperature monitoring Electronics serviced	Daily As needed
Refrigerators	Temperature monitoring Refrigerant system and electronics serviced	Twice daily As needed
Balances	Calibration Manufacturer cleaning and servicing	Daily or before use Annually
High-purity water system	Conductance monitoring	Daily

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Table B-7 Field Instrument Calibration

Parameter	Calibration Frequency	Calibration Standards	Acceptance Criteria
Gamma Walk-over Survey (Ludlum Model 44-10 detector)	Annually: At least once per year and after repairs have been performed, if applicable. ¹	NIST Traceable sources	Calibrated by the manufacturer or a rental vendor with detector efficiency optimized for gamma energies associated with Ra-226 and its decay progeny
	Daily: Field response check	Exempt sealed gamma source	Within 20% of source activity
Gamma Dose Rate Survey (Thermo Scientific™ Micro Rem r)	Annually: At least once per year and after repairs have been performed, if applicable.	NIST Traceable sources	Calibrated by the manufacturer or a rental vendor
	Daily: Field response check	Exempt sealed gamma source	Within 20% of source activity
<p>¹ Note that the detector will be correlated to a source block set (4 stacked blocks) with known total radium concentration; a 10 pCi/g total radium source block set is located at the former Kerr McGee Rare Earths Facility in West Chicago, Illinois.</p> <p>NIST – National Institute of Standards and Technology pCi/g – picoCuries/gram SSC – Supplemental Soil Characterization</p>			

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Table B-8 Analytical Instrument Calibration

Instrument and Method	Calibration Frequency	Calibration Standards	Acceptance Criteria ¹
Metals by ICP-AES SW-846 6010C	Initial: Daily	Initial: Per manufacturer's instructions. Minimum of one standard and calibration blank and instrument blank.	Initial: Highest standard within 10% of true value. % RSD 20 < RL
	Continuing: Every 10 samples	Mid-level of each metal and instrument blank	±10% of true value % RSD 20 < RL
	Ending	Mid-level of each metal and instrument blank	±10% of true value % RSD 20 < RL
Metals by ICP-MS SW-846 6020A	Instrument tune: Daily	Per manufacturer: tune solution of 10 ug/L, Be, Mg, Co, In, Pb	Manufacturer's recommended tune criteria as specified in SOP.
	Initial: Daily	Initial per manufacturer's instructions – minimum of one calibration standard, one calibration blank and interference check standards ICS-A, ICS-AB	±10% true value % RSD 20 <RL ±20% recovery
	Continuing: every 10 samples	One calibration standard and one calibration blank	±10% true value % RSD 20 <RL
	Ending	If required: run MRL standard, ICS-A and ICS-AB interference check standards, one calibration standard, one calibration blank	±20% recovery ±10% true value % RSD 20 <RL

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Instrument and Method	Calibration Frequency	Calibration Standards	Acceptance Criteria ¹
Hexavalent chromium by SW-846 7199	Initial: Daily	3 standards plus blank	$r \geq 0.999$ $\pm 10\%$ of true value < RL
	Continuing: Every 10 injections, or 24 hours, whichever is more frequent	Mid-level plus blank	$\pm 10\%$ of true value < RL
	Ending	Mid-level plus blank	$\pm 10\%$ of true value < RL
Radionuclides by Gamma Spectroscopy (DOE EML HASL 300)	Daily checks using a check source with multiple radionuclides	NIST Traceable standards	Within 2-3 sigma control limits
	Monitor FWHM and efficiency	NIST Traceable standards	Within 2-3 sigma control limits
	Peak centroid	NIST Traceable standards	± 2 keV
	Weekly or monthly background checks	NA	Within 2-3 sigma control limits

¹ If criteria are not met, corrective actions as specified in the laboratory SOPs (Attachment A) are taken.

DOE – Department of Energy

EML – Environmental Measurements Laboratory

FWHM – Full Width at Half Maximum

HASL – Health and Safety Laboratory

ICP-AES – Inductively Coupled Plasma-Atomic Emission Spectroscopy

ICP-MS – Inductively Coupled Plasma-Mass Spectrometry

ICS – Interference Check Sample

KeV – kiloelectron volt

MRL – Method Reporting Limit

NA – Not Applicable

NIST – National Institute of Standards and Technology

RL – Reporting Limit

RSD – Relative Standard Deviation

SOP – Standard Operating Procedure

ug/L – micrograms per Liter

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Pines Area of Investigation

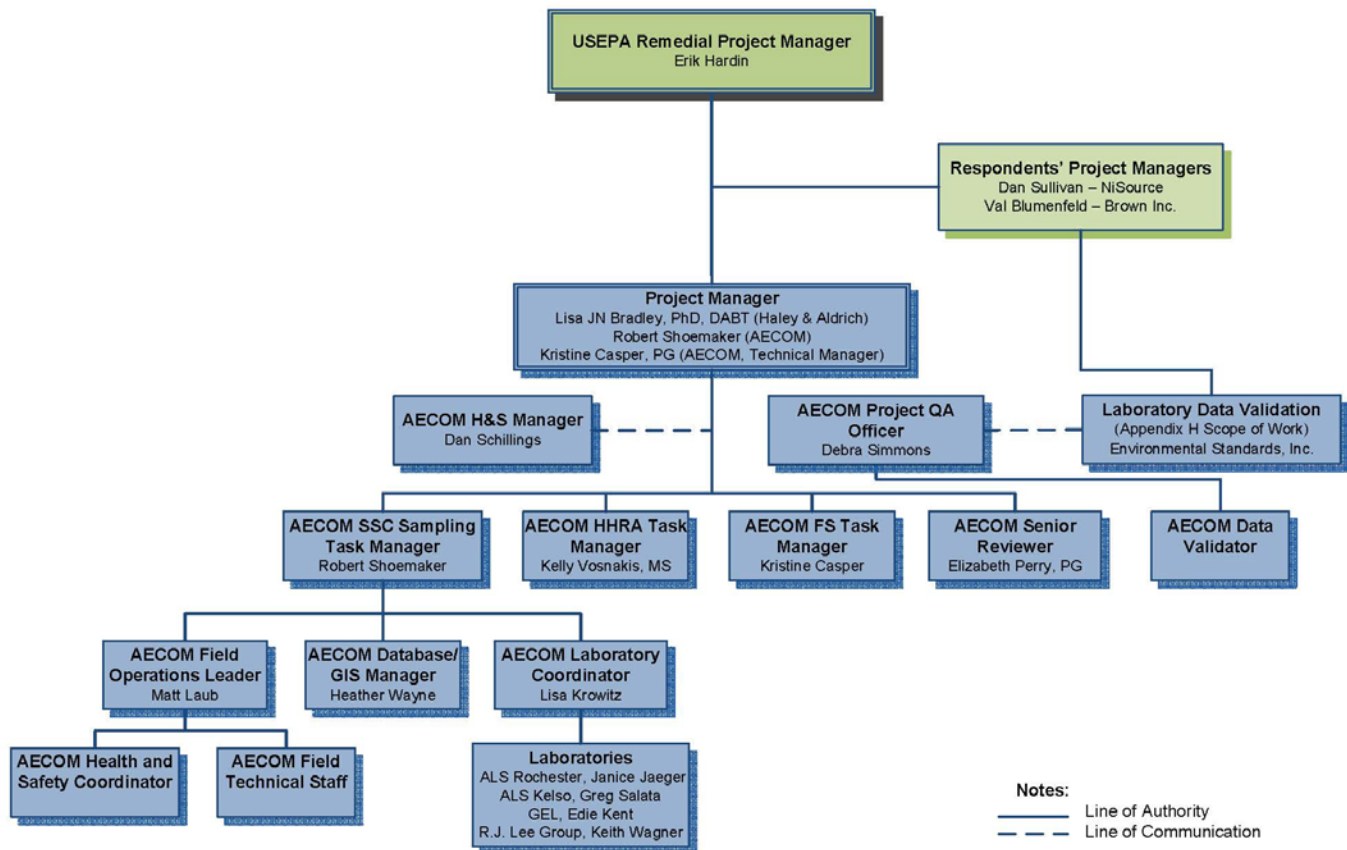
Section: Figure
 Revision: 2
 Date: November 2014
 Page: 1 of 1

Figure

Figure A-1 Project Organization Chart



**Figure A-1
 Project Organization Chart
 Quality Assurance Project Plan Addendum – Soil Investigation
 Pines Area of Investigations**



Quality Assurance Project Plan Addendum

Pines Area of Investigation

Section: Attachment A
Revision: 3
Date: May 2015

Attachment A

Laboratory Standard Operating Procedures

A-1 ALS Rochester

A-2 GEL

A-3 ALS Kelso

Quality Assurance Project Plan Addendum
Pines Area of Investigation

Section: Attachment B
Revision: 3
Date: May 2015

Attachment B

Gamma Spectroscopy Library

Quality Assurance Project Plan Addendum

Section: Attachment C
Revision: 3
Date: May 2015

Pines Area of Investigation

Attachment C

Radioisotope Calculation Spreadsheet

Quality Assurance Project Plan Addendum

Pines Area of Investigation

Section: Attachment A

Revision: 3

Date: May 2015

Attachment A

Laboratory Standard Operating Procedures

A-1 ALS Rochester

A-2 GEL

A-3 ALS Kelso

ALS Standard Operating Procedure

DOCUMENT TITLE:	DETERMINATION OF METALS AND TRACE ELEMENTS BY INDUCTIVELY COUPLED PLASMA-MASS SPECTROSCOPY (ICP-MS)
REFERENCED METHOD:	EPA 200.8, SW 6020A
SOP ID:	MET-6020A/200.8
REV. NUMBER:	2
EFFECTIVE DATE:	11/4/2013

SOP CHANGE FORM

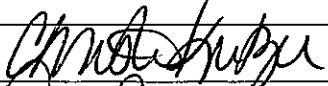
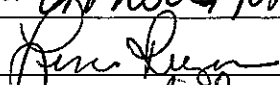
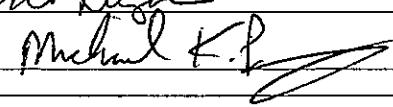
SOP Title: Determination of Metals and Trace Elements by ICP-MS
SOP Code: MET-6020A/200.8
SOP Revision No.: 2
SOP Date: 11/4/13
SOP Section(s) Affected by Change: 11.1, Appendix

<p>Description of Change:</p> <p>Add to 11.1: "The daily operating and evaluation criteria are discussed in 11.3-11.6. An example of further instrument conditions is found in the Appendix of this SOP."</p> <p>Add the attached to the Appendix.</p>
--

<p>Reason(s) for Change(s):</p> <p>PJLA Audit</p>

Change(s) Submitted by: Vicky Collom	Date: 12/5/13
--------------------------------------	---------------

Approvals:

Technical Reviewer Signature: 	Date: 12/5/13
QA Signature: 	Date: 12/5/13
Laboratory Director Signature: 	Date: 12/9/13

Change(s) Effective Date:

Distribution: Original filed with original SOP

Photocopy attached to each controlled copy

Quantitative Method Report

File Name: 6020a.mth
File Path: c:\elandata\Method\6020a.mth

Timing Parameters

Sweeps/Reading: 10
Readings/Replicate: 1
Number of Replicates: 3
Tuning File: epa.tun
Optimization File: epa2.dac
QC Enabled: Yes
Settling Time: Normal

	Analyte	Mass	Scan Mode	MCA Channels	Dwell Time	Integration Time
>	Li	6.015	Peak Hopping	1	100.0 ms	1000 ms
[Be	9.012	Peak Hopping	1	100.0 ms	1000 ms
[Al	26.982	Peak Hopping	1	100.0 ms	1000 ms
>	Sc	44.956	Peak Hopping	1	100.0 ms	1000 ms
	V	50.944	Peak Hopping	1	100.0 ms	1000 ms
	Cr	51.941	Peak Hopping	1	100.0 ms	1000 ms
[Cr	52.941	Peak Hopping	1	100.0 ms	1000 ms
[Mn	54.938	Peak Hopping	1	100.0 ms	1000 ms
	Co	58.933	Peak Hopping	1	100.0 ms	1000 ms
	Ni	59.933	Peak Hopping	1	100.0 ms	1000 ms
	Ni	61.928	Peak Hopping	1	100.0 ms	1000 ms
	Cu	62.930	Peak Hopping	1	100.0 ms	1000 ms
	Cu	64.928	Peak Hopping	1	100.0 ms	1000 ms
	Zn	65.926	Peak Hopping	1	100.0 ms	1000 ms
	Zn	66.927	Peak Hopping	1	100.0 ms	1000 ms
	Zn	67.925	Peak Hopping	1	100.0 ms	1000 ms
>	Ge	71.922	Peak Hopping	1	100.0 ms	1000 ms
	As	74.922	Peak Hopping	1	100.0 ms	1000 ms
	Se	76.920	Peak Hopping	1	500.0 ms	5000 ms
	Se	81.917	Peak Hopping	1	500.0 ms	5000 ms
	Kr	82.914	Peak Hopping	1	100.0 ms	1000 ms
	Sr	85.909	Peak Hopping	1	100.0 ms	1000 ms
	Sr	87.906	Peak Hopping	1	100.0 ms	1000 ms
[Y	88.905	Peak Hopping	1	100.0 ms	1000 ms
[Mo	94.906	Peak Hopping	1	100.0 ms	1000 ms
	Mo	96.906	Peak Hopping	1	100.0 ms	1000 ms
	Mo	97.906	Peak Hopping	1	100.0 ms	1000 ms
	Rh	102.905	Peak Hopping	1	100.0 ms	1000 ms
	Ag	106.905	Peak Hopping	1	100.0 ms	1000 ms
	Ag	108.905	Peak Hopping	1	100.0 ms	1000 ms
	Cd	110.904	Peak Hopping	1	100.0 ms	1000 ms

	Cd	113.904	Peak Hopping	1	100.0 ms	1000 ms
>	In	114.904	Peak Hopping	1	100.0 ms	1000 ms
	Sb	120.904	Peak Hopping	1	100.0 ms	1000 ms
	Sb	122.904	Peak Hopping	1	100.0 ms	1000 ms
	Ba	134.906	Peak Hopping	1	100.0 ms	1000 ms
	Ba	136.905	Peak Hopping	1	100.0 ms	1000 ms
>	Tb	158.925	Peak Hopping	1	100.0 ms	1000 ms
>	Ho	164.930	Peak Hopping	1	100.0 ms	1000 ms
	Tl	202.972	Peak Hopping	1	100.0 ms	1000 ms
	Tl	204.975	Peak Hopping	1	100.0 ms	1000 ms
	Pb	207.977	Peak Hopping	1	100.0 ms	1000 ms
	Pb	205.975	Peak Hopping	1	100.0 ms	1000 ms
	Pb	206.976	Peak Hopping	1	100.0 ms	1000 ms
>	Bi	208.980	Peak Hopping	1	100.0 ms	1000 ms
	Th	232.038	Peak Hopping	1	100.0 ms	1000 ms
	U	238.050	Peak Hopping	1	100.0 ms	1000 ms

Signal Processing

Detector Mode: Dual
 Measurement Units: Cps
 AutoLens: On
 Spectral Peak Processing: Maximum
 Signal Profile Processing: Maximum
 Blank Subtraction: After Internal Standard
 Baseline Readings: 0
 Smoothing: Yes, Factor 5

Equations

Analyte	Mass	Corrections
V	50.944	-3.127*(ClO 53-(0.113*Cr 52))
As	74.922	-3.127*(ArCl77-(0.873*Se82))
Se	81.917	-1.0087*Kr83
Sr	85.909	- 1.505657 * Kr 83
Mo	97.906	- 0.110588 * Ru 101
Cd	110.904	-1.073*(MoO 108 - (0.712*Pd 106))
Cd	113.904	- 0.026826 * Sn 118
In	114.904	- 0.014032 * Sn 118
Sb	122.904	- 0.127189 * Te 125
Pb	207.977	+1*Pb 206 + 1*Pb 207

Calibration Information

Analyte	Mass	Curve Type	Sample Units	Std Units	Std 1	Std 2	Std 3	Std 4
Li	6.015	Linear Thru Zero	ug/L	ug/L				
Be	9.012	Linear Thru Zero	ug/L	ug/L	1	20	100	
Al	26.982	Linear Thru Zero	ug/L	ug/L	10	20	100	
Sc	44.956	Linear Thru Zero	ug/L	ug/L				

V	50.944	Linear Thru Zero	ug/L	ug/L	2	20	100
Cr	51.941	Linear Thru Zero	ug/L	ug/L	2	20	100
Cr	52.941	Linear Thru Zero	ug/L	ug/L	2	20	100
Mn	54.938	Linear Thru Zero	ug/L	ug/L	1	20	100
Co	58.933	Linear Thru Zero	ug/L	ug/L	1	20	100
Ni	59.933	Linear Thru Zero	ug/L	ug/L	1	20	100
Ni	61.928	Linear Thru Zero	ug/L	ug/L	1	20	100
Cu	62.930	Linear Thru Zero	ug/L	ug/L	1	20	100
Cu	64.928	Linear Thru Zero	ug/L	ug/L	1	20	100
Zn	65.926	Linear Thru Zero	ug/L	ug/L	5	20	100
Zn	66.927	Linear Thru Zero	ug/L	ug/L	5	20	100
Zn	67.925	Linear Thru Zero	ug/L	ug/L	5	20	100
Ge	71.922	Linear Thru Zero	ug/L	ug/L			
As	74.922	Linear Thru Zero	ug/L	ug/L	1	20	100
Se	76.920	Linear Thru Zero	ug/L	ug/L	2	20	100
Se	81.917	Linear Thru Zero	ug/L	ug/L	2	20	100
Kr	82.914	Linear Thru Zero	ug/L	ug/L			
Sr	85.909	Linear Thru Zero	ug/L	ug/L	1	20	100
Sr	87.906	Linear Thru Zero	ug/L	ug/L	1	20	100
Y	88.905	Linear Thru Zero	ug/L	ug/L			
Mo	94.906	Linear Thru Zero	ug/L	ug/L	1	20	100
Mo	96.906	Linear Thru Zero	ug/L	ug/L	1	20	100
Mo	97.906	Linear Thru Zero	ug/L	ug/L	1	20	100
Rh	102.905	Linear Thru Zero	ug/L	ug/L			
Ag	106.905	Linear Thru Zero	ug/L	ug/L	1	20	100
Ag	108.905	Linear Thru Zero	ug/L	ug/L	1	20	100
Cd	110.904	Linear Thru Zero	ug/L	ug/L	1	20	100
Cd	113.904	Linear Thru Zero	ug/L	ug/L	1	20	100
In	114.904	Linear Thru Zero	ug/L	ug/L			
Sb	120.904	Linear Thru Zero	ug/L	ug/L	1	20	100
Sb	122.904	Linear Thru Zero	ug/L	ug/L	1	20	100
Ba	134.906	Linear Thru Zero	ug/L	ug/L	1	20	100
Ba	136.905	Linear Thru Zero	ug/L	ug/L	1	20	100
Tb	158.925	Linear Thru Zero	ug/L	ug/L			
Ho	164.930	Linear Thru Zero	ug/L	ug/L			
Tl	202.972	Linear Thru Zero	ug/L	ug/L	1	20	100
Tl	204.975	Linear Thru Zero	ug/L	ug/L	1	20	100
Pb	207.977	Linear Thru Zero	ug/L	ug/L	1	20	100
Pb	205.975	Linear Thru Zero	ug/L	ug/L	1	20	100
Pb	206.976	Linear Thru Zero	ug/L	ug/L	1	20	100
Bi	208.980	Linear Thru Zero	ug/L	ug/L			
Th	232.038	Linear Thru Zero	ug/L	ug/L	1	20	100
U	238.050	Linear Thru Zero	ug/L	ug/L	1	20	100

Analyte	Mass	Std 5	Std 6	Std 7	Std 8	Std 9	Std 10	Std 11	Std 12
Li	6.015								
Be	9.012								

Al	26.982
Sc	44.956
V	50.944
Cr	51.941
Cr	52.941
Mn	54.938
Co	58.933
Ni	59.933
Ni	61.928
Cu	62.930
Cu	64.928
Zn	65.926
Zn	66.927
Zn	67.925
Ge	71.922
As	74.922
Se	76.920
Se	81.917
Kr	82.914
Sr	85.909
Sr	87.906
Y	88.905
Mo	94.906
Mo	96.906
Mo	97.906
Rh	102.905
Ag	106.905
Ag	108.905
Cd	110.904
Cd	113.904
In	114.904
Sb	120.904
Sb	122.904
Ba	134.906
Ba	136.905
Tb	158.925
Ho	164.930
Tl	202.972
Tl	204.975
Pb	207.977
Pb	205.975
Pb	206.976
Bi	208.980
Th	232.038
U	238.050

AS Pos	Sample Flush	Sample Flush	Read Delay	Read Delay	Wash	Wash
--------	--------------	--------------	------------	------------	------	------

Blank	1	20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 1	2	20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 2	3	20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 3	4	20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 4		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 5		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 6		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 7		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 8		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 9		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 10		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 11		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 12		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 13		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 14		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 15		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 16		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 17		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 18		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 19		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 20		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 21		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 22		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 23		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 24		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 25		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 26		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 27		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 28		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 29		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm
Standard 30		20 s	-36 rpm	30 s	-26 rpm	120 s	-36 rpm

Reporting Options

Report Template for Printing: c:\elandata\ReportOptions\6020a qc report.rop
Send to Printer: Yes
Report Template for File: c:\elandata\ReportOptions\lims export.rop
Send to File: Yes
Report Filename: i:\instdata\metals\elan_xf\120213.rep
Create NetCDF File: No
Send to Serial Port: No
Port: COM1

Sampling Devices

Peristaltic Pump Control: Yes
Autosampler: AS-93plus
Autosampler Tray File: c:\elandata\Autosampler\as-93\as93f.try

Sampling Device Type: None
Dil. Factor: 10
Dil. to Vol. (mL): 10
1st Dil. Pos.: 1
Probe Purge Pos.: 10

FIAS Program

Step	Read	Time	Pump 1	Pump 2	Valve	A/S Loc.	Sw 2	Sw 3	Sw 4
------	------	------	--------	--------	-------	----------	------	------	------

Repeat Statement

HGA Program

Description:

Sample Volume: uL

Injection Temperature: C

Injection Speed:

Read delay: s

Closure delay: s

Modifier #1:

Modifier #2:

Step	Cell Temp	Ramp	Hold	Int. Flow	Gas Norm.	Gas Alt.	To Vent	To ICP	Read
------	-----------	------	------	-----------	-----------	----------	---------	--------	------

Pipet Seq.	Mod#1	Mod#2	Sample	Start Step	Wash	Rep. From	End S Wash	Rep. To	# R
------------	-------	-------	--------	------------	------	-----------	------------	---------	-----



STANDARD OPERATING PROCEDURE

Metals by ICP-MS
MET-6020A/200.8, Rev 2
Effective: 11/4/2013
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DETERMINATION OF METALS AND TRACE ELEMENTS BY INDUCTIVELY COUPLED PLASMA-MASS SPECTROSCOPY (ICP-MS) EPA 200.8, SW 6020A

SOPID:	MET-6020A/200.8	Rev. Number:	2	Effective Date:	11/4/2013
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Approved By:

Department Supervisor - Christine Kutzer

Date: 11/1/13

Approved By:

QA Manager - Lisa Reyes

Date: 11/1/2013

Approved By:

Laboratory Director - Michael Perry

Date: 11/1/13

Archival Date: _____

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STANDARD OPERATING PROCEDURE

Metals by ICP-MS
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DETERMINATION OF METALS AND TRACE ELEMENTS BY INDUCTIVELY COUPLED PLASMA-MASS SPECTROSCOPY (ICP-MS)

1) Scope and Applicability

This SOP uses EPA SW846 Method 6020A for the determination of the concentrations of certain elements in water, soil, aqueous and non-aqueous wastes, and sediment samples. This SOP also uses EPA 200.8 for the determination of the concentrations of certain elements in drinking water, ground water, and surface water. The scope of this document does not allow for the in-depth descriptions of the relevant spectroscopic principles required for gaining a complete level of competence in this scientific discipline.

The analytes and reporting limits are listed in Table 1.

2) Summary of Procedure

- 2.1 Prior to analysis, samples must be digested using appropriate sample preparation methods unless otherwise specified by the client. The preparation for soils is described in MET-3050 and the preparation for waters is described in MET-CLP-DIG. The digestate is analyzed for the elements of interest using ICP spectrometry.
- 2.2 This method is for the multi-elemental determination of analytes by ICP-MS. The method measures ions produced by a radio-frequency inductively coupled plasma. Analyte species originating in a liquid are nebulized and the resulting aerosol transported by argon gas into the plasma torch. The ions produced are entrained in the plasma gas and introduced, by means of an interface, into a mass spectrometer. The ions produced in the plasma are sorted according to their mass-to-charge ratios and quantified with a channel electron multiplier. Interferences must be assessed and valid corrections applied or the data flagged to indicate problems. Interference correction must include compensation for background ions contributed by the plasma gas, reagents, and constituents of the sample matrix.
- 2.3 Method Modifications
 - 2.3.1 The concentration of interfering elements in the ICSA and ICSAB solutions are spiked at levels recommended by the manufacturer due to instrument sensitivity.
 - 2.3.2 6020A says that IDLs should be determined every 3 months to monitor noise and response changes. Blanks and MRL standards are analyzed with every run and can be used for monitoring.



STANDARD OPERATING PROCEDURE

Metals by ICP-MS
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3) Definitions

- 3.1 Initial Calibration - analysis of analytical standards for a series of different specified concentrations; used to define the linearity and dynamic range of the response of the detector to the element.
- 3.2 Calibration Standard (CAL) - A solution prepared from the dilution of stock standard solutions. The CAL solutions are used to calibrate the instrument response with respect to analyte concentration.
- 3.3 Dissolved Analyte - The concentration of analyte in an aqueous sample that will pass through a 0.45 μm membrane filter assembly prior to sample acidification.
- 3.4 Initial Calibration Verification (ICV) (also called Second Source Calibration Verification) - ICV solutions are made from a stock solution which is different from the stock used to prepare calibration standards and is used to verify the validity of the initial calibration.
- 3.5 Continuing Calibration Verification Standard (CCV) - A standard analyzed at specified intervals and used to verify the ongoing validity of the instrument calibration. For the LLCCV, see MRL Standard below.
- 3.6 Matrix Spike (MS) - An aliquot of an environmental sample to which known quantities of all of the elements of client interest are added to a sample matrix prior to sample digestion and analysis. The purpose of the matrix spike is to evaluate the effects of the sample matrix on the methods used for the analyses. Percent recoveries are calculated for each of the analytes detected.
- 3.7 Duplicate (DUP) - A laboratory duplicate. Two aliquots of the same sample taken in the laboratory and analyzed separately with identical procedures. Analysis of duplicates indicates precision associated with laboratory procedures, but not with sample collection, preservation, or storage procedures.
- 3.8 Laboratory Control Sample (LCS) - A matrix blank spiked with all of the elements of client interest. The LCS is analyzed exactly like a sample, and its purpose is to determine whether the methodology is in control and whether the laboratory is capable of making accurate measurements. LCS-Soil is purchased from a vendor.
- 3.9 Method Blank (MB) - The MB is an artificial sample designed to monitor introduction of artifacts into the process. The method blank is carried through the entire preparation and analytical procedure.
- 3.10 Instrument Blank (ICB/CCB) - The instrument blank (also called initial or continuing calibration blank) is a volume of reagent water acidified with the same acid matrix as in the calibration standards. This blank is the zero standard and has a reagent composition identical to the digestates. The purpose of the ICB/CCB is to determine the levels of contamination associated with the instrumental analysis.
- 3.11 Batch - A group of no more than 20 samples analyzed together on the same day with the same reagents. See the SOP for Batches and Sequences (ADM-BATCH) for more detail.
- 3.12 Calibration Blank - A volume of reagent water acidified with the same acid matrix as in the calibration standards. The calibration blank is a zero standard and is used to the calibrate the instrument.



STANDARD OPERATING PROCEDURE

Metals by ICP-MS
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- 3.13 Internal Standard - Pure analyte(s) added to a sample, extract, or standard solution in known amount(s) and used to measure the relative responses of other method analytes that are components of the same sample or solution. The internal standard must be an analyte that is not a sample component.
- 3.14 Linear Range - The concentration range over which the instrument response to an analyte is linear.
- 3.15 Limit of Detection (LOD) - An estimate of the minimum amount of a substance that an analytical process can reliably detect. An LOD is analyte- and matrix-specific and may be laboratory - dependent. For DOD, the smallest amount or concentration of a substance that must be present in a sample in order to be detected at a high level of confidence (99%). At the LOD, the false negative rate (Type II error) is 1%.
- 3.16 Limit of Quantitation (LOQ) - The minimum levels, concentrations, or quantities of a target that can be reported with a specified degree of confidence. For DOD, the lowest concentration that produces a quantitative result within specified limits of precision and bias. The LOQ shall be set at or above the concentration of the lowest initial calibration standard.
- 3.17 Interference Check Solution (ICS) - A solution of selected method analytes of higher concentrations which is used to evaluate the procedural routine for correcting known interelement spectral interferences with respect to a defined set of method criteria. This is used for 6020A.
- 3.18 MRL Standard - Standard prepared with a known concentration of elements to check accuracy at the quantitation limit. This is also known as Low-level calibration check standard (LLCCV or LLICV). This standard is not digested.
- 3.19 LOQ Standard - also called LLQC - Standard prepared at the LOQ that undergoes digestion and preparation procedures.
- 3.20 HLCCV2 - A standard prepared slightly higher than the calibration range for metals. This is an "upper range limit" standard used to verify the upper limit of the linear dynamic range of the instrument.
- 3.21 Tune - The analysis of a standard element to verify that the mass spectrometer meets standard mass spectra abundance criteria prior to sample analysis.
- 3.22 Matrix - the predominant material, component, or substrate (e.g., surface water, soil, etc.) of which the sample to be analyzed is composed.

4) Health and Safety Warnings

- 4.1 All appropriate safety precautions for handling reagents and samples must be taken when performing this procedure. This includes the use of personal protective equipment, such as safety glasses, lab coat and the correct gloves.
- 4.2 Chemicals, reagents and standards must be handled as described in the Company safety policies, approved methods and in MSDSs where available. Refer to the Environmental, Health and Safety Manual and the appropriate MSDS prior to beginning this method.
- 4.3 Hydrochloric and Nitric Acid are used in this method. These acids are extremely corrosive and care must be taken while handling them. A face shield should be used while pouring acids and safety glasses should be worn while working with the solutions. Lab coat and gloves should always be worn while working with these solutions.



STANDARD OPERATING PROCEDURE

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-
- 4.4 The use of pressurized gases is required for this procedure. Care should be taken when moving cylinders. All gas cylinders must be secured to a wall or an immovable counter with a chain or a cylinder clamp at all times. Sources of flammable gases (e.g., pressurized hydrogen) should be clearly labeled.
- 4.5 High Voltage - The RF generator supplies up to 2000 watts to maintain an ICP. The power is transferred through the load coil located in the torch box. Contact with the load coil while generator is in operation will likely result in death. When performing maintenance on the RF generator, appropriate grounding of all HV capacitors must be performed as per manufacturer.
- 4.6 UV Light - The plasma is an intense source of UV emission, and must not be viewed with the naked eye. Protective lenses are in place on the instrument. Glasses with special protective lenses are available when direct viewing of the plasma is necessary.
- 4.7 Refer to the Safety Manual for further discussion of general safety procedures and information.
- 4.8 Waste Management and Pollution Prevention
- It is the laboratory's practice to minimize the amount of acids and reagent used to perform this method wherever feasible. Standards are prepared in volumes consistent with methodology and only the amount needed for routine laboratory use is kept on site. The threat to the environment from solvent and reagents used in this method can be minimized when disposed of properly.
- The laboratory will comply with all Federal, State and local regulations governing waste management, particularly the hazardous waste identification rules and land disposal restrictions as specified in the EH&S Manual.
- Samples with analyte concentrations exceeding TCLP regulatory limits are disposed of as hazardous waste, see the SOP for Sample Disposal (SMO-SPLDIS).

5) Cautions

Preventive maintenance activities listed below should be performed when needed as determined by instrument performance (i.e. stability, sensitivity, etc.) or by visual inspection. Other maintenance or repairs may, or may not require factory service, depending on the nature of the task.

- cone removal and cleaning
- removal and cleaning of ICP glassware and fittings
- checking air filters and cleaning if necessary
- checking the rotary pump oil and adding or changing if necessary
- removal and cleaning of extraction lens
- removal and cleaning of ion lens stack



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6) Interferences

- 6.1 Isobaric elemental interferences in ICP-MS are caused by isotopes of different elements forming atomic ions with the same nominal mass-to-charge ratio (m/z). A data system must be used to correct for these interferences. This involves determining the signal for another isotope of the interfering element and subtracting the appropriate signal from the analyte isotope signal. Attention should be given to circumstances where very high ion currents at adjacent masses may contribute to ion signals at the mass of interest. Matrices exhibiting a significant problem of this type may require resolution improvement, matrix separation, or analysis using another isotope.
- 6.2 Isobaric molecular and doubly-charged ion interferences in ICP-MS are caused by ions consisting of more than one atom or charge, respectively. Most isobaric interferences that could affect ICP-MS determinations have been identified in the literature. Refer to Method 6020A for further discussion.
- 6.3 Physical interferences are associated with the sample nebulization and transport processes as well as with ion-transmission efficiencies. Nebulization and transport processes can be affected if a matrix component causes a change in surface tension or viscosity. Changes in matrix composition can cause significant signal suppression or enhancement. Total solid levels below 2000 mg/L have been recommended to minimize solid deposition on the nebulizer tip. An internal standard can be used to correct for physical interferences, if it is carefully matched to the analyte so that the two elements are similarly affected by matrix changes.
- 6.4 Memory interferences can occur when there are large concentration differences between samples or standards which are analyzed sequentially. The rinse period between samples must be long enough to eliminate these interferences.

7) Personnel Qualifications and Responsibilities

- 7.1 It is the responsibility of the analyst to perform the analysis according to this SOP and to complete all documentation required for data review. Analysis and interpretation of the results are performed by personnel in the laboratory who have demonstrated the ability to generate acceptable results utilizing this SOP. This demonstration is in accordance with the training program of the laboratory. Final review and sign-off of the data is performed by the department supervisor/manager or designee.
- 7.2 Training – see CE-QA003.



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8) Sample Collection, Containers, Preservation, and Storage

- 8.1 Solid samples require no preservation prior to analysis other than storage at 0-6°C. Sample containers may be glass or plastic. When the laboratory provides the sample containers, the containers are glass soil jars with Teflon-lined lids.
- 8.2 For aqueous samples, glass or plastic sample containers are acceptable. When the laboratory provides the sample containers, the containers are certified clean 500 mL plastic bottles. Sample volume should be acid preserved with (1+1) nitric acid to pH <2. Samples are held at room temperature (although refrigeration is acceptable also).
- 8.3 For the determination of the dissolved elements, the sample must be filtered through a 0.45 µm pore diameter membrane filter at the time of collection or as soon thereafter as practically possible. (Glass or plastic filtering apparatus are recommended to avoid possible contamination. Only plastic apparatus should be used when the determinations of boron and silica are critical.) Samples should be filtered prior to acidification, because acidification changes the sample (usually by dissolving particulates that would be filtered out). Acidified samples received at the lab that require lab filtration must be noted in the case narrative. See digestion SOPs for filtration procedure.
- 8.4 Samples are generally received in the ICP-MS laboratory as 1% Nitric Acid digestates. Digestates are stored at room temperature in hotblock cups or B-cups. There is no specific holding time from digestion to analysis. Client samples must be analyzed within 6 months of collection.
- 8.5 Following analysis, digestates are stored until all results have been reviewed. Digestates are diluted and disposed of through the sewer system in approximately 90 days after receipt of sample.
- 8.6 For more information about custody, sample handling, and storage procedures, see SMO-GEN and SMO-ICOC.

9) Equipment and Supplies

Instrument ID	Instrument Configuration	Manufacturer Part	Serial Number	Year Acquired
ICPMS (R-ICP-MS-01)	SCIEX ICP/MS	Perkin Elmer Elan 9000	PO370203	2002
	Autosampler	PE AS93Plus		
	Computer Workstation	Dell Optiplex GX150		
	Analytical Software	ELAN v.2.4		

The system has a mass range from at least 6 to 240 amu and a data system which allows corrections for isobaric interferences and the application of the internal standard technique.

- 9.1 Peristaltic Pump and pump tubing
- 9.2 15 and 50 mL autosampler tubes
- 9.3 Pipettes (Eppendorf) – calibrated according to ADM-PCAL.
- 9.4 Volumetric flasks, class A



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10) Standards and Reagents

10.1 Standards Preparation General Information and Disclaimers

All of the preparation instructions are general guidelines. Other technical recipes may be used to achieve the same results. Example – a 20 mg/L standard may be made by adding 1 mL of 200 mg/L to 10 mLs or may be made by adding 4 mL of 50 mg/L to 10 mL. The preparation depends upon the final volume needed and the initial concentration of the stock. Reasonable dilution technique is used.

Vendors and vendors' products are sometimes listed for the ease of the analyst using this SOP, but products and purchased concentrations are examples only and subject to change at any time. All purchased standards are certified by the vendor. Certificates of Analysis are kept in the department until the standards are no longer being used – at which time they are filed with QA. Certificates of Analysis are available upon request. Purchased standards are routinely checked against an independent source for both analyte identification and analyte concentration.

All Standards must be traceable using the laboratory lot system (CE-QA007).

All preparatory information for the standards and QC samples are provided in the Controlled Forms section of the Rochester Intranet in the Metals Standards Logbooks. They are all stored in plastic at room temperature.

10.2 Argon – 99.99% purity

10.3 Trace metals grade chemicals shall be used in all tests. Each lot of acid used is to be analyzed to demonstrate that it is free of interference before use (CE-GEN007). All acids are stored at room temperature. Acids expire per the recommendations in Expiration Policy (CE-QA012) if no other indication is given.

- Hydrochloric acid (conc.), HCl. Purchased commercially
- Nitric acid (conc.), HNO₃. Purchased commercially.

10.4 Reagent Water. All references to water in this SOP refer to the water produced by the laboratory water system.

10.5 Standards

10.5.1 Mixed Calibration Standards are prepared by combining appropriate volumes of the stock solutions in volumetric flasks. Matrix match with the appropriate acid and dilute to 500 mL with water. Calibration standards should be verified using a second source quality control sample (LCS or ICV). Calibration standards expire in 1 month.

10.5.2 Initial Calibration Verification (ICV) Standard is prepared by combining compatible analytes at concentrations equivalent to the midpoint of their respective calibration curves. The ICV standard should be prepared from a separate source independent from that used in the calibration standards. ICV standard expires in 48 hours.

10.5.3 Continuing Calibration Verification (CCV) Standard is prepared by combining compatible analytes at concentrations equivalent to the midpoint of their respective calibration curves. The CCV standard is prepared from a separate source as that used in the calibration standards. CCV standard expires in 48 hours.

10.5.4 MRL Standard is prepared to contain known concentrations of elements at the MRL. MRL standard expires in 1 month.



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- 10.5.5 Interference Check Solutions A and AB are prepared to contain known concentrations of interfering analytes that will provide an adequate test of the correction factors. Used for 6020A only. ICSA / ICSAB standards expire in 1 week.
- 10.5.6 The spiking solutions for the Laboratory Control Sample and Matrix Spike are purchased as custom mixes at the concentrations recommended in the method. Solutions expire per manufacturer recommended expiration date. The LCS and MS are spiked at digestion according to the preparation SOP. All target analytes are spiked in the LCS and MS.
- 10.5.7 Internal Standards Solutions-high purity grade solutions are purchased and diluted to final concentration of 1000-5000 ppb. Diluted solutions expire in 6 months.
- 10.6 Blanks
- 10.6.1 Method Blanks must contain all the reagents and in the same volumes as used in the preparation of samples. The method blanks are blank matrix samples carried through the complete procedure and contain the same acid concentration in the final solution as the samples.
- 10.6.2 The Calibration Blank is prepared by acidifying reagent water to the same concentrations of acid found in the standards and samples.

11) Method Calibration

- 11.1 Refer to method 6020A (Section 11.0) and the instrument manuals for detailed instruction on implementation of the following procedures preceding an analytical run. All of the following are done daily prior to initial calibration unless otherwise specified.
- 11.2 Initiate plasma and allow a warm up of at least 30 minutes. The tuning procedures may be carried out during warm-up.
- 11.3 Open the EPA Tune Method
- 11.3.1 Aspirate 10 ppb tuning solution
- 11.3.2 Click on Tune Mass Spec button in Tuning Window
- 11.3.3 Check mass calibration. Measured mass must be within 0.05 amu of actual mass. The resolution must be <0.9 amu full width at 10% peak height. Save and print tune. Put one copy of the tune with the analytical run and another copy in the tune binder.
- 11.4 Open EPA Daily Method
- 11.4.1 Aspirate the 10 ppb tuning solution
- 11.4.2 Click on the Analyze button to acquire
- 11.4.3 Check that the RSDs for the five replicates are less than 5%
- 11.4.4 Monitor daily performance measures as recommended by Perkin Elmer for Rh sensitivity, background, % double charged, and % oxide levels.
- Rh>150000 cps for 10 ppb
 - Background @ mass 220<30cps (once initial operating conditions have changed, i.e. voltage on detector, RF power, etc, background should be <100 cps. Ensure that desired signal to noise ratio is achieved.)



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- % double charged <4%
- % oxides <4%

Oxides and double charged levels can be reduced by slightly decreasing the nebulizer flow rate.

- 11.4.5 If the tune and Daily Performance checks do not meet criteria, retune the instrument and reanalyze tuning solution. The tune and Daily must pass before proceeding.
- 11.5 Open the EPA Lens Calibration method and perform the autolens calibration using 10 ppb tuning solution. This is done as needed.
 - 11.5.1 Clear old calibration
 - 11.5.2 Get analytes
 - 11.5.3 Optimize (Takes about 6 minutes)
 - 11.5.4 Save file
- 11.6 Open the Neblens power oxides workspace and optimize the following parameter using the 10 ppb tuning solution containing Be, Mg, Co, Rh, In, Ba, Ce, Pb
 - 11.6.1 Set RF power to desired level
 - 1500 watts
 - 11.6.2 Optimize the nebulizer argon flow – done as needed
 - 11.6.3 Optimize the static lens voltage – done as needed
 - 11.6.4 Save the optimization file
- 11.7 Open the EPA 6020 Dual Detector calibration method and aspirate a solution twice the calibration range. All elements plus internal standard elements should be present.

Note: This procedure is required when detector voltages or a new detector is installed. The laboratory does periodically.

 - 11.7.1 Clear old calibration
 - 11.7.2 Get analytes
 - 11.7.3 Optimize (Takes about 20 minutes)
 - 11.7.4 Save file
- 11.8 Initial Calibration – follow ADM-ICAL. If these instructions conflict with ADM-ICAL, follow the instructions in this SOP.
 - 11.8.1 Number of Calibration standards – 3 standards and a calibration blank are analyzed daily before samples or QC. The correlation coefficient must be greater than 0.998 for each analyte. If the correlation is less than 0.998, recalibrate the instrument prior to analyzing any samples.
 - 11.8.2 The use of internal standards, ICAL calculation, and sample calculation is in the Calculation Section.



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12) Sample Preparation and Analysis

- 12.1 Digest samples with the appropriate digestion method. For waters, use MET-CLPDIG. For soils, use MET-3050. Digestates originating from soil samples are diluted prior to instrumental analysis to avoid interferences and allow the analysis to achieve maximum sensitivity which results in optimum Reporting Limits.
- 12.2 Analytical Run
 - 12.2.1 For soil digestates - Dilute 1 mL of digestate to 5 mL with matrix matched reagent water prior to internal standard addition and analysis.
 - 12.2.2 Each 5 mL aliquot of blanks, standards, samples, and sample dilutions are spiked with 50 uL of internal standard solution prior to analysis. Load samples on the autosampler according to the analytical sequence.
 - 12.2.3 Open desired method. Enter sample information. Enter pump control speeds for all samples and save sample file. Re-open the sample file (this must be done for batch QC to run properly) and highlight the row numbers to be analyzed. Select "analyze batch."
 - 12.2.4 Calibration is done daily using 3 standards and a blank for each analyte. Analyze QC as per the frequency described in the QC Section. See ADM-BATCH for further detail.
- 12.3 Sample Analysis and Evaluation
 - 12.3.1 Sensitivity, LOD, LOQ, precision, linear dynamic range, and interference effects must be established for each individual analyte line on each particular instrument. All measurements must be within the instrument linear range where the correction equations are valid.
 - 12.3.2 Dilute and reanalyze samples which are above the linear range of the instrument or measure an alternate, less-abundant isotope (if calibrated). See ADM-DIL for more instruction on preparing and documenting sample dilutions.
 - 12.3.3 Evaluate QC according to the QC Section. Repeat samples associated with non-compliant QC whenever possible.
 - 12.3.4 If sample concentration for Ag is greater than 100 ug/L, sample should be redigested at a dilution until the sample solution contains less than 100 ug/L.

13) Troubleshooting -

Maintenance log - All Preventive maintenance, as well as instrument repair, should be documented in the appropriate instrument maintenance log. Most routine maintenance and troubleshooting are performed by ALS staff. The laboratory maintains a service contract with the instrument manufacturer that allows for an unlimited number of service calls and full reimbursement of all parts and labor. Any maintenance performed by outside services must also be documented - either through notes in the log or through documents provided by the service. The log entries will include the date maintenance was performed, symptoms of the problem, serial numbers of major equipment upgrades or replacements. The data file name of the first acceptable run after maintenance is to be documented in the maintenance log.

See instrument manual or maintenance log for help in solving specific analytical or instrument problems.



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14) Data Acquisition

Data is electronically transferred from the instrument software to LIMS. The preparation information entered to LIMS is used by LIMS to calculate the final results – see Calculation Section.

15) Calculations and Data Reduction Requirements

15.1 Calculate sample results using the data system printouts and digestion information. The digestion and dilution information is entered into the data system. The data system then uses the calculations below to generate a sample result.

15.2 **Aqueous samples** are reported in µg/L:

$$\mu\text{g} / \text{L} (\text{Sample}) = \frac{C^* \times \text{Dilution Factor} \times \text{FinalDigestateVolume}(\text{ml})}{\text{InitialVolumeDigested}(\text{ml})}$$

C* = Concentration of analyte as measured at the instrument in µg/L (in digestate).

15.3 **Solid samples** are reported in µg/Kg:

$$\mu\text{g/Kg} (\text{Sample}) = C^* \times \text{PostDigestionDilutionFactor} \times \frac{\text{Digestion Vol}(\text{ml})}{\text{Sample wt.}(\text{g})} \times \frac{1\text{L}}{1000\text{ml}} \times \frac{1000\text{g}}{1\text{Kg}}$$

C* = Concentration of analyte as measured at the instrument in µg/L (in digestate).

15.4 Internal standards are identified by a caret and are bracketed by elements with which they are associated.

15.5 Table 2 has the recommended Isotopes. The actual mass used is on the Quantitation report.

15.6 ELAN software assumes that all elements within a standard group are similarly affected by instrument drift or matrix interferences. Changes in measured intensity of the Internal standard are used to create the ratios for correcting measured intensities of an element. A blank (called "Blank Intensity" on quantitation report) is analyzed prior to the ICAL to establish Internal standard intensities. A ratio is calculated from the measured intensity of the sample and the blank intensity for a given Internal standard. This ratio is applied to the measured intensity of the target element and adjusts the measured intensity based on performance of the sample matrix. The adjusted intensity is used to plot the ICAL. Sample concentrations are calculated from adjusted intensities.

$$15.7 \quad \text{Adjusted Intensity} = \frac{I_B}{I_{IS}} * I_E$$

Where:

I_B = Intensity of Internal Standard in Blank Solution

I_{IS} = Intensity of Internal Standard in Measured Solution

I_E = Intensity of Element in Measured Solution



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- 15.8 The adjusted intensity of the standards is plotted against concentration using the linear regression equation of a line and forced through zero intensity and zero concentration. This calibration curve is assuming that the relationship between concentration (the X values) and intensity (the Y values) is linear and that the following equation describes this relationship:

$$Y=MX$$

Where:

X=concentration

Y= adjusted intensity

M=slope of the calibration curve

The intercept is forced to be zero.

Given 2 or more data points, the values for M are calculated using the following equation.

$$M = \frac{\sum_{i=1}^n (X_i Y_i)}{\sum_{i=1}^n (X_i^2)}$$

Where: n=number of standards (includes the blank)

In this equation, the blank is subtracted from all solutions and included in the calculation of the calibration curve

- 15.9 To avoid bias at the low end of the curve, the curve is forced through zero. Forcing the curve through zero may favor the low end of the curve therefore a MRL standard and high level CCV (HLCCV2) are analyzed to verify both ends of the curve.
- 15.10 Common isobaric interferences are corrected using equations equivalent to those listed in EPA Methods 6020A and 200.8. Monitoring of multiple isotopes for a single element provides a mechanism for identifying isobaric interferences. If an element has more than one monitored isotope, examination of the concentration calculated for each isotope will provide useful information in detecting a possible spectral interference. Consideration should therefore be given to both primary and secondary isotopes in the evaluation of the sample concentration. In some cases, secondary isotopes may be less sensitive or more prone to interferences than the primary recommended isotopes, therefore differences between the results do not necessarily indicate a problem with data calculated for the primary isotopes. Refer to the Interferences section of EPA Method 6020A and 200.8 for additional descriptions of possible interferences and the mechanisms required for adequately compensating for



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their effects.

- 15.11 The ArCl correction equation adds artificial signal to the As result when selenium is high. Check to see if there is ArCl interference on As by comparing 52Cr and 53Cr (ClO interferes on 53Cr). If they look about the same, turn the correction equation off and reprocess the data. The data is valid if the QC passed before the equation was turned off. If there is Cl interference, qualify the As result as estimated or report it by ICP.
- 15.12 Data is reviewed according to ADM-DREV.

16) Quality Control, Acceptance Criteria, and Corrective Action

- 16.1 Initial Calibration Verification Standard – midlevel (ICV) Analyzed immediately after the calibration using a second source standard to verify the standards in the calibration curve. The results for each analyte must agree within 10% of the expected value. If not, the analyses should be terminated and instrument recalibrated. Investigate the sources and preparation procedures of the standards.
- 16.2 Continuing Calibration Verification Standard – midlevel (CCV) Analyzed after every 10 samples and at the end of the analytical sequence. The results of the CCV must agree within $\pm 10\%$ of the expected value. If the control limits are exceeded, correct the problem and reanalyze the CCV. If that fails, recalibrate the instrument. Repeat samples bound by the out of control CCV unless the CCV is high and the sample concentration is less than the reporting limit.
- 16.3 Initial and Continuing Calibration Blank (ICB/CCB) The ICB is analyzed at the beginning of the run. The CCB is analyzed after every 10 samples (immediately following the CCV) and at the end of the run. The CCB must be less than the reporting limit (less than the LOD for DOD). If the control limits are exceeded, correct the problem and reanalyze the CCB. If that fails, recalibrate the instrument. Repeat any samples bound by the out of control CCB. If CCB results are exceeded, these associated data shall be flagged unless the sample concentration is less than the reporting limit.
- 16.4 MRL Standard (MRL) - A standard at or near the MRL is analyzed at the beginning and end of each analytical run but not before the ICV. If the MRL is at the MRL, the limit is 70-130% of the true value for 6020A. (DOD requires a limit of $\pm 20\%$). If the MRL standard is near the MRL, the limit is 80-120%. If the limits are not met the analysis is stopped and the instrument is recalibrated.
- 16.5 LOQV (LLQC) – This digested standard spiked at the reporting limit must be analyzed quarterly for DOD (6020A requires “as needed”). The limits are 70-130% for 6020A and LCS limits for DOD. If this QC does not meet these limits, determine the source of the problem. Achieve an acceptable LOQV before continuing.
- 16.6 HLCCV – analyzed once per daily run. The limit is $\pm 10\%$. If it is out of control, client data above the high ICAL standard should be re-analyzed.
- 16.7 DUP –
 - 16.7.1 Frequency for 200.8 – 1/10 or one per batch, whichever is greater.
 - 16.7.2 Frequency for 6020A – 1/20 or per batch, whichever is greater.
 - 16.7.3 Limits - The control limits are listed in the Data Quality Objectives Table. Client specific QC recoveries may supercede the limits listed in the QA manual.
 - 16.7.4 Corrective Action - If the control limits are exceeded, the data will be reported with a qualifier.



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- 16.8 MS –
- 16.8.1 Frequency for 200.8 - 1/10 or one per batch, whichever is greater
 - 16.8.2 Frequency for 6020A - 1/20 or per batch, whichever is greater
 - 16.8.3 Limits - The control limits are listed in the Data Quality Objectives Table. These limits are 75-125% for 6020A and 70-130% for 200.8. Client specific QC recoveries may supercede the limits listed in the QA manual.
 - 16.8.4 Corrective Action - If the control limits are exceeded, analyze a post-digestion spike. DOD requires project-specific DQOs be examined or contact the client for additional measures. See Table F-8.
- 16.9 LCS is prepared at a frequency of one per preparation batch, not to exceed 20 samples. The % recovery must be within the limits in the Data Quality Objectives Table. These limits are 80-120% for 6020A waters, 85-115% for 200.8, and per the Manufacturer's Certificate of Analysis for soils. Client specific QC recoveries may supercede these limits. If the control limits are exceeded, the associated batch of samples will be redigested and reanalyzed for the out of control elements or the data is to be flagged. Exception – if the LCS recovery fails high, elements which are less than the reporting limit may be reported.
- 16.10 Linear Range Study – performed every 6 months. A high level check standard (HLCCV2) must be within 5% of the expected value. If it is not, repeat the linear range study or reduce the linear range and analyze a new high level check standard.
- 16.11 Ongoing verification of the detection and quantitation limits is required. See CE-QA011 for requirements.
- 16.12 Method Blank (MB) Method Blanks must be prepared with each batch of 20 or fewer samples of the same matrix. MB values must not exceed the LOQ (1/2 LOQ for DOD). Fresh aliquots of the samples must be prepared and analyzed again for affected analytes after the source of the contamination has been corrected and acceptable MB values have been obtained. If detections are greater than the limit, the batch needs to be redigested if sample concentration is less than 10 times the concentration found in the method blank. If the sample concentration is less than the reporting limit the sample does not require redigestion.
- 16.13 Internal standards–
- 16.13.1 Limits – Evaluate the intensity of each internal standard in each standard and QC compared to the intensity of the internal standards in the initial calibration.
 - 200.8: 60-125%
 - 6020A: > 70%
 - 16.13.2 Corrective Action - If these limits are not met in the samples, verify that the instrument is not drifting by evaluating the internal standards in the CCBs and dilute the sample five fold and reanalyze with the proper addition of internal standards. Repeat this procedure until the internal standard intensities fall within the limits. If the internal standards are not acceptable for the QC, terminate the analysis, correct the problem, recalibrate, and reanalyze any associated samples.



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- 16.14 Dilution Test (serial dilution) – One dilution test must be performed for each prep batch of 20 or fewer samples. To perform the Dilution Test, choose a sample which has a concentration within the linear range of the instrument and dilute it 1/5. The result of the dilution must agree within 10% of the original sample determination when concentrations exceed 50X LOQ. If it does not, an interference effect must be suspected and should be noted in the case narrative. (DOD requires post-digestion spike addition if the dilution test fails.)
- 16.15 Post digestion spike – Perform when the MS fails or, for DOD, if no samples had concentration >50xLOQ. (For DOD, Perform when dilution test fails or analyte concentration for all samples <50 times LOQ) The spike addition should produce a concentration of 10-100 times the LOQ. The Post Digestion Spike recovery should be 75-125% of the true value. If it is not the sample must be diluted and reanalyzed to compensate for the matrix effect. Results must agree within 10% of the original determination. The use of standard-addition may also be used to compensate for this effect, or samples may be flagged.
- 16.16 Interference Check Standard (ICS) – not required for 200.8
- 16.16.1 Frequency - Solutions A and AB are analyzed at the beginning of each analytical run or every 12 hours, which ever is more frequent.
- 16.16.2 Limits –
- ICS-A - the absolute value of concentration for all non-spiked analytes must be less than the LOD (unless they are a verified trace impurity from one of the spiked analytes).
 - ICS-AB - The analytes in ICS-AB should recover within 20% of the expected value. If the analytes are not present, monitor concentration for possible interferences.
- 16.16.3 Corrective action – Terminate analysis, locate and correct the problem, reanalyze ICS, reanalyze all affected samples. If corrective action fails, qualify the associated sample data.

17) Data Records Management

See CE-GEN003 and ADM-ARCH.

18) Contingencies for Handling Out-Of-Control Data

If data is produced that is out of control and is not to be re-analyzed due to sample volume restrictions, holding times, or QC controls can not be met, flag and narrate appropriately.

19) Method Performance

- 19.1 This method was validated through single laboratory studies of accuracy and precision. Refer to the reference method for additional available method performance data.
- 19.2 Detection and Quantitation Limits are determined for all masses utilized for each type of matrix commonly analyzed. See CE-QA011 for determination and verification of detection and quantitation limits.
- 19.3 Demonstration of Capability is performed according to the SOP for Training (CE-QA003).



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20) Summary of Changes

- Updated to new ALS format
- Removed copies of standards log and referenced Controlled Forms
- Incorporated Change form for background mass typo.
- Changed the limits for internal standards.

21) References and Related Documents

- USEPA, Test Methods for Evaluating Solid Waste, SW-846, 3rd Edition, Update IV, Method 6020A, Revision 1, February 2007.
- Method 200.8 - Determination of Trace Elements in Waters and Wastes By Inductively Coupled Plasma-Mass Spectrometry, USEPA-EMSL, Revision 5.4, 1994.
- Perkin Elmer Instrument Manuals
- DOD Quality Systems Manual for Environmental Laboratories – Version 4.2, October 2010.

22) Appendix

- DOD Summary
- Table 1 Analytes and Reporting Levels
- Table 2 Recommended Isotopes for Selected Elements
- Table F-8 DOD Data Quality Objectives from QSMv4.2



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DOD SUMMARY

For work for the Department of Defense – the DOD Quality Systems Manual must be followed. The DOD Manual is based on the NELAC Standards with some additional requirements. The exact wording is in Table F-8 (attached). The following are the requirements which are different or additional to routine analysis and must be followed for DOD work:

- The CCB must be less than the LOD.
- The Method Blank must be less than ½ the reporting limit (<RL for common laboratory contaminants).
- Apply J flag to all hits between LOD and LOQ.
- The limits for LCS and MS are 80-120%. All targets are spiked and evaluated.
- The IDL must be less than the LOD. The IDL study is only required by DoD at set-up and after significant change.
- Low Level Check Standard (MRL standard) must be spiked at or below the reporting limit and it must have a recovery of 80-120% of the true value when analyzed at the beginning of the run.
- ICS-A absolute value of concentration for all non-spiked analytes must be <LOD (unless they are a verified trace impurity from one of the spiked analytes).
- Serial Dilution Test
- Only for samples >50 times LOQ
- Corrective Action – If the limits are not met, perform a post digestion spike addition.
- Post digestion spike – Perform when dilution test fails or analyte concentration for all samples <50 times LOQ.
- Method of Standard Additions – use when matrix interference is suspected. Document use in the case narrative.



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TABLE 1

Analyte	MRL / LOQ (ug/L)	MRL/LOQ (ug/kg)
Antimony	1	0.50
Arsenic	1	0.50
Barium	1	0.50
Beryllium	1	0.50
Cadmium	1	0.50
Chromium	2	1.0
Cobalt	1	0.50
Copper	1	0.50
Lead	1	0.50
Manganese	1	0.50
Molybdenum	1	0.50
Nickel	1	0.50
Platinum	1	0.50
Selenium	2	1.0
Silver	1	0.50
Strontium	1	0.50
Thallium	1	0.50
Uranium	1	0.50
Vanadium	2	1.0
Zinc	5	2.5



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TABLE 2
RECOMMENDED ISOTOPES FOR SELECTED ELEMENTS

Element of Interest	Mass(es)
Aluminum	<u>27</u>
Antimony	121, <u>123</u>
Arsenic	<u>75</u>
Barium	138, 137, 136, <u>135</u> , 134
Beryllium	<u>9</u>
Bismuth (IS)	209
Cadmium	<u>114</u> , 112, <u>111</u> , 110, 113, 116, 106
Calcium (I)	42, 43, <u>44</u> , 46, 48
Chlorine (I)	35, 37, (77, 82) ^a
Chromium	<u>52</u> , <u>53</u> , <u>50</u> , 54
Cobalt	<u>59</u>
Copper	<u>63</u> , <u>65</u>
Germanium (IS)	74
Holmium (IS)	165
Indium (IS)	<u>115</u> , 113
Iron (I)	<u>56</u> , <u>54</u> , <u>57</u> , 58
Lanthanum (I)	139
Lead	<u>208</u> , <u>207</u> , <u>206</u> , 204
Lithium (IS)	6 ^b , 7
Magnesium (I)	24, <u>25</u> , <u>26</u>
Manganese	<u>55</u>
Mercury	202, <u>200</u> , 199, 201
Molybdenum (I)	98, 96, 92, <u>97</u> , 94, (108) ^a
Nickel	58, <u>60</u> , 62, <u>61</u> , 64
Potassium (I)	<u>39</u>
Rhodium (IS)	103
Scandium (IS)	45
Selenium	80, <u>78</u> , <u>82</u> , <u>76</u> , <u>77</u> , 74
Silver	<u>107</u> , <u>109</u>
Sodium (I)	<u>23</u>
Terbium (IS)	159
Thallium	<u>205</u> , 203
Vanadium	<u>51</u> , <u>50</u>
Tin (I)	120, <u>118</u>
Yttrium (IS)	89
Zinc	64, <u>66</u> , <u>68</u> , <u>67</u> , 70

^a These masses are also useful for interference correction (Sec. 4.2).

^b Internal standard must be enriched in the ⁶Li isotope. This minimizes interference from indigenous lithium.

NOTE: Method 6020 is recommended for only those analytes listed in Sec.1.2. Other elements are included in this table because they are potential interferents (labeled I) in the determination of recommended analytes, or because they are commonly used internal standards (labeled IS). Isotopes are listed in descending order of natural abundance. The most generally useful isotopes are underlined and in boldface, although certain matrices may require the use of alternative isotopes.



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Table F-8. Trace Metals Analysis by Inductively Coupled Plasma/Mass Spectrometry (ICP/MS) (Method 6020)

QC Check	Minimum Frequency	Acceptance Criteria	Corrective Action	Flagging Criteria	Comments
Demonstrate acceptable analytical capability	Prior to using any test method and at any time there is a change in instrument type, personnel, test method, or sample matrix.	QC acceptance criteria published by DoD, if available; otherwise method-specified criteria.	Recalculate results; locate and fix problem, then rerun demonstration for those analytes that did not meet criteria (see Section C.1.f).	NA.	This is a demonstration of analytical ability to generate acceptable precision and bias per the procedure in Appendix C. No analysis shall be allowed by analyst until successful demonstration of capability is complete.
LOD determination and verification (See Box D-13)					
LOQ establishment and verification (See Box D-14)					
Instrument detection limit (IDL) study	At initial set-up and after significant change in instrument type, personnel, test method, or sample matrix.	IDLs shall be \leq LOD.	NA.	NA.	Samples may not be analyzed without a valid IDL.
Tuning	Prior to ICAL.	Mass calibration \leq 0.1 amu from the true value; Resolution $<$ 0.9 amu full width at 10% peak height; For stability, RSD \leq 5% for at least four replicate analyses.	Retune instrument then reanalyze tuning solutions.	Flagging criteria are not appropriate.	No analysis shall be performed without a valid MS tune.
Initial calibration (ICAL) for all analytes (minimum one high standard and a calibration blank)	Daily ICAL prior to sample analysis.	If more than one calibration standard is used, $r \geq 0.995$.	Correct problem, then repeat ICAL.	Flagging criteria are not appropriate.	Problem must be corrected. No samples may be run until ICAL has passed.



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Table F-8. Trace Metals Analysis by Inductively Coupled Plasma/Mass Spectrometry (ICP/MS) (Method 6020) (continued)

QC Check	Minimum Frequency	Acceptance Criteria	Corrective Action	Flagging Criteria	Comments
Second source calibration verification	Once after each ICAL, prior to beginning a sample run.	Value of second source for all analytes within $\pm 10\%$ of true value.	Verify second source standard. Rerun second source verification. If that fails, correct problem and repeat ICAL.	Flagging criteria are not appropriate.	Problem must be corrected. No samples may be run until calibration has been verified.
Continuing calibration verification (CCV)	After every 10 field samples and at the end of the analysis sequence.	All analytes within $\pm 10\%$ of true value.	Correct problem, rerun calibration verification. If that fails, then repeat ICAL. Reanalyze all samples since the last successful calibration verification.	If reanalysis cannot be performed, data must be qualified and explained in the case narrative. Apply Q-flag to all results for the specific analyte(s) in all samples since the last acceptable calibration verification.	Problem must be corrected. Results may not be reported without a valid CCV. Flagging is only appropriate in cases where the samples cannot be reanalyzed.
Low-level calibration check standard	Daily, after one-point ICAL.	Within $\pm 20\%$ of true value.	Correct problem, then reanalyze.	Flagging criteria are not appropriate.	No samples may be analyzed without a valid low-level calibration check standard. Low-level calibration check standard should be less than or equal to the reporting limit.
Linear dynamic range or high-level check standard	Every 6 months.	Within $\pm 10\%$ of true value.	NA.	NA.	
Method blank	One per preparatory batch.	No analytes detected $> \frac{1}{2}$ RL and greater than $1/10$ the amount measured in any sample or $1/10$ the regulatory limit (whichever is greater). Blank result must not otherwise affect sample results. For common laboratory contaminants, no analytes detected $> RL$ (see Box D-1).	Correct problem, then see criteria in Box D-1. If required, reprep and reanalyze method blank and all samples processed with the contaminated blank.	If reanalysis cannot be performed, data must be qualified and explained in the case narrative. Apply B-flag to all results for the specific analyte(s) in all samples in the associated preparatory batch.	Problem must be corrected. Results may not be reported without a valid method blank. Flagging is only appropriate in cases where the samples cannot be reanalyzed.
Calibration blank	Before beginning a sample run, after every 10 samples, and at end of the analysis sequence.	No analytes detected $> LOD$.	Correct problem. Re-prep and reanalyze calibration blank. All samples following the last acceptable calibration blank must be reanalyzed.	Apply B-flag to all results for specific analyte(s) in all samples associated with the blank.	



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Table F-8. Trace Metals Analysis by Inductively Coupled Plasma/Mass Spectrometry (ICP/MS) (Method 6020) (continued)

QC Check	Minimum Frequency	Acceptance Criteria	Corrective Action	Flagging Criteria	Comments
Interference check solutions (ICS-A and ICS-AB)	At the beginning of an analytical run and every 12 hours.	ICS-A: Absolute value of concentration for all non-spiked analytes < LOD (unless they are a verified trace impurity from one of the spiked analytes); ICS-AB: Within $\pm 20\%$ of true value.	Terminate analysis, locate and correct problem, reanalyze ICS, reanalyze all samples.	If corrective action fails, apply Q-flag to all results for specific analyte(s) in all samples associated with the ICS.	
LCS containing all analytes to be reported	One per preparatory batch.	QC acceptance criteria specified by DoD, if available; see Box D-3 and Appendix G.	Correct problem, then reprep and reanalyze the LCS and all samples in the associated preparatory batch for failed analytes, if sufficient sample material is available (see full explanation in Appendix G).	If reanalysis cannot be performed, data must be qualified and explained in the case narrative. Apply Q-flag to specific analyte(s) in all samples in the associated preparatory batch.	Problem must be corrected. Results may not be reported without a valid LCS. Flagging is only appropriate in cases where the samples cannot be reanalyzed.
Matrix spike (MS)	One per preparatory batch per matrix (see Box D-7).	For matrix evaluation, use QC acceptance criteria specified by DoD for LCS.	Examine the project-specific DQOs. If the matrix spike falls outside of DoD criteria, additional quality control tests (dilution test and post-digestion spike addition) are required to evaluate matrix effects.	For the specific analyte(s) in the parent sample, apply J-flag if acceptance criteria are not met.	For matrix evaluation only. If MS results are outside the LCS limits, the data shall be evaluated to determine the source of difference and to determine if there is a matrix effect or analytical error.
Matrix spike duplicate (MSD) or sample duplicate	One per preparatory batch per matrix (see Box D-7).	MSD: For matrix evaluation use QC acceptance criteria specified by DoD for LCS. MSD or sample duplicate: RPD < 20% (between MS and MSD or sample and sample duplicate).	Examine the project-specific DQOs. Contact the client as to additional measures to be taken.	For the specific analyte(s) in the parent sample, apply J-flag if acceptance criteria are not met.	The data shall be evaluated to determine the source of difference.
Dilution test	One per preparatory batch.	Five-fold dilution must agree within $\pm 10\%$ of the original measurement.	Perform post-digestion spike addition.	Flagging criteria are not appropriate.	Only applicable for samples with concentrations > 50 x LOQ.

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Table F-8. Trace Metals Analysis by Inductively Coupled Plasma/Mass Spectrometry (ICP/MS) (Method 6020) (continued)

QC Check	Minimum Frequency	Acceptance Criteria	Corrective Action	Flagging Criteria	Comments
Post digestion spike addition	When dilution test fails or analyte concentration for all samples < 50 x LOD.	Recovery within 75-125% (see Table B-1).	Run all associated samples in the preparatory batch by method of standard additions (MSA) or see flagging criteria.	For the specific analyte(s) in the parent sample, apply J-flag if acceptance criteria are not met.	Spike addition should produce a concentration of 10 – 100 x LOQ.
Method of standard additions (MSA)	When matrix interference is confirmed.	NA.	NA.	NA.	Document use of MSA in the case narrative.
Internal standards (IS)	Every sample.	IS intensity within 30-120% of intensity of the IS in the ICAL.	Reanalyze sample at 5-fold dilution with addition of appropriate amounts of internal standards.	Flagging criteria are not appropriate.	
Results reported between DL and LOQ	NA.	NA.	NA.	Apply J-flag to all results between DL and LOQ.	

ALS Standard Operating Procedure

DOCUMENT TITLE:	METALS DIGESTION, SOILS, SEDIMENTS, AND SLUDGE FOR ICP-AES AND ICP-MS ANALYSIS
REFERENCED METHOD:	EPA SW846 3050B
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METALS DIGESTION, SOILS, SEDIMENTS, AND SLUDGE FOR ICP-AES AND ICP-MS ANALYSIS

EPA SW846 3050B

SOPID:	MET-3050	Rev. Number:	5	Effective Date:	11/4/13
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METALS DIGESTION, SOILS, SEDIMENTS, AND SLUDGE FOR ICP-AES AND ICP-MS ANALYSIS

1) Scope and Applicability

Method 3050 is an acid digestion procedure used to prepare matrices such as soils, sludges, or sediments for analysis by ICP-AES or ICP-MS using SW846 methods 6010 and 6020.

2) Summary of Procedure

A representative aliquot of sample is digested in nitric acid and hydrogen peroxide (and Hydrochloric for ICP-AES). Hydrochloric acid is used as a final reflux acid for ICP-AES analyses. Nitric Acid is used as the final reflux acid for ICP-MS analyses.

3) Definitions

- 3.1 Laboratory Duplicates - Two aliquots of the same sample taken in the laboratory and analyzed separately with identical procedures. Analyses of duplicates indicates precision associated with laboratory procedures, but not with sample collection, preservation, or storage procedures.
- 3.2 Laboratory Control Sample (LCS) - An aliquot of a purchased soil with known quantities of the method analytes. If the purchased soil (LCSS) does not contain all of the target elements needed, the laboratory prepares an LCS with the missing elements by spiking Teflon chips. The LCS is analyzed exactly like a sample, and its purpose is to determine whether the methodology is in control and whether the laboratory is capable of making accurate and precise measurements.
- 3.3 Matrix Spike - An aliquot of an environmental sample to which known quantities of the method analytes are added in the laboratory. The matrix spike is analyzed exactly like a sample, and its purpose is to determine whether the sample matrix contributes bias to the analytical results.
- 3.4 Method Blank (MB) - An aliquot of reagent water or other blank matrices that are treated exactly as a sample from digestion to analysis including exposure to all glassware, equipment, solvents, reagents, and internal standards that are used with other samples. The MB is used to determine if method analytes or other interferences are present.
- 3.5 Digestion Batch - A digestion batch is no more than 20 samples of the same matrix digested as a unit per day. See ADM-BATCH.
- 3.6 Limit of Detection (LOD) - An estimate of the minimum amount of a substance that an analytical process can reliably detect. An LOD is analyte- and matrix-specific and may be laboratory - dependent.
- 3.7 Limit of Quantitation (LOQ) - The minimum levels, concentrations, or quantities of a target that can be reported with a specified degree of confidence.



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4) Health and Safety Warnings

- 4.1 All appropriate safety precautions for handling reagents and samples must be taken when performing this procedure. This includes the use of personnel protective equipment, such as, safety glasses, lab coat and the correct gloves.
- 4.2 Chemicals, reagents, and standards must be handled as described in the company safety policies, approved methods and in the MSDSs where available. Refer to the Environmental Health and Safety Manual and the appropriate MSDS prior to beginning this method.
- 4.3 Nitric and Hydrochloric Acids are used in this method. These acids are extremely corrosive and care must be taken while handling them. A face shield should be used while pouring acids. And safety glasses should be worn while working with the solutions. Lab coat and gloves should always be worn while working with these solutions.
- 4.4 Waste Management and Pollution Prevention
 - 4.4.1 It is the laboratory's practice to minimize the amount of acids and reagents used to perform this method wherever feasible. Standards are prepared in volumes consistent with methodology and only the amount needed for routine laboratory use is kept on site. The threat to the environment from solvent and reagents used in this method can be minimized when disposed of properly.
 - 4.4.2 The laboratory will comply with all Federal, State and local regulations governing waste management, particularly the hazardous waste identification rules and land disposal restrictions as specified in the EH&S Manual.
 - 4.4.3 For further information refer to SMO-SPLDIS.

5) Cautions

- 5.1 All hoods in the Metals Prep Lab are wiped down once a week with DI water. The tops of all digestion hot plates are wiped down daily.

6) Interferences

- 6.1 Elements bound in silicate structures are not normally dissolved by this method. Such bound elements would not be mobile in the environment and are not normally of interest.
- 6.2 See appropriate analysis SOP for applicable interferences

7) Personnel Qualifications and Responsibilities

- 7.1 It is the responsibility of the analyst to perform the analysis according to this SOP and to complete all documentation required for data review. Analysis and interpretation of the results are performed by personnel in the laboratory who have demonstrated the ability to generate acceptable results utilizing this SOP. Final review and sign-off of the data is performed by the department supervisor or designee.
- 7.2 Training – see CE-QA003.



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8) Sample Collection, Preservation and Storage

For solids and non-aqueous samples, glass or plastic sample containers are acceptable. Typically our lab uses purchased, precleaned 4 or 8 oz glass jars. Samples are to be stored at 0-6°C from collection to digestion. Samples are analyzed within 6 months of sample collection. Additional sample handling policies, storage and custody procedures are in SMO-GEN and SMO-ICOC.

9) Equipment and Supplies

9.1 Hot Plate Digestion

- 9.1.1 Digestion vessel = 250 and 100 mL beakers
- 9.1.2 Ribbed watch glasses
- 9.1.3 Hot plates
- 9.1.4 Funnels
- 9.1.5 Filter paper

9.2 Hot Block Digestion

- 9.2.1 Digestion vessel = Graduated block digester cups
- 9.2.2 Reflux cap
- 9.2.3 Hot Block Digester with ETR-3200 Controller by Environmental Express, LTD
- 9.2.4 CPI MOD Block Digester
- 9.2.5 Block Digester Filters.

9.3 Graduated cylinders

9.4 Eppendorf Pipettors –Calibrated according to ADM-PCAL.

9.5 Mortar and pestle

9.6 Tongue depressors



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10) Standards and Reagents

10.1 Standards Preparation General Information and Disclaimers

- 10.1.1 All of the preparation instructions are general guidelines. Other technical recipes may be used to achieve the same results. Example – a 20 mg/L standard may be made by adding 1 mL of 200 mg/L to 10 mL or may be made by adding 4 mL of 50 mg/L to 10 mL. The preparation depends upon the final volume needed and the initial concentration of the stock. Reasonable dilution technique is used
- 10.1.2 Vendors and vendors' products are sometimes listed for the ease of the analyst using this SOP, but products and purchased concentrations are examples only and subject to change at any time. All purchased standards are certified by the vendor. Certificates of Analysis are kept in the department until the standards are no longer being used – at which time they are archived with QA. Certificates of Analysis are available upon request. Purchased standards are routinely checked against an independent source for both analyte identification and analyte concentration.
- 10.1.3 Standards and Reagent expire per the Expiration Policy (CE-QA012) unless otherwise specified.
- 10.1.4 All Standards must be traceable using the laboratory lot system (CE-QA007).

10.2 Reagent water – laboratory produced deionized water.

10.3 Purchased Reagents and standards – Store at room temperature.

- 10.3.1 Concentrated nitric acid (Baker Instra-Analyzed 69-70%): Acid should be demonstrated to be free of impurities at levels which would interfere with sample determinations. Store in the dark.
 - 10.3.2 Concentrated hydrochloric acid (Baker Instra-Analyzed 36.5-38%): Acid should be demonstrated to be free of impurities at levels which would interfere with sample determinations.
 - 10.3.3 Hydrogen peroxide (30%) - H₂O₂. Should be demonstrated to be free of impurities at levels which would interfere with sample determinations.
 - 10.3.4 ERA Soil Laboratory Control Sample (LCSS) - Concentrations and Performance Acceptance Limits distributed through vendor
 - 10.3.5 Metals spiking solutions – Purchased commercially. See Table 1. All target elements needed for client samples in the batch are added to the LCS and MS.
- 10.4 LCS – use ERA Soil Laboratory Control Sample (LCSS) as above. If the LCSS does not contain all of the target elements needed, create a second LCS sample and spike Teflon chips with the elements missing from the ERA Soil (see Table 1 for appropriate spiking solutions). Digest as a sample.
- 10.5 MS – Add appropriate spiking solutions (see Table 1) to client sample prior to the addition of heat or reagents. Digest as a sample.
- 10.6 Method Blank – Digest Teflon chips as a sample.



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11) Method Calibration

- 11.1 The uniformity of the temperature of the hotplates and hotblocks is monitored by randomly moving a temperature blank around to different positions with each run. The temperature blank is DI in a digestion vessel and the temperature is read with a thermometer held in the DI when the samples should be at the required temperature. If positions become unreliable (not within the required temperature of digestion), those positions must not be used and the unit should be serviced or replaced.
- 11.2 The graduations on the hot block cups are verified for volume whenever a new lot number of cups is received. To verify the volume, tare a cup on a toploader balance, add DI to the desired graduation (all graduations used for measuring volume must be checked) and record the mass. Repeat on 10 cups. The mean mass must be within 3% of the marked volume. The RSD must be $\leq 3\%$. If criteria is not met, the lot of cups is either rejected and removed from service or a correction factor must be employed

12) Sample Preparation

- 12.1 Be sure there is a current LOD and the analyst has a current Demonstration of Capability.
- 12.2 Be sure the hot block cups have been verified for volume. Record the lot number of the cups used in the batch through the Standards Log in LIMS.
- 12.3 Set the temperature on the Hot Plate or Block Digestor to a temperature that brings the sample temperature to 90-95°C without boiling.
- 12.4 The Hot Block is on a timer which can be set to turn on and off whenever necessary. To set timer press the timer button and choose the days M-F (Monday through Friday). Then choose the hour and minutes to start and stop the Block Digestor. The temperature of the batch is monitored by randomly placing a vessel with DI in the block or on the hotplate. Record the temperature and the ID of the hotplate or hotblock in the Comments section of the MB on the prep sheet.
- 12.5 Label digestion vessel with appropriate sample IDs for digestion.
- 12.6 See ADM-SPLPREP for instructions of how to homogenize and subsample and how to handle standing water and extraneous materials. Make note of any special sample handling in the comments section of the prep sheet.
- 12.7 Weigh (to the nearest 0.01g) 1.00g to 1.05g of sample into the digestion vessel. For sludges and sediments that have a high moisture content, use more sample. The goal is to use about 1g of dry weight sample. Record the sample weight, sample color, and sample texture directly into LIMS in real time – do not handwrite on anything to be later entered. Add the appropriate spiking solutions (see Table 1) directly onto the designated spike sample prior to addition of reagents. Record the spike volumes, spiking solutions, and reagents directly into LIMS.
- 12.8 Unless specified by project or state requirements, add the following to each sample and QC sample: 10 mL of 1:1 HNO₃ and, for ICP-AES and Silver or Antimony by ICP-MS, add 1.0 mL of 1:1 HCl. Place digestion vessel on hotplate or in hotblock. Cover with a ribbed watch glass or reflux cap and reflux for 15 minutes. The sample temperature should be 90-95°C. Allow the sample to cool. Add 5 mL of concentrated HNO₃, cover and reflux for 30 minutes. Repeat the addition of 5 mL of HNO₃ and reflux to 5 mL. Do not allow the sample to go to dryness. CAUTION: Do not boil. Antimony is easily lost by volatilization.



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- 12.9 Cool the sample and add 2 mL of DI and 3 mL of 30% H₂O₂. Cover and heat to start the peroxide reaction. Care must be taken to ensure that losses do not occur due to excessive effervescence. Heat until effervescence subsides and cool the digestion vessel.
- 12.10 If the effervescence does not subside, add 3 mLs of hydrogen peroxide with warming to each of the samples (including blanks and LCSs) in the batch. If necessary, continue to add 30% H₂O₂ in 1 mL aliquots with warming until the effervescence is minimal, or until the general sample appearance is unchanged. Do not add more than 10ml of 30% H₂O₂.
- 12.11 If the sample is being prepared for analysis by ICP-AES or Silver or Antimony by ICP-MS, add 10 mL 1:1 HCL. If the sample is being prepared for analysis by ICP-MS, no HCl is added and the sample is evaporated to approximately 5 mL.
- 12.12 Cover and reflux the ICP-AES and Silver or Antimony by ICP-MS samples for 15 minutes without boiling. Allow to cool.
- 12.13 Prepare filters by rinsing with 1:1 nitric acid and DI.
- 12.14 Filter if particulates are suspended. If particulates have settled, no filtering is necessary. Filtering is usually required if analyzed on the same day as digestion. If filtering, record the lot number of the filters through the Standards Log in LIMS. If the entire batch is filtered, also filter the QC samples. If only some of the samples are filtered, note which samples are filtered in the comments section of the benchsheet and filter an undigested LCS and MB to demonstrate the acceptability of the filters.
- 12.15 Quantitatively transfer the digestate to a graduated cylinder by pouring the sample through a prepared filter into the cylinder and rinsing the beaker and watch glass or reflux cap with DI into the filter. Rinse the filter with DI. If not filtering, quantitatively transfer digestate to a graduated cylinder rinsing beaker with DI. All samples are diluted to 100 mL with DI. Document the final volume in LIMS. Pour into a labeled B-cup.
- 12.16 Print out one copy of the Preparation Information Benchsheet. The prep sheets are reviewed and signed by a peer or supervisor within 48 hours of preparation. This prep sheet is filed in the appropriate binder (logbook). A copy remains with the samples through analysis. Copies made for client folders are to include all review signatures.

13) Troubleshooting

None

14) Data Acquisition

Preparation volumes are entered manually into the LIMS Prep Sheet and are used to calculate final results

15) Calculation and Data Reduction Requirements

Data must be reviewed by the analyst and a peer (supervisor or qualified analyst) using a Data Quality Checklist before the results are validated and reported to the client. Further data review policies and procedures are discussed in ADM-DREV.



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16) Quality Control, Acceptance Criteria, and Corrective Action

- 16.1 LCS - Digest one laboratory control sample (LCS) per digestion batch.
- 16.2 MB - Digest one Method blank per digestion batch.
- 16.3 DUP/MS - Digest one spiked sample and one duplicate sample (or matrix spiked duplicate if specified by client) with each digestion batch.
- 16.4 See appropriate analytical SOP for applicable QC limits and corrective action.

17) Data Records Management

See CE-GEN003 and ADM-ARCH

18) Contingencies for Handling Out Of Control Data

If data is produced that is out of control and is not to be re-analyzed due to sample volume restrictions, holding times, or QC controls can not be met, flag and narrate appropriately.

19) Method Performance

- Detection and Quantitation limits are determined according to the requirements in CE-QA011. The supporting information is filed with the QA office.
- Demonstration of Capability is performed according to CE-QA003.

20) Summary of Changes

- Updated to ALS format – removed CAS throughout
- Removed references to GFAA methods and CLP
- Incorporated change form for reduction of HCl

21) References and Related Documents

- “Test Methods For Evaluating Solid Waste, Physical/Chemical Methods”. EPA SW846, Third Edition, December 1996.

22) Appendix

- Table 1 Spike Concentrations
- Digestion Log Benchsheet
- SW846 Method 3050 Flow Chart



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Table 1 Spiking Concentrations for LCS and MS Samples

SPIKE SOLUTION A		1.00ml Spk A to Final Vol of 100ml	
<i>Metal</i>	<i>Conc. (ug/mL)</i>	<i>Metal</i>	<i>Conc. (ug/mL)</i>
AL	200	NI	50
AS	4	SE	1
BA	200	AG	5
BE	5	TL	200
CD	5	V	50
CR	20	ZN	50
CO	50	B	100
CU	25	CA	200
FE	100	MG	200
PB	50	NA	2000
MN	50	K	2000

SPIKE SOLUTION B		1.00ml Spk B to Final Vol of 100ml	
<i>Metal</i>	<i>Conc. (ug/mL)</i>	<i>Metal</i>	<i>Conc. (ug/mL)</i>
SB	50	TI	50
MO	50	-	-

INDIVIDUAL METALS		0.10ml Spk. to Final Volume of 100ml		INDIVIDUAL METALS		0.5ml Spk. to Final Volume of 100ml	
<i>Metal</i>	<i>Conc. (ug/mL)</i>			<i>Metal</i>	<i>Conc. (ug/mL)</i>		
SE	1000			SN	1000		

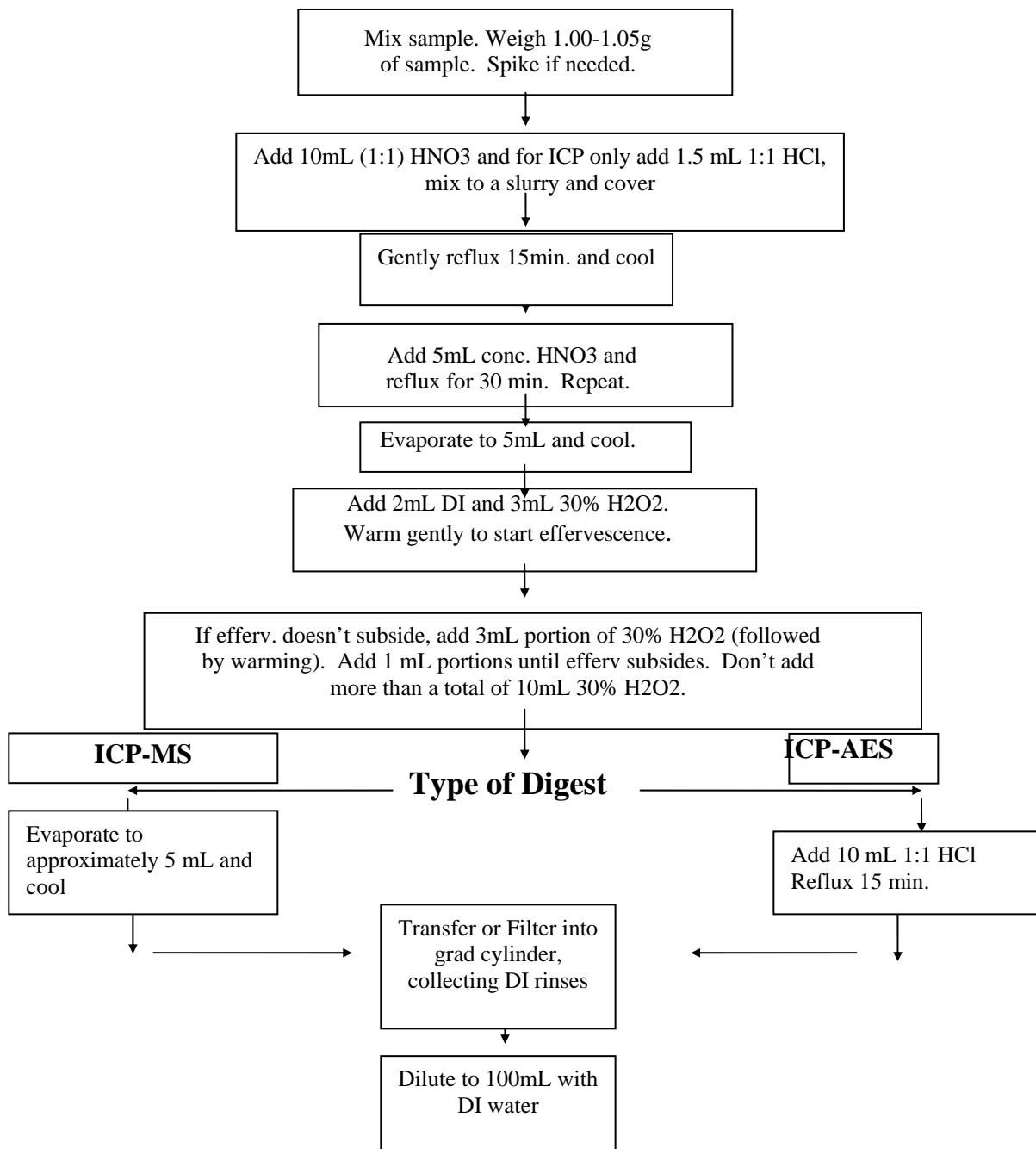


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SW846 Method 3050 Flow Chart

Soils, Sediments and Sludges



ALS Standard Operating Procedure

DOCUMENT TITLE:	DETERMINATION OF METALS AND TRACE ELEMENTS BY INDUCTIVELY COUPLED PLASMA EMISSION SPECTROSCOPY (ICP)
REFERENCED METHOD:	EPA 200.7 AND SW 846 6010C
SOP ID:	MET-200.7
REV. NUMBER:	14
EFFECTIVE DATE:	11/4/13





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EPA 200.7 AND SW 846 6010C

SOPID:	MET-200.7	Rev. Number:	14	Effective Date:	11/4/13
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Approved By:

Department Supervisor - Christine Kutzer

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DETERMINATION OF METALS AND TRACE ELEMENTS BY INDUCTIVELY COUPLED PLASMA EMISSION SPECTROSCOPY (ICP)

1) **Scope and Applicability**

This SOP uses EPA methods 200.7 and 6010C for the determination of trace metals in solution by inductively coupled plasma-atomic emission spectrometry (ICP-AES). The method is applicable to all of the elements listed in Table 1. All matrices, including ground water, aqueous samples, TCLP and EP extracts, industrial and organic wastes, soils, sludges, sediments, and other solid wastes, require digestion prior to analysis.

Detection limits, sensitivity, and the optimum and linear concentration ranges of the elements can vary with the wavelength, spectrometer, matrix and operating conditions. The Reporting Limits are listed in Table 1. The reporting limit may be adjusted if required for specific project requirements, however, the capability of achieving other reporting limits must be demonstrated.

2) **Summary of Procedure**

Sample preparation and digestion is found in the appropriate extraction and digestion SOPs. Examples include MET-3010A, MET-3050B, MET-TCLP, MET-TZHE, MET-SPLP, and MET-SPLPZHE.

This method uses multiple elemental determinations by ICP-AES using sequential or simultaneous optical systems and both axial and radial viewing of the plasma (dual view). The instrument measures characteristic emission spectra by optical spectrometry. Samples are nebulized and the resulting aerosol is transported to the plasma torch. Element-specific emission spectra are produced by a radio-frequency inductively coupled plasma. The spectra are dispersed by a grating spectrometer, and the intensities of the emission lines are monitored by photosensitive devices. Background correction is required for trace element determination. Background must be measured adjacent to analyte lines on samples during analysis. The position selected for the background-intensity measurement, on either or both sides of the analytical line, will be determined by the complexity of the spectrum adjacent to the analyte line. In one mode of analysis the position used should be as free as possible from spectral interference and should reflect the same change in background intensity as occurs at the analyte wavelength measured. Background correction is not required in cases of line broadening where a background correction measurement would actually degrade the analytical result. The possibility of additional interferences named in the Interferences Section should also be recognized and appropriate corrections made.



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3) Definitions

- 3.1 **Initial Calibration** - analysis of analytical standards for a series of different specified concentrations; used to define the linearity and dynamic range of the response of the detector to the element.
- 3.2 **Calibration Standard (CAL)** - A solution prepared from the dilution of stock standard solutions. The CAL solutions are used to calibrate the instrument response with respect to analyte concentration.
- 3.3 **Dissolved Analyte** - The concentration of analyte in an aqueous sample that will pass through a 0.45 µm membrane filter assembly prior to sample acidification.
- 3.4 **Initial Calibration Verification (ICV)** (also called Second Source Calibration Verification) - ICV solutions are made from a stock solution which is different from the stock used to prepare calibration standards and is used to verify the validity of the initial calibration.
- 3.5 **Continuing Calibration Verification Standard (CCV)** - A standard analyzed at specified intervals and used to verify the ongoing validity of the instrument calibration. For the LLCCV, see MRL Standard below.
- 3.6 **Matrix Spike (MS)** - An aliquot of an environmental sample to which known quantities of all of the elements of client interest are added in the laboratory. The matrix spike is analyzed exactly like a sample, and its purpose is to determine whether the sample matrix contributes bias to the analytical results in a given client matrix.
- 3.7 **Duplicate (DUP)** - A laboratory duplicate. Two aliquots of the same sample taken in the laboratory and analyzed separately with identical procedures. Analysis of duplicates indicates precision associated with laboratory procedures, but not with sample collection, preservation, or storage procedures.
- 3.8 **Laboratory Control Sample (LCS)** - A matrix blank spiked with all of the elements of client interest. The LCS is analyzed exactly like a sample, and its purpose is to determine whether the methodology is in control and whether the laboratory is capable of making accurate measurements. LCS-Soil is purchased from a vendor.
- 3.9 **Method Blank (MB)** - An aliquot of reagent water or other blank matrix that is treated exactly as a sample including exposure to all glassware, equipment, solvents, reagents, and internal standards that are used with other samples. The MB is used to determine if method analytes or other interferences are present in the laboratory environment, reagents, or apparatus.
- 3.10 **Instrument Blank (ICB/CCB)** - The instrument blank (also called initial or continuing calibration blank) is a volume of reagent water acidified with the same acid matrix as in the calibration standards. This blank is the zero standard and has a reagent composition identical to the digestates. The purpose of the ICB/CCB is to determine the levels of contamination associated with the instrumental analysis.



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- 3.11 **Batch** – a group of no more than 20 samples analyzed together on the same day with the same reagents. See the SOP for Batches and Sequences (ADM-BATCH) for more detail.
- 3.12 **Calibration Blank** - A volume of reagent water acidified with the same acid matrix as in the calibration standards. The calibration blank is a zero standard and is used to calibrate the ICP instrument.
- 3.13 **Instrument Detection Limit (IDL)** - The concentration equivalent to a signal due to the analyte which is equal to three times the standard deviation of a series of 21 replicate measurements of a low standard's signal at the same wavelength.
- 3.14 **Internal Standard** - Pure analyte(s) added to a sample, extract, or standard solution in known amount(s) and used to measure the relative responses of other method analytes that are components of the same sample or solution. The internal standard must be an analyte that is not a sample component.
- 3.15 **Linear Range** - The concentration range over which the instrument response to an analyte is linear.
- 3.16 **Limit of Detection (LOD)** – An estimate of the minimum amount of a substance that an analytical process can reliably detect. An LOD is analyte- and matrix-specific and may be laboratory – dependent. For DOD, the smallest amount or concentration of a substance that must be present in a sample in order to be detected at a high level of confidence (99%). At the LOD, the false negative rate (Type II error) is 1%.
- 3.17 **Limit of Quantitation (LOQ)** – The minimum levels, concentrations, or quantities of a target that can be reported with a specified degree of confidence. For DOD, the lowest concentration that produces a quantitative result within specified limits of precision and bias. The LOQ shall be set at or above the concentration of the lowest initial calibration standard.
- 3.18 **Plasma Solution** - A solution that is used to determine the optimum height above the work coil for viewing the plasma.
- 3.19 **Interference Check Solution (ICS)** - A solution of selected method analytes of higher concentrations which is used to evaluate the procedural routine for correcting known interelement spectral interferences with respect to a defined set of method criteria.
- 3.20 **MRL Standard** – Standard prepared with a known concentration of elements to check accuracy at the quantitation limit. This is also known as Low-level calibration check standard (LLCCV or LLICV). This standard is not digested.
- 3.21 **LOQ Standard – also called LLQC** – Standard prepared at the LOQ that undergoes digestion and preparation procedures.
- 3.22 **HLCCV1** – A standard prepared at the bench at a high concentration to encompass the range of the samples being analyzed. This standard is used to assess accuracy at the high end of the calibration curve.
- 3.23 **HLCCV2** – A standard prepared slightly higher than the calibration range for metals. This is an “upper range limit” standard used to verify the upper limit of the linear dynamic range of the instrument.



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- 3.24 **Matrix** - the predominant material, component, or substrate (e.g., surface water, drinking water, etc.) of which the sample to be analyzed is composed.
- 3.25 **Relative Percent Difference (RPD)** - The absolute value of the difference of two values divided by the average of the same two values. Used to compare the precision of the analysis. The result is always a positive number.
- 3.26 **Interelement Correction Factors (IECs)** - factors that the instrument uses to compensate for spectral overlap when analyzing samples with complex spectra.

4) Health and Safety Warnings

- 4.1 All appropriate safety precautions for handling solvents, reagents and samples must be taken when performing this procedure. This includes the use of personal protective equipment, such as safety glasses, lab coat and the correct gloves.
- 4.2 Chemicals, reagents and standards must be handled as described in the Company safety policies, approved methods and in MSDSs where available. Refer to the Environmental, Health and Safety Manual and the appropriate MSDS prior to beginning this method.
- 4.3 Hydrochloric and Nitric Acid are used in this method. These acids are extremely corrosive and care must be taken while handling them. A face shield should be used while pouring acids. And safety glasses should be worn while working with the solutions. Lab coat and gloves should always be worn while working with these solutions.
- 4.4 The use of pressurized gases is required for this procedure. Care should be taken when moving cylinders. All gas cylinders must be secured to a wall or an immovable counter with a chain or a cylinder clamp at all times. Sources of flammable gases (e.g., pressurized hydrogen) should be clearly labeled.
- 4.5 Refer to the Safety Manual for further discussion of general safety procedures and information.
- 4.6 High Voltage - The power unit supplies high voltage to the RF generator which is used to form the plasma. The unit should never be opened. Exposure to high voltage can cause injury or death.
- 4.7 UV Light - The plasma when lit is a very intense light, and must not be viewed with the naked eye. Protective lenses are in place on the instrument. Glasses with special protective lenses are available.
- 4.8 Waste Management And Pollution Prevention

It is the laboratory's practice to minimize the amount of acids and reagent used to perform this method wherever feasible. Standards are prepared in volumes consistent with methodology and only the amount needed for routine laboratory use is kept on site. The threat to the environment from solvent and reagents used in this method can be minimized when disposed of properly.

The laboratory will comply with all Federal, State and local regulations governing waste management, particularly the hazardous waste identification rules and land disposal restrictions as specified in the EH&S Manual..



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Samples with analyte concentrations exceeding TCLP regulatory limits are disposed of as hazardous waste, see SOP *SMO-SPLDIS*.

5) Cautions

- 5.1 Typical preventive maintenance measures include, but are not limited to, the following items:
- Changing the pump tubing as needed
 - Empty waste container, as needed
 - Cleaning the nebulizer, spray chamber, and torch, as needed
 - Replace water and vacuum filters, as needed

6) Interferences

- 6.1 There are several types of interferences by the ICPs: Spectral interferences can be from an overlap of spectral lines, background points or background from line emissions of high concentration elements. Physical interferences are effects associated with the sample introduction process, example high dissolved solids buildup on the nebulizer tip. Chemical interferences caused by the sample matrix itself. IECs aid in eliminating some of these interferences. IECs are interelement correction factors that the instrument uses to compensate for spectral overlap when analyzing samples with complex spectra. The following is the text from Method 6010B Section 3.0. Similar text is found in Method 200.7 and 6010C. Not all of the items may be applicable to this laboratory's instruments or procedures.
- 6.2 Spectral interferences are caused by background emission from continuous or recombination phenomena, stray light from the line emission of high concentration elements, overlap of a spectral line from another element, or unresolved overlap of molecular band spectra.
- 6.3 Background emission and stray light can usually be compensated for by subtracting the background emission determined by measurements adjacent to the analyte wavelength peak. Spectral scans of samples or single element solutions in the analyte regions may indicate when alternate wavelengths are desirable because of severe spectral interference. These scans will also show whether the most appropriate estimate of the background emission is provided by an interpolation from measurements on both sides of the wavelength peak or by measured emission on only one side. The locations selected for the measurement of background intensity will be determined by the complexity of the spectrum adjacent to the wavelength peak. The locations used for routine measurement must be free of off-line spectral interference (interelement or molecular) or adequately corrected to reflect the same change in background intensity as occurs at the wavelength peak. For multivariate methods using whole spectral regions, background scans should be included in the correction algorithm. Off-line spectral interferences are handled by including spectra on interfering species in the algorithm.



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- 6.4 To determine the appropriate location for off-line background correction, the user must scan the area on either side adjacent to the wavelength and record the apparent emission intensity from all other method analytes. This spectral information must be documented and kept on file. The location selected for background correction must be either free of off-line interelement spectral interference or a computer routine must be used for automatic correction on all determinations. If a wavelength other than the recommended wavelength is used, the analyst must determine and document both the overlapping and nearby spectral interference effects from all method analytes and common elements and provide for their automatic correction on all analyses. Tests to determine spectral interference must be done using analyte concentrations that will adequately describe the interference. Normally, 100 mg/L single element solutions are sufficient; however, for analytes such as iron that may be found at high concentration, a more appropriate test would be to use a concentration near the upper analytical range limit.
- 6.5 Spectral overlaps may be avoided by using an alternate wavelength or can be compensated by equations that correct for interelement contributions. Instruments that use equations for interelement correction **require** the interfering elements be analyzed at the same time as the element of interest. When operative and uncorrected, interferences will produce false positive determinations and be reported as analyte concentrations. More extensive information on interfering effects at various wavelengths and resolutions is available in reference wavelength tables and books. Users may apply interelement correction equations determined on their instruments with tested concentration ranges to compensate (off line or on line) for the effects of interfering elements. Some potential spectral interferences observed for the recommended wavelengths are given in Table 2. For multivariate methods using whole spectral regions, spectral interferences are handled by including spectra of the interfering elements in the algorithm. The interferences listed are only those that occur between method analytes. Only interferences of a direct overlap nature are listed. These overlaps were observed with a single instrument having a working resolution of 0.035 nm.
- 6.6 When using interelement correction equations, the interference may be expressed as analyte concentration equivalents (i.e. false analyte concentrations) arising from 100 mg/L of the interference element. For example, assume that As is to be determined (at 193.696 nm) in a sample containing approximately 10 mg/L of Al. According to Table 2, 100 mg/L of Al would yield a false signal for As equivalent to approximately 1.3 mg/L. Therefore, the presence of 10 mg/L of Al would result in a false signal for As equivalent to approximately 0.13 mg/L. The user is cautioned that other instruments may exhibit somewhat different levels of interference than those shown in Table 2. The interference effects must be evaluated for each individual instrument since the intensities will vary.
- 6.7 Interelement corrections will vary for the same emission line among instruments because of differences in resolution, as determined by the grating, the entrance and exit slit widths, and by the order of dispersion. Interelement corrections will also vary depending upon the choice of background correction points. Selecting a background correction point where an interfering emission line may appear should be avoided when practical. Interelement corrections that constitute a



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major portion of an emission signal may not yield accurate data. Users should not forget that some samples may contain uncommon elements that could contribute spectral interferences.

- 6.8 The interference effects must be evaluated for each individual instrument whether configured as a sequential or simultaneous instrument. For each instrument, intensities will vary not only with optical resolution but also with operating conditions (such as power, viewing height and argon flow rate). When using the recommended wavelengths, the analyst is required to determine and document for each wavelength the effect from referenced interferences (Table 2) as well as any other suspected interferences that may be specific to the instrument or matrix. The analyst is encouraged to utilize a computer routine for automatic correction on all analyses.
- 6.9 Users of sequential instruments must verify the absence of spectral interference by scanning over a range of 0.5 nm centered on the wavelength of interest for several samples. The range for lead, for example, would be from 220.6 to 220.1 nm. This procedure must be repeated whenever a new matrix is to be analyzed and when a new calibration curve using different instrumental conditions is to be prepared. Samples that show an elevated background emission across the range may be background corrected by applying a correction factor equal to the emission adjacent to the line or at two points on either side of the line and interpolating between them. An alternate wavelength that does not exhibit a background shift or spectral overlap may also be used.
- 6.10 If the correction routine is operating properly, the determined apparent analyte(s) concentration from analysis of each interference solution should fall within a specific concentration range around the calibration blank. The concentration range is calculated by multiplying the concentration of the interfering element by the value of the correction factor being tested and divided by 10. If after the subtraction of the calibration blank the apparent analyte concentration falls outside of this range in either a positive or negative direction, a change in the correction factor of more than 10% should be suspected. The cause of the change should be determined and corrected and the correction factor updated. The interference check solutions should be analyzed more than once to confirm a change has occurred. Adequate rinse time between solutions and before analysis of the calibration blank will assist in the confirmation.
- 6.11 When interelement corrections are applied, their accuracy should be verified, daily, by analyzing spectral interference check solutions. If the correction factors or multivariate correction matrices tested on a daily basis are found to be within the 20% criteria for 5 consecutive days, the required verification frequency of those factors in compliance may be extended to a weekly basis. Also, if the nature of the samples analyzed is such they do not contain concentrations of the interfering elements at \pm one reporting limit from zero, daily verification is not required. All interelement spectral correction factors or multivariate correction matrices must be verified and updated every six months or when an instrumentation change, such as in the torch, nebulizer, injector, or plasma conditions occurs. Standard solution should be inspected to ensure that there is no contamination that may be perceived as a spectral interference.
- 6.12 When interelement corrections are not used, verification of absence of interferences is required.



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- 6.13 One method is to use a computer software routine for comparing the determinative data to limits files for notifying the analyst when an interfering element is detected in the sample at a concentration that will produce either an apparent false positive concentration, (i.e., greater than) the analyte instrument detection limit, or false negative analyte concentration, (i.e., less than the lower control limit of the calibration blank defined for a 99% confidence interval).
- 6.14 Another method is to analyze an Interference Check Solution(s) which contains similar concentrations of the major components of the samples (>10 mg/L) on a continuing basis to verify the absence of effects at the wavelengths selected. These data must be kept on file with the sample analysis data. If the check solution confirms an operative interference that is > 20% of the analyte concentration, the analyte must be determined using (1) analytical and background correction wavelengths (or spectral regions) free of the interference, (2) by an alternative wavelength, or (3) by another documented test procedure.
- 6.15 Physical interferences are effects associated with the sample nebulization and transport processes. Changes in viscosity and surface tension can cause significant inaccuracies, especially in samples containing high dissolved solids or high acid concentrations. If physical interferences are present, they must be reduced by diluting the sample or by using a peristaltic pump, by using an internal standard or by using a high solids nebulizer. Another problem that can occur with high dissolved solids is salt buildup at the tip of the nebulizer, affecting aerosol flow rate and causing instrumental drift. The problem can be controlled by wetting the argon prior to nebulization, using a tip washer, using a high solids nebulizer or diluting the sample. Also, it has been reported that better control of the argon flow rate, especially to the nebulizer, improves instrument performance: this may be accomplished with the use of mass flow controllers.
- 6.16 Chemical interferences include molecular compound formation, ionization effects, and solute vaporization effects. Normally, these effects are not significant with the ICP technique, but if observed, can be minimized by careful selection of operating conditions (incident power, observation position, and so forth), by buffering of the sample, by matrix matching, and by standard addition procedures. Chemical interferences are highly dependent on matrix type and the specific analyte element.
- 6.17 Memory interferences result when analytes in a previous sample contribute to the signals measured in a new sample. Memory effects can result from sample deposition on the uptake tubing to the nebulizer and from the build up of sample material in the plasma torch and spray chamber. The site where these effects occur is dependent on the element and can be minimized by flushing the system with a rinse blank between samples. The possibility of memory interferences should be recognized within an analytical run and suitable rinse times should be used to reduce them. The rinse times necessary for a particular element must be estimated prior to analysis. This may be achieved by aspirating a standard containing elements at a concentration ten times the usual amount or at the top of the linear dynamic range. The aspiration time for this sample should be the same as a normal sample analysis period, followed by analysis of the rinse blank at designated intervals. The length of time required to reduce analyte signals to within a factor of two of the method detection limit should be



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noted. Until the required rinse time is established, this method suggests a rinse period of at least 60 seconds between samples and standards. If a memory interference is suspected, the sample must be reanalyzed after a rinse period of sufficient length. Alternate rinse times may be established by the analyst based upon their DQOs.

- 6.18 Users are advised that high salt concentrations can cause analyte signal suppressions and confuse interference tests. If the instrument does not display negative values, fortify the interference check solution with the elements of interest at 0.5 to 1 mg/L and measure the added standard concentration accordingly. Concentrations should be within 20% of the true spiked concentration or dilution of the samples will be necessary. In the absence of measurable analyte, overcorrection could go undetected if a negative value is reported as zero.
- 6.19 The dashes in Table 2 indicate that no measurable interferences were observed even at higher interfering concentrations. Generally, interferences were discernible if they produced peaks, or background shifts, corresponding to 2 to 5% of the peaks generated by the analyte concentrations.

7) Personnel Qualifications and Responsibilities

It is the responsibility of the analyst to perform the analysis according to this SOP and to complete all documentation required for data review. Analysis and interpretation of the results are performed by personnel in the laboratory who have demonstrated the ability to generate acceptable results utilizing this SOP. Final review and sign-off of the data is performed by the department supervisor or designee.

Training – see CE-QA003.

8) Sample Collection, Preservation, and Storage

- 8.1 Solid samples require no preservation prior to digestion other than storage at 0-6°C. Sample containers may be glass or plastic. When the laboratory provides the sample containers, the containers are purchased, certified clean glass soil jars with Teflon-lined lids. Samples are analyzed within 6 months of collection.
- 8.2 For aqueous samples, glass or plastic sample containers are acceptable. When the laboratory provides the sample containers, the containers are purchased, certified clean 250 or 500 mL plastic bottles. Sample volume should be acid preserved with (1+1) nitric acid to pH <2. Samples are held at room temperature (although refrigeration is acceptable also). Samples are analyzed within 6 months of sample collection.
- 8.3 For the determination of the dissolved elements, the sample must be filtered through a 0.45 µm pore diameter membrane filter at the time of collection or as soon thereafter as practically possible. (Glass or plastic filtering apparatus are recommended to avoid possible contamination. Only plastic apparatus should be used when the determinations of boron and silicon are critical.) Samples should be filtered prior to acidification, because acidification changes the sample (usually by dissolving particulates that would be filtered out). Acidified samples received at the lab that require lab filtration must be noted in the case narrative. See digestion SOPs for filtration procedure.



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- 8.4 Samples received by the ICP lab as digestates contain nitric and hydrochloric acid. Digestates are stored at room temperature in plastic B-cups or Hot Block Digestion cups.
- 8.5 Following analysis, digestates are stored until all results have been reviewed. Digestates are diluted and disposed of through the sewer system in approximately 90 days after receipt of sample.
- 8.6 For more information about custody, sample handling, and storage procedures, see SMO-GEN and SMO-ICOC.

9) Equipment and Supplies

Instrument ID	Instrument Configuration	Manufacturer Part	Serial Number	Year Acquired
ICP #3 (R-ICP-AES-03)	Instrument	Perkin Elmer 5300DV	077N6051602	2006
	Computer Workstation	Dell Optiplex GX620		
	Analytical Software	PE ICP WinLab v.3.1		

ICP #4 (R-ICP-AES-04)	Instrument	Perkin Elmer 5300DV	077N6052202	2010
	Computer Workstation	Dell Optiplex GX620		
	Analytical Software	PE ICP WinLab v.3.1		

ICP #5 (R-ICP-AES-05)	Instrument	Perkin Elmer Optima 8000	078N2072408C	2013
	Autosampler	AAS S-10	102S12031301	
	Computer Workstation	Lenova Thinkcentre		
	Analytical Software	PE ICP WinLab 32 v5.2.0.0612		

These instruments are “dual” view – they use axial and radial views simultaneously

- 9.1 Argon gas supply - high purity
- 9.2 Volumetric flasks, class A.
- 9.3 Calibrated adjustable Micropipet with disposable tips. See ADM-PCAL for calibration requirements.



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10) Standards and Reagents

- 10.1 Trace metals grade chemicals shall be used in all tests. Each lot of acid used is to be analyzed to demonstrate that it is free of interference before use (see CE-GEN007). Store all acids at room temperature. Acids expire upon manufacturer's indications or per Expiration Policy if not otherwise indicated.
- Hydrochloric acid (conc.), HCl, and Nitric acid (conc), HNO₃. Purchased commercially.
- 10.2 Reagent Water. All references to water in this SOP refer to the water produced by the laboratory water system.
- 10.3 Standards Preparation General Information and Disclaimers
- 10.3.1 All of the preparation instructions are general guidelines. Other technical recipes may be used to achieve the same results. Example – a 20 mg/L standard may be made by adding 1 mL of 200 mg/L to 10 mL or may be made by adding 4 mL of 50 mg/L to 10 mL. The preparation depends upon the final volume needed and the initial concentration of the stock. Reasonable dilution technique is used.
- 10.3.2 Vendors and vendors' products are sometimes listed for the ease of the analyst using this SOP, but products and purchased concentrations are examples only and subject to change at any time. All purchased standards are certified by the vendor. Certificates of Analysis are kept in the department until the standards are no longer being used – at which time they are filed with QA. Certificates of Analysis are available upon request. Purchased standards are routinely checked against an independent source for both analyte identification and analyte concentration.
- 10.3.3 All Standards must be traceable using the laboratory lot system. See the SOP for Making Entries Onto Analytical Records (CE-QA007) for detail.
- 10.3.4 All standards are prepared from NIST traceable stock standard solutions. Manufacturer's expiration dates are used to determine viability of standards. Preparatory procedures for standards and QC solutions vary between instruments due to the working ranges. All preparatory information for the standards and QC samples are provided in the Controlled Forms section of the Rochester Intranet in the Metals Standards Logbooks.
- 10.4 Mixed Calibration Standards are prepared by combining appropriate volumes of the stock solutions in volumetric flasks. Matrix match with the appropriate acid and dilute to 100 mL with water. Calibration standards should be verified using a second source quality control sample (LCS, ICV, or CCV). Calibration standards should be stored at room temperature in glass volumetric flasks with a shelf-life of 7 days.
- 10.5 Initial and Continuing Calibration Verification (ICV and CCV) Standards are prepared by combining compatible analytes at concentrations equivalent to the midpoint of their respective calibration curves. Matrix match with the



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appropriate acid. The ICV and CCV standards are prepared from a separate source independent from that used in the calibration standards. ICV / CCV standards should be stored at room temperature in glass volumetric flasks with a shelf-life of 48 hours.

- 10.6 Internal Standard - An internal standard solution consisting of a 10 mg/L solution of Yttrium and Cesium. The apparent concentration is 1.00 mg/L Yttrium. The Yttrium intensity is used by the instrument to ratio the analyte intensity signals for both calibration and quantitation. Cesium is used only as a stabilizer. Store the prepared solutions at room temperature for up to 6 months.
- 10.7 HLCCV1- The highest calibration standard with the same storage and expiration as the calibration standards.
- 10.8 HLCCV2 - A standard prepared to verify the linear dynamic range of the instrument. Store at room temperature for up to 14 days.
- 10.9 LOQ Standards (low level calibration check standards or MRL Standards) are prepared to contain known concentrations of elements at or near the Reporting Limit. LOQ standards should be stored in plastic containers with a shelf-life of 6 months.
- 10.10 Interference Check Solutions A and AB are prepared to contain known concentrations of interfering analytes that will provide an adequate test of the correction factors. ICSA / ICSAB standards should be stored in plastic containers with a shelf-life of 6 months.
- 10.11 Laboratory Control Sample and Matrix Spike - see preparation SOP.
- 10.12 Blanks
 - 10.12.1 Method Blanks -see preparation SOP.
 - 10.12.2 The Calibration Blank is prepared by acidifying reagent water to the same concentrations of acid found in the standards and samples.



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11) Method Calibration

- 11.1 Follow policies in ADM-ICAL. If these instructions conflict with ADM-ICAL, follow the instructions in this SOP.
- 11.2 Number of Calibration standards – At least 3 standards and a calibration blank are analyzed for each element at the beginning of the daily sequence. The low standard must be at or below the LOQ. The HLCCV2 standard will define the upper limit of the linear range for the curve.
- 11.3 Initial Calibration Curve Calculation – Internal Standard Method
 - 11.3.1 Yttrium is added to all samples and QC standards to provide a reference for individual performance of each injection. The software divides the element intensity by the internal standard intensity for each injection and provides a “corrected intensity” for each element on the print-out.

$$\text{Corrected Intensity} = \frac{\text{Intensity}_{\text{sample}}}{\text{Intensity}_{\text{internal standard}}}$$

- 11.3.2 The “corrected intensity” is plotted against concentration using the linear regression equation of a line and forced through zero intensity and zero concentration. This calibration curve assumes that the relationship between concentration (the X values) and intensity (the Y values) is linear and that the following equation describes this relationship:

$Y=MX$
Where:
X=concentration
Y= corrected intensity
M=slope of the calibration curve
The intercept is forced to be zero.

Given 2 or more data points, the values for M are calculated using the following equation.

$$M = \frac{\sum_{i=1}^n (X_i Y_i)}{\sum_{i=1}^n (X_i^2)}$$

Where: n=number of standards (includes the blank)

In this equation, the blank is subtracted from all solutions and included in the calculation of the calibration curve



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To avoid bias at the low end of the curve, the curve is forced through zero. Forcing the curve through zero may favor the low end of the curve therefore a MRL Standard (LLCCV) and high level CCV (HLCCV1 and HLCCV2) are analyzed to verify both ends of the curve.

- 11.4 Acceptance Criteria – the correlation coefficient must be 0.998 or greater for target analytes of interest and the ICV must meet limits (see QC section).
- 11.5 Corrective Action – if the ICAL does not meet acceptance criteria, correct the problem and recalibrate. The curve must be acceptable before sample analysis begins.

12) Sample Preparation and Analysis

- 12.1 All samples are digested prior to analysis. Refer to the following Metals Digestion SOPs:
 - MET-3010A Metals Digestion, Waters for ICP
 - MET-3050B Metals Digestion, Soils, Sediments and Sludges for ICP and GFAA
- 12.2 Set up the instrument with proper operating parameters established as detailed below. Operating conditions - The analyst should follow the instructions provided in Table 3.
- 12.3 Before using this procedure to analyze samples, there must be data available documenting initial demonstration of performance. The required data documents the selection criteria of background correction points; linear ranges, and the upper limits of those ranges; the Limits of Detection and Quantitation; and the determination and verification of interelement correction equations or other routines for correcting spectral interferences. This data must be generated using the same instrument, operating conditions and calibration routine to be used for sample analysis. These documented data must be kept on file and be available for review by the data user or auditor. The limits and on-going frequency of these performance demonstrations are provided in the QC Section.
- 12.4 Turn on power supply for the instrument, computer, printer and light the plasma. Allow instrument to warm-up while preparing the run (typically 45-60 minutes before operation, although only ~10 minutes are necessary). The cooling water and the argon are on when the instrument are on.
- 12.5 Profile the instrument on a daily basis, or when maintenance is done to align it optically for both horizontal and vertical optimization in either mode. The vendor recommends aspirating a 1.0 ppm source of manganese. Choose the Tools menu/Spectrometer Control/Optimize “Axial” or “Radial” The instrument automatically adjusts the torch viewing position for maximum intensity.



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12.6 Addition of Internal standard -

12.6.1 Pour 40 mL of the calibration blank, the 3 calibration standards, ICV/CCV standards, LOQ, ICSA, and ICSAB into separate 50 mL centrifuge tubes and add 0.80 mL of the 100 mg/L internal standard solution. All other samples, preparation blanks and laboratory control samples are poured up to 10 mL in a 15 mL centrifuge tube and 0.20 mL of internal standard solution is added. This will give an apparent concentration of 1.00 mg/L Yttrium. Other volumes may be used to result in the same concentration.

12.6.2 Internal standards can be added via pump and mixing block. This technique uses a solution of 10 mg/L Y and 10 mg/L Cs.

12.6.3 Place the tubes on the autosampler and program the software to analyze according to the analytical sequence.

12.7 Analytical Sequence - see the Quality Control Section and ADM-BATCH for frequency and requirements.

12.7.1 Rinse the system with the blank solution before the analysis of each sample. The rinse time will be at least one minute, depending upon the instrument. A reduction in rinse time must be demonstrated to be acceptable.

12.8 Sample Analysis and Evaluation

12.8.1 Sensitivity, LOD, LOQ, precision, linear dynamic range, and interference effects must be established for each individual analyte line on each particular instrument. All measurements must be within the instrument linear range where the correction equations are valid.

12.8.2 Samples which exceed the linear range of the instrument (greater than HLCCV2) must be diluted and reanalyzed according to ADM-DIL.

12.8.3 Evaluate QC according to the QC Section. Repeat samples associated with non-compliant QC whenever possible.

13) Troubleshooting

Maintenance log - All Preventive maintenance, as well as instrument repair, should be documented in the appropriate instrument maintenance log. Most routine maintenance and troubleshooting are performed by ALS staff. Other maintenance or repairs may, or may not require factory service, depending upon the nature of the task. Any maintenance performed by outside services must also be documented - either through notes in the log or through documents provided by the service. The log entries will include the date maintenance was performed, symptoms of the problem, serial numbers of major equipment upgrades or replacements. The datafile name of the first acceptable run after maintenance is to be documented in the maintenance log.

14) Data Acquisition

Data is electronically transferred from the instrument software to LIMS. The preparation information entered to LIMS is used by LIMS to calculate the final results - see Calculation Section.



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15) Calculations and Data Reduction Requirements

- 15.1 Calculations: If dilutions were performed, the appropriate factors must be applied to sample values.
- 15.2 The use of the internal standard is found in the Method Calibration Section.
- 15.3 Sample Calculation (water)

$$\text{Conc. (mg/L)} = \frac{\text{Instrument Reading (mg/L)} \times \text{Final digestion volume (L)}}{\text{Initial volume (L)}}$$

- 15.4 Sample Calculation (soils)

$$\text{Conc. (mg/g)} = \frac{\text{Instrument Reading (mg/L)} \times \text{Final digestion volume (L)}}{\text{Initial mass (g)} \times \text{Percent Solids expressed as a decimal}}$$

16) Quality Control, Acceptance Criteria, and Corrective Action

- 16.1 Instrument values are based on duplicate readings. Precision between the emission readings shall not exceed 20 %RSD. If RSD values exceed 20%, reanalyze the sample. Exception: Analytes <MRL are OK to report with RSD>20%.
- 16.2 Method Blanks
 - Frequency - at least one MB with preparation batch of 20 or fewer samples of the same matrix.
 - Limits - MB values must not exceed the LOQ(1/2 LOQ for DOD).
 - Corrective Action - Fresh aliquots of the samples must be prepared and analyzed again for affected analytes after the source of the contamination has been corrected and acceptable MB values have been obtained. If detections are greater than the LOQ, the batch needs to be redigested if sample concentration is less than 10 times the concentration found in the method blank. If the sample concentration is less than the LOQ the sample does not require redigestion.
- 16.3 HLCCV1 – High standard used in curve
 - Frequency - analyzed once during daily analysis.
 - Limits - Should agree within 10% of the true value.
 - Corrective Action - If HLCCV1 is > 10% different the analysis is judged to be out of control and the source of the problem should be identified and resolved before continuing analysis.



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- 16.4 HLCCV2 – standard slightly higher than calibration for some metals.
- Frequency - Analyzed once during daily analysis.
 - Limits - Should agree within 10% of the true value.
 - Corrective Action - If out of control, client data above the HLCCV1 should be re-analyzed.
- 16.5 ICV –
- Frequency - must immediately follow each calibration.
 - Limits - $\pm 5\%$ for Method 200.7 and $\pm 10\%$ for Method 6010C.
 - Corrective Action – correct the problem and recalibrate the instrument.
- 16.6 CCV
- Frequency - after every tenth sample, and at the end of the sample run
 - Limits - $\pm 10\%$
 - Corrective Action – correct the problem and analyze another CCV. If second CCV fails, recalibrate. Reanalyze affected samples since the last successful CCV. Non-detect samples associated with a high recovery CCV may be reported.
- 16.7 ICB/CCB
- Frequency – ICB after every ICV and CCB after every CCV (at the beginning and end of the run and after every 10 samples)
 - Limits - The results of the calibration blank must be less than the LOQ (less than LOD for DOD).
 - Corrective Action - If the limits are not met, terminate the analysis, correct the problem, recalibrate, and reanalyze the samples affected. Non-detect samples associated with a high blank may be reported.
- 16.8 MS –
- Frequency - one per matrix batch (max. 20 samples) or one per 10 for Method 200.7.
 - Limits - within 70-130% (80-120% for DOD) of the actual value or within the documented historical acceptance limits for each matrix. All elements of client interest are evaluated. Sample concentrations greater than four times the spike concentration are not valid and shall not be evaluated.
 - Corrective Action - If the matrix spike does not meet these criteria, analyze a Post Digestion Spike. DOD requires project-specific DQOs be examined or contact the client for additional measures. See Table F-7.



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16.9 DUP -

- Frequency - one per matrix batch (max. 20 samples) or one per 10 for Method 200.7
- Limits - A control limit of $\pm 20\%$ RPD shall be used for original and duplicate samples greater than or equal to 5X the LOQ. A control limit of \pm the LOQ shall be used if either the sample or duplicate value is less than 5 times the LOQ.
- Corrective Action - If the DUP does not meet criteria, data will be reported with a qualifier.

16.10 Laboratory Control Sample -

- Frequency - one per matrix batch (max. 20 samples).
- Limits - all elements of client interest are evaluated. Client specific QC recoveries may supercede these limits.
 - Waters - Results should be within $\pm 15\%$ of the true value for method 200.7 and $\pm 20\%$ for 6010C.
 - Soils - Results must be within the limits indicated on the Certificate of Analysis for the soil reference.
- Corrective Action - Outlying recoveries may indicate loss of analyte due to digestion procedures or laboratory contamination. If an LCS is found to be out of the specified limits, recalibrate and reanalyze. If the LCS remains out of control with a low bias, redigestion of the entire batch should occur. If the LCS remains out of control with a high bias, redigestion of all positive results (greater than the LOQ) should occur.

16.11 MRL standard (LLICV/LLCCV) - A standard less than or equal to the LOQ is analyzed

- Frequency - at the beginning and end of each daily analytical run but not before the ICV.
- Limits- There are no limits in the 200.7 method. 6010C requires 70-130%. DOD requires 80-120%.
- Corrective Action - If the limits are not met the analysis is stopped and the instrument is recalibrated.

16.12 Interference Check Samples-

16.12.1 ICSA

- Frequency - at the beginning and end of each daily analytical sequence.
- Limits - less than LOQ (less than 2 times LOQ for metals with LOQ $< 10\text{mg/L}$) - DOD requires that the ICSA be less than the LOD unless they are a verified trace impurity from one of the spiked analytes.
- Corrective Action - terminate analysis; locate and correct problem;



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reanalyze ICS and any associated samples. Samples less than LOQ will be flagged and reported.

16.12.2 ICSAB -

- Frequency - immediately after ICSA
- Limits - The analyte recoveries for the AB solution must fall within 20% of the true value
- Corrective Action - terminate analysis; locate and correct problem; reanalyze ICS and all associated samples unless analytes are not detected in the associated samples or interfering elements are not present at the ICSA level.

16.13 Serial Dilution Test -

- Frequency - one each prep batch or when a new or unusual matrix is encountered. Only applicable to samples with a concentration greater than 50 times the LOQ. If no samples in the prep batch are $>50 \times \text{LOQ}$, analyze a post-digestion spike.
- Limits - a 5-fold dilution should agree within $\pm 10\%$ of the original determination.
- Corrective Action - if the results are not within limits, flag the data with a qualifier. For DOD, if the results are not within limits, perform post-digestion spike addition.

16.14 Post Digestion Spike Addition (for 6010C analyses): The spike addition should produce a minimum level of 10 times and a maximum of 100 times the LOQ.

- Frequency - When a dilution test fails, or when no samples have a concentration 50 times the LOQ or if a matrix spike does not yield acceptable results.
- Limits - should be recovered to within 75% to 125% of the known value.
- Corrective action - If the spike is not recovered within the specified limits, a matrix effect has been confirmed and the data must be flagged.

16.15 Instrument Performance

16.15.1 InterElement Correction Factors (IEC)

- Procedure - A calibration curve is analyzed as per instrument specifications. Once completed, individual standards for Al, Ca, Fe, Mg, and Mo (Mo for Optima 4 only) are analyzed. The instrument software then creates an IEC table.
- Frequency - annually, or as needed.
- Limits and corrective action - The ICSA check standard routinely confirms the IECs. If the ICSA is persistently problematic, re-establish IECs.



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16.15.2 Linear Ranges (LR)

- Frequency - Standards (HLCCV1 and HLCCV2) are prepared at the linear range level and analyzed every quarter
- Limits - must be $\pm 5\%$ of true value.
- Corrective Action - If the linear range is too high, lower the linear range and repeat the study.

16.15.3 Instrument Detection Limits (IDL)

- Procedure - Analyze 7 replicates of a low level standard. Repeat twice more for a total of 21 replicates over 3 non-consecutive days.
- Frequency - with initial set-up and after significant change.
- Limits - the calculated IDL must be less than the LOD.
- Corrective Action - If the IDL fails, correct the problem and repeat the study or raise the LOD.

16.15.4 Ongoing verification of detection and quantitation limits is required. See CE-QA011 for requirements.

16.15.5 LOQV (LLQC) - This digested standard spiked at the reporting limit must be analyzed quarterly for DOD (6010C requires "as needed"). The limits are 70-130% for 6010C and LCS limits for DOD. If this QC does not meet these limits, determine the source of the problem. Achieve an acceptable LOQV before continuing.

17) Data Records Management

See CE-GEN003 and ADM-ARCH.

18) Contingencies for Handling Out Of Control Data

If data is produced that is out of control and is not to be re-analyzed due to sample volume restrictions, holding times, or QC controls can not be met, flag and narrate appropriately.

19) Method Performance

Detection and Quantitation limits are determined for all wavelengths utilized for each type of matrix commonly analyzed. See Table 2 for approximate wavelengths. Determine limits according to the requirements in CE-QA011. The supporting information is filed with the QA office.

Demonstration of Capability is performed according to CE-QA003.



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20) Summary of Changes

- Updated to new ALS format
- Incorporated SOP change form for reporting non-detects when MRL standard fails high.
- Added ICP #5
- Removed copies of standards log and referenced Controlled Forms
- Updated MRL for Beryllium

21) References and Related Documents

- Test Methods For Evaluating Solid Waste, Physical/Chemical Methods. USEPA SW-846, Update IV, December 1996.
- Methods For the Determination of Metals in Environmental Samples Supplement I. USEPA/600/R-94/111, May 1994.
- DOD Quality Systems Manual for Environmental Laboratories – Version 4.2, October 2010.

22) Appendix

- DOD Summary
- Table 1 List of Analytes and Practical Quantitation Limits
- Table 2 Potential Interferences
- Table 3 Recommended Wavelengths and Instrument Specifications
- Table F-7 DOD specific Requirements from DOD QSM Version 4.2 October 2010



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DOD SUMMARY

For work for the Department of Defense – the DOD Quality Systems Manual must be followed. The DOD Manual is based on the NELAC Standards with some additional requirements. The exact wording is in Table F-7. The following are the requirements which are different or additional to routine analysis and must be followed for DOD work:

- The CCB must be less than the LOD.
- The Method Blank must be less than ½ the reporting limit (<RL for common laboratory contaminants).
- Apply J flag to all hits between LOD and LOQ.
- The limits for LCS and MS are 80-120%. All targets are spiked and evaluated.
- The IDL must be less than or equal to the LOD.
- Low Level Check Standard (MRL standard) must be spiked at or below the reporting limit and it must have a recovery of 80-120% of the true value when analyzed at the beginning of the run.
- ICS-A absolute value of concentration for all non-spiked analytes must be <LOD (unless they are a verified trace impurity from one of the spiked analytes).
- Serial Dilution Test Corrective Action – If the limits are not met, perform a post digestion spike addition.



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TABLE 1

<i>Analyte</i>	<i>TypicalMRL/ LOQ**</i>	
	<i>Water-Optima</i>	<i>Soil</i>
	<i>ug/L</i>	<i>ug/g</i>
Silver	10	1.00
Aluminum	100	10.0
Arsenic	10	1.0
Boron	200	20.0
Barium	20	2.00
Beryllium	3.0	0.300
Calcium	1000	100.0
Cadmium	5.0	0.500
Cobalt	50	5.00
Chromium	10	1.0
Copper	20	2.00
Iron	100	10.0
Potassium	2000	200
Magnesium	1000	100
Manganese	10	1.0
Molybdenum	25	2.50
Sodium	1000	100
Nickel	40	4.00
Lead	50	5.00
Antimony	60	6.00
Selenium	10	1.00
Tin	500	50.0
Titanium	50	5.00
Thallium	10	1.00
Vanadium	50	5.00
Zinc	20	2.00
Lithium	100	10
Silicon	1000	100
Strontium	100	10
Gold	50	
Gallium	200	
Germanium	100	
Hafnium	100	
Indium	100	
Palladium	100	
Platinum	100	
Ruthenium	100	
Rhenium	100	
Tungsten	100	
Tantalum	100	
Zirconium	100	

**See Data Quality Objectives Table for the most current LOQs



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TABLE 2

POTENTIAL INTERFERENCES AND ANALYTE CONCENTRATION EQUIVALENTS (mg/L)
ARISING FROM INTERFERENCE AT THE 100-mg/L LEVEL

Analyte	Wavelength (nm)	Interferant ^{a,b}									
		Al	Ca	Cr	Cu	Fe	Mg	Mn	Ni	Ti	V
Aluminum	308.215	--	--	--	--	--	--	0.21	--	--	1.4
Antimony	206.833	0.47	--	2.9	--	0.08	--	--	--	0.25	0.45
Arsenic	193.696	1.3	--	0.44	--	--	--	--	--	--	1.1
Barium	455.403	--	--	--	--	--	--	--	--	--	--
Beryllium	313.042	--	--	--	--	--	--	--	--	0.04	0.05
Cadmium	226.502	--	--	--	--	0.03	--	--	0.02	--	--
Calcium	317.933	--	--	0.08	--	0.01	0.01	0.04	--	0.03	0.03
Chromium	267.716	--	--	--	--	0.003	--	0.04	--	--	0.04
Cobalt	228.616	--	--	0.03	--	0.005	--	--	0.03	0.15	--
Copper	324.754	--	--	--	--	0.003	--	--	--	0.05	0.02
Iron	259.940	--	--	--	--	--	--	0.12	--	--	--
Lead	220.353	0.17	--	--	--	--	--	--	--	--	--
Magnesium	279.079	--	0.02	0.11	--	0.13	--	0.25	--	0.07	0.12
Manganese	257.610	0.005	--	0.01	--	0.002	0.002	--	--	--	--
Molybdenum	202.030	0.05	--	--	--	0.03	--	--	--	--	--
Nickel	231.604	--	--	--	--	--	--	--	--	--	--
Selenium	196.026	0.23	--	--	--	0.09	--	--	--	--	--
Sodium	588.995	--	--	--	--	--	--	--	--	0.08	--
Thallium	190.864	0.30	--	--	--	--	--	--	--	--	--
Vanadium	292.402	--	--	0.05	--	0.005	--	--	--	0.02	--
Zinc	213.856	--	--	--	0.14	--	--	--	0.29	--	--

^a Dashes indicate that no interference was observed even when interferents were introduced at the following levels:
Al at 1000 mg/L Cu at 200 mg/L Mn at 200 mg/L
Ca at 1000 mg/L Fe at 1000 mg/L Ti at 200 mg/L
Cr at 200 mg/L Mg at 1000 mg/L V at 200 mg/L

^b The data shown above as analyte concentration equivalents are not the actual observed concentrations. To obtain those data, add the listed concentration to the interferant figure.

^c Interferences will be affected by background choice and other interferences may be present.



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Table 3

Recommended Wavelengths and Instrument Specifications

Suggested wavelengths are listed below:

Analyte	Wavelength
Ag Silver	328.068
Al Aluminum	308.215
B Boron	249.773
Ba Barium	233.527
Be Beryllium	234.861
Ca Calcium	430.253
Cd Cadmium	226.502
Co Cobalt	228.616
Cr Chromium	267.716
Cu Copper	324.754
Fe Iron	238.863
Mg Magnesium	279.079
Mn Manganese	257.610
Mo Molybdenum	202.030
Na Sodium	330.237
Ni Nickel	231.604
Pb Lead	220.353
Sb Antimony	206.833
Sn Tin	189.933
Ti Titanium	334.941
V Vanadium	292.402
Zn Zinc	206.191
Y Yttrium	371.030

Other wavelengths may be substituted if they can provide the needed sensitivity and are corrected for spectral interference. Because of differences among various makes and models of spectrometers, specific instrument operating conditions cannot be provided. The instrument operating conditions herein are recommended based upon manufacturer's instrument manuals.

Current Method Operating Conditions are as follows, these conditions may vary to optimize the instrument for different analyses:

Parameter	Radial Plasma	Axial Plasma
Resolution	Fixed	Fixed
Purge Gas Flow	Normal	Normal
Read Time (min/max sec.)	5/20	5/50
Replicates	2	2
Plasma (L/min)	15	15
Aux. (L/min)	0.5	0.3
Nebulizer Flow (L/min)	0.72	0.56
Power (watts)	1300	1450
Viewing Height (mm)	15	15



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Table F-7. Inorganic Analysis by Inductively Coupled Plasma (ICP) Atomic Emission Spectrometry and Atomic Absorption Spectrophotometry (AA) (Methods 6010 and 7000 Series)

QC Check	Minimum Frequency	Acceptance Criteria	Corrective Action	Flagging Criteria	Comments
Demonstrate acceptable analytical capability	Prior to using any test method and at any time there is a significant change in instrument type, personnel, test method, or sample matrix.	QC acceptance criteria published by DoD, if available; otherwise, method-specified criteria.	Recalculate results; locate and fix problem, then rerun demonstration for those analytes that did not meet criteria (see Section C.1.f).	NA.	This is a demonstration of analytical ability to generate acceptable precision and bias per the procedure in Appendix C. No analysis shall be allowed by analyst until successful demonstration of capability is complete.
LOD determination and verification (See Box D-13)					
LOQ establishment and verification (See Box D-14)					
Instrument detection limit (IDL) study (ICP only)	At initial set-up and after significant change in instrument type, personnel, test method, or sample matrix.	IDLs shall be \leq LOD.	NA.	NA.	Samples may not be analyzed without a valid IDL.
Linear dynamic range or high-level check standard (ICP only)	Every 6 months.	Within $\pm 10\%$ of true value.	NA.	NA.	

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Table F-7. Inorganic Analysis by Inductively Coupled Plasma (ICP) Atomic Emission Spectrometry and Atomic Absorption Spectrophotometry (AA) (Methods 6010 and 7000 Series) (continued)

QC Check	Minimum Frequency	Acceptance Criteria	Corrective Action	Flagging Criteria	Comments
Initial calibration (ICAL) for all analytes ICP: minimum one high standard and a calibration blank; GFAA: minimum three standards and a calibration blank; CVAA: minimum 5 standards and a calibration blank	Daily ICAL prior to sample analysis.	If more than one calibration standard is used, $r \geq 0.995$.	Correct problem, then repeat ICAL.	Flagging criteria are not appropriate.	Problem must be corrected. No samples may be run until ICAL has passed.
Second source calibration verification (ICV)	Once after each ICAL, prior to beginning a sample run.	Value of second source for all analyte(s) within $\pm 10\%$ of true value.	Correct problem and verify second source standard. Rerun ICV. If that fails, correct problem and repeat ICAL.	Flagging criteria are not appropriate.	Problem must be corrected. No samples may be run until calibration has been verified.
Continuing calibration verification (CCV)	After every 10 field samples and at the end of the analysis sequence.	ICP: within $\pm 10\%$ of true value; GFAA: within $\pm 20\%$ of true value; CVAA: within $\pm 20\%$ of true value.	Correct problem, rerun calibration verification. If that fails, then repeat ICAL. Reanalyze all samples since the last successful calibration verification.	If reanalysis cannot be performed, data must be qualified and explained in the case narrative. Apply Q-flag to all results for the specific analyte(s) in all samples since the last acceptable calibration verification.	Problem must be corrected. Results may not be reported without a valid CCV. Flagging is only appropriate in cases where the samples cannot be reanalyzed.
Low-level calibration check standard (ICP only)	Daily, after one-point ICAL.	Within $\pm 20\%$ of true value.	Correct problem, then reanalyze.	Flagging criteria are not appropriate.	No samples may be analyzed without a valid low-level calibration check standard. Low-level calibration check standard should be less than or equal to the reporting limit.



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Table F-7. Inorganic Analysis by Inductively Coupled Plasma (ICP) Atomic Emission Spectrometry and Atomic Absorption Spectrophotometry (AA) (Methods 6010 and 7000 Series) (continued)

QC Check	Minimum Frequency	Acceptance Criteria	Corrective Action	Flagging Criteria	Comments
Method blank	One per preparatory batch.	No analytes detected > ½ RL and greater than 1/10 the amount measured in any sample or 1/10 the regulatory limit (whichever is greater). Blank result must not otherwise affect sample results. For common laboratory contaminants, no analytes detected > RL (see Box D-1).	Correct problem, then see criteria in Box D-1. If required, reprep and reanalyze method blank and all samples processed with the contaminated blank.	If reanalysis cannot be performed, data must be qualified and explained in the case narrative. Apply B-flag to all results for the specific analyte(s) in all samples in the associated preparatory batch.	Problem must be corrected. Results may not be reported without a valid method blank. Flagging is only appropriate in cases where the samples cannot be reanalyzed.
Calibration blank	Before beginning a sample run, after every 10 samples, and at end of the analysis sequence.	No analytes detected > LOD.	Correct problem. Re-prepare and reanalyze calibration blank. All samples following the last acceptable calibration blank must be reanalyzed.	Apply B-flag to all results for specific analyte(s) in all samples associated with the blank.	
Interference check solutions (ICS) (ICP only)	At the beginning of an analytical run.	ICS-A: Absolute value of concentration for all non-spiked analytes < LOD (unless they are a verified trace impurity from one of the spiked analytes); ICS-AB: Within ± 20% of true value.	Terminate analysis; locate and correct problem; reanalyze ICS, reanalyze all samples.	If corrective action fails, apply Q-flag to all results for specific analyte(s) in all samples associated with the ICS.	
LCS containing all analytes to be reported	One per preparatory batch.	QC acceptance criteria specified by DoD, if available; see Box D-3 and Appendix G.	Correct problem, then reprep and reanalyze the LCS and all samples in the associated preparatory batch for failed analytes, if sufficient sample material is available (see full explanation in Appendix G).	If reanalysis cannot be performed, data must be qualified and explained in the case narrative. Apply Q-flag to specific analyte(s) in all samples in the associated preparatory batch.	Problem must be corrected. Results may not be reported without a valid LCS. Flagging is only appropriate in cases where the samples cannot be reanalyzed.



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Table F-7. Inorganic Analysis by Inductively Coupled Plasma (ICP) Atomic Emission Spectrometry and Atomic Absorption Spectrophotometry (AA) (Methods 6010 and 7000 Series) (continued)

QC Check	Minimum Frequency	Acceptance Criteria	Corrective Action	Flagging Criteria	Comments
Matrix spike (MS)	One per preparatory batch per matrix (see Box D-7).	For matrix evaluation, use QC acceptance criteria specified by DoD for LCS.	Examine the project-specific DQOs. If the matrix spike falls outside of DoD criteria, additional quality control tests are required to evaluate matrix effects.	For the specific analyte(s) in the parent sample, apply J-flag if acceptance criteria are not met.	For matrix evaluation only. If MS results are outside the LCS limits, the data shall be evaluated to determine the source of difference and to determine if there is a matrix effect or analytical error.
Matrix spike duplicate (MSD) or sample duplicate	One per preparatory batch per matrix (see Box D-7).	MSD: For matrix evaluation use QC acceptance criteria specified by DoD for LCS. MSD or sample duplicate: RPD \leq 20% (between MS and MSD or sample and sample duplicate).	Examine the project-specific DQOs. Contact the client as to additional measures to be taken.	For the specific analyte(s) in the parent sample, apply J-flag if acceptance criteria are not met.	The data shall be evaluated to determine the source of difference.
Dilution test (ICP and GFAA only)	One per preparatory batch.	Five-fold dilution must agree within \pm 10% of the original measurement.	ICP: Perform post-digestion spike (PDS) addition; GFAA: Perform recovery test.	Flagging criteria are not appropriate.	Only applicable for samples with concentrations $>$ 50 x LOQ.
Post-digestion spike (PDS) addition (ICP only)	When dilution test fails or analyte concentration in all samples $<$ 50 x LOD.	Recovery within 75-125% (see Table B-1).	Run all associated samples in the preparatory batch by method of standard additions (MSA) or see flagging criteria.	For the specific analyte(s) in the parent sample, apply J-flag if acceptance criteria are not met.	Spike addition should produce a concentration of 10 – 100 x LOQ.
Recovery test (GFAA only)	When dilution test fails or analyte concentration in all samples $<$ 25 x LOD.	Recovery within 85-115%.	Run all associated samples in the preparatory batch by method of standard additions (MSA) or see flagging criteria.	For the specific analyte(s) in the parent sample, apply J-flag if acceptance criteria are not met.	
Method of standard additions (MSA)	When matrix interference is confirmed.	NA.	NA.	NA.	Document use of MSA in the case narrative.
Results reported between DL and LOQ	NA.	NA.	NA.	Apply J-flag to all results between DL and LOQ.	

DOCUMENT TITLE:	HEXAVALENT CHROMIUM BY ION CHROMATOGRAPHY FOR WATER AND SOIL EXTRACTS
REFERENCED METHOD:	EPA 218.6, EPA218.7, SW 7199, EPA 0061, NIOSH 7605
SOP ID:	GEN-7199
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EPA 218.6, EPA218.7, SW 7199, EPA 0061, NIOSH 7605

SOPID:	GEN-7199	Rev. Number:	6	Effective Date:	10/14/13
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HEXAVALENT CHROMIUM BY ION CHROMATOGRAPHY FOR WATER AND SOIL EXTRACTS

1) Scope And Applicability

- 1.1 This SOP uses the following methods for the determination of Hexavalent chromium: EPA Method 7199 for water samples and soil extracts; 218.6 for drinking water and other waters; 218.7 for drinking water; NIOSH 7605 for air filter extracts, and 0061 for air impingers.
- 1.2 Although testing for Hexavalent chromium is not currently regulated in drinking water (except through total chromium limits), the EPA recommended monitoring (Dec 2010) of Hexavalent chromium using 218.6, modified to reach a MDL of 0.02 ug/L (Jan 2011). The EPA released method 218.7 in January of 2012. Method 218.7 has been included in UCMR3 and is to be used for testing of certain public water systems designated by EPA starting January 2013. The UCMR3 MRL is 0.03 ug/L.
- 1.3 This method may not be useful for the analysis of samples containing high levels of anionic species such as sulfate and chloride, since these species may cause column overload.
- 1.4 Samples containing high levels of organics or sulfides cause rapid reduction of soluble Cr(VI) to Cr(III).
- 1.5 This method should be used by analysts experienced in the use of ion chromatography and in the interpretation of ion chromatograms.
- 1.6 Current reporting limits:
- | | Drinking Water | Water | Soils |
|------------|----------------|----------------------------------|------------|
| 218.7 | 0.03 ug/L | NA | NA |
| 218.6 | 0.02 ug/L | 0.02 ug/L (LL)
0.01 mg/L (RL) | NA |
| 7199 | NA | 0.01 mg/L | 0.40 mg/kg |
| NIOSH 7605 | | see GEN-N7605 | |
- 1.7 Current MDLs:
- | | Drinking Water | Water | Soils |
|------------|----------------|-----------------------------------|-------------|
| 218.7 | 0.01 ug/L | NA | NA |
| 218.6 | 0.01 ug/L | 0.01 ug/L (LL)
0.001 mg/L (RL) | NA |
| 7199 | NA | 0.001 mg/L | 0.026 mg/kg |
| NIOSH 7605 | | see GEN-N7605 | |

2) Summary of Procedure

- Waters by 218.6 and 218.7 are preserved with a combined buffer/dechlorinating reagent which complexes free chlorine and increases the pH to a specified value (see Preservation Section). Water samples by 7199 are not chemically preserved with the buffer. Soils are digested by EPA 3060A (GEN-3060A). Air filters are digested by NIOSH 7605 (GEN-N7605).
- A measured volume of the sample is introduced into an ion chromatograph. CrO₄²⁻ is separated from other matrix components on an anion exchange column. CrO₄²⁻ is derivatized with 1,5-diphenylcarbazide in a post-column reaction and is detected spectrophotometrically at a wavelength of 530 nm. Cr(VI) is qualitatively identified via retention time, and the concentration of CrO₄²⁻ in the sample is calculated using the integrated peak area and the external standard technique.



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3) Definitions

- 3.1 Initial Calibration - analysis of analytical standards for a series of different specified concentrations; used to define the linearity and dynamic range of the response of the system.
- 3.2 Matrix Spike (MS) - In the matrix spike analysis, a predetermined quantity of standard solution is added to a sample matrix prior to sample preparation and analysis. The purpose of the matrix spike is to evaluate the effects of the sample matrix on the methods used for the analyses. Percent recovery is calculated for the analyte detected. EPA methods 218.6 and 218.7 call the matrix spike a Laboratory Fortified Sample Matrix (LFM or LFSM).
- 3.3 Soluble Matrix Spike (MS-Sol) - Only applicable to soil preparation batches digested by GEN-3060A. In the soluble matrix spike analysis, a predetermined quantity of a soluble standard solution of the analyte is added to a sample matrix prior to sample digestion and analysis. The purpose of the soluble matrix spike is to evaluate the effects of the sample matrix on the methods used for the analyses. Percent recovery is calculated for the analyte detected.
- 3.4 Insoluble Matrix Spike (MS-Insol) - Only applicable to soil preparation batches digested by GEN-3060A. In the insoluble matrix spike analysis, a predetermined quantity of an insoluble standard of the analyte is added to a sample matrix prior to sample digestion and analysis. The purpose of the insoluble matrix spike is to evaluate the dissolution during the digestion process and effects of the sample matrix on the dissolution. Percent recovery is calculated for the analyte detected.
- 3.5 Duplicate Sample (DUP) - A laboratory duplicate. The duplicate sample is a separate field sample aliquot that is processed in an identical manner as the sample proper. The relative percent difference between the samples is calculated and used to assess analytical precision. EPA methods 218.6 and 218.7 use the abbreviation LD (Laboratory Duplicate).
- 3.6 Relative Percent Difference (RPD) - The absolute value of the difference of two values divided by the average of the same two values. Used to compare the precision of the analysis. The result is always a positive number.
- 3.7 Method Blank (MB) - The method blank is an artificial sample designed to monitor introduction of artifacts into the process. The method blank is carried through the entire preparation and analytical procedure. For waters by 218.7, the MB is the Laboratory Reagent Blank (LRB) or CCB and contains the method preservative.
- 3.8 Laboratory Control Sample (LCS) - In the LCS or blank spike analysis, a predetermined quantity of standard solution is added to a blank prior to sample analysis. Percent recovery is calculated for the analyte detected. This LCS is a check on the calibration only and has not undergone digestion. The LCS-Insol is a check on the preparation batch.
- 3.9 Blank Spike (LCS-Insol) - Only applicable to soil preparation batches digested by GEN-3060A. In the LCS-Insol analysis, a predetermined quantity of an insoluble standard solution is added to a blank prior to sample digestion and analysis. Percent recoveries are calculated for the analyte detected.



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- 3.10 Post Verification Spike (PVS) – The PVS (sometimes referred to as PDS for Post Digestion Spike by New Jersey) analysis is designed to verify that neither a reducing nor a chemical interference is affecting the analysis. In the PVS analysis, a predetermined quantity of a soluble standard solution is added to a sample after sample digestion and analysis. Percent recoveries are calculated for the analyte detected.
- 3.11 Independent Calibration Verification/ Continuing Calibration Verification (ICV / CCV) – ICV/CCV solutions are made from a stock solution which is different from the stock used to prepare calibration standards and is used to verify the validity of the standardization. This is sometimes referred to by New Jersey as CCS (Calibration Check Standard), or QCS (Quality Control Sample). Method 218.7 calls the ICV a QCS and the CCV a CCC (Continuing Calibration Check).
- 3.12 Initial Calibration Blank (ICB) - A blank run immediately after calibration to determine if the instrument is adequately zeroed.
- 3.13 Continuous Calibration Blank (CCB) - A blank run periodically (every 10 samples after the CCV) to ensure the instrument is still zeroed adequately. The CCB may be used for the MB for waters.
- 3.14 Preparation Batch - Samples digested together as a unit, not to exceed 20 investigative samples. The Digested QC (PB, LCSSs, MSs, DUP) are associated with the preparation batch and may be analyzed in separate analytical batches. See ADM-BATCH for further discussion.
- 3.15 Analytical Batch - Samples analyzed together as a unit, not to exceed 10 injections for 218.7 or 20 injections for 7199 and 218.6. This batch must contain all of the undigested or instrument QC and may not necessarily contain all of the digested QC associated with the samples in the analytical batch. Typically, the preparation batch QC is analyzed prior to the analysis of client samples to verify that the associated samples are valid for use. See ADM-BATCH for further discussion. The analytical sequence for 218.7 is attached.
- 3.16 Method Detection Limit (MDL): a statistically derived value representing the lowest level of target analyte that may be measured by the instrument with 99% confidence that the value is greater than zero.
- 3.17 Method Reporting Limit (MRL): The minimum amount of a target analyte that can be measured and reported quantitatively.
- 3.18 Limit of Detection (LOD): An estimate of the minimum amount of a substance that an analytical process can reliably detect. An LOD is analyte- and matrix- specific and may be laboratory-dependent. For DOD, the smallest amount or concentration of a substance that must be present in a sample in order to be detected at a high level of confidence (99%). At the LOD, the false negative rate (Type II error) is 1%.
- 3.19 Limit of Quantitation (LOQ)/Reporting Limit: The minimum levels, concentrations, or quantities of a target that can be reported with a specified degree of confidence. For DOD, the lowest concentration that produces a quantitative result within specified limits of precision and bias. The LOQ shall be set at or above the concentration of the lowest calibration standard.



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4) Health and Safety Warnings

- 4.1 All appropriate safety precautions for handling reagents and samples must be taken when performing this procedure. This includes the use of personnel protective equipment, such as, safety glasses, lab coat and the correct gloves.
- 4.2 Chemicals, reagents and standards must be handled as described in the company safety policies, approved methods and in MSDSs where available. Refer to the Environmental, Health and Safety Manual and the appropriate MSDS prior to beginning this method.
- 4.3 Refer to the Safety Manual for further discussion of general safety procedures and information.

4.4 Waste Management And Pollution Prevention

Hexavalent chromium solutions should be dumped in the red inorganic carboys which will later be emptied by qualified personnel. All other waste can be flushed down the drain with large amounts of water. See SMO-SPLDIS for further information on sample disposal.

It is the laboratory's practice to minimize the amount of acids and reagent used to perform this method wherever feasible. Standards are prepared in volumes consistent with methodology and only the amount needed for routine laboratory use is kept on site. The threat to the environment from solvent and reagents used in this method can be minimized when recycled or disposed of properly.

The laboratory will comply with all Federal, State and local regulations governing waste management, particularly the hazardous waste identification rules and land disposal restrictions as specified in the EH&S Manual.

5) Cautions

- See SOP GEN-300.0 for preventive maintenance of the instrument.
- Clean labware according to GEN-GC. NOTE: Chromic acid must not be used for the cleaning of labware.

6) Interferences

- 6.1 Contamination - A trace amount of Cr is sometimes found in reagent grade salts. Since a concentrated buffer solution is used in this method to adjust the pH of samples, reagent blanks are analyzed to assess for potential Cr(VI) contamination. Contamination can also come from improperly cleaned glassware or contact or caustic or acidic reagents of samples with stainless steel or pigmented material.
- 6.2 OXIDATION-REDUCTION (REDOX) CONCERNS - To ensure sample integrity, Cr(VI) must be protected from reduction, and Cr(III), if present, must not oxidize to Cr(VI) during sample storage or processing.
 - 6.2.1 Within the normal pH range in drinking water, Cr(VI), present as a result of pollution or oxidation of Cr(III) in source water during treatment, forms oxyanions, which are typically represented as HCrO_4^- and CrO_4^{2-} . The very stable CrO_4^{2-} anion dominates above pH 8; therefore, the method preservative is designed to buffer samples to at least pH 8. Chromate compounds are quite soluble, mobile and stable, particularly in an oxidizing environment. In contrast, soluble Cr(III) species oxidize to Cr(VI) in the presence of free chlorine, although natural organic matter in surface water sources may



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complex Cr(III), slowing its oxidation even in a highly oxidizing environment. The rate of Cr(III) oxidation increases with chlorine concentration and is pH-dependent. For these reasons, the preservation includes ammonium ions to complex free chlorine. The resulting formation of chloramines minimizes, but does not completely prevent, the oxidation of Cr(III). EPA experiments have demonstrated the ability of the method preservative to minimize the oxidation of Cr(III) and to prevent the reduction of Cr(VI) for at least 14 days in drinking water from ground and surface water sources.

- 6.2.2 A reducing tendency of the sample matrix may change Cr (VI) to Cr(III). The reducing/oxidizing tendency of each sample may be characterized using additional analytical parameters, such as pH, ferrous iron, sulfides, and oxidation/reduction potential. Other indirect indicators of reducing/oxidizing tendency include Total Organic Carbon (TOC), Chemical Oxygen Demand (COD), and Biochemical Oxygen Demand (BOD). Analysis of these additional parameters establishes the tendency of Cr(VI) to exist in the unspiked sample(s) and may be necessary to assist in the interpretation of QC data for matrix spike recoveries outside conventionally accepted criteria. Reduction of Cr(VI) to Cr(III) can occur in an acidic medium. However, at a pH of 6.5 or greater, CrO_4^{2-} (which is less reactive than the HCrO_4^-) is the predominant species.
- 6.3 Sample ionic strength may enhance or suppress Cr(VI) response; however, the 4-mm column systems used tolerate typical concentrations of common anions in drinking water in combination with method preservative. Acceptable method performance has been demonstrated by EPA for samples with hardness up to 350 mg/L as CaCO_3 and total organic carbon content of 3 mg/L.
- 6.4 Overloading of the analytical column capacity with high concentrations of anionic species, especially chloride and sulfate, will cause a loss of Cr(VI). The column can handle samples containing up to 5% sodium sulfate or 2% sodium chloride. Poor recoveries from fortified samples and tailing peaks are typical manifestations of column overload.

7) Personnel Qualifications and Responsibilities

Personnel must be trained according to CE-QA003.

It is the responsibility of the Project Manager to obtain information from the client before sampling. The laboratory needs to know whether the client will require additional investigation of the sample matrix in the case of matrix spike failures. A reducing condition in the sample matrix will reduce Cr(VI) to Cr(III) causing low bias. Additional parameters – sulfide, pH, REDOX, ferrous irons, BOD, COD, TOC may be used to demonstrate a reducing condition in the sample matrix. If any or all of these parameters are to be analyzed, additional aliquots are to be sampled and the tests are to be scheduled when hexavalent chromium is scheduled due to holding time limitations. If pH and REDOX are to be analyzed, they are to be analyzed from the same DI extract. It is the responsibility of the Project Manager to appropriately flag the data and make notes in the case narrative about the nature of the matrix, if applicable (see the attached flowchart).

It is the responsibility of the analyst to perform the analysis according to this SOP and to complete all documentation required for data review. Analysis and interpretation of the results are performed by personnel in the laboratory who have demonstrated the ability to generate acceptable results utilizing this SOP. Final review and sign-off of the data is performed by the department supervisor or designee.

8) Sample Containers, Collection, Preservations, and Storage



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- 8.1 Bottleware - If the lab provides the bottleware, waters are sampled in certified clean 250-mL narrow-mouth, high-density polypropylene containers, or equivalent. Soils are sampled in certified clean 4 or 8 oz. glass soil jars. The client must provide additional aliquots of soil sample if further investigation into the reducing/oxidizing nature of the sample is to be done.
- 8.2 Pretesting - Pre-test each lot of the the preservation solution (buffer). Add 2.5 mL of the buffer to a 250 mL sample bottle and fill with DI. Preservative is acceptable if result is <MDL of the low level analysis.
- 8.3 Temperature Requirements - Samples are typically shipped and stored at 0-6°C. Drinking water samples for UCMR3 method 218.7 received within 48 hours of collection must be ≤10°C or they must be rejected. If UCMR3 samples are received after 48 hours of collection they must be ≤6°C or they must be rejected. The project manager is to notify the client that a new sample must be collected.

8.4 Preservation and Holding Time:

	Drinking Water 218.7	Drinking Water 218.6	Non- Potable Water 218.6	Waters 7199	Soil 7199	Air filters NIOSH 7605	Air (impingers) collected by 0061
Filter	No	Required see 6.4.1	Required see 6.4.1	Optional – Field or Lab	NA	NA	Field
Preservative	NH ₄ OH/ (NH ₄) ₂ SO 4 liquid	NH ₄ OH/ (NH ₄) ₂ SO ₄ liquid	NH ₄ OH/ (NH ₄) ₂ SO ₄ liquid	None	None	None	None
pH	>8.0	9.0-9.5	9.3-9.7	unpreserved	NA	NA	NA
Residual Chlorine	<0.1 mg/L	NA	NA	NA	NA	NA	NA
Holding Time	14 days	5 days	28 days	24 hours	7 days from extraction	28 days (recommend ed)	14 days

Sample pH and residual chlorine (if applicable) is measured and recorded upon receipt (see attached preservation sheet and SMO-GEN for procedures). The preservation sheet and associated steps may be completed by SMO or Wetchem personnel. SMO also records the lot number of the bottle and preservative upon receipt.

- For 218.7, the sample is only valid if received pH>8.0 and Residual chlorine <0.1 mg/L. The sample must be rejected if it does not meet receipt requirements. It may not be reported for UCMR3.



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- For 218.6, the holding times listed only apply if the sample is properly filtered and buffered within 24 hours of sampling. If the buffering requirements (pH) are not met, and the sample is still within 24 hours of collection, the pH is adjusted with the appropriate buffer or ammonium hydroxide.

8.4.1 For 218.6, samples shall be filtered and preserved with buffer solution within 24-hours of collection (preferably in the field at the time of collection). Procedures are as follows, listed in order of preference

8.4.1.1 Filter and Buffer in the Field during time of collection:

Filter sample aliquot using an in-line filter or plastic syringe filtration unit equipped with a 0.45um membrane filter (mfr. Gelman, Millipore, or equivalent).

- Non-Potable Water - Adjust the pH of the filtered sample to pH 9.3 - 9.7 using buffer solution provided by the laboratory. pH meter should be capable of ± 0.03 SU. Holding time is 28-days from collection.
- Drinking Water - Adjust the pH of the filtered sample to pH 9.0 - 9.5 using buffer solution provided by the laboratory. pH meter should be capable of ± 0.03 SU. Holding time is 5-days from collection.

The bottle set shall include unpreserved 125 or 250 mL plastic bottles. A minimum of 25 mL filtered sample should be provided for analysis.

8.4.1.2 Filter and Buffer in the Field during time of collection –No pH meter available:

Filter sample aliquot using an in-line filter or plastic syringe filtration unit equipped with a 0.45um membrane filter (mfr. Gelman, Millipore, or equivalent). Filtered sample shall be collected in a pre-preserved sample bottle containing buffer solution. Samples must be received by the laboratory within 24-hr of collection to ensure proper pH has been achieved by the buffer solution. The lab will adjust the pH, if necessary, within 24-hr of collection. Holding time is 28-days.

The bottle set shall include pre-preserved 125-ml plastic bottles with 1 mL buffer solution, or equivalent. A minimum of 25 mL filtered sample should be provided for analysis.

8.4.1.3 Filter and Buffer at the Laboratory –No filtration unit or pH meter available:

Collect sample in an un-preserved plastic bottle. Samples must be received by the laboratory within 24-hr for immediate analysis, or in-lab filtration and pH adjustment with buffer solution. Holding time is 28-days for non-potable water and 5 days for drinking water if properly filtered and buffered.

The bottle set shall include unpreserved 125 or 250-mL plastic bottles.



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- 8.5 Method 7199 water sample: If it is impossible to complete 2 simultaneous injections of all samples within the 24 hour holding time, perform the injections separately in order to meet one injection within holding time of as many samples as possible.
- 8.6 If investigation is requested, the investigative tests (such as sulfide, pH, REDOX and Fe2+) must be scheduled with the Cr6+ test because some of the investigative test have shorter holding times than hexavalent chromium. These tests must be completed prior to the evaluation of the Cr6+ QC so that the results are available if needed.
- 8.7 Sample handling, storage, receipt, and custody procedures are discussed in SMO-GEN and SMO-ICOC.

9) Equipment and Supplies

Instrument ID	Instrument Configuration	Manufacturer Part	Serial Number	Year Acquired
IC #8 (R-IC-08)	Ion Chromatograph	Dionex ICS-2100	12030901	2012
	Heated Conductivity Cell	DS6	12030664	
	Reagent Pump	AXP	20045075	
	Variable Wavelength Detector	ICS Series VWD	12031294	
	Autosampler	AS-AP	12031171	
	Loop	1000 uL		
	Analytical Column	Dionex AS-7 2x250mm		
	Computer Workstation	Dell Optiplex 790	15105322945	
	Analytical Software	Chromeleon 7.0	151838	

- 9.1.1 Reaction Coil: Dionex P/N 042631 750 uL
- 9.1.2 Column Heater - integrated
- 9.1.3 Pressurized eluent container, plastic, two liter size.
- 9.1.4 Nitrogen Tanks
- 9.2 Labware
 - 9.2.1 Class A volumetric flasks, and graduated cylinders.
 - 9.2.2 Assorted pipettes - of acceptable precision and accuracy – calibrated according to ADM-PCAL.
 - 9.2.3 Disposable syringes - 50-mL, with male luer-lock fittings.
 - 9.2.4 Dionex ONGuard-P sample pretreatment cartridges - p/n 39597
 - 9.2.5 Syringe filters - 0.45-µm, Millipore, p/n SLHV 025 NK.



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- 9.2.6 Storage bottles - high density polypropylene or amber glass, 1-L capacity.
- 9.2.7 Orion Model 720A pH meter or Orion SA 520 pH meter, or equivalent, calibrated according to ADM-phSupport, with accuracy ± 0.05 pH.
- 9.2.8 Analytical and TopLoading Balances – calibrated according to ADM-DALYCK.

10) Standards and Reagents

- 10.1 All standards must be traceable using the laboratory lot system (CE-QA007)
- 10.2 All purchased standards are certified by the vendor. Certificates of Analysis are kept in the department until the standards are no longer being used – at which time they are filed with QA. Certificates of Analysis are available upon request.
- 10.3 Purchased Reagents and Standards - store at room temperature and expire per Expiration Policy (CE-QA012) unless otherwise indicated.
 - 10.3.1 Ammonium hydroxide, NH_4OH . EM Science cat. #AX1303-14, or equivalent, sp.gr. 0.902, CAS RN 1336-21-6.
 - 10.3.2 Ammonium sulfate, $(\text{NH}_4)_2\text{SO}_4$. EM Science cat #SX0597-1, or equivalent, CAS RN 7783-20-2
 - 10.3.3 1,5-Diphenylcarbazide. EM Science cat. #DX2205-1, or equivalent, CAS RN 140-22-7.
 - 10.3.4 Methanol, HPLC grade, EM Science cat #MX0475-1, or equivalent
 - 10.3.5 Sulfuric acid, concentrated, EM Science, Omnitrace, cat #SX1247-2 or equivalent.
 - 10.3.6 Cr(VI) Standards Stock Solution (1000 mg/L)
 - 10.3.7 Cr(VI) Reference Stock Solution (1000 mg/L) – the Reference Stock is to be from a separate manufacturer of the Standard Stock.
 - 10.3.8 Chlorine Residual Test Strips capable of reading to 0.1 mg/L. HF Scientific Micro Check Test Strips.
- 10.4 Prepared Reagents
 - 10.4.1 Buffer Solution. Dissolve 3.3 g of ammonium sulfate in 75 mL of reagent water in a 100 mL volumetric flask. Add 6.5 mL of ammonium hydroxide. Dilute to volume (100 mL) with reagent water. Degas the solution with helium gas for 5-10 minutes prior to use. Store at 0-6°C for up to one year.
 - 10.4.2 Buffered DI (Dilution Water). A batch of reagent grade water must be prepared by adjusting the pH within the range of 9-9.5 using the buffer solution. Use this solution for diluting working standards and high level samples. Prepare fresh before use. pH range of 9.3-9.7 shall be used for Method 218.6 waters as per Footnote 20 of EPA Method Update Rule and FAQ-Cr6.
 - 10.4.3 Eluent.- Dissolve 33 g of ammonium sulfate in 500 mL of DI and add 6.5 mL of ammonium hydroxide. Dilute to one liter with reagent water. Degas the solution with helium gas for 5-10 minutes prior to use. Expires 1 month from preparation.
 - 10.4.4 Post-column reagent. Dissolve 0.5 g of 1,5 diphenylcarbazide in 100 mL of HPLC grade methanol in a 1000 mL volumetric flask. In a separate container, add about 500 mL DI, then add 28 mL of 98% sulfuric acid, mix and degas with helium gas for 5-10 minutes. Carefully combine the degassed acid with the



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diphenylcarbazide/methanol solution, introducing as little air as possible. This method of preparation reduces the frequency and intensity of air spikes in the chromatography. Dilute to volume with reagent water. Store in amber glass at 0-6°C. This reagent should be made fresh for the low level analysis, and may be kept no longer than 5 days.

10.5 Standards: 7199/218.6 Regular Level. Prepare fresh before use.

10.5.1 Regular Level Intermediate Standards Working Stock (10 mg/L). Do two 1/10 serial dilutions of the 1000 ppm standard stock solution, using buffered DI.

10.5.2 Initial Calibration Standards: Prepare a series of standards and a blank by pipetting suitable volumes of the Intermediate Standards Working Stock and buffered DI into a dispo cup.

The typical calibration for regular level (7199 and 218.6) is as follows:

Standard #	Volume (mL) of 10.0 mg/L Standard	Volume buffered DI (mL)	Final Concentration (mg/L)
6	1.00	9.0	1.00
5	0.70	9.3	0.70
4	0.50	9.5	0.50
3	0.10	9.9	0.10
2	1/10 of #3	-	0.01
1	0.0	10	0.000

10.5.3 Regular Level Intermediate Reference Working Stock (10 mg/L) - Do two 1/10 serial dilutions of the 1000 ppm reference stock solution, using buffered DI (as above).

10.5.4 Regular Level ICV/CCV (0.5 mg/L): In a 10mL dispo cup add 9.5 mL buffered DI (as above) and 0.5 mL Intermediate Reference Working Stock.

10.5.5 ICB / CCB / Method Blank / LRB (waters): DI preserved as a sample.

10.5.6 LCS - Regular Level Waters (0.20 mg/L): Add 0.20 mL of 10 mg/L Intermediate Standards Working Stock to 9.8 mL buffered DI. Analyze as a sample.

10.5.7 MS Regular Level Waters (0.2 mg/L): Add 0.20 mL of 10 mg/L Intermediate Standards Working Stock to 10 mL sample.

10.5.8 MB/LCS/MS Soils: prepared and digested according to GEN-3060A.

10.5.9 Post Verification Spike (PVS) – soil extracts only - after digestion, filtration, pH adjustment to 9.0-9.5, and dilution to 100 mLs (as described in GEN-3060), add 0.45 mL of 100 mg/L Cr(VI) standard stock solution to a 45 mL aliquot (this is equal to 40 mg/Kg) OR a concentration twice the original sample, whichever is higher.



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10.6 Standards: 218.7 Prepare fresh before use.

10.6.1 1000 ug/L Standards Working Stock - Add 75 mL of DI to a 100 mL volumetric flask. To this add 1.0 mL Buffer Solution. Add 0.1 mL 1000 mg/L Cr(VI) Standard Stock Solution. Bring to volume with DI. Prepare fresh before use. Mix thoroughly.

10.6.2 100 ug/L Standards Working Stock - Add 75 mL of DI to a 100 mL volumetric flask. To this add 1.0 mL Buffer Solution. Add 10 mL 1000 ug/L Cr(VI) Standard Working Stock Solution. Bring to volume with DI. Prepare fresh before use. Mix thoroughly.

10.6.3 Initial calibration standards: For each standard, add 75 mL DI to a 100 mL volumetric flask. To this add 1/0 mL Buffer Solution. Then follow the recipe below:

Standard #	Volume (mL) of 1000 µg/L Standard	Final Volume with DI (mL)	Final Concentration (µg/L)
8	0.5	100	5.00
7	0.1	100	1.00
6	0.7 mL of 100 ug/L Std	100	0.70
5	0.5 mL of 100 ug/L Std	100	0.50
4	0.3 mL of 100 ug/L Std	100	0.30
3	0.1 mL of 100 ug/L Std	100	0.10
2	10 mL Std#4	100	0.030
1	10 mL Std#3	100	0.010

10.6.4 MS Spiking Stock (10 µg/L). Perform two 1/10 serial dilutions of the 1000 µg/L Standard Stock using DI.

10.6.5 1000 ug/L Reference Working Stock - Add 75 mL of DI to a 100 mL volumetric flask. To this add 1.0 mL Buffer Solution. Add 0.1 mL 1000 mg/L Cr(VI) Reference Stock Solution. Bring to volume with DI. Prepare fresh before use. Mix thoroughly.

10.6.6 ICV/CCVs:

10.6.6.1 Low Level (LL-CCV): 0.03 ug/L -

- 100 ug/L Reference Working Stock - Add 75 mL of DI to a 100 mL volumetric flask. To this add 1.0 mL Buffer Solution. Add 10 mL 1000 ug/L Reference Stock Solution. Bring to volume with DI. Prepare fresh before use. Mix thoroughly.
- 0.30 ug/L Reference Working Stock - To a 100 mL volumetric flask add about 75 mL DI. To this add 1.0 mL Buffer Solution and 0.3 mL of the 100 ug/L Reference stock. Bring to volume with DI. Prepare fresh before use. Mix thoroughly.
- To a separate 100 mL volumetric flask, add about 75 mL DI. To this add 1.0 mL Buffer Solution and 10 mL of the 0.30 ug/L Reference Working Stock. Bring to volume with DI. Prepare fresh before use. Mix thoroughly.

10.6.6.2 Mid Level CCV (ML-CCV) / ICV: 2.5 ug/L - To a 100 mL volumetric flask, add about 50 mL DI. To this add 1.0 mL Buffer Solution and 0.25 mL of 1000 ug/L Reference stock. Bring to volume with DI and mix. Prepare fresh before use.



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- 10.6.6.3 High Level CCV (HL-CCV): 4.0 ug/L - to a 100 mL volumetric flask, add about 50 mL DI. To this add 1.0 mL Buffer Solution and 0.4 mL of 1000 ug/L Reference. Bring to volume with DI and mix. Prepare fresh before use.
- 10.6.7 ICB / CCB / Method Blank / LRB (waters): In a 250 mL volumetric flask, add 2.5 mL liquid preservative to about 100 mL DI. Bring to volume with DI and transfer to a 250 mL HDPE sample bottle, received from SMO.
- 10.6.8 MS/LFSM - Add 0.20 mL of 10 µg/L Low Level LCS/MS Spiking Stock to 10 mL pH adjusted sample.
- 10.7 Standards: 218.6 Low Level -prepare fresh before use
- 10.7.1 1000 ug/L Standards Working and Reference Stocks - same as 218.7.
- 10.7.2 100 ug/L Standard Working Stock - make a 1/10 dilution of the 1000 ug/L Standard Working stock.
- 10.7.3 ICAL standards -
- | Standard # | Volume (mL) of 1000 µg/L Standard | Final Volume with buffered DI (mL) | Final Concentration (µg/L) |
|------------|-----------------------------------|------------------------------------|----------------------------|
| 6 | 0.1 | 100 | 1.00 |
| 5 | 0.7 mL of 100 ug/L Std | 100 | 0.70 |
| 4 | 0.5 mL of 100 ug/L Std | 100 | 0.50 |
| 3 | 0.2 mL of 100 ug/L Std | 100 | 0.20 |
| 2 | 10 mL of Std#6 | 100 | 0.10 |
| 1 | 10 mL Std#3 | 100 | 0.020 |
| 0 | 0 | 100 | 0 |
- 10.7.4 LCS/MS Spiking Stock (10 µg/L). Perform two 1/10 serial dilutions of the 1000 µg/L Standard Stock.
- 10.7.5 ICV/CCV (0.5 µg/L): To 5 mL buffered DI (as above) add 5 mL 1.0 µg/L Low Level Intermediate Reference Working Stock.
- 10.7.6 LCS (0.20 µg/L): Add 0.20 mL of 10 µg/L Low Level LCS/MS Spiking Stock to 9.8 mL buffered DI. Analyze as a sample.
- 10.7.7 Matrix Spike(0.20 µg/L): Add 0.20 mL of 10 µg/L Low Level LCS/MS Spiking Stock to 10 mL pH adjusted sample.
- 10.7.8 ICB/CCB/MB - same as 218.7



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11) Method Calibration

- 11.1 Follow policies in ADM-ICAL unless otherwise specified in this SOP.
- 11.2 Initial Calibration
 - 11.2.1 For New Jersey - Inject a calibration blank. Be sure this calibration blank is less than the MDL before continuing.
 - 11.2.2 Number of Standards - Calibrate the instrument using the standards prepared as per the Standards and Reagents Section. Six or more standards must be used for 218.7. Three or more standards must be used for 218.6. A blank and three or more standards must be used for 7199. If a quadratic fit is to be used, analyze at least 6 standard levels plus blank.
 - 11.2.3 Frequency - Initially and whenever continuing calibration verification criteria cannot be met.
 - 11.2.4 Calibration Fit- Construct a calibration curve of analyte response (peak area) versus analyte concentration. The curve may be first order ($y=mx+b$), or second order (quadratic). Weighting may be used ($1/x$).
 - 11.2.5 Limits -
 - 11.2.5.1 For first order polynomial, the coefficient of correlation must be 0.999 or greater. File the printout of the linear regression with the ICAL.
 - 11.2.5.2 For second order, mark the linear regression printout so that it is clear that the linear was for verification only. The accuracy of the quadratic calibration is verified by assessing the recovery of each standard. The recovery of each standard above the LOQ must be within 10% for 218.6 and 7199.
 - 11.2.5.3 The LOQ standard must be within $\pm 50\%$ of the true value.
 - 11.2.5.4 For 218.7, all standards above the LOQ must be within $\pm 15\%$ of the true value (regardless of curve type).
 - 11.2.6 ICV(QCS) - Analyze an ICV immediately after the calibration standards. The ICV must be within limits to use the curve. The limits are 90-110% for 7199, 95-105% for 218.6, and 85-115% for 218.7. If the correctly prepared ICV is not compliant, the second source standard does not verify the curve. Fix the problem and recalibrate. If the ICV was prepared incorrectly, the curve may be used if a correctly prepared ICV verifies the curve.



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11.3 Continuing Calibration Verification and Blank

11.3.1 Frequency - run a CCV and CCB set to start a daily run, every ten injections (or 24 hours, whichever is more frequent), and at the close of the run. For water samples, one blank counts as both the CCB and the MB. For 218.7, the CCV concentration varies (see attached Analytical Sequence and Standards and Reagents Section).

11.3.2 Limits -

Method	CCV % Recovery	Method	ICB/CCB/MB Limit
7199	90-110	7199 7605	<MRL
218.6	95-105	7199 NJ	<MDL
218.7	85-115	218.6	<LOD
NIOSH 7605	85-115	218.7	<0.01 ug/L
		NIOSH 7605	<MRL

11.3.3 CCV Corrective Action - If a CCV fails, corrective actions must be performed. If routine corrective action fails to produce a second consecutive (immediate) acceptable CCV, then either the lab has to demonstrate acceptable performance after corrective action with two consecutive CCVs or a new ICAL must be performed.

11.3.4 Blank Corrective Action - Sample results greater than 10 times the result of the contaminated blank may be reported without qualification. Samples <10x blank contamination must be reanalyzed with a compliant blank whenever possible. If samples associated with a contaminated blank are not repeated (due to holding time restrictions, etc.), the data must be qualified on the report.

12) Sample Preparation and Analysis

12.1 Sample preparation.

12.1.1 Waters - Allow pH-adjusted samples to equilibrate to ambient temperature prior to analysis. Samples that have not been pH adjusted should be adjusted to pH 9-9.5 by dropwise addition of buffer solution. Record the pH adjustment and the pH meter ID on the benchsheet. If salts are formed as a result of the pH adjustment, the filtrate must be filtered again prior to analysis. pH range for 218.6 non-potable waters shall be 9.3-9.7 due to footnote 20 of EPA method update rule and FAQ-Cr6. All drinking waters are to be pH 9.0-9.5.

12.1.2 Soils - Digest according to GEN-3060A except pH adjust the sample to 9.0-9.5 instead of 7.5±0.5.

12.1.3 Air Filters - Prepare according to GEN-7605.



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12.2 Establish Operating Conditions:

	IC #5	IC #8
Warm up	45-60 minutes	45-60 minutes
Eluent flow rate	1.0 mL/min	0.36 mL/min
Post-column flow rate	~0.33 mL/min	~0.12 mL/min
Column heater	30°C	30°C
Sample loop	2000 uL (LL) or 100 uL (RL)	1000 uL
Wavelength	530 nm	530 nm

Check flow rate of waste after the flow cell prior to calibration and sample analysis.

12.3 Sample Analysis.

12.3.1 Standards and samples are injected onto the column using an autosampler.

12.3.2 Color Interference - The guard column should be removing any inherent color in samples. If, however, a sample appears colored after elution through the columns, it is possible that not all sample color/organic material was removed and a false positive due to color could occur. At this point, pass a sample aliquot through Dionex ONGuard-P syringe filters and re-analyze. Alternatively, re-analyze a sample aliquot, but replace the post-column reagent with the matrix match reagent and subtract the matrix-match reagent result from the post-column reagent result.

12.3.3 Double Injection – 7199 requires that samples are injected twice (double injections are not required for 218.6 nor 218.7). The RPD between the samples must be less than 20 if the sample concentration is \geq four times the reporting limit. A control limit of \pm the reporting limit is used when the sample concentration is $<$ four times the reporting limit. If it is impossible to meet holding times for both injections, it is best to inject samples once within holding time and make the second injection out of holding time. Report both results. Label the comments field R1, R2, R3, R4, etc.

12.4 Sample Evaluation –

12.4.1 Each chromatogram is reviewed for compliance and initialed by the reviewer. All chromatographic baselines should be examined by a knowledgeable analyst to ensure that proper integrations have been made by the analytical software. Especially for the low level analysis, care must be taken to ensure proper identification and integration of all low level peaks approaching a signal to noise ratio of 3:1. When the data system incorrectly quantitates or identifies analytes, manual integration is necessary. Data must be integrated consistently between standards, samples, and QC. See CE-QA002 for integration requirements and manual integration documentation.

12.4.2 Samples or extracts exceeding the highest calibration standard (including PVS) must be diluted using buffered DI and re-analyzed.

12.4.3 Sample must be bound by acceptable analytical QC and from a batch with acceptable batch QC. Double injection must meet limits.



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13) Troubleshooting

- 13.1 See Instrument manual or maintenance log for help with solving specific analytical or instrument problems.
- 13.2 Maintenance log - All Preventive maintenance, as well as instrument repair, should be documented in the appropriate instrument maintenance log. Most routine maintenance and troubleshooting are performed by laboratory staff. Other maintenance or repairs may, or may not require factory service, depending upon the nature of the task. Any maintenance performed by outside services must also be documented – either through notes in the log or through documents provided by the service. The log entries will include the date maintenance was performed, symptoms of the problem, serial numbers of major equipment upgrades or replacements. The datafile name of the first acceptable run after maintenance is to be documented in the maintenance log.

14) Data Acquisition

Data is uploaded electronically from the instrument to LIMS. As applicable, sample volumes, weights, and dilutions are entered to LIMS and final results are adjusted by LIMS accordingly. Calculations are presented in the following section.

15) Calculation and Data Reduction Requirements

- 15.1 Determine the concentration of the injected sample from the calibration curve.
 - 15.1.1 For waters, multiply the injected concentration by dilution (use 1 if there is no dilution). Report in mg/L except report in µg/L for low level 218.6 and 218.7.
 - 15.1.2 For digested soils:

$$\text{Concentration (mg/Kg dry wt.)} = \frac{A \times D \times E}{B \times C}$$

where: A = Concentration observed in the digest (mg/L)
 B = Initial moist sample weight (g)
 C = % Solids/100
 D = Dilution factor
 E = Final digest volume (mL)
 - 15.1.3 For air stips (NIOSH 7605) – see GEN-N7605.
- 15.2 Report both of the results from the double injection (if applicable).
- 15.3 For NJ - Data is “R” flagged if the MS is outside of 50-150%, if CCV or LCS is out of control, if calibration CC<0.999, if calibration blank >MDL, if PB >MDL for samples >MDL, if a water sample is run beyond 48 hours from sampling, if required QC is not performed, or if a soil sample is not redigested as required.
- 15.4 For NJ - Data is “J” flagged if the result is run between 24 and 48 hours from sampling (waters only), if QC is not performed at the correct frequency, if the MS is outside of limits but within 50-150%, if the RPD of a double injection or a DUP is greater than 20, if digested QC fails the initial and the redigestion, or if PDS <85%.
- 15.5 If samples are redigested, report as replicates. Both original and redigested data is reported.
- 15.6 All sample data and QC data, including calibration verification must reference the name



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(date or filename) of the ICAL on the raw data report. The current system lists the ICAL under the heading of "Quantif. Method" and the convention is to use the IC# and date in the name. For Example: "5-121610" is IC#5 on 12/16/10.

- 15.7 Data must be reviewed by the analyst and a peer (supervisor or qualified analyst) using a Data Quality Checklist before the results are validated and reported to the client. Further data review policies and procedures are discussed in ADM-DREV.

16) Quality Control, Acceptance Limits and Corrective Action

- 16.1 ICV/CCV and ICB/CCB/MB requirements are in the Calibration Section.

- 16.2 For Water Samples

- 16.2.1 Matrix Spike -

- 16.2.1.1 Frequency - A minimum of one matrix spike sample per sample batch (1/10 for 218.6) must be analyzed to check for matrix interference.

- 16.2.1.2 Limits - The recovery of the matrix spike must be within limits in the Data Quality Objectives Table. For 218.7 - the MS must be within 15%.

Method	MS Limit
7199	DQO Table
7199 NJ	DQO Table
218.6	90-110
218.7	85-115

- 16.2.1.3 Corrective Action - If the matrix spike recovery fails these limits, report with appropriate qualifiers.

- 16.2.2 DUP/MSD

- 16.2.2.1 Frequency - A minimum of one duplicate sample per sample batch must be analyzed to check for precision. Alternatively, a Matrix Spike Duplicate may be used instead of a Duplicate. The MSD is required for UCMR3.

- 16.2.2.2 Limits:

	RPD	Other
7199	<20	+/-MRL if <4xMRL
218.6	<20	+/-MRL if <4xMRL
218.7	<15	NA

- 16.2.2.3 Corrective Action - If the RPD is out of limits, repeat the sample and duplicate unless there is assignable matrix interference, historical failures, or lack of volume. If an out of control duplicate is not repeated, note the reason on the data quality checklist. If, at the time when the problem is discovered, the sample exceeds twice the holding time, discuss with supervisor or Project Manager prior to repeating the samples. Report all of the replicates and explain in the checklist for the case narrative.



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16.2.3 LCS -

16.2.3.1 Frequency -An LCS must be analyzed with every batch for 7199 and 218.6. An LCS is not required for 218.7 (performance measured with CCV). See GEN-N7605 for air strips.

16.2.3.2 Limits -

Method	Limits
7199	DQO Table
218.6	90-110%
218.7	NA

16.2.3.3 Corrective Action - If the LCS fails these limits, correct the problem and reanalyze the affected samples or flag the associated data.

16.3 For Samples Digested by GEN-3060A – see also the attached flowchart. Undigested QC (LCS, MB, DUP, MS) is not required to be analyzed with the digested QC in a 7199 run which only has digested samples.

16.3.1 MB - A preparation blank must be prepared and analyzed with each digestion batch. Detected Cr(VI) concentrations must be less than the reporting limit (less than the MDL for New Jersey) or the batch must be redigested and reanalyzed. If the samples are out of holding time, redigest and reanalyze and both sets of data will be reported. If insufficient sample volume necessitate the use the data, flag the data associated with the non-compliant PB (the entire preparation batch – not just those in the analytical batch).

16.3.2 LCS Insoluble – An insoluble LCS must be prepared and analyzed with each digestion batch. Recovery must be within 80-120% or the entire sample batch must be redigested and reanalyzed. If the samples are out of holding time, redigest and reanalyze and both sets of data will be reported. If insufficient sample volume necessitate the use the data, flag the data associated with the non-compliant LCS (the entire preparation batch – not just those in the analytical batch)

16.3.3 DUP - A separately prepared duplicate soil sample must be analyzed at a frequency of one per batch. Duplicate samples must have a Relative Percent Difference (RPD) of $\leq 20\%$, if the sample concentration is \geq four times the reporting limit. A control limit of \pm the reporting limit is used when the sample concentration is $<$ four times the reporting limit. If the RPD of the duplicates is out of limits, repeat the sample and duplicate unless there is assignable matrix interference, historical failures, or lack of volume. If an out of control duplicate is not repeated, note the reason on the data quality checklist. If, at the time when the problem is discovered, the sample exceeds twice the holding time, discuss with supervisor or Project Manager prior to repeating the samples. Report all of the replicates and explain in the checklist for the case narrative.

16.3.4 MS- Both soluble an insoluble digested matrix spikes must be analyzed at a frequency of one each per batch. Both matrix spike recoveries must be within 75-125% of the true value. If either of the matrix spike recoveries are not within these recovery limits the entire batch must be redigested and reanalyzed. If the reanalysis also fails the 75-125% limit, sample data is flagged. Exception – if the sample concentration is greater than 4 times the spike concentration, the spike is “diluted out” and is not used to evaluate the batch (redigestion is not required). The client should be notified of the possible condition of the sample and further investigation is needed using the oxidation/reduction parameters discussed in 4.1. If the samples are out of



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holding time for the reanalysis, redigest and reanalyze and both sets of data will be reported. If insufficient sample volume necessitate the use the data, flag the data associated with the non-compliant LCS (the entire preparation batch – not just those in the analytical batch).

- 16.3.5 A post-digestion Cr(VI) matrix spike must be analyzed per batch, whether or not the MS passed or failed. Use the attached PVS Calculation sheet. The criteria for the post digestion matrix spike recovery is 85-115% recovery. If spike recovery is outside limits, and the matrix spike also failed, no further action is needed apart from the corrective action for the MS. The corrective actions will show whether these digestates may contain soluble reducing agents for Cr(VI). If the PVS fails and the MS passes, reanalyze the PVS.

17) Data Records Management

See CE-GEN003 and ADM-ARCH

18) Contingencies for Handling Out of Control Data

If data is produced that is out of control and is not to be re-analyzed due to sample volume restrictions, holding times, or QC controls can not be met, data is flagged with the appropriate data qualifiers.

19) Method Performance

- 19.1 Reporting limits are based upon an MDL study performed according to CE-QA011 and filed in the MDL binders in the QA office.
- 19.2 Demonstration of Capability is performed according to CE-QA003.
- 19.3 From the EPA Method 7199:
- Single Laboratory Precision and Accuracy is available in Table 3
 - Single Analyst Precision, overall precision and Recovery from Multilaboratory Study is available in Table 4.

20) Summary of Changes

- Added NIOSH 7605 throughout.
- Removed IC-5
- Updated to current ALS format.

21) References and Related Documents

- Test Methods for Evaluating Solid Waste Physical/Chemical Methods, USEPA SW-846.
- NJDEP Standard Operating Procedure (SOP No.5.A.10) Dated August 15, 2005: Standard Operating Procedure for Analytical Data Validation of Hexavalent Chromium.
- Methods for the Determination of Metals in Environmental Samples, Supplement I. EPA/600/R-94/111. May 1994. Method 218.6 revision 3.3.
- Method 218.7: Determination of Hexavalent Chromium in Drinking Water by Ion Chromatography with Post-Column Derivatization and UV-Visible Spectroscopic Detection. EPA Document Number EPA 815-R-11-005. Version 1.0, November 2011.
- NIOSH Manual of Analytical Methods, Fourth Edition, Method 7605 Issue 1, March 15, 2003.



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22) Attachments

- Table 3 Single Laboratory Precision and Accuracy
- QC Flowchart
- PVS Calculation Sheet
- PH Adjustment Sheet – controlled separately on the Controlled Forms section of the Rochester Intranet.
- eH/pH diagram
- Analytical Sequence



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TABLE 3
 SINGLE-LABORATORY PRECISION AND ACCURACY

Sample Type	Cr(VI) (µg/L) ^(a)	Percent Mean Recovery	RPD ^(b)
Reagent Water	100	100	0.8
	1000	100	0.0
Drinking Water	100	105	6.7
	1000	98	1.5
Ground Water	100	98	0.0
	1000	96	0.8
Primary Sewage	100	100	0.7
Wastewater	1000	104	2.7
Electroplating	100	99	0.4
Wastewater	1000	101	0.4

^(a) Sample spiked at this concentration level.

^(b) RPD - relative percent difference between duplicates.

TABLE 4
 SINGLE-ANALYST PRECISION, OVERALL PRECISION AND RECOVERY
 FROM MULTILABORATORY STUDY

	Reagent Water (6-960 µg/L)	Matrix Water (6-960 µg/L)
Mean Recovery	$X = 1.020C + 0.592$	$X = 0.989C - 0.411$
Overall Standard Deviation	$S_R = 0.035X + 0.893$	$S_R = 0.059X + 1.055$
Single-Analyst Standard-Deviation	$S_R = 0.021X + 0.375$	$S_R = 0.041X + 0.393$

X = Mean concentration; µg/L, exclusive of outliers.

C = True value, µg/L.

S_R = Overall standard deviation.

S_R = Single-Analyst standard deviation.

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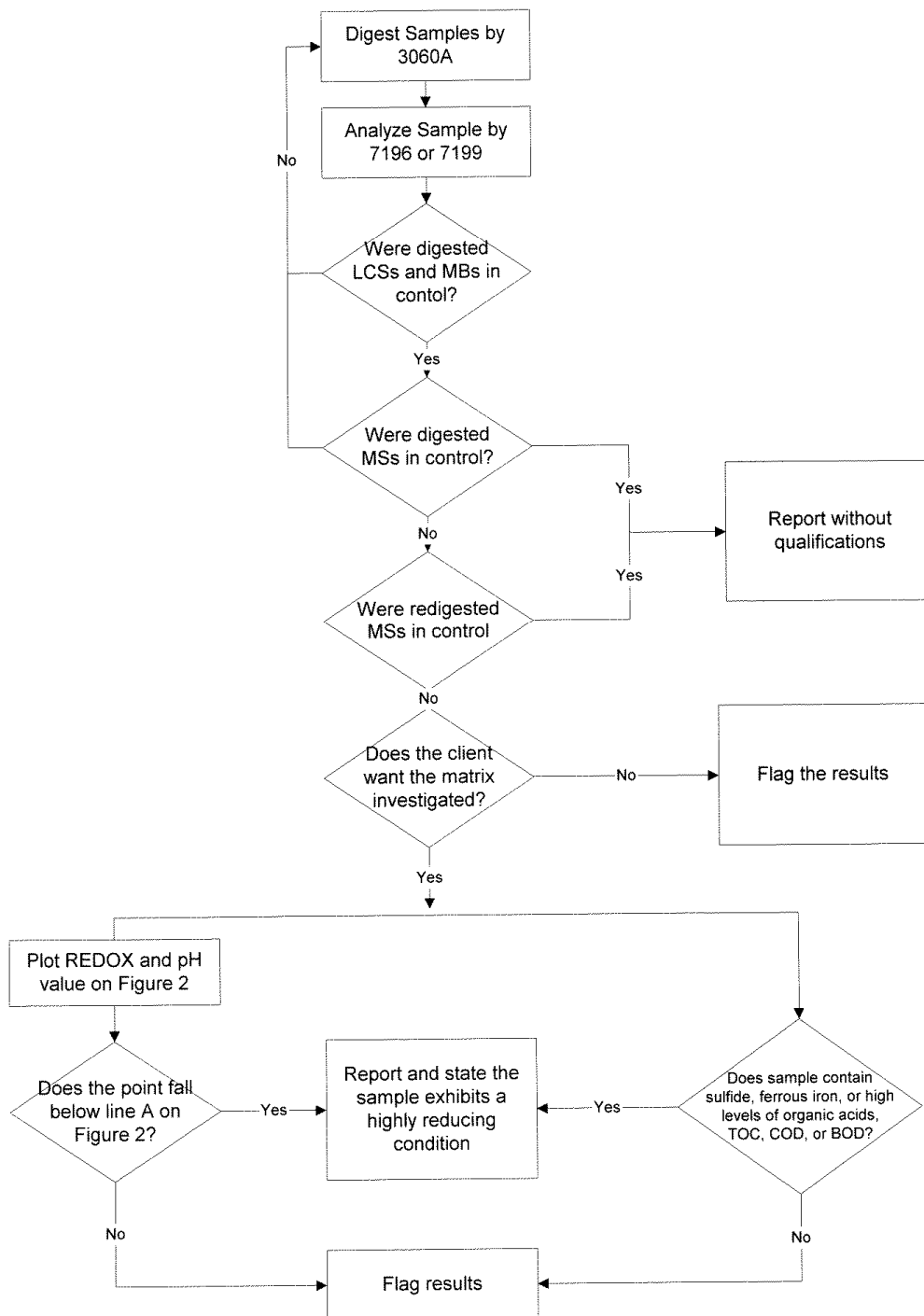
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HEXAVALENT CHROMIUM QUALITY CONTROL FLOW CHART



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$$\text{True Value} = \frac{A \times B}{C}$$
[illegible]



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Date/Time Received	Sample ID	Analysis	Matrix	Date/Time Sampled	Sample Filtered	Filter Lot ID	Chlorine Residual (mg/L) 218.7 only	pH at Receipt	pH Adjustment	Analyst/ Date/ Time pH Adjustment	Solution Used For pH Adjust	Solution Lot ID
		7199 218.6 RL 218.6 LL 218.7	Water Drinking Water		Yes No Field						Buffer 10%H ₂ SO ₄ 10%NH ₄ OH NH ₄ OH(conc)	
		7199 218.6 RL 218.6 LL 218.7	Water Drinking Water		Yes No Field						Buffer 10%H ₂ SO ₄ 10%NH ₄ OH NH ₄ OH(conc)	
		7199 218.6 RL 218.6 LL 218.7	Water Drinking Water		Yes No Field						Buffer 10%H ₂ SO ₄ 10%NH ₄ OH NH ₄ OH(conc)	
		7199 218.6 RL 218.6 LL 218.7	Water Drinking Water		Yes No Field						Buffer 10%H ₂ SO ₄ 10%NH ₄ OH NH ₄ OH(conc)	
		7199 218.6 RL 218.6 LL 218.7	Water Drinking Water		Yes No Field						Buffer 10%H ₂ SO ₄ 10%NH ₄ OH NH ₄ OH(conc)	
		7199 218.6 RL 218.6 LL 218.7	Water Drinking Water		Yes No Field						Buffer 10%H ₂ SO ₄ 10%NH ₄ OH NH ₄ OH(conc)	
		7199 218.6 RL 218.6 LL 218.7	Water Drinking Water		Yes No Field						Buffer 10%H ₂ SO ₄ 10%NH ₄ OH NH ₄ OH(conc)	
		7199 218.6 RL 218.6 LL 218.7	Water Drinking Water		Yes No Field						Buffer 10%H ₂ SO ₄ 10%NH ₄ OH NH ₄ OH(conc)	
		7199 218.6 RL 218.6 LL 218.7	Water Drinking Water		Yes No Field						Buffer 10%H ₂ SO ₄ 10%NH ₄ OH NH ₄ OH(conc)	
		7199 218.6 RL 218.6 LL 218.7	Water Drinking Water		Yes No Field						Buffer 10%H ₂ SO ₄ 10%NH ₄ OH NH ₄ OH(conc)	
		7199 218.6 RL 218.6 LL 218.7	Water Drinking Water		Yes No Field						Buffer 10%H ₂ SO ₄ 10%NH ₄ OH NH ₄ OH(conc)	

Drinking water	218.7	Drinking water	218.6	Non-Pot Water	218.6	Water	7199
Filter	No	Required	Required	Required	Optional unpreserved - adjust to 9.0-9.5		
pH	>8.0	9.0-9.5	NA	9.3-9.7	NA		
Res Chlorine	<0.1 mg/L	NA	NA	NA	NA		
Holding Time	14 days	5 days	28 days	28 days	24 hours		

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SMO Cr6+ preservation log r1 1/10/13
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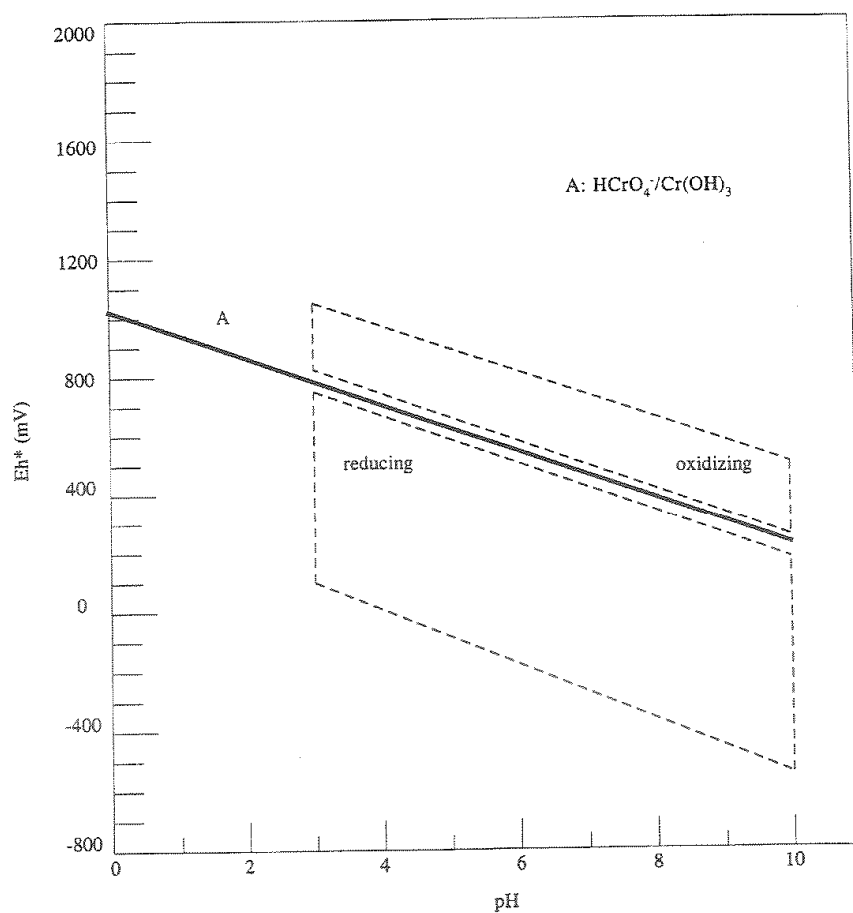


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FIGURE 2
Eh/pH PHASE DIAGRAM

The dashed lines define Eh-pH boundaries commonly encountered in soils and sediments.



* Note the Eh values plotted on this diagram are corrected for the reference electrode voltage: 244 mV units must be added to the measured value when a separate calomel electrode is used, or 199 mV units must be added if a combination platinum electrode is used.

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Analytical Sequence for Method 218.7

Injection Number	Sample ID	Injection Number	Sample ID
1	Calibration Standard 1	22	Analytical Sample 9
2	Calibration Standard 2	23	Analytical Sample 10
3	Calibration Standard 3	24	Analytical Sample 11
4	Calibration Standard 4	25	Analytical Sample 12
5	Calibration Standard 5	26	Analytical Sample 13
6	Calibration Standard 6	27	Analytical Sample 14
7	Calibration Standard 7	28	Analytical Sample 15
8	LL-CCV	29	Analytical Sample 16
9	CCB	30	Analytical Sample 16 Spike
10	Analytical Sample 1	31	Analytical Sample 16 Spike Dup
11	Analytical Sample 2	32	HL-CCV
12	Analytical Sample 3	33	CCB
13	Analytical Sample 4	34	Analytical Sample 17
14	Analytical Sample 5	35	Analytical Sample 18
15	Analytical Sample 6	36	Analytical Sample 19
16	Analytical Sample 7	37	Analytical Sample 20
17	Analytical Sample 8	38	Analytical Sample 20 Spike
18	Analytical Sample 8 Spike	39	Analytical Sample 20 Spike Dup
19	Analytical Sample 8 Spike Dup	40	ML-CCV
20	ML-CCV	41	CCB
21	CCB		



DOCUMENT TITLE:

*ALKALINE DIGESTION FOR
HEXAVALENT CHROMIUM IN SOIL*

REFERENCED METHOD:

SW846 3060A

SOP ID:

GEN-3060

REV. NUMBER:

3

EFFECTIVE DATE:

3/1/2013



ALKALINE DIGESTION FOR
HEXAVALENT CHROMIUM IN SOIL

SOPID: GEN-3060

Rev. Number: 3

Effective Date: 3/1/2013

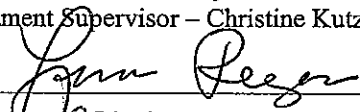
Approved By:


Department Supervisor – Christine Kutzer

Date:

2/13/13

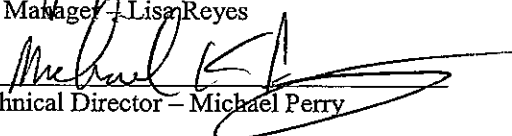
Approved By:


QA Manager – Lisa Reyes

Date:

2/15/2013

Approved By:


Technical Director – Michael Perry

Date:

2/15/13

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1) Scope and Applicability

- 1.1 This SOP uses USEPA Method 3060A for extracting hexavalent chromium [Cr(VI)] from soluble, adsorbed, and precipitated forms of chromium compounds in soil, sludge, sediment, and some industrial waste materials. To quantify total Cr(VI) in a solid matrix, three criteria must be satisfied: (1) the extracting solution must solubilize all forms of Cr(VI), (2) the conditions of the extraction must not induce reduction of native Cr(VI) to Cr(III), and (3) the method must not cause oxidation of native Cr(III) contained in the sample to Cr(VI).
- 1.2 The quantification of Cr(VI) in the digestates produced by this SOP is performed using Method 7196A (See GEN-7196A) or Method 7199 (see GEN-7199).
- 1.3 The reporting limits are listed in the analytical SOPs.

2) Summary of Procedure

- 2.1 This procedure is an alkaline digestion to solubilize Cr(VI) compounds in solid samples. The pH of the digestate must be carefully adjusted during the digestion procedure. Failure to meet the pH specifications will necessitate redigestion of the samples. The sample is digested using a Na₂CO₃/NaOH solution and heated to dissolve the Cr(VI) and stabilize it against reduction to Cr(III).
- 2.2 Analysis of Cr(VI) solubilized in the alkaline digestate is accomplished by reaction with diphenylcarbohydrazide (Method 7196A or 7199). It is highly selective for Cr(VI), and few interferences are encountered when it is used on alkaline digestates.

3) Definitions

- 3.1 Soluble Matrix Spike (MS-Sol) - In the soluble matrix spike analysis, a predetermined quantity of a soluble standard solution of the analyte is added to a sample matrix prior to sample digestion and analysis. The purpose of the soluble matrix spike is to evaluate the effects of the sample matrix on the methods used for the analyses. Percent recovery is calculated for the analyte detected.
- 3.2 Insoluble Matrix Spike (MS-Insol) - In the insoluble matrix spike analysis, a predetermined quantity of an insoluble standard of the analyte is added to a sample matrix prior to sample digestion and analysis. The purpose of the insoluble matrix spike is to evaluate the dissolution during the digestion process and effects of the sample matrix on the dissolution. Percent recovery is calculated for the analyte detected.
- 3.3 Duplicate Sample (DUP) - A laboratory duplicate. The duplicate sample is a separate field sample aliquot that is processed in an identical manner as the sample proper. The relative percent difference between the samples is calculated and used to assess analytical precision.
- 3.4 Method Blank (MB) - The method blank is an artificial sample designed to monitor introduction of artifacts into the process. The method blank is carried through the entire analytical procedure.
- 3.5 Insoluble Laboratory Control Sample (LCS-Insol) - In the LCS-Insol analysis, a predetermined quantity of an insoluble standard solution is added to a blank prior to sample digestion and analysis. Percent recovery is calculated for the analyte detected.



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- 3.6 Post Verification Spike (PVS) – A predetermined quantity of standard solution is added to a sample matrix after digestion and after the sample is pH adjusted and brought up to 100 mLs. Percent recoveries are calculated for the analyte added. This is described in the analytical procedures.
- 3.7 Preparation Batch - Samples digested together as a unit, not to exceed 20 samples. The Digested QC samples are associated with the preparation batch and may be analyzed in separate analytical batches. See ADM-BATCH for further discussion.

4) Responsibilities

- 4.1 It is the responsibility of the Project Manager to obtain information from the client before sampling. The laboratory needs to know whether the client will require additional investigation of the sample matrix in the case of matrix spike failures. A reducing condition in the sample matrix will reduce Cr(VI) to Cr(III) causing low bias. Additional parameters – sulfide, pH, REDOX, ferrous irons, BOD, COD, TOC may be used to demonstrate a reducing condition in the sample matrix. If any or all of these parameters are to be analyzed, additional aliquots are to be sampled and the tests are to be scheduled when hexavalent chromium is scheduled due to holding time limitations. It is the responsibility of the Project Manager to appropriately flag the data and make notes in the case narrative about the nature of the matrix, if applicable (see the attached flowchart).
- 4.2 It is the responsibility of the analyst to perform the analysis according to this SOP and to complete all documentation required for data review. Analysis and interpretation of the results are performed by personnel in the laboratory who have demonstrated the ability to generate acceptable results utilizing this SOP. Final review and sign-off of the data is performed by the department supervisor or designee.

5) Interferences

- 5.1 A reducing tendency of the sample matrix may change Cr (VI) to Cr(III). The reducing/oxidizing tendency of each sample may be characterized using additional analytical parameters, such as pH, ferrous iron, sulfides, and oxidation/reduction potential. Other indirect indicators of reducing/oxidizing tendency include Total Organic Carbon (TOC), Chemical Oxygen Demand (COD), and Biochemical Oxygen Demand (BOD). Analysis of these additional parameters establishes the tendency of Cr(VI) to exist in the unspiked sample(s) and may be necessary to assist in the interpretation of QC data for matrix spike recoveries outside conventionally accepted criteria.
- 5.2 Substances not typically found in the alkaline digests of soils may interfere in the analytical techniques for Cr(VI) following alkaline extraction if the concentration of these interferences are high relative to the Cr(VI) concentration. Refer to EPA methods 7196A and 7199 for a discussion of the specific agents that interfere with Cr(VI) quantification.
- 5.3 For materials suspected of containing high concentrations (greater than four times the laboratory Cr(VI) reporting limit) of soluble Cr(III), Cr(VI) results obtained using Method 3060A may be biased high due to method induced oxidation. The addition of Mg²⁺, and a phosphate buffer, to the alkaline extraction solution has been shown to suppress oxidation.



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- 5.4 One of the most insoluble forms of chromate in alkaline solution, barium chromate, may require additional heating time to affect complete dissolution in some soil matrices.
- 5.5 Reduction of Cr(VI) to Cr(III) can occur in an acidic medium. However, at a pH of 6.5 or greater, CrO_4^{2-} (which is less reactive than the HCrO_4^-) is the predominant species.

6) Safety

- All appropriate safety precautions for handling solvents, reagents and samples must be taken when performing this procedure. This includes the use of personnel protective equipment, such as, safety glasses, lab coat and the correct gloves.
- Chemicals, reagents and standards must be handled as described in the company safety policies, approved methods and in MSDSs where available. Refer to the Environmental, Health and Safety Manual and the appropriate MSDS prior to beginning this method.
- Sodium Hydroxide (NaOH) is a strong caustic and a severe health and contact hazard. Use nitrile or latex gloves while handling pellets or preparing solutions.
- Nitric and sulfuric acids are used in this method. These acids are extremely corrosive and care must be taken while handling them. A face shield should be used while pouring acids. And safety glasses should be worn while working with the solutions. Lab coat and gloves should always be worn while working with these solutions.
- Refer to the Safety Manual for further discussion of general safety procedures and information.

7) Sample Collection, Containers, Preservation, and Storage

- 7.1 Samples should be collected using devices that do not contain stainless steel. Sample containers provided by the lab are purchased, pre-cleaned, certified 8 oz. wide mouth clear glass jars with Teflon lined lids. The client must provide additional aliquots of sample if further investigation into the reducing/oxidizing nature of the sample is to be done.
- 7.2 Samples should be stored at 0-6°C until analysis.
- 7.3 If investigation is requested, the investigative tests (such as sulfide, pH, REDOX and Fe^{2+}) must be scheduled with the Cr6+ test because some of the investigative tests have shorter holding times than hexavalent chromium. These tests must be completed prior to the evaluation of the Cr6+ QC so that the results are available if needed.
- 7.4 Holding Times
 - 7.4.1 Samples may be held for 30 days from collection to digestion.
 - 7.4.2 7199: Once the samples have been digested, they may be held overnight in tightly capped B-cups in the cooler at 0-6 °C. Holding the samples overnight facilitates settling and subsequent filtering. Samples must be filtered, pH adjusted, and analyzed within 7 days of digestion.
 - 7.4.3 7196A - Once the extracts are pH adjusted to 7.5 ± 0.5 , samples are required to be analyzed within 1-hour. Samples must be filtered, pH adjusted, and analyzed within 7 days of digestion.



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- 7.5 For further sample handling, storage and custody procedures, see SMO-GEN and SMO-ICOC.

8) Apparatus and Equipment

- 8.1 Glassware: Beakers (150 mL), watch glass covers, graduated cylinders (50 mL), volumetric flasks (Class A, 1000 mL), rinsed with 50/50 Nitric Acid.
- 8.2 Filtration apparatus and filter membranes (0.45, μm), preferably cellulosic or polycarbonate membranes.
- 8.3 Hot block – Environmental Express SC154 (SN7021CECW3005) – with Stirbase and controller. Capable of maintaining the digesting samples at 90-95°C, and Teflon coated stir bars. The temperature of the sample is monitored as described in the procedure. This is the primary equipment used.
- 8.4 Stirring hot plates, capable of maintaining the digesting samples at 90-95°C, and Teflon coated stir bars. The hot plates currently in use are all single - place stirring. Plates are to be studied and the position marked on the temperature control dial where the dial should be adjusted to maintain 90-95 °C. The temperature of the sample is monitored as described in the procedure. The manufacturers of the hot plates are VWR, Thermolyne and Corning.
- 8.5 pH meter: Orion Model 720A and/or Orion Model SA520 with Thermo-Orion combination pH electrodes. Meters are to be calibrated and used according to ADM-pHSupport.
- 8.6 Balances: Top-Loading – American Scientific products Model DTL 2500g; Analytical – Mettler AE240. Calibrated according to ADM-DALYCK.
- 8.7 Thermometers – calibrated according to ADM-DALYCK.
- 8.8 B-cups - 130 mL plastic beakers.
- 8.9 Micropipettes – Eppendorf 100-1000 uL and 10-100 uL. Calibrated according to ADM-PCAL. IDs of pipettes used during analysis are documented on the digestion and analysis benchsheets so that pipettes can be traced back to pipette calibration logs.
- 8.10 Computer Hardware and Software – Any computer in the lab which is connected to the LAN and has access to the LIMS and the Excel benchsheet.



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9) Standards, Reagents, and Consumable Materials

- 9.1 All Standards must be traceable using the laboratory lot system (CE-QA007). All purchased standards are certified by the vendor. Certificates of Analysis are kept in the department until the standards are no longer being used – at which time they are filed with QA. Certificates of Analysis are available upon request. Purchased standards are routinely checked against an independent source for analyte concentration.
- 9.2 Purchased Reagents and Standards – Store at room temperature and expire per the Expiration Policy unless otherwise indicated.
 - 9.2.1 Nitric acid: HNO_3 , concentrated, analytical reagent grade or spectrograde quality. Store in the dark. Expires sooner than assigned date if the solution develops a yellow color.
 - 9.2.2 Sodium carbonate: Na_2CO_3 , anhydrous, analytical reagent grade.
 - 9.2.3 Sodium hydroxide pellets: NaOH , analytical reagent grade.
 - 9.2.4 Magnesium Chloride: MgCl_2 (anhydrous), analytical reagent grade. 392.18 mg MgCl_2 is equivalent to 100 mg Mg^{2+} .
 - 9.2.5 K_2HPO_4 : Analytical reagent grade.
 - 9.2.6 KH_2HPO_4 : analytical reagent grade.
 - 9.2.7 Sulfuric acid (H_2SO_4), concentrated, reagent grade.
 - 9.2.8 Cr (VI) standard solution (1000 mg/L): purchased commercially - certified primary standard solution.
 - 9.2.9 Cr (VI) reference solution (1000 mg/L): purchased commercially - certified primary standard solution from a separate source as the standard solution.
 - 9.2.10 Lead Chromate, (PbCrO_4), powder
- 9.3 Prepared Reagents – Store at 0-6 °C and expire per the Expiration Policy unless otherwise indicated.
 - 9.3.1 Phosphate Buffer - 0.5M K_2HPO_4 /0.5M KH_2PO_4 buffer at pH 7 (also called 1M Phosphate buffer): Dissolve 87.09 g K_2HPO_4 and 68.04 g KH_2PO_4 into 700 mL of DI. Transfer to a 1L volumetric flask and dilute to volume.
 - 9.3.2 Digestion solution: Dissolve 20.0 ± 0.05 g NaOH pellets and 30.0 ± 0.05 g Na_2CO_3 in DI in a one-liter volumetric flask and dilute to the mark. Store the solution in a tightly capped polyethylene bottle. The pH of the digestion solution must be checked before using. Expires in 1 month or if the pH is not 11.5 or greater.
 - 9.3.3 10% Sulfuric acid (v/v) (1.8M): Add 10 mL of concentrated H_2SO_4 to approximately 70 mL of DI. Mix well and let cool. Dilute to a final volume of 100 mL with DI.
 - 9.3.4 50/50 Nitric acid wash solution: Slowly and carefully combine 1 part concentrated Nitric Acid with 1 part DI.



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9.4 Prepared Standards – Prepare fresh daily.

9.4.1 Cr (VI) intermediate standard solution (100 mg/L): dilute 1.00 mL of 1000 mg/L standard solution to 10.0 mL with DI.

9.4.2 Cr (VI) intermediate reference solution (100 mg/L): dilute 1.00 mL of 1000 mg/L reference solution to 10.0 mL with DI.

9.4.3 Calibration Standards for 7196A:

<u>Conc. (mg/L)</u>	<u>Vol. 100 mg/L stock (mLs)</u>	<u>Volume of digest solution</u>	<u>Final Volume (mL)</u>
2.0	2.00	50	100
1.5	1.50	50	100
1.0	1.00	50	100
0.5	0.50	50	100
0.1	0.10	50	100
0.0	0.00	50	100

7196A standards are adjusted to a pH of 7.5 +/- 0.5 with concentrated nitric acid before they are brought to a final volume of 100 mLs with DI. Must be analyzed within 1 hour after pH adjustment.

9.4.4 Calibration Standards for 7199 – as described in the 7199 SOP. These standards do not undergo these procedures. They are made with buffered DI (pH 9.0-9.5).

9.4.5 ICV/CCV for 7196A (1.0 mg/L): add 1.0 mL 100 mg/L reference solution to 50 mL digest solution. Adjust pH to 7.0 - 8.0 with concentrated nitric acid. Bring to 100 mL with DI. Must be analyzed within 1 hour after pH adjustment.

9.4.6 ICV/CCV for 7199- as described in the 7199 SOP. This standard does not undergo these procedures. It is made with buffered DI (pH 9.0-9.5).

9.5 Soluble MS (1.0 mg/L or 40 mg/Kg): add 1.0 mL 100 mg/L standard solution to sample or blank (Ottawa sand) and 50 mL digest solution. Sample is digested, filtered, and completed as a sample including pH and volume adjustment.

9.6 Insoluble LCS – Add approximately 10 mg PbCrO₄ to 50 mLs digest solution. Digest and analyze as a sample. The true value will depend on the amount used. Lead chromate is 0.161% chrome. The 0.0025 is as if it was spiked into 2.5 g sample weight.

$$TrueValue(mg / kg) = \frac{mgPbCrO_4}{0.0025} \times 0.161\%$$

9.7 Insoluble MS - Add 10 mg PbCrO₄ to sample. Digest and analyze. Calculate true value as above.

9.8 PVS – described in the analytical SOPs since it is not done during digestion.

9.9 Method Blank – Analyze Ottawa Sand as a sample.



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10) Preventive Maintenance

- 10.1 Inspect thermometers before use. Be sure the liquid column is not separated.
- 10.2 Be sure glassware is scrupulously cleaned according to GEN-GC.
- 10.3 Inspect the pH probe for cracks or scratches. Replace the filling solution weekly.
- 10.4 Troubleshooting – see the manual of the pH meter for specific problems regarding the pH meter.

11) Procedure

- 11.1 Be sure the analyst has a current Demonstration of Capability and the system has current detection and quantitation studies.
- 11.2 Print a Responsibility Report from LIMS to determine what samples are available to be run. Prioritize samples to be run based on holding time and due dates. Scan the samples to be analyzed out of storage according to SMO-ICOC. Plan the digestion batch according to ADM-BATCH and the frequency requirements in section 12 of this SOP.
- 11.3 Heated Digestion
 - 11.3.1 Homogenize and subsample according to ADM-SPLPREP. Place 2.5 ± 0.10 g of the sample into a clean and labeled beaker (for digestion using hot plates) or a labeled digestion vessel (for digestion using stirring hot block). Record the actual weight. Document the ID of the balance used to weigh the sample.
 - 11.3.2 Spike LCSs and MSs (soluble and insoluble) at this time as described in Section 9 and at the frequency in Section 12. Document the spikes added on the benchsheet. Record the ID of the pipette used for spiking.
 - 11.3.3 Measure the pH of the digestion solution with a pH meter. It must be 11.5 or greater to be used. If it is not, discard the solution and make fresh. Record the pH of the solution and the ID of the pH meter on the benchsheet. Add 50 mL of digestion solution to each sample, blank, MS and LCS.
 - 11.3.4 Add approximately 400 mg magnesium chloride and 0.5 mL of phosphate buffer to each sample.
 - 11.3.5 Add the appropriate stir bar to each sample.
 - 11.3.6 Mix each sample thoroughly before placing the beaker on the hotplate.
 - 11.3.6.1 For digestion using hot plates - Turn on the heat to the previously calibrated setting. Place the beakers on the hotplates. Turn on the stirrer. Maintain a temperature range of 90 - 95°C for 60-65 minutes with constant stirring. Record the start time of the digestion. Check the digestion solution temperature of each sample at 0, 30 and 60 minutes using a calibrated alcohol thermometer placed directly in the digestion beaker so that the thermometer is in the beaker spout and the watch glass is placed over the entire beaker. Record the temperature on the benchsheet. Record the end time of the digestion.



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- 11.3.6.2 For digestion using stirring hot block - Turn on the heat to the previously calibrated setting. For this Environmental Express brand hot block it has been determined that a set point temperature of 112°C provides the proper temperature for samples to be between 90 - 95°C. Place the digestion vessels in the hot block. Turn on the stirrer. Maintain a temperature range of 90 - 95°C for 60-65 minutes with constant stirring. Record the start time of the digestion. Check the digestion solution temperature of each sample at 0, 30 and 60 minutes using a calibrated alcohol thermometer placed directly in the digestion vessel. Plastic thermometer holders (purchased from Environmental Express) can be used to suspend the thermometer in the solution. Record the temperature on the benchsheet. Record the end time of the digestion.
- 11.3.7 Gradually cool each solution to room temperature. Transfer to b-cups (if digested in beakers), cap, and place in the cooler overnight to allow settling and aid filtration.
- 11.4 Filtration
- 11.4.1 Using a 0.45 µm membrane filter, follow the assembly and use procedure for the vacuum filter funnels in GEN-FILTER. Document the filter brand, type, and pore size on the benchsheet. Transfer the digested sample quantitatively to the vacuum filtration apparatus with DI rinses. Rinse the inside of the filter flask and filter pad with DI. Transfer the filtrate and the rinses to a clean, labeled 130-mL B-cup. Record the color of the filtrate on the benchsheet. Cap the cup.
- 11.4.2 If analysis is not to immediately follow, place the filtered samples in the cooler (0-6°C) and store for up to 7 days (from day of digestion) unless the client has other requirements. On the day of analysis remove the samples from the cooler and allow to warm to room temperature. Continue.
- 11.5 Standards Preparation
- 11.5.1 For 7196A - Prepare the standards and references in B-cups. Record the pipettes used. Standards must be prepared and analyzed with each daily run. They do not need to be digested, but they do need the next 2 steps (pH adjustment and dilution).
- 11.5.2 For 7199 - standards will be prepared by the 7199 analyst with buffered DI and will not need to be pH adjusted and diluted.



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- 11.6 PH adjustment – record the ID of the pH meter on the benchsheet
- 11.6.1 For analysis by 7199 - With constant stirring, slowly add concentrated sulfuric acid solution to the beaker dropwise. Adjust the pH of the solution to 9.0-9.5 and monitor the pH with a calibrated pH meter. The buffering capacity of the digestion solution is high, so the first 50 drops of acid will bring the pH down to only about 9.5 to 10.0. Record the pH and the date of adjustment on the bench sheet.
- 11.6.2 For analysis by 7196A - With constant stirring, slowly add concentrated nitric acid solution to the beaker dropwise. Adjust the pH of the solution to 7.5 ± 0.5 and monitor the pH with a calibrated pH meter. The buffering capacity of the digestion solution is high, so the first 50 drops of acid will bring the pH down to only about 9.5 to 10.0. Go very slowly once the pH reaches about 8.5; at that point it will require only a few drops to go to a pH of 7.5 ± 0.5 - the analyst may wish to switch to 1:1 nitric acid at this point to complete adjustment. If the pH of the digest should drop below 7.0, discard the solution and redigest, and adjust the pH with a dilute nitric acid solution. Record the pH and the date of adjustment on the bench sheet.
- Note: The pH adjustment and subsequent steps is time consuming. A single analyst should only adjust about 10 digested samples and complete the colorimetry before adjusting more digested samples. This ensures the samples are analyzed within 1 hour of adjustment. Additional analysts speeds the process and production can be adjusted accordingly.
- CAUTION: CO₂ will be evolved. This step should be performed in a fume hood.
- 11.7 Volume Adjustment – Transfer the sample to a graduated cylinder. Adjust the sample volume to 100 mL with DI. Transfer back to the same b-cup. Cap and mix well. This applies to 7196A and 7199 samples and digested QC samples.
- 11.8 Quantitative Analysis - The sample digestates are now ready to be analyzed. Determine the Cr(VI) concentration in mg/kg by Method 7196A (GEN-7196A) or Method 7199 (GEN-7199). For 7196A, be sure to analyze samples within 1 hour of the time that they are adjusted to a pH of 7.5 ± 0.5 .

12) Quality Assurance/Quality Control Requirements

- 12.1 Frequency – One of each of the following are required to be digested with each preparatory batch: MB, LCS (insoluble), DUP, MS (soluble), MS (insoluble).
- 12.2 Acceptance and corrective action are given in the analytical SOPs.
- 12.3 As per EPA Method 3060A, if there are QC failures, the entire batch may need re-extracted and reanalyzed. See the analytical SOPs for further details.



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13) Data Reduction and Reporting

- 13.1 A Copy of the digestion bench sheet is attached.
- 13.2 Calculations are given in the analytical SOPs.
- 13.3 Data must be reviewed by the analyst and a peer (supervisor or qualified analyst) using a Data Quality Checklist before the results are validated and reported to the client. Further data review policies and procedures are discussed in ADM-DREV.

14) Method Performance

Single Laboratory Method Evaluation Data from the EPA Method is available in Table 1.

Reporting limits are based upon an MDL study performed according to ADM-MDL and filed in the MDL binders in the QA office.

Demonstration of Capability is performed upon instrument set-up, whenever a new analyst begins independent analysis, and annually thereafter according to Section 18 below. The documentation of this method performance is retained by the Quality Assurance office.

15) Waste Management and Pollution Prevention

- Hexavalent chromium solutions should be dumped in the red inorganic carboys which will later be emptied by qualified personnel. All other waste can be flushed down the drain with large amounts of water. See SMO-SPLDIS for further information.
- It is the laboratory's practice to minimize the amount of acids and reagent used to perform this method wherever feasible. Standards are prepared in volumes consistent with methodology and only the amount needed for routine laboratory use is kept on site. The threat to the environment from solvent and reagents used in this method can be minimized when disposed of properly.
- The laboratory will comply with all Federal, State and local regulations governing waste management, particularly the hazardous waste identification rules and land disposal restrictions as specified in the EH&S Manual.

16) Corrective Action for Out-of-Control Data

Failure to meet established analytical controls, such as the quality control objectives, prompts corrective action. A QC Flowchart provided in the analytical SOPs describes the corrective actions for out of control QC.

17) Contingencies for Handling Out-of-Control or Unacceptable Data

If a quality control measure is found to be out of control, and the data is to be reported, all samples associated with the failed quality control measure shall be reported with the appropriate data qualifier(s).



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18) Training

- 18.1 Read current SOP and applicable methodologies. Demonstrate a general understanding of the methodology and chemistry. Follow Training Policies in the SOP for Training (CE-QA003).
- 18.2 Observe Sample Preparation and Analysis. Follow Training Plan Form.
- 18.3 Participate in the methodology, documentation, and data reduction with guidance.
- 18.4 Perform the analysis independently and show Initial Demonstration of Capability (IDC) by analyzing 4 replicates of a known mid-range standard in succession before client samples are analyzed. If recovery is within acceptable limits, complete IDC certificate and Training Plan Form and file with QA. Continuing Demonstration of Capability (CDC) will be demonstrated annually using a PE sample, single blind, or a new 4 replicate study. Demonstrate Competency by performing the analysis independently.

19) Method Modifications

The method says to mix the sample 5 minutes on the hotplate before turning on the heat. Instead, the lab mixes the sample thoroughly with the reagents before placing the sample beaker on the pre-heated hotplate. This modification reduces the amount of time spent with samples on the hotplates and increases the number of samples which may be analyzed in a day without affecting the quality of the analysis.

20) Summary of Changes

- Updated to ALS format – removed CAS throughout.
- Added information for stirring Hotblock.

21) References

- Test Methods for Evaluating Solid Waste, Physical/Chemical Methods, USEPA SW-846, 3060A, December 1996.

22) Attachments

- Digestion Benchsheets
- Table 1 – Single Laboratory Method Evaluation Data
- 3060 Flowchart



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Hexavalent Chromium Soils

Columbia Analytical Services
-Now Part of the ALS Group-
Rochester, NY 14623

Method: EPA 3060A FOR ANALYSIS BY 7196A - INITIAL DIGESTION (Page 1)

Analyst: _____ Date: _____ Pipet ID: _____

A) Digest Time Start: _____ Stop: _____ B) Digest Time Start: _____ Stop: _____

Filters: Brand: _____ Type: _____ Pore Size: _____ um. Digest Solution: pH: _____

pH Meter ID: _____ Balance ID: _____

#	Method	Order #	Sample Amt. (g.)	Digest Sol. (mLs.)	Final Vol. (mLs.)	pH Adjust	pH Adjust Date	Filtered Digestate Color/Comments	Digest Temp. Check (deg. C)			Thermometer ID
									@ 0 min.	@ 30 min.	@ 60 min.	
1												
2												
3												
4												
5												
6												
7												
8												
9												
10												
11												
12												
13												
14												
15												
16												
17												
18												
19												

Cr⁶⁺ Stocks: Standard: _____ Reference: _____
 Insoluble LCS: added _____ mgs of PbCrO₄ per _____ kg sample = _____ mg/kg x 0.161 (%Cr) = _____ mg/Kg Cr6+
 Spike Witness: _____ PbCrO₄ Log = _____

J:\ACQUATAWetChem\Cr6-soils\TEMPLATE-cr6Digest-r0

Template-Cr6Digest-r0 1/3/13



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Hexavalent Chromium Soils

Columbia Analytical Services
-Now Part of the ALS Group-
1565 Jefferson Road.
Rochester, NY 14623

Method: EPA 3060A

Analyst: _____ Date: _____ Pipet ID: _____

THESE STANDARDS ARE NOT DIGESTED

Filters: Brand: _____ Type: _____ Pore Size: _____ um.

Digest Solution: pH: _____

pH Meter ID: _____

#		Digest Sol. (mLs.)	Final Vol. (mLs.)	pH Adjust	pH Adjust Date
1	2.00	50	100		
2	1.50	50	100		
3	1.00	50	100		
4	0.50	50	100		
5	0.10	50	100		
6	0.00	50	100		
7	I/CCV	50	100		
8	I/CCB	50	100		
9					
10					
11					
12					
13					
14					
15					
16					
17					
18					
19					
20					

Cr⁶⁺ Stocks: _____ Standard: _____
Reference: _____

J:\ACQDATA\WetChem\Cr6-soils\TEMPLATE-cr6Digest-r0

Template-Cr6Digest-r0 1/3/13



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TABLE 1
SINGLE LABORATORY METHOD EVALUATION DATA

Sample Type	Eh (mV) _b	pH _d	S ²⁻ (ppm) ^c	Mean Native Cr(VI) Conc. (mg/kg)	Mean Cr(VI) Spike Conc. (mg/kg)	Matrix Spike Recovery Range, %
COPR ^a /Soil Blends	550	7.4	<10.0	4.1	42.0	89.8-116
Loam	620	6.4	<10.0	ND	62.5	65.0-70.3
Clay	840	3.0	<10.0	ND	63.1	37.8-71.1
COPR ^a	460	7.4	<10.0	759	813	85.5-94.8
Anoxic Sediment	-189	7.2	25.0	ND	381	0
Quartz Sand	710	5.3	<10.0	ND	9.8	75.5-86.3

Source: Reference 10.3

Notes:

- ND - Not detected
- a - COPR - chromite ore processing residue
- b - Corrected for the reference electrode, laboratory field moist measurement
- c - Field measurement
- d - Laboratory field moist measurement

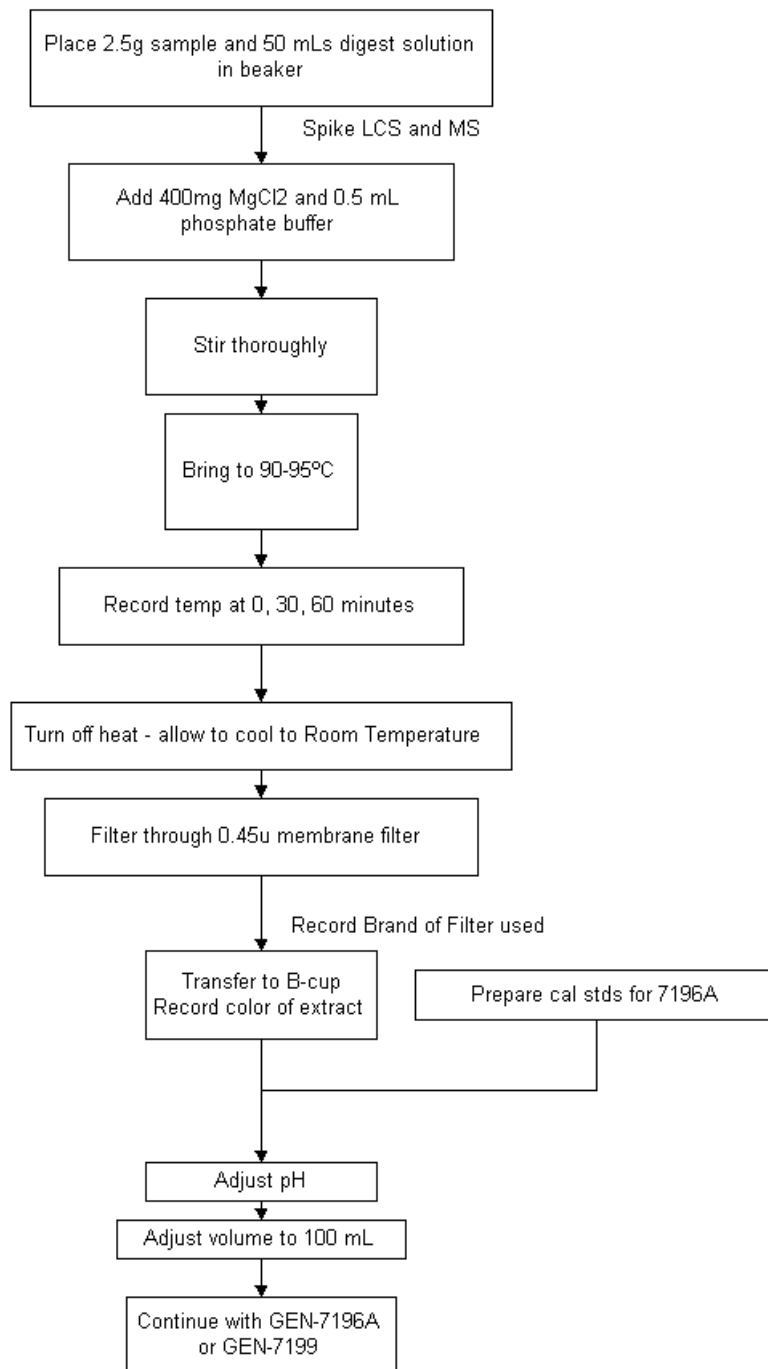


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METHOD 3060 (EPA)

HEXAVALENT CHROMIUM DIGESTION



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Environmental 

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30-Sep-2013

VERIFY THE VALIDITY OF THIS SOP EACH DAY IN USE

STANDARD OPERATING PROCEDURE
FOR
SOIL SAMPLE PREPARATION
FOR THE DETERMINATION OF RADIONUCLIDES

(GL-RAD-A-021 REVISION 20)

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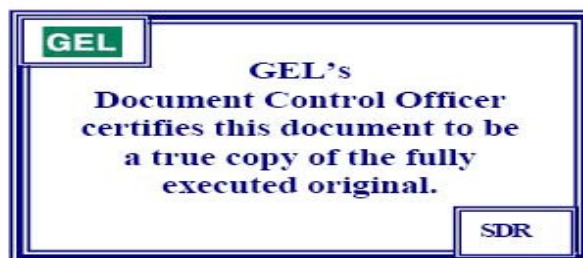


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1.0 STANDARD OPERATING PROCEDURE FOR THE SOIL SAMPLE PREPARATION FOR THE DETERMINATION OF RADIONUCLIDES**2.0 METHOD OBJECTIVE AND APPLICABILITY**

This standard operating procedure provides the necessary instructions to conduct the preparation of soil samples for radionuclide determination.

3.0 SUMMARY

This procedure involves drying the soil at a temperature between 103 and 105 °C. If that temperature would volatilize any components for which an analysis has yet to be run, a separate aliquot must be set aside for such analyses.

4.0 SAFETY PRECAUTIONS AND HAZARD WARNINGS

- 4.1 Personnel performing this analytical procedure are trained in and follow the safe laboratory practices outlined in the Safety, Health, and Chemical Hygiene Plan, GL-LB-N-001.
- 4.2 Personnel handling radioactive materials are trained in and follow the procedures outlined in GL-RAD-S-004 for Radioactive Material Handling.
- 4.3 Personnel handling biological materials are trained in and follow the procedures outlined in GL-RAD-S-010 for The Handling of Biological Materials.
- 4.4 If there is any question regarding the safety of any laboratory practice, **stop immediately**, and consult qualified senior personnel such as a Group or Team Leader.

5.0 APPARATUS AND MATERIALS**5.1 Apparatus and Equipment**

- 5.1.1 Metal cans, approximately quart and pint size
- 5.1.2 Steel balls, approximately 1" and 3/4" diameter
- 5.1.3 Sieve screens, 28 mesh
- 5.1.4 Paper funnels
- 5.1.5 100 cc aluminum cans
- 5.1.6 Aluminum loaf pans
- 5.1.7 SPEX steel grinding containers (various sizes)
- 5.1.8 Assorted tools and labware

5.2 Reagents, Chemicals, and Standards

- 5.2.1 Sand, clean
- 5.2.2 Deionized water (DI water)

5.3 Instrumentation

- 5.3.1 Paint can shaker, heavy duty
- 5.3.2 Analytical balance
- 5.3.3 SPEX Model 8515-115 Shatterbox
- 5.3.4 Drying oven

5.3.5 Hydraulic press

5.3.6 Retsch, Model BB-51, Jaw Crusher

5.3.7 Automatic or manual can sealer

6.0 SAMPLE COLLECTION AND PRESERVATION

A representative sample must be collected from a source of soil and should be large enough (50 to 100 g) so that adequate aliquots can be taken to obtain the required sensitivity. The container of choice should be plastic over glass to prevent loss due to breakage during handling. No preservation is required for solid samples.

7.0 EQUIPMENT AND INSTRUMENT MAINTENANCE

7.1 Refer to the technical manual provided with the paint shaker for information regarding equipment maintenance.

7.2 Refer to Instruction Manual for SPEX Shatterbox and Retsch Jaw Crusher operating instructions.

7.3 The analytical balance should be cleaned after use.

7.4 Refer to operating instructions for automatic or manual can sealer for information regarding equipment maintenance.

7.4.1 After receipt of new can sealer or when maintenance is performed on canner that could affect the proper sealing of the 100 cc gamma cans, perform a leak test on a can as follows:

7.4.1.1 Fill gamma can with water and seal.

7.4.1.2 Apply pressure on top and bottom of can and inspect for any visible signs of leakage.

8.0 SAMPLE PREPARATION PROCEDURES

8.1 Label a clean metal container with the laboratory sample number.

8.2 Weigh container. Record weight into the computerized soil prep balance log.

8.3 Transfer a representative aliquot from the sample to the labeled container.

8.3.1 When the amount of sample to be dried is not the entire contents of the sample container, refer to section 6.6 of SOP GL-LB-E-029 to ensure a representative sample aliquot is taken. This also ensures that the sample remaining is representative of the whole sample.

8.3.2 If the sample contains extraneous materials (i.e. rocks, twigs, vegetation) this shall be documented in the batch case narrative.

NOTE: Sample is dried at 103 to 105 °C. If this temperature will volatilize any component for an analysis that has yet to be run, a separate aliquot must be set aside for such analysis.

8.4 Enter the pre-oven sample weight into the computerized soil prep balance log. This weight represents the wet sample weight and container weight.

8.5 Place the container in a drying oven at a temperature between 103 and 105 °C for a minimum of four hours. (Normally overnight.)

NOTE: The time required to obtain a dry sample will vary depending on the type of material, size of sample, oven type and capacity, and other factors. The influence of these factors generally can be established by good judgment, experience with the materials being tested, and the apparatus being used.

- 8.6 Using heat resistant gloves, remove the sample from the oven and allow to cool.
- 8.7 Record weight into the computerized soil prep balance log. Replace sample in drying oven for a minimum of one hour.
- 8.8 Repeat step 8.7 until a constant weight is obtained. A constant weight is defined as a weight difference of less than about 0.1% (i.e. for a 20 g aliquot, the change in weight should be less than 0.02 g).
- 8.9 Homogenize the sample. This is normally accomplished by placing a lid on the container and placing the container in the industrial paint shaker. Depending on the matrix of the soil, it may be necessary to add several stainless steel balls inside the container to assist in homogenizing the sample. The length of time the sample remains on the shaker is dependent on the matrix of the sample, and normally ranges from 5 to 10 minutes. For solid samples that are composed of large particles, it may be necessary to reduce the particle size before homogenizing. This can be accomplished by placing the larger particles in the hydraulic press and applying enough pressure to break the particles into smaller pieces.
- 8.10 Remove the metal container from the shaker and allow to settle for several minutes.
- 8.11 Place the container in the sample preparation hood and remove the lid. If stainless steel balls were added to the container, they should now be removed. The stainless steel balls are discarded.
- 8.12 Determine an appropriate aliquot based on the analysis required. Normally, depending upon the required analysis, the sample will be passed through a 28 mesh sieve screen. For clients that require a smaller particle size, continue homogenizing samples per section 8.15.
- 8.13 Discard the unused portion of sample into the appropriate waste container (i.e. rocks or organic material). The soil sample is now ready for radiochemical analysis.
- 8.14 Place sample in appropriate containers (i.e. plastic bottle or vial, gamma can).

NOTE: If preparing a 100 cc gamma can, it is important that the can is properly sealed to ensure Ra-226 is quantified correctly. Ra-226 in soil samples is quantified by one of its daughter products (Bi-214). Ra-226 decays to Bi-214 through Rn-222, which is a gas, and must be isolated inside the can in order for equilibrium to be re-established.

8.14.1 For proper sealing of gamma can:

8.14.1.1 Place lid on 100 cc can and place can on base plate of the automatic or manual can sealer.

- 8.14.1.2 Raise the can until it is clamped firmly between the base plate and chuck by turning the can lifter handle as far as possible to the right until the handle locks itself against the frame.
 - 8.14.1.2.1 If using the automatic can sealer push the on button. The flywheel will turn until the can sealing is complete.
 - 8.14.1.2.2 If using the manual can sealer, turn the flywheel 21 turns until the second operational roll returns to its normal position, away from the chuck.
- 8.14.1.3 Lower the can by turning the can lifter handle to the left. Remove the sealed can from the base plate.
- 8.14.1.4 Visually inspect can after sealing for any defects. If defects are noticed that may affect the proper sealing of the can, remove contents from can and start over with a new can.

8.15 Shatter Box (200 mesh sample)

- 8.15.1 Take a portion of remaining sample (more than enough to complete the requested analysis) and further homogenize sample to approximately 200 mesh. This is accomplished by pulverizing sample in the shatterbox.

NOTE: The approximately 200 mesh is determined based on particle size study in shatterbox instruction manual.

- 8.15.2 Pulverize sample for a minimum of 5 minutes.

NOTE: Refer to Shatterbox Instruction Manual for operating procedure.

- 8.15.3 To prevent cross-contamination the dish and puck must be decontaminated prior to pulverizing the next sample.
 - 8.15.3.1 Decontaminate the dish and puck by filling with approximately 50 g of sand. Replace the lid.
 - 8.15.3.2 Place the shatterbox in operation for approximately 1 to 2 minutes. Empty the sand from the container.
 - 8.15.3.3 Dampen a clean paper towel with DI water and wipe the dish, puck and lid to ensure that all traces of sand are removed.
 - 8.15.3.4 To check for contamination, perform a smear survey of dish and puck after each dish and puck decontamination with sand, and submit to Radiation Safety for counting.
 - 8.15.3.5 A decontamination blank is analyzed monthly for gross alpha, beta, and gamma activity. The results of these analyses will be compared to historical data. When a decontamination blank is analyzed, it is pulverized post cleaning.

NOTE: To prevent corrosion, keep the entire container assembly dry.

8.15.4 Record all samples and decontamination blanks in the shatterbox logbook.

8.16 Pulverizer

8.16.1 Use the sample jaw crusher to prepare medium to extremely hard substances having a maximum input grain size of 35 mm.

NOTE: Refer to Retsch Jaw Crusher Type BB-51 Operating Instructions manual for operating procedure.

8.16.2 To prevent cross-contamination, the jaw crusher must be decontaminated prior to processing the next sample. Decontaminate the jaw crusher by processing approximately 200 g of pre-dried lava rock having a maximum input grain size of 35 mm. A decontamination blank is analyzed monthly for gross alpha, beta, and gamma activity. The results of these analyses will be compared to historical data.

9.0 CALCULATIONS AND DATA REDUCTION METHODS

The electronic balance program provides documentation of all necessary raw data.

Weights in AlphaLIMS are recorded to three places past the decimal point (Ex: 6.738, 314.197...).

9.1 If client requests results to be reported “as received” (based on wet weight), analysts will correct the “dry” aliquots back to wet weights using the “weight/loss aliquot correction report” link in AlphaLIMS. This report will be included, if necessary, in each analytical batch’s raw data.

10.0 QUALITY CONTROL REQUIREMENTS

10.1 Method Specific Quality Control Requirements

10.1.1 When possible (i.e., there is sufficient sample available) for gamma analysis, a separate container will be prepared for counting to meet the duplicate sample requirement.

10.1.2 Refer to the specific isotope operating procedure for instructions concerning method quality control requirements.

10.2 Actions Required if the Quality Control Requirements Are Not Met

If any of the quality criteria cannot be satisfied, the analyst should inform the group leader and initiate a Data Exception Report as outlined in GL-QS-E-004 for Documentation of Nonconformance Reporting and Dispositioning and Control of Nonconforming Items.

11.0 CALIBRATION

11.1 Balances are calibrated annually and verified daily in accordance with GL-LB-E-002.

11.2 Temperature monitoring devices are verified in accordance with GL-QS-E-007.

12.0 RECORDS MANAGEMENT AND DOCUMENT CONTROL

All data are maintained as quality records in accordance with GL-QS-E-008 for Quality Records Management and Disposition.

13.0 LABORATORY WASTE HANDLING AND DISPOSAL

Laboratory waste is handled and disposed in accordance with the Laboratory Waste Management Plan, GL-LB-G-001.

14.0 REFERENCES

- 14.1 ASTM C999-05, "Standard Practice for Soil Sample Preparation for Determination of Radionuclides," 1993 Annual Book of ASTM Standards, Vol 12.01, 2005.
- 14.2 Laboratory Sub-Sampling Procedure, GL-LB-E-029.
- 14.3 ASTM D6323-98 (2003) "Standard Guide for Laboratory Subsampling of Media Related to Waste Management Activities."
- 14.4 EPA's "Guidance for Obtaining Representative Laboratory Analytical Subsamples from Particulate Laboratory Samples", EPA/600/r-03/027, November 2003.

15.0 HISTORY

Revision 20: Section 9.1 revised for clarification to comply with DOECAP audit finding.

Revision 19: Added client specific prep procedure as an appendix.

Revision 18: Changed 8.14-8.17 to 8.15-18.17 in section 8.12.

APPENDIX I: HGEO GROSS SAMPLE PRE-TREATMENT

This proposed procedure is for the preparation of SSFL soil samples for Hydrogeologic, Inc.

- 1.) Cover an area of bench top, within a HEPA filtered enclosure, with clean paper. Transfer the total raw sample to the paper and spread sample evenly across the surface. Samples may be contained in more than one sample container.
- 2.) Remove cultural/man-made materials from the sample if applicable, photograph, and place in a labeled container for storage. If no cultural/man-made materials are found, no photograph is required, provided that the laboratory documentation clearly notes that no such objects were found. Notify Project Manager if cultural/man-made materials are found in the samples.
- 3.) Label a clean metal container with the laboratory sample number. When large amounts of soil are being processed, samples may be dried in aluminum pans to improve effectiveness of complete drying. Note: Samples may be split into more than one drying vessel. Record weights separately until the sample is recombined.
- 4.) Weigh the containers and record weights into the soil prep balance log.
- 5.) Transfer the entire sample to the labeled container. When the amount of sample to be dried is not the entire contents of the sample container, document via a Data Exception Report. Any sample removed in this manner shall be done by taking a grab sample to minimize any loss of potentially volatile radionuclides.
- 6.) Enter the pre-oven sample weight into the soil prep balance log. This weight represents the wet sample weight and container weight.
- 7.) Place the container(s) in a drying oven at a temperature between 103 and 105 °C for a minimum of four hours. Record the time the sample was placed in the oven.
- 8.) Using protective heat resistant gloves remove the sample from the oven and allow cooling. Record the time the sample was removed from the oven.
- 9.) Record the weight in the soil prep balance log and place the sample back in the drying oven for a minimum of one hour. Record the time the sample was returned to the oven. Record the time the sample was removed from the oven for each interval.
- 10.) Repeat steps 8 and 9 until a constant weight is obtained. A constant weight is achieved when two subsequent weights agree within 1%.

- 11.) Break up the sample in preparation for passing through the sieves with the process. Transfer sample to an appropriate number of 1 gallon paint cans. Add 6- 1" stainless steel balls to each paint can to aid in the breakup of the samples. NOTE: Paint cans should not be filled above approximately 3/4 full. Overfilling will reduce the effectiveness of sample blending. Place the paint cans on the industrial paint can shaker for 5-10 minutes.
- 12.) Remove the container(s) from the paint can shaker and allow settling for several minutes.
- 13.) Place the container(s) in the sample in a ventilated preparation hood and remove the lid. If stainless steel balls were added to the container(s), they should now be removed and discarded.
- 14.) Pass entire dried sample through a 4-mesh sieve. Transfer material that will not pass through into a labeled container for storage. The analyst will evaluate the material to determine if they may be further broken up in order to be passed through the 4-mesh sieve.



- 15.) Pass the entire dried sample through a 28-mesh sieve. Any sample not passing through the 28-mesh sieve must be processed until it passes through the 28-mesh sieve. Contact the Group leader if these criteria cannot be met.
- 16.) Place sample back into the paint cans and mix on the paint can shaker for 5-10 minutes.

- 17.) From the <28-mesh sample, use the '*Cone and Quartering*' method described below to remove a representative sample fraction for gamma analysis (~2000 grams).
- 18.) ***Cone and Quartering***' method - Sample is emptied out onto a non-contaminating smooth surface. Material is piled into a cone with a flattened top surface. Two top-to-bottom cuts are made through the cone at perpendicular angles to form four equal portions, or quarters. Two opposite quarters are compiled into a new cone, and the process is repeated until the proper sample mass is obtained. When sub-sampling using this method it is important to remove the entire quarters to be used. Process enough sample to prepare a new 1-Liter Marinelli beaker (~1600-1700 grams) and 100cc gamma can (~160-180 grams) geometries for gamma analysis.



- 19.) Record the weight of the gamma aliquots in the soil prep balance log. Some samples may need an equal portion of sample for archiving as noted by the project manager.
- 20.) Take a portion of remaining sample using the 'Cone and Quartering' method (more than enough to complete the requested analysis) and further homogenize sample to approximately 200 mesh. This is accomplished by pulverizing sample in the shatterbox (puck mill).ⁱ

Note: Some samples may be designated to prepare an equal portion of sample for archiving.

ⁱ The approximately 200 mesh is determined based on particle size study in shatterbox instruction manual. The number of replicates included in the study was determined based on guidance provided in chapter 6 of the MARLAP manual (Level B validation).

- 21.) Pulverize sample for 5 minutes. Refer to Shatterbox Instruction Manual for operating procedure.
- 22.) To prevent cross-contamination the dish and puck must be decontaminated prior to running the next sample. Decontaminate the dish and puck by filling with approximately 50 g of sand. Replace the lid and operate the shatterbox for 2 minutes. Empty the sand from the container.
- 23.) Dampen a clean paper towel with DI water and wipe the dish, puck and lid to ensure that all traces of the sand are removed. To check for gross contamination, perform a smear survey of dish and puck and submit to radiation safety office for counting.
- 24.) To monitor for low level contamination, process blanks will be analyzed. The process blanks will consist of an ICP/MS analysis for U-238 on DI water rinses of the grinding containers (500 mLs). Acceptable results of the blanks shall be less than MDL. Should a blank indicate the presence of U-238 at a level >MDL, it shall be documented and a review conducted to determine the impact on any data produced using the container in question.

Record all samples and decontamination blanks in the shatterbox logbook.

30-Sep-2013

SOP Effective Date: 2/4/92
Revision 25 Effective February 2013

The Determination of Gamma Isotopes

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STANDARD OPERATING PROCEDURE

FOR

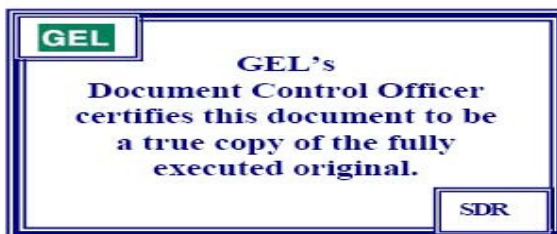
THE DETERMINATION OF GAMMA ISOTOPES

(GL-RAD-A-013 REVISION 25)

APPLICABLE TO METHODS:
EPA 600/4-80-032 Method 901.1
DOE EML HASL-300 Section 4.5.2.3
DOE EML HASL-300 Ga-01-R

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30-Sep-2013

The Determination of Gamma Isotopes

SOP Effective Date: 2/4/92
Revision 25 Effective February 2013

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GEL LABORATORIES, LLC

2040 Savage Road Charleston, SC 29407

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1.0 STANDARD OPERATING PROCEDURE FOR THE DETERMINATION OF GAMMA ISOTOPES**2.0 METHOD OBJECTIVE, PURPOSE, AND SUMMARY**

- 2.1 This standard operating procedure (SOP) provides the necessary instructions to conduct the analysis for gamma isotopes in water, soil, urine, filters, drinking water and miscellaneous matrices.
- 2.2 Water samples are typically counted in Marinelli beakers. Soil samples are typically sealed in aluminum cans, which can be counted immediately if Ra-226 is not desired. If Ra-226 is desired, the sealed can is set aside for minimum of 20 days to allow equilibrium between Rn-222 and Bi-214 to become re-established. Ra-226 is then quantified using the 609 keV line of Bi-214.
- 2.3 This method is based on the source method EPA 600/4-80-032 "Prescribed Procedures for Measurement of Radioactivity in Drinking Water," August 1980, Method 901.1, and the Department of Energy (DOE) EML Procedures Manual source method for Gamma PHA in environmental samples, HASL-300 Section 4.5.2.3 and Ga-01-R, Gamma Radioassay.
- 2.4 This SOP is applicable for analyzing samples that contain radionuclides emitting gamma photons with energies ranging from about 5 to 2000 keV (including I-131).

3.0 METHOD SCOPE, APPLICABILITY, AND DETECTION LIMIT

- 3.1 Minimum Detectable Activity (MDA): The MDA is based upon sample volume, Compton background, instrument efficiency, count time, and other statistical factors, as well as specific isotopic values such as abundance and half-life. A typical detection limit is 10 pCi/L or 0.1 pCi/g (based on Cs-137). The MDA for drinking water samples is 10 pCi/L (based on Cs-137).
- 3.2 Method Precision: Typical Relative Percent Difference (RPD) is 20% or less or 100% or less if the activity is less than five times the MDA.
- 3.3 Method Bias (Accuracy): The method accuracy requirement for gamma, measured by running a Laboratory Control Sample (LCS) with each batch, is 25% of the true value. For drinking water samples, laboratory fortified blanks (LFB, equivalent to LCS) recoveries should be between 90-110% of the known value.
- 3.4 Procedures contained in this SOP may be used to analyze REMP samples.
- 3.5 Analysts training records are maintained as quality records as outlined in GL-QS-E-008. Analysts training and proficiency in the method is outlined in the Employee Training SOP GL-HR-E-002.
- 3.6 For drinking water samples, analyst initial and ongoing demonstrations of proficiency will follow critical elements for radiochemistry, chapter VI, section 1.5, of The Manual for the Certification of Laboratories Analyzing Drinking Water (reference 20.5).
- 3.7 Sensitivity studies will follow critical elements for radiochemistry, chapter VI, section 7.3 of The Manual for the Certification of Laboratories Analyzing Drinking Water (reference 20.5).

4.0 METHOD VARIATIONS

- 4.1 Some variations may be necessary due to special matrices encountered in the lab. These variations may be used with approval from a Group Leader or Team Leader. Variations to a method will be documented with the analytical raw data.
- 4.2 Filter samples can either be counted directly, or digested prior to counting. If filters are digested, they are digested in accordance with GL-RAD-A-026.
- 4.3 No method modifications are permitted for drinking water samples.

5.0 DEFINITIONS

- 5.1 National Institute of Standards and Technology (NIST): For the purpose of this method, the national scientific body responsible for the standardization and acceptability of analyte solutions.
- 5.2 Deionized (DI) water: Type I water. Refer to GL-LB-E-016.
- 5.3 AlphaLIMS: GEL's Laboratory Information Management System.
- 5.4 Batch: Environmental samples that are prepared and/or analyzed together with the same process and personnel, using the same lot(s) of reagents.
- 5.5 Method Blank (MB): A sample of a matrix similar to the batch of associated samples (when available) that is free from the analytes of interest and is processed simultaneously with and under the same conditions as samples containing an analyte of interest through all steps of the analytical procedures.
- 5.6 Laboratory Duplicate (DUP): For soils, when sufficient sample is available, a separate duplicate will be prepared. For liquid samples and when sufficient sample is not available for solids, an independent count of the sample container will be performed to show precision.
- 5.7 Laboratory Control Sample (LCS): A sample matrix, similar to the batch of associated samples (when available) that is free from the analytes of interest, spiked with verified known amounts of analytes from a source independent of the calibration standards or a material containing known and verified amounts of analytes. The LCS is equivalent to a Fortified Blank in the EPA drinking water compliance manual (See to section 20.5).

6.0 INTERFERENCES

- 6.1 Some gamma isotopes emit gamma lines that may overlap with other isotopes. If the energies of the two isotopes are within the energy tolerance setting, the peaks may not be resolvable and may give a positive bias to the result. This problem is minimized by careful review of the peak search.
- 6.2 Soil samples may vary in density from the standard used for calibration. A density correction is applied to the "CAN" geometry. This correction was determined using solids with weights varying between 54 g and 192 g.

7.0 SAFETY PRECAUTIONS AND WARNINGS

- 7.1 Keep hands free from moving parts of canning device and gamma shields.

- 7.2 Personnel performing this analytical procedure are trained in and follow the safe laboratory practices outlined in the Safety, Health and Chemical Hygiene Plan, GL-LB-N-001.
- 7.3 Personnel handling radioactive materials are trained in and follow the procedures outlined in GL-RAD-S-004 for Radioactive Material Handling.
- 7.4 Personnel handling biological materials are trained in and follow the procedures outlined in GL-RAD-S-010 for The Handling of Biological Materials.
- 7.5 If there is any question regarding the safety of any laboratory practice, **stop immediately**, and consult qualified senior personnel such as a Group or Team Leader.

8.0 APPARATUS, EQUIPMENT, AND INSTRUMENTATION

- 8.1 Ancillary Equipment
 - 8.1.1 100 cc aluminum cans with lids for soil and miscellaneous samples
 - 8.1.2 10 cc Gelman Sciences Petri dish for soil, filters and miscellaneous samples
 - 8.1.3 2 L and 500 mL Marinelli beakers for water samples
 - 8.1.4 Air displacement pipettes
 - 8.1.5 Can sealing tool
 - 8.1.6 Graduated cylinder
 - 8.1.7 25 cc VWR Petri for soil and miscellaneous samples
 - 8.1.8 250 mL plastic jar for filters, soil, and miscellaneous samples
 - 8.1.9 Hot plate
 - 8.1.10 Teflon beakers and lids
 - 8.1.11 1 L Marinelli beaker for soil samples
- 8.2 Instrumentation
 - 8.2.1 High purity germanium detector, with associated electronics and data reduction software
 - 8.2.2 NaI Detector with associated electronics and data reduction software
 - 8.2.3 Top loader balance

9.0 REAGENTS, CHEMICALS, AND STANDARDS

- 9.1 NIST traceable mixed gamma standard in 100 cc aluminum can
- 9.2 NIST traceable mixed gamma standard in 2.0 L Marinelli beaker
- 9.3 NIST traceable mixed gamma standard in 0.5 L Marinelli beaker
- 9.4 NIST traceable mixed gamma standard in Gelman Sciences 10 cc Petri dish
- 9.5 NIST traceable mixed gamma standard in 13, 47 mm glass fiber filter composites in Gelman Sciences Petri dish.
- 9.6 NIST traceable mixed gamma standard in 0.4 L jar
- 9.7 NIST traceable mixed gamma standard in 0.25 L jar
- 9.8 NIST traceable mixed gamma standard in 1, 47 mm glass fiber filter

- 9.9 NIST traceable mixed gamma standard in Impregnated Charcoal Sample Cartridge.
- 9.10 NIST traceable mixed gamma standard in VWR (53 mm x 15 mm) Petri dish (approximately 25 cc)
- 9.11 NIST traceable mixed gamma standard in aqueous solution
- 9.12 NIST traceable mixed gamma standard in 1.0 L Marinelli beaker
- 9.13 NIST traceable mixed gamma standard in 20 mL liquid scintillation vial
- 9.14 16 M Nitric acid, reagent grade (HNO_3)
- 9.15 49% Hydrofluoric acid (HF)
- 9.16 12 M Hydrochloric acid, reagent grade (HCl)
- 9.17 5% Boric acid: Dissolve 50 g of H_3BO_3 per liter of DI water.
- 9.18 Nitric acid (8 M HNO_3): Prepare by cautiously adding a measured volume of concentrated nitric acid to an equal volume of DI water.

10.0 SAMPLE HANDLING AND PRESERVATION

- 10.1 For soil samples, 500 g of sample should be collected, preferably in a plastic container to avoid breakage.
- 10.2 For water samples, 2 L of sample should be collected in a plastic container and preserved to a $\text{pH} < 2$ with nitric acid.
 - 10.2.1 Before beginning an analysis, the analyst should check the sample pH by removing a minimal amount of sample with a transfer pipette and placing it on a pH strip. DO NOT insert pH strip into sample container. If the sample is received with a pH greater than 2, the analyst should contact the Group Leader or Team Leader.
- NOTE:** If the analysis is requesting I-131 (or any other iodine isotopes) Analysis without preserving is acceptable. If a sample is preserved with acid without stabilizing the iodine, Iodine may volatilize and escape the solution as a gas.
- 10.2.2 If approved by the client, the analyst should adjust the pH with nitric acid to a $\text{pH} < 2$. If the sample pH is adjusted, let the sample sit in the original container for a minimum of 24 hours before analysis. This acidification should be documented on a batch history sheet and attached to the batch paperwork.
- 10.3 For filters no preservation is necessary.

11.0 SAMPLE PREPARATION

- 11.1 Solid Sample Preparation.
 - 11.1.1 Prepare the sample for gamma counting in accordance with SOP GL-RAD-A-021, Soil Sample Preparation for the Determination of Radionuclides.
 - 11.1.2 Fill the appropriate container with sample prepared from step 11.1.1 using the following steps as a guideline:
 - 11.1.2.1 If Ra-226 analysis is required, the sample is placed in a 100 cc can for in-growth.

NOTE: It is recommended that in-growth be allowed 20 days to quantify Ra-226. Shorter ingrowth periods can be used at the request of the client. However, shorter in-growth periods may decrease the accuracy of the data. If there is insufficient mass of sample to fill the 100 cc can, contact the Team or Group Leader.

- 11.1.2.2 If sufficient mass is available, homogenized samples should be placed in the 100 cc can. Determine the net weight of the sample. If the net weight is less than 54 g or greater than 192 g, contact the Team or Group Leader to determine the appropriate counting container. Record sample weight and date in AlphaLIMS and on sample container.
- 11.1.2.3 If there is insufficient sample to fill the 100 cc can, place sample in the 10 cc or 25 cc Petri dish, cap and seal. Record sample weight and date in AlphaLIMS and on sample container.
- 11.1.2.4 If there is insufficient sample to fill the 10 cc Petri dish, perform the following digestion process:
 - 11.1.2.4.1 Weigh out an appropriate aliquot into a labeled Teflon beaker. Record this weight on the Queue sheet.
 - 11.1.2.4.2 Add 10 mL of concentrated nitric acid to each sample.
 - 11.1.2.4.3 Place samples on medium heat (approximately 300 °F) and cover each sample with a Teflon lid. Reflux all samples for 30 minutes.
 - 11.1.2.4.4 Remove Teflon lids and add 5 mL concentrated hydrochloric acid and 10 mL hydrofluoric acid to each sample. Cover samples and reflux for 120 minutes.
 - 11.1.2.4.5 Remove Teflon lids and allow samples to evaporate to dryness.
 - 11.1.2.4.6 Add 5 mL of concentrated nitric acid and evaporate to dryness.
 - 11.1.2.4.7 Repeat Step 11.1.2.4.6.
 - 11.1.2.4.8 Add 5 mL of concentrated nitric acid to the dry samples. Add 1 mL of 5% boric acid. Place the samples back on the hot plate long enough so that the dried sample dissolves into solution.
 - 11.1.2.4.9 Transfer solution to a 250 mL gamma container and dilute to 200 mL. Record the original sample mass and diluted volume on sample container.

Record the original sample mass on batch Queue sheet.

11.2 Water Sample Preparation

11.2.1 Place the appropriate labeled Marinelli beaker (typically 500 mL or 2 L) on a balance and tare the balance.

11.2.2 If less than approximately 1.1 L is available, sample should be poured into a 500 mL Marinelli beaker.

11.2.3 Transfer the appropriate volume to the tared container and record the volume of the sample on the Queue sheet.

NOTE: If there is insufficient sample to fill the Marinelli, record the exact amount of sample volume on the container and on the Queue sheet. Dilute the sample to the appropriate volume to maintain the calibration geometry. Record the volume the sample was diluted to on the sample container, also.

11.2.4 The MB should be recorded on the Queue sheet to be the same aliquot as the largest sample in the batch. An empty Marinelli beaker should be labeled as the MB and submitted with each batch of samples.

11.2.5 Submit the Marinellis and completed paperwork to the count room for gamma counting analysis.

11.3 Urine Sample Preparation

11.3.1 Refer to GL-RAD-B-030.

11.4 Preparation of Miscellaneous Matrices

11.4.1 Prepare the sample in accordance with GL-RAD-A-026 for The Preparation of Special Matrices for the Determination of Radionuclides.

11.4.2 If sample(s) was (were) received from the client in a container that matches a calibrated geometry, a direct count of the sample can be performed.

12.0 QUALITY CONTROL SAMPLES AND REQUIREMENTS

Refer to GL-RAD-D-003.

13.0 INSTRUMENT CALIBRATION, STANDARDIZATION, AND PERFORMANCE

Refer to GL-RAD-I-001.

14.0 ANALYSIS AND INSTRUMENT OPERATION

Refer to GL-RAD-I-001.

15.0 EQUIPMENT AND INSTRUMENT MAINTENANCE

Refer to GL-RAD-I-010.

16.0 DATA RECORDING, CALCULATION, AND REDUCTION METHODS

Data recording, calculation and reduction take place in accordance with SOP GL-RAD-D-003 and GL-RAD-D-006.

17.0 DATA REVIEW, APPROVAL, AND TRANSMITTAL

Data are reviewed and packaged in accordance with GL-RAD-D-003 for Data Review, Validation, and Data Package Assembly.

18.0 RECORDS MANAGEMENT

Records generated as a result of this procedure are maintained as Quality Documents in accordance with GL-QS-E-008 for Quality Records Management and Disposition.

19.0 LABORATORY WASTE HANDLING AND DISPOSAL

Radioactive samples and material shall be handled and disposed of as outlined in the Laboratory Waste Management Plan, GL-LB-G-001.

20.0 REFERENCES

- 20.1 USEPA. Prescribed Procedures for Measurement of Radioactivity in Drinking Water, Method 901.1, August 1980.
- 20.2 Canberra Nuclear Genie System Spectroscopy, Applications and Display User's Guide, Vol. I and II, May 1991.
- 20.3 DOE EML Procedures Manual, HASL-300, 27th Edition.
- 20.4 DOE EML Procedures Manual, HASL-300, 28th Edition.
- 20.5 Manual for the Certification of Laboratories Analyzing Drinking Water. Criteria and Procedures Quality Assurance. Fifth Edition EPA 815-R-05-004 January 2005.

21.0 HISTORY

Revision 25: Type II to type I water.

Revision 24: Changed recovery limit for laboratory fortified blank from 90-100% to 90-110% in section 3.3.

Revision 23: Procedure updated to include requirements for drinking water samples.

Revision 22: Updated ingrowth period for Ra-226 to 20 days.

Revision 21: SOP revised to add Ra-226, NIST traceable gamma standards, and other clarifications.

VERIFY THE VALIDITY OF THIS SOP EACH DAY IN USE

STANDARD OPERATING PROCEDURE
FOR
DETERMINATION OF METALS BY ICP-MS

APPLICABLE TO METHODS:
EPA Method 200.8
EPA SW-846 Method 6020 and 6020A

(GL-MA-E-014 REVISION 25)

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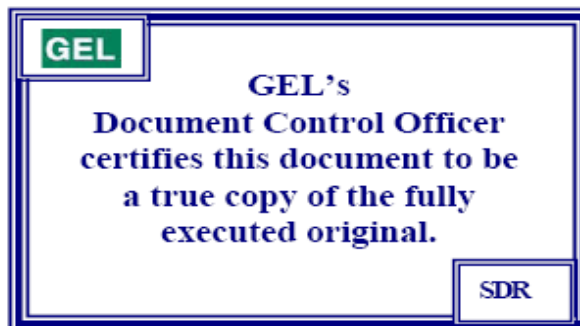


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1.0 STANDARD OPERATING PROCEDURE FOR DETERMINATION OF METALS BY ICP-MS**2.0 METHOD CODES**

- 2.1 EPA SW-846 Method 6020
- 2.2 EPA SW-846 Method 6020A
- 2.3 EPA Method 200.8
- 2.4 ASTM D4698-92 Total Dissolution

3.0 METHOD OBJECTIVE AND PURPOSE

This standard operating procedure (SOP) describes the determination of metals using a Perkin Elmer ELAN ICP-MS Model 6100 Spectrometer or a Perkin Elmer ICP-MS Model 9000 Spectrometer. Prior to analysis, samples must be digested using appropriate sample preparation methods (such as Methods 3005, 3010, 3050, or 200.2) and other applicable requests.

4.0 METHOD APPLICABILITY AND METHOD SUMMARY

- 4.1 Refer to Appendix 3 for analyte lists and masses.
- 4.2 Applicable Matrices: These methods are applicable to the determinations of any of the analytes listed above for various matrices including waters, oils, soils, sludges, biological tissues, Toxicity Characteristic Leaching Procedure (TCLP) extracts and other more unusual types of sample which are generally classified as a miscellaneous matrix.
- 4.3 General Method Summary: After the samples are prepared in accordance with the sample preparation SOP, they are analyzed by Inductively Coupled Plasma-Mass Spectrometry (ICP-MS) as follows:
 - 4.3.1 The instrument is calibrated with a minimum of two calibration points for each element to be analyzed. The points consist of a calibration blank solution to define the lower calibration point and at least one standard calibration solution at the analyte concentrations to define the higher calibration point(s). A correlation coefficient of 0.995 or better (0.998 or better for SW-846 Method 6020A) is required for each analyte if multiple standards are used or the instrument is recalibrated for the analyte of interest.
 - 4.3.2 Prepared client samples and numerous check standards and quality control samples, identified in Section 22.1 are then analyzed. The check standards and quality control samples are used to determine the quality and acceptability of the analytical data.
 - 4.3.3 Continuing Calibration Verification standards (CCV) and Continuing Calibration Blanks (CCB) are analyzed a minimum of every 10 samples to ensure that the instrument is continuing to perform correctly. For 6020A, low level continuing calibration verification standards are analyzed in conjunction with the CCVs and CCBs.

4.4 Method Codes: Analyses must conform to SW-846 Method 6020, SW-846 Method 6020A, EPA Method 200.8, and/or customer contract specifications.

4.5 Radiochemistry conversion calculations for the uranium isotopes are included in Appendix 4.

5.0 METHOD SCOPE AND PERFORMANCE CHARACTERISTICS

5.1 Calibration Range: The range of concentrations between the calibration blank, typically 0, and that of the highest calibration standard for each analyte. Calibration standards vary according to method and equipment. A minimum of two, a blank and value standard, are required.

5.2 Linear Dynamic Range standards (LRS) are analyzed with each calibration. The linear calibration range that may be used for the analysis of samples should be judged by the analyst from the resulting data. The instrument is calibrated. The target linear range should be prepared and analyzed. The LRS results must fall within $\pm 10\%$ of the target value. This LRS value is entered into the instrument's software. Any hits below this value will be valid. Hits at or above this value will be flagged by the system and must be diluted to fall within the linear dynamic range. If a linear range standard is not used for a specific calibration, the highest calibration standard becomes the upper limit of reporting.

5.3 Instrument Detection Limits (IDLs) in $\mu\text{g/L}$ are determined by calculating the average of the standard deviation of the three runs on three non-consecutive days from the analysis of a reagent blank solution with seven consecutive measurements per day. Each measurement must be performed as though it were a separate analytical sample (i.e., each measurement must be followed by a rinse and/or any other procedure normally performed between the analysis of separate samples). IDLs must be determined at least every three months.

5.4 Method Detection Limit (MDL) studies for each analyte are performed and/or verified at least annually. These studies are conducted and calculated in accordance with SW-846, Chapter 1, paragraph 5.0, and GL-LB-E-001 for the Determination of Method Detection Limits. The relevant quantitation limits are established based on the most current MDL study. The current MDLs are maintained and can be found in the AlphaLIMS database.

5.5 Method Precision: To assure analytical precision of methods used, Laboratory Control Samples (LCS) are analyzed with each batch. LCS duplicates are analyzed with each batch when requested.

5.6 Method Bias (Accuracy) is determined by calculating recoveries of LCS of a similar matrix.

5.7 If uncertainty and total propagated uncertainty measurements are needed, they may be determined using GL-QS-E-014 for Quality Assurance Measurement Calculations and Processes.

6.0 DEFINITIONS

6.1 AlphaLIMS: The Laboratory Information Management System used at GEL Laboratories, LLC.

- 6.2 Analysis Date/Time: The date and military time (24-hour clock) of the introduction of the sample, standard, or blank into the analysis system.
- 6.3 Analytical Sample: Any solution of media introduced into an instrument on which an analysis is performed excluding instrument calibration, initial calibration verification (ICV), initial calibration blank (ICB), continuing calibration verification (CCV), and continuing calibration blank (CCB).
- 6.4 Calibration Standard (CAL): A solution prepared from the primary dilution standard solution or stock standard solutions and the internal standards surrogate analytes. The CAL solutions are used to calibrate the instrument response with respect to analyte concentration.
- 6.5 Continuing Calibration Blank (CCB): An aliquot of reagent water or other blank matrix that is analyzed after each CCV. The CCB is used to determine whether the analytical sequence is in control during sample analysis.
- 6.6 Continuing Calibration Verification (CCV) Standard: An aliquot of reagent water or other blank matrix to which known quantities of the method analytes are added. The CCV is analyzed exactly like a sample, periodically throughout the run sequence. Its purpose is to determine whether the analytical sequence is in control during the sample analysis. It may be prepared from the same source as the calibration standards, and is usually of varied concentrations.
- 6.7 Contract Required Detection Limit (CRDL): Minimum level of detection acceptable under the client project requirements.
- 6.8 Control Limits: A range within which specified measurement results must fall to be compliant. Control limits may be mandatory, requiring corrective action if exceeded, or advisory, requiring that noncompliant data be flagged.
- 6.9 Correlation Coefficient: A number (r) that indicates the degree of dependence between two variables (concentration-absorbance). The more dependent they are, the closer the value to one. Determined on the basis of the least squares line.
- 6.10 Data Qualifiers: The following qualifiers should be used in order to identify analytical situations that might need additional information stated in narrative before the release of the data.
- U - Non-Detect. Below the Instrument or Method Detection Limit (depending upon specific project requirements)
 - B - Sample concentration value is between the MDL (or IDL) and the CRDL or analyte was detected in the Method Blank (Client Specific)
 - J - Sample concentration is between the MDL (or IDL) and the CRDL-client specific qualifier.
 - Blank - Concentration value is above the CRDL
 - * - An RPD value in the duplicate sample is out of criteria
 - N - A percent recovery value in the spike sample is out of criteria
 - E - A percent difference in the serial dilution sample is out of criteria because of the presence interference.

- 6.11 Duplicate: A second aliquot of a sample that is treated the same as the original sample in order to determine the precision of the method.
- 6.12 Initial Calibration Blank (ICB): An aliquot of reagent water or other blank matrix that is analyzed after each ICV. The ICB is used to determine whether there is carryover contamination.
- 6.13 Initial Calibration Verification (ICV): A solution of method analytes of known concentrations. The ICV is obtained from a source external to the laboratory and different from the source of calibration standards. It is used to check laboratory performance with externally prepared test materials.
- 6.14 Instrument Performance Check Solution (IPC): A solution of one or more method analytes, surrogates, internal standards, or other test substances used to evaluate the performance of the instrument system with respect to a defined set of criteria.
- 6.15 Interferants: Substances that affect the analysis for the element of interest.
- 6.16 Internal Standard: Pure analyte(s) added to a sample, extract, or standard solution in known amounts and used to measure the relative responses of other method analytes.
- 6.17 Laboratory Control Standard (LCS): An aliquot of reagent water or other blank matrix to which known quantities of the method analytes are added in the laboratory. The LCS is analyzed exactly like a sample, and its purpose is to determine whether the methodology is in control and whether the laboratory is capable of making accurate and precise measurements.
- 6.18 Linear Calibration Range (LCR): The concentration range over which the instrument response is linear.
- 6.19 Method Blank (MB): An aliquot of reagent water or other blank matrix that is treated exactly as a sample including exposure to all glassware, equipment, solvents, reagents, internal standards, and surrogates that are used with other samples. The MB is used to determine if method analytes or other interferences are present in the laboratory environment, the reagents, or the apparatus.
- 6.20 Method Detection Limit (MDL): The minimum concentration of an analyte that can be identified measured and reported with 99% confidence that the analyte concentration is greater than zero.
- 6.21 Serial Dilution: The dilution of a sample by a known factor. When corrected by the dilution factor, the diluted sample should agree with the original undiluted sample within the specified limits. Serial dilution may reflect the influence of interferants.
- 6.22 Spike (Matrix Spike or Post Spike): An aliquot of an environmental sample to which known quantities of the method analytes are added in the laboratory. The MS or PS is analyzed exactly like a sample, and its purpose is to determine whether the sample matrix contributes bias to the analytical results. The background concentrations of the analytes in the sample matrix must be determined in a separate aliquot and the measured values in the MS or PS corrected for background concentrations.

- 6.23 Limit of Detection (LOD): An analyte, method and matrix specific estimate of the minimum amount of a substance that can be reliably detected. GEL has established $LOD = 2 \times MDL$.
- 6.24 Limit of Quantitation (LOQ): An analyte, method and matrix specific estimate of the minimum amount of a substance that can be reported with a specific level of confidence. The LOQ is set at or above the concentration of the lowest initial calibration standard. The laboratory must empirically demonstrate precision and bias at the LOQ. The LOQ and associated precision and bias must meet client requirements and must be reported. GEL uses the following guidance ($LOD < LOQ$):
- When $LOD < PQL$, $PQL = LOQ$
- When $LOD > PQL$, LOQ is raised to next lowest calibration standard.
- 6.25 Practical Quantitation Limit (PQL): The PQL is typically at or above the lowest point on an acceptable initial calibration curve. It may also be determined by multiplying the MDL by approximately 2 to 10. Concentrations of a target analyte determined to be greater than its PQL are defined as quantitative results. This limit is not used in DoD ELAP reporting.
- 6.26 Statistical Process Control (SPC) Limits: Statistically derived limits that establish acceptable ranges for recoveries of analytes of interest, including LCS, MS, MSD, PS and PSD.
- 6.27 Stock Standard Solution: A concentrated solution containing one or more method analytes prepared in the laboratory using certified reference materials or purchased from a reputable commercial source.
- 6.28 10% Frequency: A frequency specification during an analytical sequence allowing for not more than 10 analytical samples between required calibration verification measurement, as specified by the EPA methodology.

7.0 INTERFERENCES TO THE METHOD

- 7.1 Chemical interferences are minimal in ICP-MS Spectroscopy because the extremely high energy of the plasma breaks nearly all the chemical bonds. However, ICP-MS analysis is subject to the following three types of interferences:
- 7.1.1 Physical interferences are those physical properties of a sample solution that prevent their introduction to the plasma with efficiency equal to that of the calibration standards. This type of interference can be corrected via the bias correction calculation in SW-846, Chapter 1, paragraph 5.0, through the use of an internal standard in accordance with the instrument operating manual or by diluting the sample in reagent blank solution until the percent recovery falls with method guidelines.
- 7.1.2 Isobaric elemental interferences are caused by isotopes of different elements that form singly or doubly charged ions of the same nominal mass-to-charge ratio and that cannot be resolved by the mass spectrometer. If analytical isotopes are selected that may have an isobaric interference, then all data obtained under such conditions must be corrected by

measuring the signal from another isotope of the interfering element and subtracting the appropriate signal ratio from the isotope of interest.

- 7.1.3 Isobaric polyatomic ion interferences are caused by ions consisting of more than one atom that have the same nominal mass-to-charge ratio as the isotope of interest, and that cannot be resolved by the mass spectrometer in use. These ions are commonly formed in the plasma or interface system from support gases or sample components. Most of the common interferences have been identified and are listed in Method 200.8, Table 2 together with the method elements affected. Such interferences must be recognized, and when they cannot be avoided by the selection of alternative analytical isotopes or sample prep procedures, appropriate corrections must be made to the data.

8.0 SAFETY PRECAUTIONS AND HAZARD WARNINGS

- 8.1 PREVENT SKIN AND EYE CONTACT BY USING SPECIFIED PERSONAL PROTECTIVE EQUIPMENT WHEN MAKING STOCK REAGENTS.
- 8.2 WORK UNDER A HOOD TO PREVENT INHALATION WHEN MAKING STOCK REAGENTS.
- 8.3 Sample digestates are not extremely volatile or spontaneously combustible, but they are normally acidic and should be handled with care. Small spills may generally be wiped up with paper towels that can be disposed of in the trash. Larger spills may require the use of a mop, and the mop head may have to be disposed of as potentially hazardous waste in accordance with the Laboratory Waste Management Plan (GL-LB-G-001). If the spilled digestates begin any obvious fuming or reacting, pour a generous amount of the acid neutralizer, which is located in each lab, onto the spill before attempting to clean it up.
- 8.3.1 Gloves should be worn to avoid skin contact with digestate during clean-up.
- 8.3.2 Eye protection is required when handling samples and an eyewash station is located in each analysis lab.
- 8.3.3 Do not persist in cleaning up a spill in the presence of strong fumes. Move out of the area, try to isolate the area and notify your supervisor immediately.
- 8.4 Handling radioactive samples requires the use of gloves, a lab coat or an apron in addition to eye protection. Refer to GL-RAD-S-004 for Radioactive Material Handling
- 8.5 These instruments use high voltage electricity and therefore, should be shut completely down any time electronic components may be exposed to personnel or any liquids.
- 8.6 Wear eye protection with side shields while performing procedures in the lab.
- 8.7 All chemicals and samples should be treated as potential health hazards, and exposure to these chemicals must be reduced to the lowest level possible. GEL maintains a current awareness file of Occupational Safety and Health Administration (OSHA) regulations regarding the safe handling of the chemicals in

the laboratory as well as a reference file of Material Safety Data Sheets (MSDS). These documents are maintained in the laboratory. Individual sample MSDS forms provided by the clients are kept in Login.

8.8 Personal protective equipment

8.8.1 Gloves are required when handling the chemicals in this procedure.

8.8.2 Work under a hood when using concentrated acids.

8.9 Prior to handling radioactive samples, analysts must have had radiation safety training and must understand their full responsibilities in radioactive sample handling. Some general guidelines follow:

8.9.1 Wear a lab coat when working with radioactive samples.

8.9.2 Prohibit admittance to immediate work area.

8.9.3 Protect counter tops with counter paper or work from radioactive sample handling trays.

8.9.4 Post signs indicating radioactive samples are in the area.

8.9.5 Take swipes of the counter tops upon completion of work. Deliver those swipes to the nearest swipe count box.

8.9.6 Segregate radioactive wastes. Radioactive waste containers are obtained from Waste Management.

8.10 All samples, chemicals, extracts, and extraction residues must be transferred, delivered, and disposed of safely according to all related SOPs.

8.10.1 Segregate solid wastes from liquid wastes in the satellite area containers.

8.10.2 Segregate oil wastes from water-soluble wastes in the satellite area containers

8.11 Never leave gas cylinders unchained or untied, including when they are on the moving carts.

8.12 In the event of an accident or medical emergency, call for help immediately. When time and safety permit, an accident report form should be completed and turned in to the safety committee.

8.13 Fire escape routes are posted in the lab; all personnel should be familiar with them. In addition, fire safety equipment such as fire extinguishers is located in the lab. Training is available on the proper operation of this equipment.

9.0 CAUTION WARNINGS

9.1 Because they can be health hazards, the exhaust gases from the plasma and vacuum systems must be eliminated through the laboratory's ventilation duct, which is attached to the instrument's exhaust vent. If inadequate ventilation occurs, pump-fluid vapor, ozone, and other toxic products of combustion can accumulate in the laboratory and cause bodily harm. Hydrofluoric acid (HF) fumes, if inhaled, extensively burn lung tissue. Ensure that the exhaust system established at installation continues to operate effectively.

9.2 Prepare sample and transfer acids using a hood to avoid fumes.

9.3 Store and prepare sample away from the instrument to minimize corrosion.

- 9.4 Clean up any spills quickly.
- 9.5 The drain vessel contains the spray chamber's effluent, which can be toxic. Corrosion of the vessel and connecting tube can result in leaks that damage the instrument or cause bodily harm.
 - 9.5.1 Use the capped plastic drain vessel that was provided with the instrument. Never use glass.
 - 9.5.2 Place the drain vessel on the instrument table below the peristaltic pump where the container is easy to check.
 - 9.5.3 Check the drain vessel frequently. Empty it before it is three-fourths full.
 - 9.5.4 Check the tubing and vessel for deterioration. If the tubing becomes brittle or cracked, replace it. Organic solvents cause more rapid deterioration than aqueous solutions.
- 9.6 The torch and interface remain hot after the plasma is turned off. Do not touch the torch box or interface cones for 10 minutes after the plasma has been shut off.
- 9.7 High voltages and radio frequencies are potential hazards of the ICP-MS. Shut the instrument down completely before removing any of the outside panels (to clean air filters, replace a fuse, etc.).
- 9.8 When changing the rotary pump oil, remember that the pump oil may be hot. The oil can cause a burn if allowed to contact the skin.
- 10.0 APPARATUS, MATERIALS, REAGENTS, EQUIPMENT, AND INSTRUMENTS**
 - 10.1 Apparatus and Equipment
 - 10.1.1 Replacement special glass parts for the ICP-MS such as quartz torch bodies, injector tips and spray chambers may be ordered from a qualified vendor through the GEL Purchasing Agent.
 - 10.1.2 Replacement ICP-MS interface parts such as sampling cones, skimmer cones and ion-optics can be purchased from a qualified vendor through the GEL Purchasing Agent.
 - 10.1.3 Consumable materials such as tubing are often attainable from various scientific product companies. The GEL Purchasing Agent can help to find the best prices. These items may also be ordered from the instrument manufacturer if necessary. All orders must be placed through the GEL Purchasing Agent.
 - 10.2 Reagents, Chemicals, and Standards
 - 10.2.1 Reagents: Refer to Reagent Logbook
 - 10.2.2 Standards: Refer to GL-LB-E-007 for Laboratory Standards Documentation and GL-LB-E-015 for Control of Laboratory Standards.
 - 10.2.3 Other Chemicals: Additional compounds, surfactants, oils, cleaning agents, etc., may be routinely ordered through the GEL Purchasing Agent.
 - 10.3 Instrumentation
 - 10.3.1 Perkin Elmer ICPMS ELAN Model 6100 with IBM compatible PC, Monitor, Printer.

- 10.3.2 Perkin Elmer ICP-MS ELAN Model 9000 with IBM compatible PC Monitor, Printer.
- 10.3.3 CETAC Model ASX-500 Autosampler (PE 6100)
- 10.3.4 CETAC Model ASX-510 Autosampler with accessory autodiluter (PE 9000)
- 10.3.5 Neslab CFT75 recirculating bath provides cooling to the ICP-MS (PE 6100/ PE 9000)
- 10.3.6 CETAC ASXpress Rapid Sample Introduction System

11.0 SAMPLE HANDLING AND PRESERVATION REQUIREMENTS

- 11.1 Aqueous samples should be preserved with nitric acid to a pH of < 2 prior to receipt by the analyst. Solid samples should be kept at $0^{\circ} \leq 6^{\circ} \text{ C}$ prior to digestion.
- 11.2 Refer to GL-SR-E-001 for Sample Receipt, Login and Storage.

12.0 SAMPLE PREPARATION TECHNIQUES

- 12.1 All samples except drinking water with Turbidity < 1 NTU and samples specifically exempted by contract, are prepared in accordance with the following SOPs:
 - 12.1.1 GL-MA-E-016 for Sample Preparation for Total Recoverable Elements by EPA Method 200.2 (USEPA Method 200.2)
 - 12.1.2 GL-MA-E-006 for Acid Digestion of Total Recoverable or Dissolved Metals in Surface and Groundwater Samples for Analysis by ICP or ICP-MS (USEPA SW-846 Method 3005A)
 - 12.1.3 GL-MA-E-008 for Acid Digestion of Total Metals in Aqueous Samples and Extracts for Analysis by ICP or ICP-MS (USEPA SW-846 Method 3010A)
 - 12.1.4 GL-MA-E-009 for Acid Digestion of Sediments, Sludges and Soils (USEPA SW-846 Method 3050B)
 - 12.1.5 GL-MA-E-021 for Total Digestion of Sediment Samples for Analysis by ICP or ICP-MS (ASTM D4698-92)
- 12.2 Additional filtration may be required to prevent clogging of sample introduction system.
- 12.3 All sample preparation records are stored in AlphaLIMS.
- 12.4 Sample spills should be handled as stated in Section 8.3.

13.0 EQUIPMENT AND INSTRUMENT MAINTENANCE

- 13.1 Routine Preventative and Special Operational (Failure)
- 13.2 Routine Preventative Maintenance (PM) Procedures are done as follows:

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Frequency	Procedure
When Needed	Clean nebulizer tip after use. Replace peripump sample introduction tubing. Change pump hoses on drain systems. Check drain waste collection containers, and empty as necessary. Check Neslab water level and add water if required. Clean/replace interface cones. Clean/replace nebulizer. Clean/replace torch. Check/replace water filter.
Quarterly	Change oil in interface rotary pump (or as needed). Clean ion lenses 4-6 months (or as needed).
6 Months	Clean air filters.
12 months	Change pump oil in backing rotary pump. Evaluate/replace EM (electron multiplier)

13.3 Non-Routine Maintenance Procedures (Special, Operational or Failure Mode Maintenance)

13.3.1 If the instrument will not function properly see the trouble shooting section in the appropriate Maintenance Manual.

13.3.2 If the analyst is unable to determine cause/fix instrument, the GEL Service Engineer is called, and if needed, the manufacturer's customer support number may be found in the owner's maintenance manuals.

13.4 Refer to ICP-MS Maintenance Logbook for routine records. Service call records are also available.

14.0 PREPARATION OF STANDARD SOLUTION AND QUALITY CONTROL SAMPLES

14.1 Source standards records are recorded in AlphaLIMS.

14.2 Recommended Suppliers: Refer to source log and use Approved Vendors List maintained in Procurement.

14.3 Standards are receipted, labeled, prepared and stored in accordance with the GL-LB-E-007 for Laboratory Standards Documentation.

15.0 INSTRUMENT CALIBRATION

15.1 Samples may be analyzed manually or automatically.

15.1.1 Tuning for each instrument is performed daily according to the following directions and criteria.

15.1.2 PE 6100/ 9000: Aspirate a tuning solution consisting of 10 µg/L each of ⁹Be, ²⁴Mg, ⁵⁹Co, ¹¹⁵In and ²⁰⁸Pb. Perform 5 replicates. Manufacturer's recommended tune criteria are as follows:

Parameter	Starting Point
⁹ Be	± 0.10 amu
²⁴ Mg	± 0.10 amu
⁵⁹ Co	± 0.10 amu
¹¹⁵ In	± 0.10 amu
²⁰⁸ Pb	± 0.10 amu
Ba ⁺⁺ net intensity mean	< 0.05 or 5%
CeO net intensity mean	< 0.03 or 3%
Be, Mg, Co, In, Pb net intensity RSD	< 5%
Resolution at 10% peak height	< 0.9 amu

15.1.3 Conduct additional tuning procedures as specified by client contract or other methodology.

15.1.4 If any of the preceding tune criteria does not meet the recommended requirements, investigate the problem, correct the situation, and reanalyze the tune sequence.

15.1.5 Standardization: Standardization is required on a daily basis.

15.2 Internal standards are used as appropriate for the analytes of interest. The internal standards are made in the appropriate acid and contain varying concentrations of such elements as ⁴⁵Sc, ⁷⁴Ge, ¹¹⁵In, ¹⁷⁵Lu, and/or ¹⁸¹Ta. The internal standard solution is mixed in-line with the sample stream using a dedicated channel of the peristaltic pump. Internal standards may be added at the time of analysis as an alternative to in-line mixing. Alternate internal standards may be used to meet client needs.

15.3 Calculations are described in the instrument manual.

15.4 Calibration standards vary according to method and equipment. A minimum of two, a blank and value standard, is required.

15.5 For required quality control standards refer to Section 21.0.

15.6 For continuing calibration requirements refer to Section 21.0.

15.7 For what to do when initial or continuing calibrations fail to meet requirements, refer to Section 21.0.

16.0 INSTRUMENT PERFORMANCE REQUIREMENTS

16.1 Before samples may be analyzed to generate reportable data, the instrument must have been tuned and calibrated. Also, the Initial Calibration Verification (ICV), which is prepared from an independent source, Initial Calibration Blank (ICB), the Reportable Detection Limit (CRDL), the Interference Check Standards (ICS-A, ICS-AB), and the Linear Range Standards (LRS) must meet the requirements stated in Section 21.2 for each analyte being reported, unless otherwise required by methodology, clients or contracts.

- 16.2 The instrument calibration and all continuing verification data is maintained in the printed hard copy file. The printouts are kept in chronological order by instrument. Recent files are in the metals laboratory; older records are archived in short or long-term storage.

17.0 ANALYST AND METHOD VERIFICATION REQUIREMENTS

- 17.1 Analyst training is conducted and certified in accordance with the GL-HR-E-002 for Employee Training.
- 17.2 Method performance is verified by the conductance of MDL studies in accordance with GL-LB-E-001 for The Determination of Method Detection Limits, and by the evaluation of LCS and LCS duplicates for each batch of samples.

18.0 ANALYSIS PROCEDURES AND INSTRUMENTAL OPERATION

- 18.1 All samples are introduced in a set measured quantity, via a peristaltic pump, through a nebulizer into a spray chamber and carried with argon gas through the Radio Frequency (RF) field to generate analyte ions that are selected and measured by the quadruple mass spectrometer.
- 18.2 Run Sequence
- 18.2.1 Instrument Calibration executed in accordance with Section 16.
- 18.2.2 Initial Calibration Verification steps include:
- 18.2.2.1 ICV
 - 18.2.2.2 ICB
 - 18.2.2.3 CRDL (low level ICV for 6020A)
 - 18.2.2.4 ICS-A
 - 18.2.2.5 ICS-AB
 - 18.2.2.6 CCV
 - 18.2.2.7 CCB
 - 18.2.2.8 LRS
 - 18.2.2.9 CCV
 - 18.2.2.10 CCB
- 18.2.3 Sample Run (10 samples or less)
- 18.2.4 Continuing Calibration Verification:
- 18.2.4.1 CCV
 - 18.2.4.2 CRDL (for SW-846 Method 6020A only and analyzed at end of batch).
 - 18.2.4.3 CCB
- 18.2.5 Repeat steps 18.2.3 and 18.2.4 until end of run or verification is out of specification. When the latter occurs, repeat step 18.2.1 and continue with steps 18.2.2 and 18.2.3.
- 18.2.6 Repeat steps 18.2.3 through 18.2.5 until the end of the run or verification is out of specification. When latter condition occurs, repeat 18.2.1 and continue sequentially through 18.2.7.

18.2.7 Final Verification Steps

- 18.2.7.1 ICS-A (if required)
- 18.2.7.2 ICS-AB (if required)
- 18.2.7.3 CCV
- 18.2.7.4 CRDL (for SW-846 Method 6020A only)
- 18.2.7.5 CCB

NOTE: Method 6020A only requires there to be a closing low level continuing calibration standard (CRDL). For ease of analysis, the CRDL standard can be analyzed every 10 samples.

18.3 For data storage refer to Section 24.1.

18.4 General operation of the instrument.

18.4.1 The Perkin Elmer Model 6100 and Perkin-Elmer Model 9000 are inductively-coupled argon plasma mass spectrometers. The ICP-MS is capable of determining analytes from $m/z = 6$ (Li) through $m/z = 238$ (U). The operating software of the system is based on Microsoft Windows.

18.4.2 Set-up

- 18.4.2.1 Attach the nebulizer argon line to the quick-connect on the nebulizer's argon tube and ensure that it is tightly in place.
- 18.4.2.2 Attach the nebulizer and endcap to the spray chamber.
- 18.4.2.3 Attach sample pump tubing to the front peristaltic pump (various sizes may be used depending on need), and attach feed end to endcap.
- 18.4.2.4 If running manually, place the suction end of the sample tubing in acidified rinse water; otherwise attach it to the autosampler.
- 18.4.2.5 Ensure that drain hose is connected to the spray chamber and properly plumbed through the drain peristaltic pump.

18.4.3 Run Procedure

- 18.4.3.1 Refer to the respective owner's manuals for specific instructions for tuning, sequence loading, and method development on each instrument; ELAN 6000/6100 software guide; ELAN 9000 software guide.
- 18.4.3.2 The owner's manual for each instrument is located in the ICP-MS laboratory.

18.4.4 Typical Analysis Problems

- 18.4.4.1 Sample Overage: If a requested element is overrange for a sample, it is necessary to dilute the sample and rerun. Dilution factors must be taken into account when reporting final values.
- 18.4.4.2 Interferences: Although the ICP-MS can compensate for many interferences with appropriate correction factors, unpredicted interferences may still occur. If the analyst suspects this, then

the sample should be diluted and rerun. Dilution factors must be taken into account when reporting final values.

- 18.4.4.3 Torch/Sample Introduction Drift: Various changes in torch plasma conditions or sample introduction can have a great effect on the detector counts. They must be corrected or the instrument must be re-standardized.

18.4.4.3.1 If oils or solid samples have been run, and the CCV is not acceptable, allow the instrument to rinse 15 to 20 minutes. If CCV is then acceptable, reanalyze the samples and continue. If the CCV again fails its requirements, re-standardize as necessary.

18.4.4.3.2 If sample introduction drift is suspected, check peristaltic pump tubing for collapse and replace as needed.

18.4.4.3.3 Tubing connections may become loose allowing air to bubble into the sample path. Tighten or replace tubing. If necessary, seal with Parafilm.

- 18.4.4.4 Serial dilution: If the analyte concentration is sufficiently high (minimally, a factor of 100 times above the IDL for non-DOE Clients, 100 x the MDL for DOE-Alb Clients, and 50 times the LOQ for DoD QSM in the original sample), an analysis of a five fold (1 + 4) dilution should agree within 10% of the original determination. If not, a chemical or physical interference effect should be suspected.

18.5 Power switches and auxiliaries

- 18.5.1 Each instrument has one main power switch. Refer to the respective maintenance manuals for the location (ELAN 6100 and ELAN 9000 Hardware Guide). This switch remains on except while servicing the instrument.

- 18.5.2 The computer system has 3 power cords all connected to a switchable power strip: the computer main, the monitor and the printer power cords. The power to these is left on while the instrument is not in use for short times (e.g. overnight).

18.5.2.1 The main computer switch is on the front of the CPU.

18.5.2.2 The **monitor switch** is on the front of the monitor itself.

NOTE: If the computer is left in use without an analyst present for an extended period of time, turn the monitor off to prevent screen burn-in.

18.5.2.3 The **printer switch** is on the lower front of the printer. Refer to printer manual for operating instructions.

- 18.5.3 Argon gas is provided through a manifold from the storage tank. Under normal conditions a constant argon flow is available.

- 18.5.4 Start-up

- 18.5.4.1 Before starting, ensure that the power is on and the vacuum system is switched on.
- 18.5.4.2 The owner's manual for each instrument provides specific instructions for ignition of the plasma (refer to Section 18.5.1 for software manuals).
- 18.5.4.3 The plasma should be steady and should not flicker.
- 18.5.4.4 Once ignition has occurred, levels may be adjusted; it is best to set levels appropriate to the method by loading or editing an appropriate tune file and allow 30 minutes for warm-up.
- 18.5.5 Shutdown
 - 18.5.5.1 When the plasma is off, the instrument is either in Standby or Shutdown mode. The ICP-MS should be completely shut down only in case of major maintenance, relocation of the instrument, or when the lab is closed for an extended time.
 - 18.5.5.2 Daily shutdown (if required): Refer to the respective owner's manual shutdown procedures (Refer to Section 18.5.1 for software manuals.)
- 18.6 Sample quantity requirements are approximately 5 mL for each run with the nebulizer. If out of specification or range, further sample may be necessary for reruns.
- 18.7 The autosampler can be used by attaching introduction tube to sample introduction system, and rinsing tubing to peristaltic pump and following autosampler table set-up and procedures. To define a sequence, refer to the appropriate software manual (refer to Section 18.5.1).
- 19.0 CALCULATIONS AND DATA REDUCTION METHODS**
 - 19.1 Any dilutions, concentrations, or preparation factors must be taken into account prior to reporting data to a client.

$$\text{Relative Percent Difference} = \frac{100 * |\text{Sample 1 Value} - \text{Sample 2 Value}|}{(\text{Sample 1 Value} + \text{Sample 2 Value}) / 2}$$

$$\text{Matrix Spike Recovery} = \frac{100 * (\text{Spike Value} * \text{DF} * \text{PF} - \text{Sample Value} * \text{DF} * \text{PF})}{\text{Spike Nominal Concentration} * \text{DF} * \text{PF}}$$

$$\text{Post Spike Recovery} = \frac{100 * (\text{Spike Value} - \text{Sample Value})}{\text{Spike Nominal Concentration}}$$

$$\text{LCS Recovery} = \frac{100 * (\text{Sample Value} * \text{DF} * \text{PF})}{\text{Nominal Concentration} * \text{DF} * \text{PF}}$$

Where: Sample Value = instrument reading for the sample
 Spike Value = instrument reading for the spiked sample
 DF = Dilution Factor
 PF = Preparation Factor

Relative Error Ratio (RER) 2 sigma equation:

$$RER = \frac{| \text{Sample Activity} - \text{Duplicate Activity} |}{\sqrt{(\text{Sample 2 sigma TPU}/1.96)^2 + (\text{Duplicate 2 sigma TPU}/1.96)^2}} \leq 3$$

NOTE: Activity calculations can be found in Appendix 4 of this SOP. Two sigma TPU calculations can be found in SOP GL-QS-E-014.

19.2 Care must be taken that the correct units are being employed.

20.0 DATA RECORDING

20.1 ICP-MS data are generally stored to the hard disk drive of the ICP-MS computer system and printed at the instrument as each sample is analyzed.

20.2 ICP-MS samples are generally analyzed in three replicates (minimum of 2) and the reported value is the average of the replicates. The data for individual replicates is stored in the computer.

20.3 Data are processed locally by manual or programmable procedures to eliminate unused data, to enter dilution factors, and to enter relevant conversion factors prior to uploading to AlphaLIMS.

21.0 QUALITY CONTROL REQUIREMENTS

21.1 Frequency of Quality Control Activities (also refer to Appendix 1)

21.1.1 Initial Calibration Verification (ICV) is performed immediately following each calibration and Continuing Calibration Verification (CCV) is performed after at least every 10 samples.

21.1.2 Initial Calibration Blank (ICB) is performed immediately following the ICV and Continuing Calibration Blanks (CCB) must run with each CCV.

21.1.3 An Interference Check Standard (ICS) is analyzed at the beginning of each analytical run and at least once every twelve hours (if required). Additional requirements may be specified by client contract or methodology.

21.1.4 The PQL standard is analyzed after each calibration and recommended at least every 10 samples between the CCV and CCB analyses for SW-846 Method 6020A. Method 6020A requires analysis at the end of every batch and only recommends analysis every 10 samples. This standard may also be labeled as a CRDL standard.

21.1.5 A method blank (MB) is performed for each batch of 20 or fewer samples or per client requirement.

- 21.1.6 A matrix spike (MS) and a duplicate (DUP), or matrix spike (MS) and matrix spike duplicate (MSD) are analyzed for each batch of 20 or fewer samples or per client requirements (5% frequency). For EPA 200.8, the same QC are analyzed for each batch of 10 or fewer samples (10% frequency).
- 21.1.7 A laboratory control sample (LCS) is analyzed with each batch of 20 or fewer samples. An LCS duplicate (LCS DUP) may be added if required by the client.
- 21.1.8 Serial dilutions or analytical spikes are analyzed to confirm the presence or absence of interferences when analyzing a new matrix type. The serial dilution is generally performed at a 5x of the test sample.
- 21.1.9 A post spike (PS) is required for DoD-QSM or SW 846 6020A if the matrix spike (MS) or matrix spike duplicate (MSD) recoveries fall outside of the limits in section 21.2.8. Post spikes can also be performed at client request.
- 21.2 Acceptance Limits (also refer to Appendix 2)
 - 21.2.1 ICV results must be between 90% and 110% of the true values for work under EPA SW-846 Method 6020A or EPA Method 200.8. CCV results must be between 90% and 110% of the true values for work under EPA Method 200.8 or SW846 Method 6020A.

NOTE: The ICV is the second source standard and may be used as the CCV as it also will show calibration verification.
 - 21.2.2 ICB and CCB results must have an absolute value less than the Reporting Limit (RL). For DoD QSM analysis, the absolute value must be less than the LOD. If this is not the case, the reason for the out-of-control condition must be found and corrected, or any data reported must be 10 times greater than the absolute value for the element or less than the RL.
 - 21.2.3 Interference Check Sample results must be monitored at the beginning of an analytical run or once every 12 hours, whichever is more frequent, for work under SW-846. The ISCA and ICSAB must recover 80-120% the reporting level for the spiked analysis and must have an absolute value less than 2x the reporting level for the non-spiked analyte. For DoD QSM, the ICSA must have an absolute value of less than the absolute value of the LOD analytes.
 - 21.2.4 The Linear Range Standard (LRS) is analyzed within the calibration verification read back and must fall between 90% and 110% of the true values. Meeting these criteria allows target analyte concentration to be reported up to the LRS concentration thus extending the calibration range of the instrument. Any sample concentrations above the LRS concentration will be diluted to fall below the concentration of the linear calibration range standard.
 - 21.2.5 The intensities of all internal standards must be monitored for every analysis. When the intensity of any internal standard fails to fall between

30% and 120% (or 70% and 130% for SW-846 Method 6020A) of the intensity of that internal standard in the initial calibration then the sample must be diluted five fold (1 + 4) and reanalyzed with the addition of appropriate amounts of internal standards. The intensity levels of the internal standards for the calibration blanks (ICB and CCB) and instrument check standards (ICV and CCV) must agree within $\pm 20\%$ of the intensity level of the internal standard of the original calibration solution. For work done under EPA Method 200.8, the internal standard responses of any one internal standard must not deviate more than 60% to 125% of the original response in the calibration blank. Five internal standards (^{45}Sc , ^{74}Ge , ^{115}In , ^{175}Lu , and/or ^{181}Ta) are used to cover the mass ranges reported. Refer to Appendix 3 for list of internal standard/ analyte associations. Other exotic analytes may be used as needed. Refer to Section 15.2.

- 21.2.6 Method blank results must be lower than the PQL or less than 10% of the determined value of all samples in the batch. When performing work under EPA Method 200.8, if LRB (laboratory reagent blank) values are 10% or more of the analyte level determined for a sample or are 2.2 times the analyte MDL, then fresh aliquots of the samples must be prepared and analyzed again for the affected analytes after the source of contamination has been corrected and acceptable LRB values have been obtained. For DoD QSM work, the absolute value must be less than $\frac{1}{2}$ RL or less than 10% of the determined value of all samples in the batch. Al, Fe, Mg, Ca, Na, and K must be less than the RL.
- 21.2.7 LCS results, and LCS duplicate (if performed), must be within process control limits as established by Statistical Process Control, manufacturer's certification, or method requirements.
- 21.2.8 Matrix spikes with recoveries between 75% and 125% suggest the absence of interference for work under EPA Method 200.8. Matrix spikes with recoveries between 75% and 125% suggest the absence of interference for work under SW-846 Method 6020 and SW-846 Method 6020A. Matrix spikes with recoveries between 80% and 120% suggest the absence of interferences for work under DoD QSM.
- 21.2.9 The relative percent difference (RPD) between a sample and a sample duplicate should be within $\pm 20\%$ if the analyte concentration in the sample or duplicate is greater than 5 times the RL. If either the sample or duplicate concentration is less than 5 times the RL, the results should agree within the absolute value of the RL. Results less than the MDL or IDL are not evaluated. The relative error ratio (RER) between a sample and a sample duplicate should be $\leq 3\%$.
- 21.2.10 Serial dilution results that agree within 10% of the original analytical results, if the original results are greater than 100 times the instrument detection limit or greater than 50 times the LOQ, suggest the absence of interference.

- 21.2.11 Post spikes with recoveries between 75% and 125% under EPA Method 200.8 and SW-846 Method 6020, and between 80% and 120% under SW-846 6020A suggest the absence of interference.
- 21.3 Out-of-Control Situations
- 21.3.1 ICV and/or CCV failure requires recalibration of the instrument and/or preparation of new standard solutions. Samples analyzed prior to or after calibration verifications that are not acceptable for required analytes must be reanalyzed. An ICV or CCV that has failed may be rerun once only if there is an attributable cause known to have affected the CCV only and not the previous samples. Examples of an acceptable cause may be a sample tip out of solution during analysis, an incorrectly prepared CCV, or obvious carryover in the CCV from a very high sample immediately prior to the CCV. If a CCV is reanalyzed, the data must be lined through, initialed and dated, and the reason for the rerun must be documented on the raw data. In addition, corrective action should be taken to eliminate the cause of the initial CCV failure to prevent future occurrence.
- 21.3.2 ICB and CCB failure requires recalibration of the instrument and/or calibration blank solution to be remade. The CCB is acceptable if the level of analyte in the corresponding sample(s) is 10 times greater or less than the PQL for the failing element. For DoD QSM work, the absolute values must be less than the LOD or less than 10% of the determined value of all samples in the batch.
- 21.3.3 ICS failure requires that the instrument be re-calibrated or the interferences be corrected, via recalculation, of Interelement Correction Factors so that the ICS can be read within the required limits before samples are analyzed. ICS failure at the end of an analysis period will require that the samples' ICSA run for the affected analyte(s) during that period to be reanalyzed. The ICSA and ICSAB must recover 80-120% for the spiked analytes and must have an absolute value less than 2x the reporting level for the non-spiked analytes. For DoD QSM, the ICSA must have an absolute value of less than the LOD for the non-spiked analytes.
- 21.3.4 LRS failures limit the reportable calibration range to the high standard in the calibration curve. Any sample concentration that falls above the high calibration standard will be diluted to fall within the calibration range.
- 21.3.5 Internal Standard failure requires one or more of the following: five-fold dilution of the sample, correction of the problem, termination of analysis, recalibration of the instrument, and/or reanalysis of the affected samples depending on whether the failure is due to the samples or the instrumental drift.
- 21.3.6 Method blank results higher than the PQL and greater than 10% of any sample value in that batch that has concentrations above the PQL require that batch be redigested and reanalyzed. If the method blank results are less than -2x PQL there may be significant interference, calibration, or contamination problems with the sample, instrument, or calibration

- standards that must be resolved before the batch can be analyzed. For DoD QSM work, the absolute value must be less than ½ RL or less than 10% of the determined value of all samples in the batch. Al, Fe, Ca, Mg, Na, and K must be less than RL.
- 21.3.7 Matrix spikes, duplicates and spike duplicates are used only as indicators of method effectiveness on that sample and will not be used as acceptability criteria for the process, unless a special requirement of the client.
 - 21.3.8 LCS and/or LCS duplicate results outside of established acceptance limits require the batch to be redigested and reanalyzed.
 - 21.3.9 When analytical results suggest the presence of interference, one of the methods listed in Section 18.4.4.2 should be employed.
 - 21.3.10 The CRDL standard should be evaluated, but no action is required if the results fall outside of the 70-130% advisory window. For DoD QSM, the CRDL standard must be 80-120% of the true value or recalibration is required. For SW-846 Method 6020A, the CRDL standard must be 70-130% of the true value or recalibration is required. This also includes the low level continuing calibration verification standard analyzed every 10 samples. Sample results can be evaluated for reporting if they are at least 2x the CRDL for a given analyte.
- 21.4 Corrective actions taken for data not conforming to the requirements in Section 21.2 are stated in Section 21.3. If these corrective actions can be taken by the analyst prior to the acceptance of the data, then no nonconformance documentation is required. However, if these corrective actions include redigestion of the batch or sample, if the data have already been accepted, or if the corrective action requires an instrument service call, then a nonconformance and/or corrective action report should be completed. This report includes the date, person requesting the action, sample(s) or batch(s) affected, and action requested, all provided by the requester. The person taking the action will provide any pertinent comments, their signature, and the date the action is completed. The disposition of the nonconformance will then be verified by the Quality Systems specialist. These reports will be kept on file. Refer to GL-QS-E-002 for Documentation of Nonconformance Reporting and Dispositioning and Control of Nonconforming Items.
- 21.5 Analytical data are evaluated for conformance with the requirements stated in Section 21.2 by the analyst during and/or after the analysis, but before the data are entered into AlphaLIMS. Data may be accepted or rejected by the analyst at this point or by the data reviewer(s) as stated in Section 22.0.
- 22.0 DATA REVIEW, VALIDATION, AND APPROVAL PROCEDURES**
- 22.1 After samples are analyzed, the data must go through the review process before it can be reported out of the lab. The analyst who performed the analysis will review the raw data prior to uploading it into AlphaLIMS. The upload process may be handled by the analyst or by a data entry clerk.
 - 22.2 After the upload is complete, an AlphaLIMS data report is generated that will be passed (along with the batch sheet, data review checklist, and the raw data) to

- another analyst for a peer review. Discrepancies found in this review will be resolved before the batch data are passed and the status updated.
- 22.3 When this peer review is completed and all of the data are found to be acceptable, the reviewer signs and dates the batch data report and updates the status of the batch to done. If the reviewer determines that the reviewed data are not acceptable and requires additional work or correction, data are returned to the analyst or representative with an appropriate explanation.
- 22.4 Listed below are the data review responsibilities of the Analyst and Peer Reviewer:
- 22.4.1 Analyst Review is performed by the analyst who generated the data before the data are submitted for entry into AlphaLIMS. Analyst completes and attaches the run data cover sheet to the printout.
- 22.4.2 Peer Analyst Review is performed by an individual who did not perform the analysis but is familiar with the analytical method used and the reporting requirements. This person will review the complete data report after all data are entered, will ensure that any data entry corrections are made before data are approved, and will complete the reviewer portion of the data review check list. If the data do not meet the necessary quality requirements and need further analysis, they are returned to the analyst or representative.
- 22.5 A Third Review is performed for Data Packages (level 3 to level 6 CLP-Like) and/or by client requirement. This review is by the Metals Team Validator or other qualified person.
- 22.6 Specific items that are reviewed at each level include the following:
- 22.6.1 Analyst Review: Before data is submitted for entry into AlphaLIMS.
- 22.6.1.1 All analyses in the batch are completed.
- 22.6.1.2 Any corrections and comments on the data are properly initialed and dated.
- 22.6.1.3 Proper standard identification numbers appear on the runlog to ensure traceability from the data to original source standards.
- 22.6.1.4 All data are complete and accurate in the AlphaLIMS data report.
- 22.6.1.5 Data acceptance limit criteria identified in Section 21.2 are met or an explanation given.
- 22.6.2 Peer Analyst Review: Before data are submitted for further review or update:
- 22.6.2.1 All data are complete and accurate in the AlphaLIMS Data Report.
- 22.6.2.2 Any exceptions or shortcomings have been sufficiently explained or corrected.
- 22.6.2.3 Data are reported in the proper units or an explanation is given.
- 22.6.2.4 Prep factor and dilution calculations by AlphaLIMS are present in the data report and AlphaLIMS calculations are correct.

22.6.2.5 LCS and Spike Recoveries and RPD calculations by AlphaLIMS are correct, and the values are within control limits.

22.7 When the data review is completed and the data have been reported out of the lab, they are bound with the batching sheet and kept on file in the lab.

22.8 The complete data review process requires the use of the Prep Log Book, Prep data report, batch sheet, AlphaLIMS data report, raw instrument data, and runlog.

23.0 DATA REPORTING

23.1 To report data after the review process has been completed:

23.1.1 Enter the AlphaLIMS program through an available terminal and select DATA MENU, BATCH ITEMS, CHANGE BATCH STATUS.

23.1.2 Enter the Batch Number and "Submit."

23.1.3 Use the down arrow key to move the cursor down the new status column to the end. Change status from REVW to DONE and "Save."

24.0 RECORDS MANAGEMENT AND DOCUMENT CONTROL

24.1 Records of the instrumental analysis, operation, and maintenance are maintained as follows:

24.1.1 Run logs are an accurate chronology of what the instrument did during a specified period of time. The logs detail the standard name or sample number for each standardization or analysis, the analyst identification, and the date and time of each analysis or standardization. This information is periodically retrieved from the instrument data files, printed in chronological sequence, and maintained as documentation of the sequence of events and as a reference for the raw data.

24.1.2 Extraction/Digestion Logs are maintained in accordance with the established procedures in the areas where these processes are performed.

24.1.3 Instrument Maintenance Logs are chronological representations of all maintenance activities involving the instrument operation. This record is kept in bound composition books and consists of the details of the action, who performed it, when it was performed, and when instrument was returned to operation.

24.1.4 Batch Sheets accompany the batch of samples through digestion and analysis and then accompany the raw analytical data through data entry and data review until the batch status is changed to "DONE" in the laboratory. This record is maintained in the Metals group files.

24.1.5 Batch Data Reports are generated to be used in the peer data review. After the batch is reviewed, corrected if necessary, and updated to "DONE," the batch data report is stored with the Batch sheet.

24.1.6 Raw Instrument Data are generated as the analyses are performed and from AlphaLIMS after the data are entered and attached to the batch sheet to be used in the data review process and kept in the Metals group files.

25.0 LABORATORY WASTE HANDLING AND DISPOSAL: SAMPLES, EXTRACTS, DIGESTATES, AND REAGENTS

- 25.1 Standard solutions that must be disposed are taken to the Waste Disposal coordinator for disposal in accordance with Laboratory Waste Management Plan, GL-LB-G-001.
- 25.2 Sample digestates are stored in the lab for a specified period of time following analysis. At this time, they are composited into a waste container that is picked up by the Waste Management Technician for proper disposal.
- 25.3 Radioactive Waste:
 - 25.3.1 Samples returned to sample storage
 - 25.3.2 Drain waste collected in the radioactive waste carboy is dumped when full into the appropriate 55 gallon drum sitting outside the ICPMS laboratory. Ultimate disposal of liquid radioactive waste done by waste management department.
 - 25.3.3 Implements, vials, gloves, etc., are wrapped and labeled with radioactive tape and placed in the radioactive waste container in high bay area.
 - 25.3.4 Expired Standard Solutions: Refer to Section 25.1.

26.0 REFERENCES

- 26.1 Perkin Elmer ELAN 9000 Hardware Guide
- 26.2 Perkin Elmer ELAN 6100 Hardware Guide
- 26.3 Perkin Elmer ELAN 9000 Software Guide
- 26.4 Perkin Elmer ELAN 6000/6100 Software Guide
- 26.5 Test Methods for Evaluating Solid Waste: Laboratory Manual Physical/Chemical Methods. Volume 1A, USEPA SW-846, Third Edition, Revision 2, September 1994.
 - 26.5.1 Method 6020A, "Inductively Coupled Plasma – Mass Spectrometry," Revision 1, February 2007.
 - 26.5.2 Method 6020, "Inductively Coupled Plasma – Mass Spectrometry," Revision 0, September 1994.
 - 26.5.3 Method 3005A, "Acid Digestion of Waters for Total Recoverable or Dissolved Metals for Analysis by FLAA or ICP Spectroscopy", Revision 1, July 1992.
 - 26.5.4 Method 3010A, "Acid Digestion of Aqueous Samples and Extracts for Total Metals for Analysis by FLAA or ICP Spectroscopy," Revision 1, July 1992.
 - 26.5.5 Method 3050B, "Acid Digestion of Sediments, Sludges, and Soils", Revision 2, December 1996.
- 26.6 USEPA Method 200.8, "Determination of Trace Elements in Waters and Wastes by Inductively Coupled Plasma – Mass Spectrometry," Revision 5.4, May 1994.
- 26.7 2001 Annual Book of ASTM Standard, D4698-92 "Standard Practice for Total Digestion of Sediment Sampler for Chemical Analysis of Various Metals."

- 26.8 Conversion Constants for Uranium Isotopes, Dr. Robert Litman, April 2009.
- 26.9 Department of Defense Quality Systems Manual for Environmental Laboratories, Version 4.2, October 25, 2010.
- 26.10 U.S. Department of Energy, Quality Systems for Analytical Services (DOE QSAS). Rev 2.6, November 2010.

27.0 HISTORY

Revision 25: Clarifications on analyte lists, internal standards, and method requirements.

Revision 24: Updated section on sample preparation techniques to include procedure GL-MA-E-016

Revision 23: Updated sections 15.2, 21.2.5 and Appendix 3 to four internal standards. Removed outdated SOP references.

Revision 22: Editorial corrections only.

APPENDIX 1: FREQUENCY OF QUALITY CONTROL ACTIVITIES

(For illustrative purposes only)

Frequency of Quality Control Activities

Method/ Frequency	SW-846 6020	EPA 200.8	SW-846 6020A
Calibration Std readbacks	Not required	Not required	Not required
ICV	Per calibration	Per calibration	Per calibration
ICB	Per calibration	Per calibration	Per calibration
PQL	Per calibration	Per calibration	Per calibration and at end of each analytical batch.
ICSA	Per calibration	Per calibration	Per calibration
ICSAB	Per calibration; every 12 hours after	Per calibration	Per calibration; every 12 hours after
Linear Range Standard (LRS)	Per calibration, if applicable	Per calibration, if applicable	Per calibration, if applicable
CCV	Every 10 instrument runs	Every 10 instrument runs	Every 10 instrument runs
CCB	Every 10 instrument runs	Every 10 instrument runs	Every 10 instrument runs
Method Blank	5% or per batch	5% or per batch	5% or per batch
LCS – liquid LCS – soil	5% or per batch	5% or per batch	5% or per batch
Matrix Spikes	5% or per request	10% or per request	5% or per request
Sample Duplicates	5% or per request	5% or per request	5% or per request
Serial Dilutions	5% or per request	5% or per request	5% or per request
Matrix Spike Duplicates	5% or per request	10% or per request	5% or per request
Post-Digestion Spikes	5% or per request	10% or per request	5% or per request

APPENDIX 2: ACCEPTANCE LIMITS

Method/ Acceptance Criteria	SW-846 6020	EPA 200.8	DoD QSM	SW-846 6020A
Calibration Std readbacks	Not required	Not required	Not required	Not required
ICV	90% - 110%	90% - 110%	90%-110%	90% - 110%
ICB	< absolute value of RL	< absolute value of RL	< LOD	< absolute value of RL
PQL/CRI (CLP)	70% - 130% advisory limits only	70% - 130% advisory limits only	80%-120% or investigate and recalibrate	70% - 130% or investigate and recalibrate
ICSA	80-120% for major components; $\pm 2x$ RL for non-spiked	80-120% for major components; $\pm 2x$ RL for non-spiked	80%-120% for major compounds; \pm LOD for non-spiked	80-120% for major components; $\pm 2x$ RL for non-spiked
ICSAB	80%-120%	80%-120% (may be requested)	80%-120%	80%-120%
Linear Range Standard (LRS)	90%-110%, or up to the high calibration standard	90%-110%, or up to the high calibration standard	90%-110%, or up to the high calibration standard	90%-110%, or up to the high calibration standard
CCV	90% - 110%	90% - 110%	90%-110%	90% - 110%
CCB	\pm RDL	\pm RDL	< LOD	\pm RDL
Method Blank	\pm RDL	\pm RDL	$\pm \frac{1}{2}$ RL except for Al, Fe, Mg, Ca, Na, and K	\pm RDL
LCS - liquid LCS - soil	80% - 120% current SPC limits	85% - 115% current SPC limits	80%-120%	80% - 120% current SPC limits
Matrix Spikes	75% - 125%, when applicable	75% - 125%, when applicable	80%-120%	75% - 125%, when applicable
Sample Duplicates	0% - 20% when greater than 5X RL, \pm RL when less than 5X RL	0% - 20% when greater than 5X RL, \pm RL when less than 5X RL	0% - 20% when greater than 5X RL, \pm RL when less than 5X RL	0% - 20% when greater than 5X RL, \pm RL when less than 5X RL
Serial Dilutions	0% - 10% of initial raw value, when applicable	0% - 10% of initial raw value, when applicable	0% - 10% of initial raw value, when applicable (> 50x LOQ)	0% - 10% of initial raw value, when applicable
Post-digestion spikes	75%-125%, when applicable	75%-125%, when applicable	75%-125%	80%-120%, when applicable
Internal Standards	30%-120%, samples 80%-120% for ICB, ICV, CCV, CCB	60%-125% for all	30%-120%	70%-130%, for all
Matrix Spike Duplicate	0% - 20% when greater than 5X RL, \pm RL when less than 5X RL	0% - 20% when greater than 5X RL, \pm RL when less than 5X RL	0% - 20% when greater than 5X RL, \pm RL when less than 5X RL	0% - 20% when greater than 5X RL, \pm RL when less than 5X RL

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Determination of Metals by ICP-MS

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Revision 25 Effective March 2013

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APPENDIX 2-Cont'd

Method/ Acceptance Criteria	SW-846 6020	EPA 200.8	DoD QSM	SW-846 6020A
Sample Duplicates RER activity	$\leq 3\%$	$\leq 3\%$	$\leq 3\%$	$\leq 3\%$

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APPENDIX 3: INTERNAL STANDARDS WITH ASSOCIATED ANALYTES/ISOTOPES

IS ⁴⁵ Sc	IS ⁷⁴ Ge	IS ¹¹⁵ In	IS ¹⁷⁵ Lu ¹⁸¹ Ta
⁷ Li	⁶⁶ Zn	⁸⁸ Sr	¹³³ Cs
⁹ Be	⁶⁷ Zn (calc)	⁹⁰ Zr	¹³⁵ Ba (calc)
¹¹ B	⁶⁸ Zn (calc)	⁹⁸ Mo	¹³⁷ Ba
²³ Na	⁷⁵ As	¹⁰⁵ Pd	¹³⁹ La
²⁴ Mg	⁷⁷ Se (calc)	¹⁰⁷ Ag	¹⁴⁰ Ce
²⁷ Al	⁸² Se	¹¹¹ Cd	¹⁴¹ Pr
³¹ P	⁸³ Kr (calc)	¹¹⁴ Cd (calc)	¹⁴² Nd
³⁹ K		¹²⁰ Sn	¹⁵² Sm
⁴³ Ca		¹²¹ Sb	¹⁵³ Eu
⁴⁷ Ti		¹²³ Sb (calc)	¹⁵⁸ Gd
⁵² Cr			¹⁵⁹ Tb
⁵³ Cr (calc)			¹⁷⁸ Hf
⁵⁵ Mn			¹⁸⁷ Re
⁵⁷ Fe			¹⁹⁵ Pt
⁵⁹ Co			¹⁹⁷ Au
⁶⁰ Ni			²⁰⁵ Tl
⁶³ Cu (calc)			²⁰⁸ Pb
⁶⁵ Cu			²⁰⁹ Bi
			²³² Th
			²³³ U
			²³⁴ U
			²³⁵ U
			²³⁶ U
			²³⁸ U

*(calc) – isotope used in calculations

**APPENDIX 4: RADIOCHEMISTRY CONVERSION CALCULATIONS FOR
URANIUM ISOTOPES**Conversion for liquids ($\mu\text{g/L} \times \text{CF} = \text{pCi/L}$)

$$^{233}\text{U} (\mu\text{g/L to pCi/L}) = 9640.6$$

$$^{234}\text{U} (\mu\text{g/L to pCi/L}) = 6224.9$$

$$^{235}\text{U} (\mu\text{g/L to pCi/L}) = 2.1615$$

$$^{236}\text{U} (\mu\text{g/L to pCi/L}) = 64.698$$

$$^{238}\text{U} (\mu\text{g/L to pCi/L}) = 0.33627$$

Conversion for solids ($\text{mg/kg} \times \text{CF} = \text{pCi/g}$)

$$^{233}\text{U} (\text{mg/kg to pCi/g}) = 9640.6$$

$$^{234}\text{U} (\text{mg/kg to pCi/g}) = 6224.9$$

$$^{235}\text{U} (\text{mg/kg to pCi/g}) = 2.1615$$

$$^{236}\text{U} (\text{mg/kg to pCi/g}) = 64.698$$

$$^{238}\text{U} (\text{mg/kg to pCi/g}) = 0.33627$$

30-Sep-2013

SOP Effective 8/93
Revision 22 Effective May 2013

Acid Digestion of Sediments, Sludges, and Soils

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VERIFY THE VALIDITY OF THIS SOP EACH DAY IN USE

STANDARD OPERATING PROCEDURE

FOR

ACID DIGESTION OF

SEDIMENTS, SLUDGES, AND SOILS

(GL-MA-E-009 REVISION 22)

APPLICABLE TO METHODS:
EPA SW-846 3050B Modified

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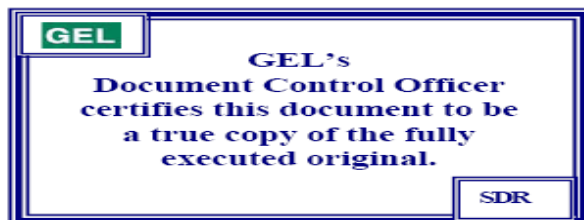


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1.0 STANDARD OPERATING PROCEDURE FOR ACID DIGESTION OF SEDIMENTS, SLUDGES, AND SOILS**2.0 METHOD CODE**

2.1 EPA SW-846 3050B Modified

3.0 METHOD OBJECTIVE/PURPOSE

To describe the manner in which sediments, sludges, and soils for Inductively Coupled Plasma (ICP) and Inductively Coupled Plasma-Mass Spectrometry (ICP-MS) analysis are digested by EPA SW-846 Method 3050B Modified. Samples digested by this procedure are applicable for analysis by SW-846 methods 6010B and 6020. Included in this standard operating procedure is guidance for paint and jewelry preparation.

4.0 METHOD SUMMARY

A representative portion of sample is digested with nitric acid and hydrogen peroxide. This digestate is then refluxed with hydrochloric acid for ICP analysis only. Samples prepared by this method may be analyzed for all the listed metals. Other metals may be analyzed if they pass control standard criteria:

Aluminum	Copper	Palladium	Terbium
Antimony	Europium	Phosphorus	Tin
Arsenic	Gold	Platinum	Thorium
Barium	Hafnium	Praseodymium	Titanium
Beryllium	Iron	Rhenium	Thallium
Bismuth	Lanthanum	Ruthenium	Tungsten
Boron	Lead	Samarium	Uranium
Boron-10	Lithium	Selenium	Uranium-233
Cadmium	Magnesium	Silica	Uranium-234
Calcium	Manganese	Silicon	Uranium-235
Cerium	Molybdenum	Sulfur	Uranium-236
Cesium	Neodymium	Sodium	Uranium-238
Chromium	Potassium	Silver	Vanadium
Cobalt	Nickel	Strontium	Zinc
			Zirconium

This method is not a “total” digestion technique for most samples. It is a very strong acid digestion that will dissolve all elements that could become “environmentally available” by design; elements bound in silicate structures (boron, silicon, silica) are not normally dissolved by this procedure as they are not usually mobile in the environment.

5.0 APPLICABLE MATRICES

5.1 Soils

5.2 Sludges

- 5.3 Sediments
- 5.4 Solid debris/powders
- 5.5 Heavy oils
- 5.6 Filters
- 5.7 Paints
- 5.8 Metal jewelry

6.0 HOLD TIME

Holding time is 180 days from the time and date of collection until the start of analysis unless otherwise specified by contract.

7.0 SAMPLE CONTAINER/PRESERVATION/COLLECTION/STORAGE REQUIREMENTS

Solid samples are not preserved but should be stored at 0°- 6° C.

8.0 INTERFERENCES

There are rarely any interferences with this digestion. If any are encountered, consult the group leader or quality officer before continuing.

9.0 PERFORMANCE CHARACTERISTICS

Method detection limits (MDLs) are performed annually and method detection limit verification (MDLVs) are performed quarterly.

10.0 DEFINITIONS

- 10.1 Blank: Type I water that has been taken through the digestion process. The blank is used to determine the amount of background contamination.
- 10.2 Laboratory Control Sample (LCS): A certified reference material that has been taken through the digestion process. The LCS is used to determine digestion accuracy and to determine if the digestion process is in control.
- 10.3 Laboratory Control Sample Duplicate (LCS DUP): A duplicate of the LCS. The LCS DUP is used to determine reproducibility and to indicate precision.
- 10.4 Matrix Spike (MS): A sample that has added to it a known amount of solution containing known concentrations of analytes. The MS is used to determine the presence or absence of interferences and matrix effects in the digested sample.
- 10.5 Matrix Spike Duplicate (MSD): A duplicate of the MS. The MSD indicates reproducibility.
- 10.6 Sample Duplicate (DUP): A duplicate of a sample. The DUP indicates reproducibility.
- 10.7 AlphaLIMS: The Laboratory Information Management System used at GEL.
- 10.8 National Institute of Standards and Technology (NIST): For the purpose of this method, the national scientific body responsible for the standardization and acceptability of analyte solutions.

11.0 ANALYST VERIFICATION

Before a technician/analyst is allowed to analyze samples without supervision, he or she is trained by qualified personnel and is required to successfully analyze a proficiency sample. Training records are maintained as quality records (Refer to GL-QS-E-008).

12.0 DOCUMENTATION OF DATA

Sample preparation data are recorded in AlphaLIMS.

13.0 SAFETY, HEALTH, AND ENVIRONMENTAL HAZARDS

WARNING

CONCENTRATED HYDROCHLORIC ACID AND NITRIC ACID ARE EXTREMELY CORROSIVE AND CAN CAUSE SEVERE BURNS TO THE SKIN.

CONCENTRATED 30% HYDROGEN PEROXIDE IS A VIOLENT OXIDIZER. KEEP AWAY FROM OPEN FLAMES, AND RINSE WITH WATER IF SKIN CONTACT OCCURS.

- 13.1 Wear eye protection with side shields while performing procedures in the lab.
- 13.2 Treat all chemicals and samples as potential health hazards, and reduce exposure to these chemicals to the lowest level possible. GEL maintains a current awareness file of OSHA regulations regarding the safe handling of the chemicals. A reference file of Material Safety Data Sheets (MSDS) and individual client sample MSDSs are also maintained.
- 13.3 Personal protective equipment
 - 13.3.1 Disposable gloves are worn and changed frequently when working with acids, glassware, or samples. Dirty gloves pose a contamination hazard to the samples. Gloves that have holes can be dangerous to the wearer by allowing acids and toxic metals to come in contact with skin.
 - 13.3.2 Hood doors are pulled down partially while digesting samples. Acidified samples can splash and pop as they are being heated.
 - 13.3.3 To protect clothes and skin from exposure to corrosive material, wear a lab jacket.
- 13.4 Prior to handling radioactive samples, analysts must have had radiation safety training and must understand their full responsibilities in radioactive sample handling. Some general guidelines follow:
 - 13.4.1 Wear plastic apron over lab coat when working with radioactive samples.
 - 13.4.2 Protect counter tops with counter paper, or work from radioactive sample handling trays.
 - 13.4.3 Prohibit admittance to immediate work area.
 - 13.4.4 Post signs indicating radioactive samples are in the area.
 - 13.4.5 Take swipes of the counter tops upon completion of work. Deliver those swipes to the designated swipe count box.
 - 13.4.6 Segregate radioactive wastes. Radioactive waste containers are obtained from the Waste Management.

- 13.5 All samples, chemicals, extracts, and extraction residues must be transferred, delivered, and disposed of safely according to all related SOPs.
 - 13.5.1 Segregate solid wastes from liquid wastes in the satellite area containers.
 - 13.5.2 Segregate oil wastes from water-soluble wastes in the satellite area containers.
- 13.6 In the event of an accident or medical emergency, call for help immediately. When time and safety permit, an accident report form should be completed and turned in to the safety committee.
- 13.7 Fire escape routes are posted in the lab, and all personnel should be familiar with them. In addition, fire safety equipment such as fire extinguishers is located in the lab. Training is available on the proper operation of this equipment.

14.0 SAMPLE RECEIPT FOR ANALYSIS

- 14.1 The analyst/technician submits the list of samples needed to the sample custodian group. The sample custodian removes the appropriate sample from the cooler and scans it using the barcode scanner to the appropriate area of the lab. The analyst then takes custody of the samples and scans them to the sample batch. The samples are now ready to be prepared or analyzed.
- 14.2 Analysts/technicians are responsible for retrieving their own samples when the sample custodian is unavailable.

15.0 INSTRUMENT/EQUIPMENT/GLASSWARE

- 15.1 Equipment
 - 15.1.1 Air displacement pipettes
 - 15.1.1.1 5-10 mL with disposable tips
 - 15.1.1.2 0.5-5 mL with disposable tips
 - 15.1.1.3 100-1000 µL with disposable tips
 - 15.1.1.4 10-100 µL with disposable tips
 - 15.1.2 Environmental Express hot blocks or equivalent
 - 15.1.3 Analytical balance capable of reading to three decimal places
 - 15.1.4 Certified disposable 50 mL digestion tubes (polypropylene)
 - 15.1.5 Ribbed disposable watch glasses (polypropylene)
 - 15.1.6 Water resistant lab markers
 - 15.1.7 Styrofoam trays to handle up to 25 digestion tubes
 - 15.1.8 500 mL Nalgene squirt bottle
 - 15.1.9 Teflon chips
 - 15.1.10 1-inch white laboratory tape
 - 15.1.11 Borosilicate beakers (various sizes)
 - 15.1.12 Borosilicate watch glasses (various sizes)

16.0 REAGENTS

- 16.1 Nitric acid (HNO_3), concentrated high purity grade 70% nitric acid
- 16.2 Hydrochloric acid (HCl), concentrated high purity grade 37% hydrochloric acid
- 16.3 Hydrogen peroxide (H_2O_2), concentrated 30% hydrogen peroxide
- 16.4 Type I deionized (DI) water.
- 16.5 Multi-element spiking solutions are purchased from NIST-traceable vendors.

17.0 PREPARATION OF SAMPLES

A batch consists of samples of the same matrix and quality control (QC) samples that are digested together. Each of the quality control samples listed in steps 10.1 through 10.6 must be included in each batch at the frequency listed or as per client request. The blank, LCS, and/or LCS DUP are digested at a frequency of one in 20 or per batch, whichever is more frequent. The MS, MSD, and/or DUP are digested at a frequency of one in 20 or per batch, whichever is more frequent, or per specified client/program requirements.

17.1 Glassware preparation:

- 17.1.1 Glassware that has been cleaned according to GL-LB-E-003 for Glassware Preparation is soaked in a water and acid mixture for at least 30 minutes.
- 17.1.2 After soaking, the glassware is rinsed with copious quantities of Type I water and then inverted over clean, absorbent paper or onto a rack for drying.

17.2 Label the Teflon beakers or centrifuge tube with the sample numbers in the batch. If centrifuge tube is to be used for measuring initial and final volumes it must be calibrated before usage. Refer to GL-LB-E-026 for centrifuge tube testing procedure.

17.3 Refer to GL-LB-E-029 for subsampling instructions. Mix the sample to achieve homogeneity. Weigh approximately 0.5 g of sample. Transfer the weighed sample to the appropriately labeled Teflon beaker or centrifuge tube.

- 17.3.1 Sample aliquots should not be taken from the top of an unmixed sample because large particles tend to rise in solid matrixes and heavy materials tend to sink in liquid matrixes.
- 17.3.2 Powdered samples may be homogenized by gently rocking the sample side to side. Then a representative aliquot may be taken from the center of the powder.
- 17.3.3 Other matrixes must be stirred, turned or mixed before sampling.

17.4 Quality control samples are prepared prior to digestion.

- 17.4.1 The beaker or tube to be used for the blank, MS, MSD, and/or DUP, LCS, and/or LCS DUP is labeled.
- 17.4.2 Weigh approximately 0.5 g of sample and transfer to the MS, MSD, and/or DUP beaker or tube.

The MS, MSD, LCS, and/or LCS DUP are spiked with known amounts of spiking solution.

- 17.4.3 Select a LCS. The LCS is purchased for an outside vendor and comes with a certificate of certified values and recovery ranges. The LCS is logged into the AlphaLIMS system for traceability and for the use of nominal calculations. For analytes not certified in the Solid Reference Material (SRM), a series of 20 preparations and analyses are conducted and an average concentration is determined. A value of 3 times the standard deviation is used as control limits for the LCS recovery for these analytes. Mix the LCS to achieve homogeneity. Weigh approximately 0.5 g of the sample and transfer to the LCS and/or LCS DUP Teflon beaker or centrifuge tube. For non-soil solid samples, a liquid LCS is used in combination with approximately 0.5 g of Teflon chips.
 - 17.4.4 The blank beaker or tube is labeled and no water, spike, or sample is added to it. Approximately 0.5 g of Teflon chips is used.
 - 17.5 If the samples are being prepared for ICP-MS analysis:
 - 17.5.1 Add 2.5 mL nitric acid and Type I DI water to the samples and quality control samples.
 - 17.5.2 Gently swirl the sample and acid mixture.
 - 17.5.3 Cover the sample with a watch glass and heat the sample on a hot plate/block to $95^{\circ} \pm 5^{\circ} \text{C}$. Reflux the sample for 10 to 15 minutes.
 - 17.5.4 Remove the sample from the hot plate or block and allow the sample to cool.
 - 17.5.5 Add 2.5 mL of concentrated nitric acid, replace the watch glass, and reflux for 30 minutes. If brown fumes are generated indicating oxidation of the sample by nitric acid, repeat step 17.5.5 over and over until no brown fumes are given off by the sample.
 - 17.5.6 Using a ribbed watch glass or vapor recovery system, allow the solution to evaporate to approximately 0.5 mL without boiling, or heat for 2 hours.
 - 17.5.6.1 Remove the sample from the hot plate or block and allow the sample to cool.
 - 17.5.6.2 Add 1.5 mL of hydrogen peroxide and 1.0 mL of Type I water. Return the sample to the hot plate or block and allow the peroxide reaction to occur. Continue to add hydrogen peroxide to the sample until the effervescence subsides. Do not add more than 5 mL hydrogen peroxide.
 - 17.5.6.3 Cover the sample with a ribbed watchglass, heating the acid-peroxide digestate until the volume is reduced to approximately 2.5 mL, or heat at $95^{\circ} \pm 5^{\circ} \text{C}$ without boiling for 2 hours.
 - 17.5.6.4 Do not allow the sample to evaporate to dryness.
 - 17.5.6.5 Remove the sample from step 17.5.6.3 from the hot plate or block.

- 17.5.6.6 Allow the sample to cool.
- 17.5.6.7 Dilute the sample to 50 mL with Type I water.
- 17.5.6.8 Cap and shake the sample.
- 17.5.6.9 Filter each sample with a 2.0 µm pore size plunger type filter (PTF grade) or allow to settle overnight.
- 17.5.6.10 Organize the samples in a storage container, and label the container with the batch number of the sample group.
- 17.6 If the samples are being prepared ICP analysis:
 - 17.6.1 Add 1.25 mL nitric acid and 10 mL hydrochloric acid to the samples and quality control samples.
 - 17.6.2 Gently swirl to mix.
 - 17.6.3 Cover the sample with a watch glass and heat the sample on a hotplate/block to 95° ± 5° C. Reflux the sample for 30 minutes.
 - 17.6.4 Remove the sample from the hotplate/block and allow to cool.
 - 17.6.5 Dilute the sample to 50 mL with Type I water.
 - 17.6.6 Cap and shake the sample.
 - 17.6.7 Filter each sample with 2.0 µm pore size plunger type filter (PTF grade) or allow to sit overnight.
 - 17.6.8 Organize the samples in a storage container, and label the container with the batch number of the sample group.
- 17.7 If the sample contains particulate material that could clog the nebulizer, you may filter or centrifuge the sample if necessary.
- 17.8 Be advised that filtration is a common cause of contamination. If a sample is filtered, any QC associated with the sample must also be filtered. Additionally, if any sample in the batch is filtered the method blank and laboratory control sample must also be filtered.
- 17.9 Filters may be prepared via this method. If the filters are small enough to fit inside the 50 mL digestion tubes, they can be treated as any solid prep materials. If the filters are too big to undergo adequate digestion using the 50 mL digestion tube, a borosilicate beaker will need to be used. All reagents and standards will need to be adjusted for any extra volumes needed. All filter analyses should be discussed and the process verified with the group/team leader prior to digestion. The group leader or project manager may have to contact the client to get the full description of what is required.

18.0 PREPARATION OF STANDARDS

Documentation of standards and their preparation is maintained in AlphaLIMS in accordance with GL-LB-E-007 for Laboratory Standards Documentation.

19.0 INSTRUMENT/EQUIPMENT START-UP PROCEDURE

Hot plates/blocks are allowed to come to the proper temperature before digestions are started. The temperatures are monitored before and after a daily digestion session.

20.0 QUALITY CONTROL (QC) REQUIREMENTS

20.1 Frequency of QC

20.1.1 A matrix spike (MS) and a matrix spike duplicate (MSD) or a sample duplicate (DUP) and a matrix spike are prepped for every batch of ≤ 20 samples

20.1.2 A method blank (MB) and a laboratory control standard (LCS) are prepped for every batch of ≤ 20 samples. A laboratory control standard duplicate (LCSD) is prepared if matrix QC is unavailable or upon client request.

20.2 Makeup of QC Samples

20.2.1 Sample duplicate (DUP) is a separate aliquot taken through the prep process exactly the same as the original sample.

20.2.2 Matrix spike and/or matrix spike duplicate is a separate aliquot of the sample to which appropriate spike volumes and solutions are added. The ID numbers and volumes of the spikes are recorded in the prep logbook.

20.2.3 The method blank (MB) is a reagent blank taken through the same prep process as the samples. Teflon chips are used to approximate matrix weights of 0.5 g.

20.2.4 The laboratory control standard (LCS) is a standard performed two different ways. For DOE-ALB clients, a purchased SRM is used at approximately 0.5 g and is taken through the same process as the samples. For all other clients, Teflon chips weighted to approximately 0.5 g are used. The chips and acid solution is spiked with the appropriate spike volumes and solutions. The ID number and volumes of the spikes are recorded in the prep logbook.

20.3 Handling Out-Of-Control Situations

If sample reactions cause popping or splattering of the digestate, discontinue the prep and contact team leader or group leader.

21.0 RUN SEQUENCE

Not applicable

22.0 PROCEDURE

Refer to section 17.0, Preparation of Samples

23.0 INSTRUMENT/EQUIPMENT SHUT-DOWN PROCEDURE

Before turning off the hotplate/blocks at the end of the day, a final monitoring temperature is recorded for each plate/block that was utilized.

24.0 METHOD VARIATION

- 24.1 This procedure deviates from method 3050B in that sample volumes are half the method recommendations.
- 24.2 The ICP procedure references a modified 3050B section 7.5 procedure. The modification eliminates the use of the Whatman 41 filters, thus eliminating contamination of common minerals.

25.0 DATA REVIEW, VALIDATION, AND APPROVAL PROCEDURE

- 25.1 Upon completion of batch preparation, digestion data shall be entered into the AlphaLIMS Prep Logbook (refer to Appendix 1) following the guidelines in GL-LB-E-008 for Basic Requirements for the Use and Maintenance of Laboratory Notebooks, Logbooks, Forms, and Other Recordkeeping Devices.
- 25.2 Data to be entered into the electronic logbook include analyst name, prep data and time, initial volume or weight with units, and final volume with units.
- 25.3 Standards and reagents may also be entered into the logbook and fall under the guidelines of GL-LB-E-015 for Control of Laboratory Standards and GL-LB-E-007 for Laboratory Standards Documentation.
- 25.4 Upon entry of prep data, obtain a printout of the logbook. The analyst listed on the logbook should sign and date the page near their printed initials. The logbook page is kept with the samples with which it is associated.
- 25.5 The entry of correct prep data is peer reviewed (correct dates, times, weights, volumes, SOP/revision, spikes, spike amounts, and reagent information, etc.) Once data are reviewed, the batch is statused to DONE in AlphaLIMS, the logbook is signed and dated by the reviewer, and the batch is ready for analysis. A copy of the prep logbook sheet is kept in the metals prep lab and is bound and given a control number when sufficient numbers of sheets are collected.

26.0 RECORDS MANAGEMENT

Records generated as a result of this procedure are maintained as quality documents in accordance with GL-QS-E-008 for Quality Records Management and Disposition.

27.0 LABORATORY WASTE

For the proper disposal of sample and reagent wastes from this procedure, refer to the Laboratory Waste Management Plan, GL-LB-G-001.

28.0 REFERENCES

- 28.1 Test Method for Evaluating Solid Waste; Laboratory Manual Physical/ Chemical Methods, Method 3050B, "Acid Digestion of Sediments, Sludges, and Soils," Revision 2, December 1996.
- 28.2 1992 Annual Book of ASTM Standards, Standard D1193-91, "Standard Specification for Reagent Water."
- 28.3 16 CFR Part 1303

30-Sep-2013

Acid Digestion of Sediments, Sludges, and Soils

SOP Effective 8/93

Revision 22 Effective May 2013

GL-MA-E-009 Rev 22

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29.0 HISTORY

Revision 21: Updated performance characteristics section; MDLs are performed annually and MDLVs are performed quarterly.

Revision 22: Removed 16 CFR Part 1303 reference.

Revision 22: Clarified DI water. Updated metals list of elements in method summary section.

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30-Sep-2013

Acid Digestion of Sediments, Sludges, and Soils

SOP Effective 8/93
Revision 22 Effective May 2013

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APPENDIX 1: SAMPLE PREP LOGBOOK

(For illustrative purposes only)

Prep LogBook								
Analyte:	AsM	Type	Sample Id	Loc. Id	Spike Amount	Spike Units		
Batch:	10607	LCS	10000231S2	S385	.25	ml.		
Prep Date:	07-FEB-2000 14:00	LCS	10000231S2	S386	.25	ml.		
Lab SOP:	GL-MA-E-013	MS	10000231S5	S385	.25	ml.		
		MS	10000231S5	S386	.25	ml.		
		MSD	10000231S4	S385	.25	ml.		
		MSD	10000231S4	S386	.25	ml.		
Type	Sample Id	Parent Sample	Method	Initial Wt.	Pipet Volume	Prep Factor	Comments	Matrix
MB	10000231S1		200.2/200.7 Full List For QC	50ml.	50ml.	1		Winter
SAMPLE	21426001		200.2/200.7 Selenium	50ml.	50ml.	1		Winter Winter
LCS	10000231S2		200.2/200.7 Full List For QC	50ml.	50ml.	1		Winter
SAMPLE	21426002		200.2/200.7 Selenium	50ml.	50ml.	1		Winter Winter
SAMPLE	21426003		200.2/200.7 Selenium	50ml.	50ml.	1		Winter Winter
SAMPLE	21426004		200.2/200.7 Selenium	50ml.	50ml.	1		Winter Winter
SPLIT	10000231S3	21426004	200.2/200.7 Full List For QC	50ml.	50ml.	1		Winter
MSD	10000231S4	21426004	200.2/200.7 Full List For QC	50ml.	50ml.	1		Winter
MS	10000231S5	21426004	200.2/200.7 Full List For QC	50ml.	50ml.	1		Winter

General Engineering Laboratories

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30-Sep-2013

SOP Effective December 1999
Revision 19 Effective January 2013

Gamma Spectroscopy System Operation

GL-RAD-I-001 Rev 19
Page 1 of 11

VERIFY THE VALIDITY OF THIS SOP EACH DAY IN USE

STANDARD OPERATING PROCEDURE

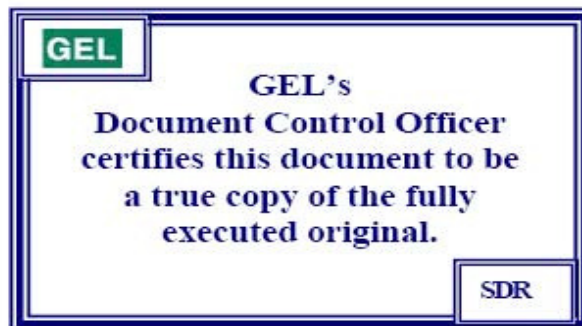
FOR

GAMMA SPECTROSCOPY SYSTEM OPERATION

(GL-RAD-I-001 REVISION 19)

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1.0 STANDARD OPERATING PROCEDURE FOR GAMMA SPECTROSCOPY SYSTEM OPERATION**2.0 METHOD OBJECTIVE, PURPOSE, CODE AND SUMMARY**

- 2.1 This standard operating procedure provides the necessary instructions to conduct the analysis for gamma isotopes using the Gamma Spectroscopy System.
- 2.2 Gamma emitting isotopes within the sample matrix are identified and quantified using gamma spectrometry. A sample aliquot is placed in a calibrated geometry and placed in the detector chamber. The germanium crystal therein produces a corresponding electrical pulse for the gamma photons that interact with the detector. The cumulative pulses are analyzed using software capable of quantifying gamma-emitting isotopes from the spectral data.

3.0 APPLICABLE MATRIX OR MATRICES

This is a nondestructive test for the measurement of gamma emitting isotopes in all matrices for which there is an available calibration standard.

4.0 METHOD SCOPE, APPLICABILITY AND DETECTION LIMIT

- 4.1 The aliquoted sample activity or sample position should be adjusted so that the detector system dead time remains less than 15%.
- 4.2 Method Detectable Activity: The MDA is based upon sample volume, instrument background, detector efficiency, count time and other statistical factors, as well as specific isotopic values such as abundance and half-life.

5.0 METHOD VARIATIONS

Not applicable

6.0 DEFINITIONS

- 6.1 Abundance: The combination of the isotopic decay branching ratio and the expected gamma emissions per disintegration of an isotope at a particular energy.
- 6.2 Key Line: The line chosen by the builder of the library to be the prominent line of the isotope. This line is used for the purposes of calculating activity, error and MDA.
- 6.3 AlphaLIMS: The Laboratory Information Management System used to store and report data.
- 6.4 National Institute of Standards and Technology (NIST): For the purpose of this method, the national scientific body responsible for the standardization and acceptability of analyte solutions.

7.0 INTERFERENCES/LIMITATIONS

- 7.1 Some gamma isotopes emit gamma lines that may overlap with those from other isotopes. If the energies of the two isotopes are within the energy tolerance setting, the peaks may not be resolvable and may give a positive bias to the result. This problem is minimized by careful review of the peak search.

8.0 SAFETY PRECAUTIONS AND WARNINGS

Follow safety precautions as outlined in GL-LB-N-001 for the Safety, Health and Chemical Hygiene Plan.

9.0 APPARATUS, EQUIPMENT AND INSTRUMENTATION

9.1 Apparatus and Equipment

- 9.1.1 Compaq/DEC Alpha Station with OpenVMS
- 9.1.2 Canberra Genie-ESP Application Software
- 9.1.3 High purity germanium detector
- 9.1.4 Pulse processing electronics

10.0 REAGENTS AND STANDARDS

10.1 Standards

- 10.1.1 NIST traceable mixed gamma standards in geometries and densities, closely approximating analytical samples, used to calibrate the instrument.

11.0 SAMPLE HANDLING AND PRESERVATION

Refer to GL-RAD-A-013 The Determination of Gamma Isotopes.

12.0 SAMPLE PREPARATION

Refer to GL-RAD-A-013 The Determination of Gamma Isotopes.

13.0 QUALITY CONTROL SAMPLES AND REQUIREMENTS

Refer to GL-RAD-A-013 The Determination of Gamma Isotopes.

14.0 INSTRUMENT CALIBRATION, STANDARDIZATION AND PERFORMANCE

14.1 Calibration Standard

- 14.1.1 Mixed Gamma calibrations typically use a standard with 8-12 photons emitted over a range from approximately 45 keV to approximately 2000 keV.
- 14.1.2 Single nuclide calibrations typically use a standard comprised of the nuclide of interest.

14.2 Verification Standard

- 14.2.1 Mixed Gamma calibrations- A second source (from different manufacturer or if from the same manufacturer, a different lot number) is used for verification. The lines from Am-241, Cs-137 and Co-60 are used to verify the efficiency curve. These encompass the low, middle and high portions of the energy range.
- 14.2.2 Single nuclide calibrations – A second source (from a different manufacturer or if from the same manufacturer, a different lot number) is used for verification.

14.3 Standardization

14.3.1 High Voltage Adjust

- 14.3.1.1 See appropriate instrument manual for operation of electronics.

- 14.4 Calibration – Energy and efficiency calibrations are performed annually, upon initial instrument setup, after major repair or service, or when performance checks indicate a need.

NOTE: Expiration dates will match the last day of the month in which the calibration data was acquired.

14.4.1 Count Calibration Spectrum

14.4.1.1 Place the radioactive source on the detector.

14.4.1.2 Select **Calibration** | **Count a Calibration Standard** from the *Calibration* menu and click **OK**.

14.4.1.3 Enter the **Preset Live (secs):** in seconds and click **OK**. Count the standard until a minimum of 10,000 counts is acquired in each peak of interest.

14.4.2 Initial Energy & Shape Calibration

14.4.2.1 Select **Calibration** | **Initial Energy & Shape Calibration** from the *Calibration* menu and click **OK**.

14.4.2.2 Select the detector.

14.4.2.3 Select the **Certificate File** from the drop down list. Click **OK**.

14.4.2.4 From the *Energy Calibration* dialog box highlight one of the energy lines listed.

14.4.2.5 Move the cursor in the MCA window to the corresponding channel expected for that energy line.

14.4.2.5.1 The apex of the peak of interest should be at the expected channel.

14.4.2.5.2 From the *Energy Calibration* dialog box click the **Cursor** button.

14.4.2.6 Repeat the previous step until all energy lines listed have been referenced with a corresponding channel.

14.4.2.7 From the *Energy Calibration* dialog box select the **OK** button.

14.4.2.8 The system will ask “Do you want to do a full energy and shape calibration?” Select **YES**.

14.4.2.9 The energy and shape calibrations will now be performed with all of the lines from step 14.4.2.6. Verify the energy and shape curve generated. Select **OK** to continue or **Cancel** to abort the calibration.

14.4.2.10 A new page will appear with the Energy Calibration Report and the FWHM Calibration Report. Review the columns marked difference. For the energy calibration, the absolute value of the difference must be less than 1.0 and for the FWHM calibration, the absolute value of the difference must be less than 0.5. Regardless of the results, select **Dismiss**.

- 14.4.2.11 A new pop-up screen will appear. If the results from the previous step were less than a 0.2 keV difference, select **OK**. If the results were greater than a 0.2 keV difference select **Cancel** and begin the energy calibration process again at step 14.4.1.
- 14.4.3 Energy Re-Calibrate
 - 14.4.3.1 Select **Calibrate | Re-Calibrate | Energy and Shape Calibration** from the main menu and click **OK**.
 - 14.4.3.2 Select the detector.
 - 14.4.3.3 Select the certificate file and select the **OK** button.
 - 14.4.3.4 The energy and shape calibrations will now be performed with all of the lines from step 14.4.2.6. Verify the energy and shape curve generated. Select **OK** to continue or **Cancel** to abort the calibration.
 - 14.4.3.5 A new page will appear with the Energy Calibration Report and the FWHM Calibration Report. Review the columns marked difference. For the energy calibration, the absolute value of the difference must be less than 1.0 and for the FWHM calibration, the absolute value of the difference must be less than 0.5. Regardless of the results, select **Dismiss**.
 - 14.4.3.6 A new pop-up screen will appear. If the results from the previous step were less than a 0.2 keV difference, select **OK**. If the were greater than a 0.2 keV difference select **Cancel** and begin the energy calibration process again. If it fails after re-calibration contact Group or Team Leader for further instructions.
- 14.4.4 Efficiency Calibrate
 - 14.4.4.1 Select **Calibrate | Efficiency Calibrate** from the main menu.
 - 14.4.4.2 Select the geometry that represents the standardized radioactive source and click **OK**. If the geometry doesn't exist select **Create New Geometry**, enter the name of the new geometry and select **OK**.
 - 14.4.4.3 Select the certificate for the calibration standard and select the **OK** button.
 - 14.4.4.4 The efficiency calibration curve will be displayed for review. Select Empirical fit, and Log scale.
 - 14.4.4.5 To accept the calibration select **OK**, or select **Cancel** to abort.
 - 14.4.4.6 Dismiss the Calibration report displayed to complete the calibration procedure.

14.4.4.7 In the DECterm type **EFFPlot**, then press ENTER the type **EFFPRINT** then hit ENTER. This will print the efficiency curve.

14.4.5 Efficiency Verifications

14.4.5.1 Verification counts are performed as a normal sample count starting at step 15.2.3 of this SOP.

14.4.5.2 No batch ID is assigned to verification counts, typically “VER” is used.

14.4.5.3 Select the only sample identification available regardless of how it is named.

14.4.5.3.1 You may be asked if you would like to extend the count. Select **NO**.

14.4.5.3.2 When the screen to enter the sample information appears (step 15.2.8), use the date and time indicated on the manufacturer’s certificate file for decay correction and change the sample identification using the following naming convention:
VER_DETECTOR_GEOMETRY, for example VER_GAM01_CAN.

14.4.5.4 Once the count has completed, in the DECterm, type “@print_virtual sample, where sample equals the same identification used in step 14.4.5.3.2. This will print out the raw data of the verification count.

14.4.5.5 Several pages will print out. The only pages needed are the background-subtracted peak report, which should be the first page, and the nuclide line activity report.

14.4.5.6 Place the results from the “Decay Corr” column into the appropriate Master Verification Spreadsheet located at S:\RAD\FORMS\EFF_VER where S:= sdrive on ‘radserver’ under the column named **Measured Activity**.

14.4.5.7 If necessary, enter the emission rate for the standard used for verification on the Master Verification Spreadsheet. This can be found on the manufacturer’s certificate file for the standard. The spreadsheet will then calculate the **Calibrated Activity**. If a column for the emission rate does not exist on the spreadsheet, the **Calibrated Activity** can be calculated by using the values from the Decay Correct Source page in Alpha LIMS

(http://prodsvr01.gel.com:7778/pls/lims/de_ref_material.decay_correction).

- 14.4.5.8 The percent difference between the **Calibrated Activity** and **Measured Activity** is calculated by the spreadsheet and is displayed under the column marked **Difference**. The verification is considered acceptable if all values in the **Difference** column are less than 10%. If the **Difference** is 10% or greater, the verification is considered invalid and must be performed again. If two verifications fail notify Group Leader or Team Leader for further action.

14.5 Performance Checks

14.5.1 Daily Quality Control Calibration Check (QCC)

- 14.5.1.1 The QCC should be counted daily or prior to sample counting. If no samples are being counted this check is not required.
- 14.5.1.2 Load the QCC check source on the detector(s). If multiple QCC checks are being started skip to step 14.5.1.5.
- 14.5.1.3 From the PROcount window, select **QC | Calibration Check**.
- 14.5.1.4 Select the detector and select **OK**.
- 14.5.1.5 To start multiple QCC checks at once, select **QC | Multi Calibration Checks** from the PROcount window.
- 14.5.1.6 Highlight each detector you wish to start by clicking once on the detector name. Once you have highlighted all of the detectors you wish to start, select **OK**.

14.5.2 Daily Quality Control Background Check (QCB)

- 14.5.2.1 The QCB should be counted daily or prior to sample counting. If no samples are being counted this check is not required.
- 14.5.2.2 Ensure the detector shield(s) are empty prior to running the QCB. If multiple QCB checks are being started, skip to step 14.5.2.5.
- 14.5.2.3 From the PROcount window, select **QC | Background Check**.
- 14.5.2.4 Select the detector and select **OK**.
- 14.5.2.5 To start multiple QCB checks at once, select **QC | Multi Background Checks** from the PROcount window.
- 14.5.2.6 Highlight each detector you wish to start by clicking once on the detector name. Once you have highlighted all of the detectors you wish to start, select **OK**.

14.5.3 Weekly Environmental Background

- 14.5.3.1 Ensure the detector shield(s) is (are) empty. The same process will be used to start single and multiple weekly environmental background counts.

14.5.3.2 Select **Count | Start Multiple Backgrounds** from the PROcount window.

14.5.3.3 highlight each detector you wish to start by clicking once on the detector name. Once you have highlighted all of the detectors you wish to start, select **OK**.

14.5.4 Generating the Daily and Weekly Check Reports

14.5.4.1 Daily check reports will generate every day following the completion of the QCC and QCB counts for each detector that will be in operation.

14.5.4.2 In the DECterm, type the command “**@QA_REPORT D**” then hit ENTER.

14.5.4.3 Weekly check reports will be completed once per week, typically Monday, following the completion of the weekly background subtraction counts.

14.5.4.4 In the DECterm, type the command “**@QA_REPORT B**” then hit ENTER.

15.0 PROCEDURE FOR ANALYSIS AND INSTRUMENT OPERATION

15.1 Prepare the sample as outlined in GL-RAD-A-013 for The Determination of Gamma Isotopes.

15.2 Sample Counting

15.2.1 Prior to starting a sample count the detector used must be scanned into AlphaLIMS. In a web browser, enter the following address:

http://prodsvr01.gel.com:7778/pls/lims/inst_instrument.start_count

15.2.2 Each sample and detector are labeled with a Universal Product Code (UPC). First scan the UPC code for the detector and then scan the UPC code for the sample. Continue doing so for any additional sample counts. Once this has been done, select Submit on the web page.

15.2.3 Load the sample on the detector.

15.2.4 Select **Count | Start a Count** from the *ProCount Main Menu*.

15.2.5 Select a detector and select **OK**.

15.2.6 Enter the batch to be started and select OK.

15.2.7 Select the sample to be counted.

15.2.8 Enter the sample specific information into the Sample Information screen and select OK.

15.2.9 Select the Analysis Sequence file used for analysis and select OK.

15.2.10 Select the counting geometry and select OK.

16.0 EQUIPMENT AND INSTRUMENT MAINTENANCE

Refer to GL-RAD-I-010 for Counting Room Instrumentation Maintenance.

17.0 DATA RECORDING, CALCULATION AND REDUCTION METHODS

Data recording, calculation and reduction take place in accordance with GL-RAD-D-003 and GL-RAD-D-006.

18.0 POLLUTION/CONTAMINATION

Ensure all samples are bagged prior to counting to prevent instrument contamination.

19.0 DATA REVIEW, APPROVAL AND TRANSMITTAL

Refer to GL-RAD-D-003 for Data Review, Validation, and Data Package Assembly.

20.0 CORRECTIVE ACTION FOR OUT-OF-CONTROL OR UNACCEPTABLE DATA

Corrective action for out-of-control data might require instrument maintenance, re-analysis, using a new spike mix, or a more complex set of actions. When troubleshooting measures (refer to Section 21) fail to bring an analytical process or data into control, a data exception report and/or corrective action should be initiated in accordance with GL-QS-E-004.

21.0 CONTINGENCIES FOR HANDLING THESE SITUATIONS

Troubleshooting the instrument is a function of analyst experience. In-house service is obtained from GEL's Group Leader or other qualified personnel. If vendor assistance is needed, then the appropriate vendor is contacted. Maintenance logbooks are kept for each instrument and contain entries for both routine and non-routine maintenance procedures.

22.0 RECORDS MANAGEMENT

- 22.1 Each sample analysis that is performed is documented in the instrument run log in accordance with GL-LB-E-009 for Run Logs.
- 22.2 All raw data printouts, calculation spreadsheets, and batch checklists are filed with the sample data for archival in accordance with GL-RAD-D-003 for Data Review, Validation, and Data Package Assembly.
- 22.3 Instrument maintenance is recorded in accordance with GL-LB-E-008 for Basic Requirements for the Use and Maintenance of Laboratory Notebooks, Logbooks, Forms and Other Recordkeeping Devices.
- 22.4 Records generated as a result of this procedure are maintained as quality documents in accordance with GL-QS-E-008 for Quality Records Management and Disposition.

23.0 LABORATORY WASTE HANDLING AND DISPOSAL

Laboratory waste is disposed in accordance with the Laboratory Waste Management Plan, GL-LB-G-001.

24.0 REFERENCES

- 24.1 United States Department of Energy, Environmental Measurements Laboratory, HASL-300 The Procedures Manual of the Environmental Measurements Laboratory, 28th Edition, "Gamma Radioassay," Ga-01-R (Vol. 1), February 1997.
- 24.2 United States Environmental Protection Agency, Prescribed Procedures for Measurement of Radioactivity in Drinking Water, Method 901.1, August 1980.

- 24.3 American National Standards Institute, American National Standard for Calibration and Use of Germanium Spectrometers for the Measurement of Gamma-Ray Emission Rates of Radionuclides, ANSI N42.14-1999.
- 24.4 Canberra Model 480720 ProCount-ESP Users Manual, September 2000.
- 24.5 Canberra Model 480726 Genie-ESP System Users Manual, September 2000.
- 24.6 Canberra Model 480198 Genie VMS Users Manual, 2000.
- 24.7 ASTM, International, Standard Practice for Setup, Calibration, and Quality Control of Instruments Used for Radioactive Measurements, D7282-6, Nov. 2010.

25.0 HISTORY

Revision 15: Procedural updates made to SOP to reflect the process currently being used.

Revision 16: Added criteria for acceptance of FWHM calibration.

Revision 17: Updated sections 14.4.3.6 and 14.4.5.2 for clarification.

Revision 18: Added note to section 14.4 to clarify instrument calibration expiration dates.

Revision 19: Revised to include new GL-RAD-D-006 for calculations.

ALS Standard Operating Procedure

DOCUMENT TITLE:
REFERENCED METHOD:
SOP ID:
REVISION NUMBER:
EFFECTIVE DATE:

METALS DIGESTION
EPA 3050B
MET-3050B
14
2/15/2015





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METALS DIGESTION

ALS-KELSO

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Approved By:  Date: 2/3/15
Department Manager/Technical Director - Jeff Coronado

Approved By:  Date: 2/3/15
QA Manager - Lee Wolf

Approved By:  Date: 2/3/15
Laboratory Director - Jeff Grindstaff

Issue Date: _____ Doc Control ID#: _____ Issued To: _____

ANNUAL REVIEW

SIGNATURES BELOW INDICATE NO PROCEDURAL CHANGES HAVE BEEN MADE TO THE SOP SINCE THE APPROVAL DATE ABOVE. THIS SOP IS VALID FOR TWELVE ADDITIONAL MONTHS FROM DATE OF THE LAST SIGNATURE UNLESS INACTIVATED OR REPLACED BY SUBSEQUENT REVISIONS.

Signature _____	Title _____	Date _____
Signature _____	Title _____	Date _____
Signature _____	Title _____	Date _____
Signature _____	Title _____	Date _____



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METALS DIGESTION

1. SCOPE AND APPLICATION

- 1.1. This procedure uses techniques described in method 3050B for acid digestion of sediments, sludges, and soil samples designated for “Total Metals” analysis. One technique is designed for the preparation of samples for analysis by flame AA (Methods 7420-Pb, 7742-Se, and 7062-As) or ICP-OES (methods 6010 and 200.7). Another technique is given for the preparation of samples for analysis by GFAA (see SOP MET-GFAA for methods) or ICP-MS (methods 6020 and 200.8). This procedure is not a *total digestion* technique, but extracts “environmentally available” elements from the sample of interest.

2. METHOD SUMMARY

- 2.1. One-gram equivalent dry weight sediment, sludge, or soil samples are digested with repeated additions of nitric acid (HNO₃) and hydrogen peroxide (H₂O₂). For GFAA and ICP-MS analysis the resultant digestate is reduced in volume while heating and then diluted to a final volume of 100 mL. For ICP-OES and flame AA analysis, hydrochloric acid (HCl) is added to the initial digestate and the sample is refluxed prior to dilution to a final volume of 100 mL.

3. DEFINITIONS

- 3.1. **Batch** - A batch of samples is a group of environmental samples that are prepared and/or analyzed together as a unit with the same process and personnel using the same lot(s) of reagents. It is the basic unit for analytical quality control.
- 3.2. **Preparation Batch** - A preparation batch is composed of one to twenty field samples, all of the same matrix, and with a maximum time between the start of processing of the first and last samples in the batch to be 24 hours.
- 3.3. **Sample**
- 3.3.1. **Field Sample** - An environmental sample collected and delivered to the laboratory for analysis; a.k.a., client’s sample.
- 3.3.2. **Laboratory Sample** - A representative portion, aliquot, or subsample of a field sample upon which laboratory analyses are made and results generated.
- 3.4. **Quality System Matrix** - The *matrix* of an environmental sample is distinguished by its physical and/or chemical state and by the program for which the results are intended. The following sections describe the matrix distinctions. These matrices shall be used for purpose of batch and quality control requirements.
- 3.4.1. **Solids** - Any solid sample such as soil, sediment, sludge, and other materials with >15% settleable solids.
- 3.5. **Laboratory Control Sample (LCS)** - A laboratory blank that has been fortified with target analyte and used to determine that the analysis is in control.



-
- 3.6. **Matrix Spike (MS)** - In the matrix spike analysis, predetermined quantities of target analytes are added to a sample matrix prior to sample preparation and analysis. The percent recovery is calculated. The MS is used to evaluate the effects of the sample matrix on the method used for the analysis. The concentration of the spike should be at three to five times the sample result or at levels specified by a project analysis plan.
- 3.7. **Duplicate Sample (DUP)** - A laboratory duplicate. The duplicate sample is a separate field sample aliquot that is processed in an identical manner as the sample proper. The relative percent difference between the samples is calculated and used to assess analytical precision.
- 3.8. **Method Blank (MB)** - The method blank is an artificial sample composed of analyte-free water or solid matrix and is designed to monitor the introduction of artifacts into the analytical process. The method blank is carried through the entire analytical procedure.

4. INTERFERENCES

- 4.1. Refer to the determinative method for a discussion of interferences.

5. SAFETY

- 5.1. All appropriate safety precautions for handling solvents, reagents and samples must be taken when performing this procedure. This includes the use of personnel protective equipment, such as, safety glasses, lab coat and the correct gloves.
- 5.2. Chemicals, reagents and standards must be handled as described in the ALS safety policies, approved methods and in MSDSs where available. Refer to the ALS Environmental, Health and Safety Manual and the appropriate MSDS prior to beginning this method.
- 5.3. Hydrochloric and/or Nitric Acid are used in this method. These acids are extremely corrosive and care must be taken while handling them. A face shield must be used while pouring concentrated acids. And safety glasses should be worn while working with the solutions. Lab coat and gloves should always be worn while working with these solutions.

6. SAMPLE COLLECTION, CONTAINERS, PRESERVATION AND STORAGE

- 6.1. Samples may be collected in plastic or glass jars. Non-aqueous samples are refrigerated at $4 \pm 2^{\circ}\text{C}$ from receipt until analysis.
- 6.2. The recommended holding time is 6 months from the day of sampling.

7. STANDARDS, REAGENTS, AND CONSUMABLE MATERIALS

- 7.1. Reagent grade chemicals shall be used in all tests. Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lowering the accuracy of the determination. The preparation for all laboratory prepared reagents and solutions must be documented in a laboratory logbook. Refer to the SOP *Reagent/Standards Login and Tracking (ADM-RTL)* for the complete procedure and documentation requirements.
- 7.2. Reagent water: ASTM Type I water (resistivity $\geq 18 \text{ M}\Omega\text{-cm}$, conductivity $\leq 0.056 \text{ uS/cm}$).



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- 7.3. Concentrated Nitric Acid: J.T. Baker “Instra-analyzed”, Trace Metals Grade
 - 7.4. Concentrated Hydrochloric Acid: EMD GR ACS
 - 7.5. Hydrogen Peroxide (30%): EMD GR ACS
 - 7.6. Standards
 - 7.6.1. Stock standards may be purchased from a number of vendors. All reference standards, where possible, must be traceable to SI units or NIST certified reference materials. The vendor assigned expiration date is used.
 - 7.6.2. Metals spiking solutions: Five spiking solutions are needed to prepare the matrix spike sample; SS1, SS2, SS3, SS4, and SS5.
 - 7.6.3. Follow the formulations laid out on the “Metals Spike Form” (see attached Table A). These solutions are prepared in acid rinsed Class A volumetric flasks using purchased custom mixed standards or 1000 ppm single analyte standards. Aliquots are made using acid rinsed Class A volumetric pipettes of the appropriate size.
 - 7.6.4. SS1 (Al, Ag, Ba, Be, Cd, Co, Cr, Cu, Fe, Pb, Mn, Ni, Sb, V, and Zn): Fill a 1000 mL volumetric flask approximately half full with reagent water, add 50 mL of nitric acid and mix. Next add 100 mL of the custom mixed standard (CAS-CAL-14) purchased from “Inorganic Ventures”. In addition add 50 mL of 1000 ppm Antimony (use the Antimony standard that does not contain HCL.) Dilute to volume with reagent water, mix thoroughly and transfer to a 1000 mL Teflon bottle for storage. The solution expiration date is determined by the earliest expiration date of any single component in the solution.
 - 7.6.5. SS2 (GFAA As, Cd, Cu, Pb, Se, Tl): Fill a 500 mL volumetric flask approximately half full with reagent water, add 25 mL of nitric acid and mix. Next add 2.0 mL each of 1000 ppm Arsenic, Cadmium, Copper, Lead, Selenium, and Thallium. Dilute to volume with reagent water, mix thoroughly and transfer to a 500 mL Teflon bottle for storage. The solution expiration date is determined by the earliest expiration date of any single component in the solution.
 - 7.6.6. SS3 (As, Se, Tl, and Hg): Fill a 500 mL volumetric flask approximately half full with reagent water, add 25 mL of nitric acid and mix. Next add 50 mL each of 1000 ppm Arsenic, Selenium, and Thallium. Add 6.0 mL of 1000 ppm Hg. Dilute to volume with reagent water, mix thoroughly and transfer to a 500 mL Teflon bottle for storage. The solution expiration date is determined by the earliest expiration date of any single component in the solution.
 - 7.6.7. SS4 (B, Mo): Fill a 500 mL volumetric flask approximately half full with reagent water, add 25 mL of nitric acid and mix. Next add 50 mL each of 1000 ppm Boron and Molybdenum. Dilute to volume with reagent water, mix thoroughly and transfer to a 500 mL Teflon bottle for storage. The solution’s expiration date is determined by the earliest expiration date of any single component in the solution.



- 7.6.8. SS5 (K,Na,Mg,Ca): Fill a 200 mL volumetric flask approximately half full with reagent water add 10.0 mL of nitric acid and mix. Next add 20 mL each of 10,000 ppm Potassium, Sodium, Magnesium and Calcium. Dilute to volume with reagent water, mix thoroughly and transfer to a 250 mL Teflon bottle for storage. The solution's expiration date is determined by the earliest expiration date of any single component in the solution.
- 7.7. Metals reference material (ERA Priority PollutnT/CLP Inorganic Soil) for use as the laboratory control sample. The expiration date is assigned by the manufacturer.
- 7.8. Teflon beads, Teflon boiling chips, or other suitable blank material.

8. APPARATUS AND EQUIPMENT

- 8.1. 125 mL plastic cup beaker cup, calibrated at 50mL and 100mL
- 8.2. Borosilicate watch glasses
- 8.3. Block Digester, calibrated to maintain $95^{\circ}\text{C} \pm 2^{\circ}\text{C}$
- 8.4. Hot Plates: "Thermolyne Cimerac 3", calibrated to maintain $95^{\circ}\text{C} \pm 2^{\circ}\text{C}$
- 8.5. Laboratory balance, top-loader capable of reading 0.01g
- 8.6. Digestion tubes, 125 mL - Environmental Express. An accuracy and precision verification check must be made with each new vendor lot prior to use. Refer to the SOP for *Checking Volumetric Labware ADM-VOLWARE*, for further detailed instructions. Performance data must meet the accuracy and precision requirements specified in Table 1 (*ADM-VOLWARE*) for non-volumetric labware used for measuring initial and/or final digestate volumes.
- 8.7. USS # 10 sieve.

9. PREVENTIVE MAINTENANCE

- 9.1. All maintenance activities are recorded in a maintenance logbook. Pertinent information must be in the logbook. Maintenance entries should include date, symptom of problem, corrective actions, and description of maintenance, date, and name. The log should contain a reference to return to analytical control.
- 9.2. Maintenance for this procedure is generally limited to glassware cleaning, pipet monitoring, and hot plate calibration. Procedures for glassware washing are described in the SOP for Metals Laboratory Glassware Cleaning (MET-GC). Procedures for pipet monitoring are given in the SOP for Checking Volumetric Labware, (ADM-VOLWARE).
- 9.3. Each hotplate or block digester is uniquely identified and the temperature is verified with each batch of samples. To perform the verification, a certified thermometer is placed in a container half filled with mineral oil, which is then placed in the center of the hotplate or block digester. The thermometer does not touch the bottom of the container. The temperature is turned to the 95°C setting and the mineral oil is allowed to come to temperature. The analyst will verify that the hotplate gives a temperature of $95^{\circ}\text{C} \pm 2^{\circ}\text{C}$. If not, the thermostat is adjusted until the thermometer reads and maintains $95^{\circ}\text{C} \pm 2^{\circ}\text{C}$. The thermostat is then marked to clearly



indicate the correct setting to be used during sample digestion (when using Hot Plates.). Each hot Block has an assigned calibrated thermometer. The Temperature and the correction factor of the assigned thermometer is recorded on the digestion bench sheet.

10. RESPONSIBILITIES

- 10.1. It is the responsibility of the analyst to perform the analysis according to this SOP and to complete all documentation required for data review. Analysis and interpretation of the results are performed by personnel in the laboratory who have demonstrated the ability to generate acceptable results utilizing this SOP. This demonstration is in accordance with the training program of the laboratory. Final review and sign-off of the data is performed by the department supervisor/manager or designee.
- 10.2. It is the responsibility of the department supervisor/manager to document analyst training.

11. PROCEDURE

- 11.1. Record all digestion and sample information on the applicable benchsheet.
- 11.2. Mix the sample thoroughly to achieve homogeneity. Sieve if necessary using a USS #10 sieve.
- 11.3. It can be difficult to obtain a representative sample with wet or damp materials. As per Method 3050B, wet samples may be dried, crushed, and ground to reduce subsample variability, however, drying is not recommended since drying may affect the extraction of the analytes of interest in the sample.
- 11.4. Weigh approximately 1g of sample into a 125ml plastic beaker cup and record the weight to the nearest 0.01g. For sludge's and sediments that have high moisture content, use more sample. A plastic 10.0 mL disposable pipette is used to measure 10.0 mL of sample. The volume and weight of the pipetted sample is recorded. In cases where the sludge is very thick a 10.0 mL graduated cylinder may be used. The objective is to use about 1g of dry weight sample. For analysis of Lead by Flame AA, use about 2.5g of dry wt. sample and change the final dilution volume to 50ml. This will achieve a lower detection limit needed for most projects. At this point add the appropriate spiking solutions directly onto the designated spike sample prior to addition of reagents.
- 11.5. Add 5ml reagent water and 5ml concentrated HNO_3 . Place in a hot block, cover and reflux (without boiling) at 95°C for 10 to 15 minutes. Allow the sample to cool. Add 5ml of concentrated HNO_3 , cover and reflux for 30 minutes. If brown fumes are generated, indicating oxidation of the sample by HNO_3 , repeat the addition of 5ml of HNO_3 and reflux over and over until no brown fumes are given off. Reduce the digestate volume to approximately 5 mL without boiling or digest for two hours maintaining a covering of solution over the bottom of the beaker at all times. If this occurs discard the digestate and begin with a new sample aliquot.

Note: The 95°C hot block temperature must be monitored and documented on a per-batch basis. The actual measured temperature, thermometer correction factor, and corrected temperature must all be recorded.



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Note: All Wisconsin samples must digest for 2 hours after generation of brown fumes has ceased.

- 11.6. Cool the sample and add 3 mL of 30% H_2O_2 . Cover and heat to start the peroxide reaction. Care must be taken to ensure that losses do not occur due to excessive effervescence. Heat in the hot block until effervescence subsides. Remove from hot block and cool the beaker.
- 11.7. Continue to add 30% H_2O_2 in 3ml aliquots with warming until the effervescence is minimal, or until the general sample appearance is unchanged. Do not add more than 10ml of 30% H_2O_2 . When the peroxide additions are complete cover the sample with a watch glass and continue heating the acid-peroxide digestate until the volume has been reduced to approximately 5 mL or heat at $95 \pm 5^\circ\text{C}$ without boiling for 2 hours. Do not let the samples go to dryness, by ensuring the solution covers the bottom of the vessel at all times.

Note: All Wisconsin samples must digest for 2 hours after the final peroxide addition.

If the sample is being prepared for analysis by ICP-OES or Flame AA, add 10 mL of concentrated HCl. If the sample is being prepared for ICP-MS or GFAA analysis no HCl is added. Dilute the sample to 100 mL with reagent water: ASTM Type I water (resistivity $\geq 18 \text{ M}\Omega\text{-cm}$, conductivity $\leq 0.056 \text{ uS/cm}$) in a 125 mL plastic beaker cup.

Note: For method 7062 and 7742 samples, the 3050B soil digestion is modified as follows: After the final peroxide addition (i.e. before the final reduction stage) add 5.0mL of concentrated hydrochloric acid and reduce the digestate volume to less than 5.0mL, but not to dryness. After cooling, dilute the digestate to 100mL with reagent water.

- 11.8. Cover and reflux the Flame AA and ICP samples for 15 minutes at 95°C . After cooling, the samples may be diluted to 100ml for ICP analysis, or 50ml for Flame AA analysis.
- 11.9. Particulates in the digestates that may clog the nebulizer are allowed to settle overnight, or the digestates may be centrifuged.
- 11.10. To improve the solubility for Antimony, Barium, Lead and Silver, the following modification of the digestion procedure may be used as directed by the client or project chemist.
 - 11.10.1. Weigh (to the nearest 0.01g) 1.00 g of sample into a 125ml plastic cup. For sludges and sediments that have high moisture content, use more sample. The objective is to use about 1g of dry weight sample.
 - 11.10.2. Add 2.5mL HNO_3 and 10mL HCl and cover with a watch glass. Reflux for 15 minutes.
 - 11.10.3. Filter the digestate through Whatman No. 41 or equivalent filter paper and collect in a 100mL volumetric flask. Wash the filter paper, while still in the funnel, with no more than 5mL of hot (95°) HCl, and then with 20mL of hot (95°) reagent water. Collect washing in the same volumetric flask.



-
- 11.10.4. Remove the filter and residue from the funnel, and place them back in the beaker. Add 5mL HCl, cover and heat at $95^{\circ} \pm 5^{\circ}$ until the filter paper dissolves. Remove from the heat and wash the cover and sides with reagent water.
- 11.10.5. Filter the residue and collect the filtrate in the same 100mL flask. Allow to cool, then dilute to volume.
- 11.10.6. If precipitation occurs in the flask upon cooling, do not dilute to volume. Instead, add up to 10mL of HCl to dissolve the precipitate. After precipitate is dissolved, dilute to volume with water.

12. QA/QC REQUIREMENTS

12.1. Initial Precision and Recovery Validation

- 12.1.1. The accuracy and precision of the procedure must be validated before analyses of samples begin, or whenever significant changes to the procedures have been made. To do this, four blank matrix samples are spiked with the LCS spike solution, then prepared and analyzed.

12.2. Monitor Hot Blocks and Hotplates on a per batch basis. Report all deficiencies to the Lab Manager. Corrective action must be taken.

12.3. Digest one laboratory control sample with each batch. Weigh 1.00 g of the current lot of Environmental Resource Associates PriorityPollutnT/CLP Inorganic Soil prepared reference material into a 150 mL beaker and digest as per the procedure.

12.4. Digest one preparation blank (method blank) per digestion batch, or per 20 samples whichever is more frequent. For the method blank, use Teflon beads, Teflon boiling chips, or other suitable solid blank material and follow the digestion procedures.

12.5. Digest one duplicate and one spiked sample with each sample matrix. Prepare one duplicate and spike sample per each digestion batch, or per twenty samples whichever is more frequent. At times, specific samples will be assigned as duplicates of spikes depending on client requirements.

12.6. Soil spikes for ICP and ICP-MS are prepared by adding 2.0 mL of SS1, and 1.0 mL of SS3, SS4 and SS5 directly to the sample aliquot, prior to the addition of any water or acid. Fill out a spiking data sheet and keep it with the digestion data sheets.

- 12.6.1. For GFAA digestions 2.0 mL of SS2 is added to the sample aliquot designated as the matrix spike sample. The matrix spike sample is then digested as per the procedure.

13. DATA REDUCTION AND REPORTING

13.1. Digestion data sheets including weights and volumes used and reagents/acids are completed and a prep run number or batch lot number is assigned and attached to the data sheet. The lot numbers for the reagents used are added to the digestion data sheet (see Attachments).

13.2. Spiking sheets are included (See Attachments).



13.3. Data Review and Assessment

- 13.3.1. Refer to the *SOP for Laboratory Data Review Process* for general instructions for data review.
- 13.3.2. It is the supervisor's responsibility to ensure that digestions data is reviewed to ensure that all quality control requirements have been met and documentation is complete.

14. CONTINGENCIES FOR HANDLING OUT-OF-CONTROL OR UNACCEPTABLE DATA

- 14.1. Refer to the SOP for *Nonconformity and Corrective Action* (CE-QA008) for procedures for corrective action. Personnel at all levels and positions in the laboratory are to be alert to identifying problems and nonconformities when errors, deficiencies, or out-of-control situations are detected.
- 14.2. Handling out-of-control or unacceptable data
- 14.2.1. On-the-spot corrective actions that are routinely made by analysts and result in acceptable analyses should be documented as normal operating procedures, and no specific documentation need be made other than notations in laboratory maintenance logbooks, runlogs, for example.
- 14.2.2. Some examples when documentation of a nonconformity is required using a Nonconformity and Corrective Action Report (NCAR):
- Quality control results outside acceptance limits for accuracy and precision
 - Method blanks or continuing calibration blanks (CCBs) with target analytes above acceptable levels
 - Sample holding time missed due to laboratory error or operations
 - Deviations from SOPs or project requirements
 - Laboratory analysis errors impacting sample or QC results
 - Miscellaneous laboratory errors (spilled sample, incorrect spiking, etc.)
 - Sample preservation or handling discrepancies due to laboratory or operations error

15. METHOD PERFORMANCE

- 15.1. This method was validated through single laboratory studies of accuracy and precision. Refer to the reference method for additional method performance data available.
- 15.2. The method detection limit (MDL) is established using the procedure described in the SOP CE-QA011, *Performing Method Detection Limit Studies and Establishing Limits of Detection and Quantification*. Method Reporting Limits are established for this method based on MDL studies and as specified in the ALS Quality Assurance Manual.

16. POLLUTION PREVENTION AND WASTE MANAGEMENT

- 16.1. It is the laboratory's practice to minimize the amount of solvents, acids, and reagents used to perform this method wherever feasibly possible. Standards are prepared in volumes consistent with methodology and only the amount needed for routine laboratory use is kept



on site. The threat to the environment from solvents and/or reagents used in this method can be minimized when recycled or disposed of properly.

- 16.2. The laboratory will comply with all Federal, State, and local regulations governing waste management, particularly the hazardous waste identification rules and land disposal restrictions as specified in the ALS Environmental Health and Safety Manual.
- 16.3. This method uses acid. Waste acid is hazardous to the sewer system and to the environment. All acid waste must be neutralized to a pH of 2.5-12 prior to disposal down the drain. The neutralization step is considered hazardous waste treatment and must be documented on the treatment by generator record. See the ALS EH&S Manual for details.

17. TRAINING

17.1. Training outline

- 17.1.1. Review literature (see references section). Read and understand the SOP. Also review the applicable MSDS for all reagents and standards used. Following these reviews, observe the procedure as performed by an experienced analyst at least three times.
- 17.1.2. The next training step is to assist in the procedure under the guidance of an experienced analyst. During this period, the analyst is expected to transition from a role of assisting, to performing the procedure with minimal oversight from an experienced analyst.
- 17.1.3. Perform initial precision and recovery (IPR) study as described above for water samples. Summaries of the IPR are reviewed and signed by the supervisor. Copies may be forwarded to the employee's training file. For applicable tests, IPR studies should be performed in order to be equivalent to NELAC's Initial Demonstration of Capability.

17.2. Training is documented following the *SOP ADM-TRAIN*.

NOTE: When the analyst training is documented by the supervisor on internal training documentation forms, the supervisor is acknowledging that the analyst has read and understands this SOP and that adequate training has been given to the analyst to competently perform the analysis independently.

18. METHOD MODIFICATIONS

- 18.1. The method uses 2 mL of water and 3 mL of H₂O₂ in step 11.6. The lab does not add the 2 mL of water. 3.0 mL aliquots of 30% H₂O₂ in lieu of 1.0 mL aliquots are added subsequently.

19. REFERENCES

- 19.1. Test Methods for Evaluating Solid Waste, Physical/Chemical Methods. EPA SW-846, 3rd Edition, Final Update III, Method 3050B, December 1996.
- 19.2. Table A – METALS SPIKING SOLUTIONS CONCENTRATIONS FORM

20. CHANGES SINCE THE LAST REVISION



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- 20.1. Reformatted SOP to current ALS format and style.
- 20.2. Updated internal SOP references.
- 20.3. Few minor changes (correct typos and errors, etc.).
- 20.4. Section 7.6.6 – revised to include Mercury in SS3 solution.
- 20.5. Section 8.6 – revised to update tubes to those in use.
- 20.6. Section 11.5 – Added second note regarding Wisconsin samples.
- 20.7. Section 11.7 – Inserted first note regarding Wisconsin samples.
- 20.8. Section 12.6 – revised to list spiking with current solution formulations.
- 20.9. Section 13.3 – New section.
- 20.10. Section 14 – updated to current standard language for the section.
- 20.11. Section 15 – updated to current standard language for the section.
- 20.12. Section 16 – updated to current standard language for the section.
- 20.13. Table A – updated



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TABLE A
METALS SPIKING SOLUTIONS CONCENTRATIONS FORM

Solution Name	Element	mL of 1000ppm Solution	Final Volume	Solution Conc. mg/L	Enter ml Added
K-MET SS1 *** Add after HNO3 and before cas cal -14 when making the solution	HNO3	50.0	1000ml	-	
	Al	100*	1000ml	200	
	Ag	100*	1000ml	5	
	Ba	100*	1000ml	100	
	Be	100*	1000ml	5	
	Cd	100*	1000ml	5	
	Co	100*	1000ml	50	
	Cr	100*	1000ml	20	
	Cu	100*	1000ml	25	
	Fe	100*	1000ml	100	
	Pb	100*	1000ml	50	
	Mn	100*	1000ml	50	
	Ni	100*	1000ml	50	
	Sb***	50	1000ml	50	
	V	100*	1000ml	50	
	Zn	100*	1000ml	50	
K-MET SS2	HNO3	25.0	500ml	-	
	As	2.0	500ml	4	
	Cd	2.0	500ml	4	
	Pb	2.0	500ml	4	
	Se	2.0	500ml	4	
	Tl	2.0	500ml	4	
	Cu	2.0	500ml	4	
K-MET SS3	HNO3	25.0	500ml	-	
	As	50.0	500ml	100	
	Se	50.0	500ml	100	
	Tl	50.0	500ml	100	
	Hg	6	500ml	12	
K-MET SS4	HNO3	25	500ml	-	
	B	50	500ml	100	
	Mo	50	500ml	100	
K-MET SS5	HNO3	10.0	200ml	-	
	K**	20	200ml	1000	
	Na**	20	200ml	1000	
	Mg**	20	200ml	1000	
	Ca**	20	200ml	1000	



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K-MET GFLCSW	HNO3	10.0	1000ml	-	
	As, Pb, Se, Tl	5.0	1000ml	2.5	
	Cd	-	-	1.25	
	Cu	2.5	1000ml	2.5	
K-MET QCP-CICV-1	Ca, Mg, Na, K	no dilution	-	2500	
	Al, Ba	no dilution	-	1000	
	Fe	no dilution	-	500	
	Co, Mn, Ni, V, Zn	no dilution	-	250	
	Cu, Ag	no dilution	-	125	
	Cr	no dilution	-	100	
	Be	no dilution	-	25	
K-MET QCP-CICV-2	Sb	no dilution	-	500	
K-MET QCP-CICV-3	As, Pb, Se, Tl	no dilution	-	500	
	Cd	no dilution	-	250	

* Denotes volume of mixed stock standard.

** Denotes 10,000 ppm individual stock standards.

ALS Standard Operating Procedure

DOCUMENT TITLE:	DETERMINATION OF METALS AND TRACE ELEMENTS BY INDUCTIVELY COUPLED PLASMA-MASS SPECTROMETRY (METHOD 6020)
REFERENCED METHOD:	EPA 6020, 6020A
SOP ID:	MET-6020
REVISION NUMBER:	16
EFFECTIVE DATE:	1/01/2015



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DETERMINATION OF METALS AND TRACE ELEMENTS BY INDUCTIVELY COUPLED PLASMA-MASS SPECTROMETRY (METHOD 6020)

ALS-KELSO

SOP ID:	MET-6020	Rev. Number:	16	Effective Date:	1/01/2015
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Approved By:  Date: 12/9/14
Department Manager/Technical Director - Jeff Coronado

Approved By:  Date: 2/9/14
QA Manager - Lee Wolf

Approved By:  Date: 12/9/14
Laboratory Director - Jeff Grindstaff

Issue Date: _____ Doc Control ID#: _____ Issued To: _____

ANNUAL REVIEW

SIGNATURES BELOW INDICATE NO PROCEDURAL CHANGES HAVE BEEN MADE TO THE SOP SINCE THE APPROVAL DATE ABOVE. THIS SOP IS VALID FOR TWELVE ADDITIONAL MONTHS FROM DATE OF THE LAST SIGNATURE UNLESS INACTIVATED OR REPLACED BY SUBSEQUENT REVISIONS.

Signature _____	Title _____	Date _____
Signature _____	Title _____	Date _____
Signature _____	Title _____	Date _____
Signature _____	Title _____	Date _____



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DETERMINATION OF METALS AND TRACE ELEMENTS BY INDUCTIVELY COUPLED PLASMA-MASS SPECTROMETRY (METHOD 6020)

1. SCOPE AND APPLICATION

- 1.1. This procedure is used to determine the concentrations of certain elements in water, soil, tissues, aqueous and non-aqueous wastes, and sediment samples using EPA Method 6020 or 6020A. Table 1 indicates analytes that are typically determined by this procedure and lists the standard Method Reporting Limits (MRLs) for each analyte in water and soil. Project-specific MRLs may apply, and if lower than standard MRLs, it is demonstrated through method detection limit determinations and analysis of MRL standards that the MRL is achievable. Method Detection Limits (MDLs) that have been achieved are listed in Table 1. These may change as new studies are performed.
- 1.2. The complexity of the technique generally requires outside study of appropriate literature as well as specialized training by a qualified spectroscopist. The scope of this document does not allow for the in-depth descriptions of the relevant spectroscopic principles required for gaining a complete level of competence in this scientific discipline.

2. METHOD SUMMARY

- 2.1. Prior to analysis, samples must be digested using appropriate sample preparation methods. The digestate is analyzed for the elements of interest using ICP-mass spectrometry (ICP-MS).
- 2.2. Methods 6020 and 6020A describe the multi-elemental determination of analytes by ICP-MS. The method measures ions produced by a radio-frequency inductively coupled plasma. Analyte species originating in a liquid are nebulized and the resulting aerosol transported by argon gas into the plasma torch. The ions produced are entrained in the plasma gas and introduced, by means of an interface, into a mass spectrometer. The ions produced in the plasma are sorted according to their mass-to-charge ratios and quantified with a channel electron multiplier. Interferences must be assessed and valid corrections applied or the data flagged to indicate problems. Interference correction must include compensation for background ions contributed by the plasma gas, reagents, and constituents of the sample matrix.

3. DEFINITIONS

- 3.1. **Batch** - A batch of samples is a group of environmental samples that are prepared and/or analyzed together as a unit with the same process and personnel using the same lot(s) of reagents. It is the basic unit for analytical quality control.
 - 3.1.1. Preparation Batch - A preparation batch is composed of one to twenty field samples, all of the same matrix, and with a maximum time between the start of processing of the first and last samples in the batch to be 24 hours.
 - 3.1.2. Analysis Batch - Samples are analyzed in a set referred to as an analysis sequence. The sequence begins with instrument calibration (initial or continuing verification) followed by sample extracts interspersed with calibration standards (CCBs, CCVs, etc.) The



sequence ends when the set of samples has been analyzed or when qualitative and/or quantitative QC criteria indicate an out-of-control situation.

3.2. **Sample**

3.2.1. Field Sample - An environmental sample collected and delivered to the laboratory for analysis; a.k.a., client's sample.

3.2.2. Laboratory Sample - A representative portion, aliquot, or subsample of a field sample upon which laboratory analyses are made and results generated.

3.3. **Quality System Matrix** - The *matrix* of an environmental sample is distinguished by its physical and/or chemical state and by the program for which the results are intended. The following sections describe the matrix distinctions. These matrices shall be used for purpose of batch and quality control requirements.

3.3.1. Aqueous - Any groundwater sample, surface water sample, effluent sample, and TCLP or other extract. Specifically excluded are samples of the drinking water matrix and the saline/estuarine water matrix.

3.3.2. Drinking water - Any aqueous sample that has been designated a potable or potential potable water source.

3.3.3. Saline/Estuarine water - Any aqueous sample from an ocean or estuary or other salt-water source.

3.3.4. Nonaqueous Liquid - Any organic liquid with <15% settleable solids.

3.3.5. Animal tissue - Any tissue sample of an animal, invertebrate, marine organism, or other origin; such as fish tissue/organs, shellfish, worms, or animal material.

3.3.6. Solids - Any solid sample such as soil, sediment, sludge, and other materials with >15% settleable solids.

3.3.7. Chemical waste - Any sample of a product or by-product of an industrial process that results in a matrix not described in one of the matrices in Sections 3.4.1 through 3.4.6. These can be such matrices as non-aqueous liquids, solvents, oil, etc.

3.3.8. Miscellaneous matrices - Samples of any composition not listed in 3.4.1 - 3.4.7. These can be such matrices as plant material, paper/paperboard, wood, auto fluff, mechanical parts, filters, wipes, etc. Such samples shall be batched/grouped according to their specific matrix.

3.4. **Matrix Spike/Duplicate Matrix Spike (MS/DMS) Analysis** - In the matrix spike analysis, predetermined quantities of target analytes are added to a sample matrix prior to sample preparation and analysis. The purpose of the matrix spike is to evaluate the effects of the sample matrix on the method used for the analysis. Duplicate samples are spiked, and analyzed as a MS/DMS pair. Percent recoveries are calculated for each of the analytes detected. The relative percent difference (RPD) between the duplicate spikes (or samples) is calculated and used to assess analytical precision. The concentration of the spike should be at the midpoint of the calibration range or at levels specified by a project analysis plan.



- 3.5. Laboratory Duplicates (DUP) – Duplicates are additional replicates of samples that are subjected to the same preparation and analytical scheme as the original sample. The relative percent difference (RPD) between the sample and its duplicate is calculated and used to assess analytical precision.
- 3.6. Surrogate - Surrogates are organic compounds which are similar to analytes of interest in chemical composition, extraction and chromatography, but which are not normally found in environmental samples. The purpose of the surrogates is to evaluate the preparation and analysis of samples. These compounds are spiked into all blanks, standards, samples and spiked samples prior to extraction and analysis. Percent recoveries are calculated for each surrogate.
- 3.7. Method Blank (MB) - The method blank is an artificial sample composed of analyte-free water or solid matrix and is designed to monitor the introduction of artifacts into the analytical process. The method blank is carried through the entire analytical procedure.
- 3.8. Laboratory Control Samples (LCS) – The LCS is an aliquot of analyte free water or analyte free solid to which known amounts target analytes are added. The LCS is prepared and analyzed in exactly the same manner as the samples. The percent recovery is compared to established limits and assists in determining whether the batch is in control.
- 3.9. Independent Verification Standard (ICV) - A pre-mixed, purchased, second-source standard analyzed after the calibration curve. This is used to verify the validity of the initial calibration standards
- 3.10. Continuing Calibration Verification Standard (CCV) - A mid-level standard analyzed at specified intervals. Used to verify that the initial calibration curve is still valid for quantitative purposes.
- 3.11. Duplicates and Duplicate Matrix Spikes are additional replicates of samples that are subjected to the same preparation and analytical scheme as the original sample. Depending on the method of analysis, either a duplicate analysis (and/or a matrix spiked sample) or a matrix spiked sample and duplicate matrix spiked sample (MS/DMS) are analyzed.
- 3.12. Standard Reference Material (SRM) – A material with specific certification criteria and is issued with a certificate or certificate of analysis that reports the results of its characterizations and provides information regarding the appropriate use(s) of the material. An SRM is prepared and used for three main purposes: (1) to help develop accurate methods of analysis; (2) to calibrate measurement systems used to facilitate exchange of goods, institute quality control, determine performance characteristics, or measure a property at the state-of-the-art limit; and (3) to ensure the long-term adequacy and integrity of measurement quality assurance programs.

4. INTERFERENCES

- 4.1. Isobaric elemental interferences in ICP-MS are caused by isotopes of different elements forming atomic ions with the same nominal mass-to-charge ratio (m/z). A data system must be used to correct for these interferences. This involves determining the signal for another isotope of the interfering element and subtracting the appropriate signal from the analyte isotope signal. Attention should be given to circumstances where very high ion currents at adjacent masses may contribute to ion signals at the mass of interest. Matrices exhibiting a



significant problem of this type may require resolution improvement, matrix separation, or analysis using another isotope.

- 4.2. Isobaric molecular and doubly-charged ion interferences in ICP-MS are caused by ions consisting of more than one atom or charge, respectively. Most isobaric interferences that could affect ICP-MS determinations have been identified in the literature. Refer to Method 6020/A for further discussion.

5. SAFETY

- 5.1. All appropriate safety precautions for handling solvents, reagents and samples must be taken when performing this procedure. This includes the use of personnel protective equipment, such as, safety glasses, lab coat and the correct gloves.
- 5.2. Chemicals, reagents and standards must be handled as described in the ALS safety policies, approved methods and in MSDSs where available. Refer to the ALS Environmental, Health and Safety Manual and the appropriate MSDS prior to beginning this method.
- 5.3. Hydrochloric and/or Nitric Acid are used in this method. These acids are extremely corrosive and care must be taken while handling them. A face shield should be used while pouring acids. And safety glasses should be worn while working with the solutions. Lab coat and gloves should always be worn while working with these solutions.
- 5.4. High Voltage - The RF generator supplies up to 2000 watts to maintain an ICP. The power is transferred through the load coil located in the torch box. Contact with the load coil while generator is in operation will likely result in death. When performing maintenance on the RF generator, appropriate grounding of all HV capacitors must be performed as per manufacturer.
- 5.5. UV Light - The plasma is an intense source of UV emission, and must not be viewed with the naked eye. Protective lenses are in place on the instrument. Glasses with special protective lenses are available when direct viewing of the plasma is necessary.

6. SAMPLE COLLECTION, CONTAINERS, PRESERVATION AND STORAGE

- 6.1. Aqueous samples are typically collected in plastic containers. Aqueous samples are preserved with nitric acid ($\text{pH} < 2$), then refrigerated at $4 \pm 2^\circ\text{C}$ from receipt until digestion. Soil or solid samples may be collected in plastic or glass jars. Non-aqueous samples are refrigerated at $4 \pm 2^\circ\text{C}$ from receipt until digestion.
- 6.2. Samples are prepared via procedures in SOPs MET-DIG, MET-3020A, or MET-3050 depending on matrix and project specifications.
- 6.3. Digestates are stored in the appropriate volumetric containers. Following analysis, digestates are stored until all results have been reviewed. Digestates are neutralized prior to disposal through the sewer system, 2 weeks after data is reviewed.

7. STANDARDS, REAGENTS, AND CONSUMABLE MATERIALS



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- 7.1. All standards are prepared from NIST traceable standards. The expiration dates are assigned according to the EPA method and the vendor's assigned expiration dates. For example, working ICS solutions are prepared weekly in accordance with Method 6020, Section 5.6.1.
- 7.1.1. 1000 ppm Single Element Stock Standard Solutions: Each stock standard is store at room temperature on shelves located in room 113 of the metals lab. The manufacturer, lot number, and expiration date of each stock standard is recorded in a bound logbook also located in room 113. Additionally each stock standard is given a unique, identifying name.
- 7.1.2. Intermediate Standard Solutions: Intermediate mixed stock solutions are made from the individual stock standards described above. The individual component of each mixed solution is recorded in a bound logbook located in the ICP-MS laboratory and mixed solution is given a unique, identifying name. The expiration date for the intermediate standard is the earlier of any one of its stock components.
- 7.1.3. Calibration Standards: Calibration standards are made fresh daily from the intermediate standard solutions. Each individual intermediate standard used in the calibration standard is recorded in a bound logbook located in the ICP-MS laboratory, and the calibration standard solution is given a unique, identifying name. The calibration standards unique name is used on the raw data to link the data to the subsequent prepared standards and ultimately the original purchased stock standard.
- 7.2. Standards Preparation
- 7.2.1. Expiration of all standard solutions defaults to the earliest expiration date of an individual component unless otherwise specified.
- 7.2.2. Calibration Standards
- The calibration standard is prepared from two intermediate stock solutions. These solutions are prepared in acid rinsed 1000 mL Class A volumetric flasks following the formulations laid out on the attached example standard sheet (see Attachments). The working calibration standard is made daily by aliquoting 2.5 mL of each of the intermediate solutions in to a 100 mL Class A volumetric flask and diluting to volume with 1% HNO₃. This standard is also used as the Continuing Calibration Verification (CCV).
- 7.2.3. Initial Calibration Verification (ICV)
- 7.2.3.1. The ICV intermediate stock solution is prepared in an acid rinsed 100 mL Class A volumetric flask. The solution is prepared by adding 2.0 mL of Inorganic Ventures QCP-CICV-1, 1.0 mL each of QCP-CICV-2 and QCP-CICV-3, 0.5 mL of 1000 ppm Molybdenum stock solution, 0.5 mL of 1000 ppm Uranium stock solution, and 0.5mL of 1000ppm B, Bi, Sr, Ti solution and diluting to volume with 1% HNO₃.
- 7.2.3.2. The working ICV solution is prepared by aliquoting 0.5 mL of the mixed ICV intermediate solution into an acid rinsed 100 mL Class A volumetric flask and diluting to volume with 1% HNO₃.
-



NOTE: The ICV solution is not at the midpoint of the linear range which may be as high as 1000 µg/L for some elements. The ICV solution used is a premixed standard purchased from Inorganic Ventures and contains the elements of interest between 2.5 and 100 µg/L. This solution provides calibration confirmation at more representative levels, given that most ICP-MS analyses are quantifying analytes in the low-ppb to sub-ppb range.

7.2.4. Interference Check Solutions (ICSA and ICSAB)

7.2.4.1. The ICSA is prepared in an acid rinsed 50 mL Class B volumetric flask by aliquoting 1.0 mL of Elements ICSAm (CS-CAK02) solution and diluting to volume with 1% HNO₃.

7.2.4.2. The ICSAB is prepared in an acid rinsed 50 mL Class B volumetric flask by aliquoting 1.0 mL of Elements ICSAm (CS-CAK02), 0.125 mL of Inorganic Ventures 6020ICS-9B, and 0.250 mL of 10 ppm Molybdenum solutions and diluting to volume with 1% HNO₃.

7.2.5. Post-digestion spikes are performed by adding appropriate amounts of the calibration intermediate solutions to aliquots of the sample digestate. The volumes of each standard used vary based on the native concentrations found in the field samples. Refer to the post-digestion spike in Section 12 for details.

7.2.6. Refer to the appropriate digestion SOP for details of LCSW and matrix spike solution composition and preparation.

7.2.7. Tuning / Mass Calibration Solution

7.2.7.1. A 1 ppm intermediate solution containing Be, Bi, Ce, Co, In, Li, Pb, Mg, and U is prepared by adding 1.0 mL of each from 1000 ppm stock standards to an acid rinsed 1000 mL volumetric flask and diluting to volume with 1% nitric acid. The expiration date for the intermediate solution is the earliest of any one of its stock components.

7.2.7.2. The working solution is prepared in three ways:

- For the Agilent: a 1.0 ppb tune/mass calibration solution is prepared by adding 1.0 mL of intermediate solution to an acid rinsed 1000 mL volumetric flask and diluting to volume with 1% nitric acid.
- For the X-Series (K-ICP-MS-03) instrument a 5.0 ppb tune/mass calibration solution is prepared by adding 5.0 mL of intermediate solution to an acid rinsed 1000 mL volumetric flask and diluting to volume with 1% nitric acid.
- For the NexION (K-ICP-MS-04) instrument a 2.0 ppb tune/mass calibration solution is prepared by adding 2.0 mL of intermediate solution to an acid rinsed 1000 mL volumetric flask and diluting to volume with 1% nitric acid.
- The expiration date for this solution is taken from the intermediate stock above.

7.3. Internal Standards Stock Solution – Prepare solutions by adding appropriate amounts of each 1000 ppm single element stock solution to a acid rinsed 1000 mL volumetric flask and diluting to volume with 1% nitric. Use this solution for addition to blanks, calibration standards



and samples at a ratio of 0.5 mL of internal standard to 100 mL of solution, or dilute by an appropriate amount using 1% (v/v) nitric acid, if the internal standards are being added by peristaltic pump. The typical solutions are:

- XSeries instrument: 50ppb Li; 25ppb Sc, Ga, Y; 10ppb Rh, In, Lu, Tm, Th.
- Agilent instrument: 2ppm Li, Sc, Y, Ga, Ge, Ce, Tm, In, Lu, Th
- NexION instrument: 30ppb In, Tm, Lu, Th; 60ppb Li, Rh, Au; 75ppb Sc; 100ppb Ga,Y; 500ppb Ge

7.4. Additional Reagents

7.4.1. Reagent water, ASTM Type II

7.4.2. "OmniTrace Ultra" Concentrated Nitric Acid (EM Science # NX0408-2)

7.4.3. Argon (Airgas Industrial Grade – 99.999% pure, bulk delivered)

8. APPARATUS AND EQUIPMENT

8.1. ICP/MS instruments:

- | | |
|--------------------|---|
| 8.1.1. Instrument: | Thermo Electron X-Series |
| Nebulizer: | Conikal |
| Spray Chamber: | VG Peltier-cooled |
| Cones: | Nickel Sampler (1.0 mm orifice)
Nickel Skimmer (0.75 mm orifice) |
| 8.1.2. Instrument: | NexION 300D |
| Nebulizer: | PFA-ST Microflow |
| Spray Chamber: | Cyclonic, Peltier-cooled |
| Cones: | Nickel Sampler (1.0 mm orifice)
Nickel Skimmer (0.75 mm orifice) |
| 8.1.3. Instrument: | Agilent 7700 |
| Nebulizer: | MicroMist |
| Spray Chamber: | Double Pass quartz spray chamber |
| Cones: | Nickel Sampler (1.0 mm orifice)
Nickel Skimmer (0.75 mm orifice) |

9. PREVENTIVE MAINTENANCE

- 9.1. All maintenance is documented in the instrument logbook. ALS/Kelso maintains a service contract with the instrument manufacturer that allows for an unlimited number of service calls and full reimbursement of all parts and labor.
- 9.2. Most routine maintenance and troubleshooting is performed by ALS staff. Preventive maintenance activities listed below should be performed when needed as determined by instrument performance (i.e. stability, sensitivity, etc.) or by visual inspection. Other maintenance or repairs may, or may not require factory service, depending on the nature of the task.



- cone removal and cleaning
- removal and cleaning of ICP glassware and fittings
- checking and cleaning RF contact strips
- checking air filters and cleaning if necessary
- checking the oil mist filters and cleaning if necessary
- checking the rotary pump oil and adding or changing if necessary
- removal and cleaning of extraction lens
- removal and cleaning of ion lens stack
- replace the electron multiplier as necessary

10. RESPONSIBILITIES

- 10.1. It is the responsibility of the analyst to perform the analysis according to this SOP and to complete all documentation required for data review. Analysis and interpretation of the results are performed by personnel in the laboratory who have demonstrated the ability to generate acceptable results utilizing this SOP. This demonstration is in accordance with the training program of the laboratory. Final review and sign-off of the data is performed by the department supervisor/manager or designee.
- 10.2. It is the responsibility of the department supervisor/manager to document analyst training. Documenting method proficiency, as described in the SOP for Documentation of Training, is also the responsibility of the department supervisor/manager.

11. PROCEDURE

- 11.1. Refer to method 6020 (or 6020A) and the instrument manuals for detailed instruction on implementation of the following daily procedures preceding an analytical run.
- 11.2. After the instrument has been placed in the "Operate" mode, begin completing the daily instrument log (see Attachments). Refer to the instrument manuals for the optimum settings for each instrument.
- 11.3. The following parameters are monitored to assure awareness of changes in the instrumentation that serve as signals that optimum performance is not being achieved, or as indicators of the physical condition of certain consumable components (i.e. EMT and cones).
- 11.3.1. Multiplier Voltages
 - 11.3.2. Gas Flows - Coolant Ar
 - 11.3.3. The nebulizer and auxiliary flows are adjusted later as part of the optimizing procedure.
- 11.4. Optimization
- 11.4.1. Gas Flows
 - 11.4.1.1. Allow a period of not less than 30 minutes for the instrument to warm up.



11.4.1.2. Aspirate a mixed tune solution into the plasma and monitor the instrument output signal of In at mass 115 on the ratemeter. Adjust the nebulizer and auxiliary flows to obtain maximum signal. Adjust the tension screw on the peristaltic pump to obtain minimum noise in the analytical signal. Record flow rates and note any large variances.

Note: Significant differences in flow rates will be observed for different torches and cones.

11.4.2. Tuning

11.4.2.1. Ion Lens Setting - While monitoring the output signal of a mixed tune solution at mass 115 on the ratemeter, adjust the ion lenses to obtain maximum sensitivity. Refer to the instrument manual for details on performing the adjustments.

11.4.2.2. Mass Calibration - Aspirate the tune / mass calibration solution described in section 7.2 and perform the mass calibration using the instrument's Mass Calibration program. (Refer to the instrument manual for details pertaining to the mass calibration procedure.) The acceptance criteria for the mass calibration is <0.1 amu from the true value. If the mass calibration fails criteria re-tune the instrument and perform the mass calibration procedure again.

11.4.2.3. Resolution Check - Using the spectra created during the mass calibration procedure; perform the resolution check to assure the resolution is less than 0.9 AMU at 5% peak height. If the resolution does not pass criteria adjust the instrument's resolution settings, run a new scan of the mass calibration solution and recheck.

11.4.2.4. Stability Check - Using the tune / mass calibration solution, perform a short-term stability check as per EPA Method 6020 or 6020A. The relative standard deviations of five scans for each element in the tune solution must be $< 5\%$. If the test does not pass criteria determine the cause (i.e. dirty cones, improper tune, etc.) correct the problem and re-run the test.

11.5. Analytical Run

11.5.1. Calibrate the instrument using a calibration blank (Standard 0), composed of reagent water and 1% nitric acid, and the working calibration standard (8.2.2). The masses typically monitored and those used for quantification are listed in Table 2. These masses are set as defaults in the instrument's analytical procedures. To begin select the correct method. Nebulize Standard 0 (Blank) into the plasma. Allow 1-2 minutes for system to equilibrate prior to establishing baseline. Follow directions on computer screen to perform standardization. Nebulize the working calibration standard into the plasma. The operator must sign and date the first page of standardization.

11.5.2. After the first CCB and before the ICS standards a CRA (MRL / LLICV / LLCCV) standard is analyzed. Method 6020 requires the detection to be $>$ the MDL but $< 2x$ the MRL. For 6020A, the criteria are 70-130% recovery. For DoD projects, the CRA criteria are 80-120%.



Note: For 6020A the LLCCV must also be analyzed at the end on the analytical run sequence.

- 11.5.3. Perform the analysis in the order listed below. A daily run log of all samples analyzed is maintained.

Initial Calibration Verification (ICV)
Continuing Calibration Verification (CCV)
Initial Calibration Blank (ICB)
Continuing Calibration Blank (CCB)
CRA (MRL / LLICV / LLCCV)
ICSA
ICSAB
Analyze 10 Samples
CCV
CCB
Analyze 10 Samples
CCV
CCB

Repeat sequence as required to complete analytical run, analyzing CCVs/CCBs every 10 analyses and at the end of the run.

12. QA/QC REQUIREMENTS

12.1. Initial Precision and Recovery Validation

The accuracy and precision of the procedure must be validated before analysis of samples begins, or whenever significant changes to the procedures have been made. To do this, four LCS aliquots are prepared and analyzed. The average percent recovery of for each analyte must be 85-115% (for water, and within the LCS limits for soils) and the RSD <20%.

12.2. Method Detection Limits

12.2.1. A method detection limit (MDL) study must be undertaken before analysis of samples can begin. To establish detection limits that are precise and accurate, the analyst must perform the following procedure. Spike a minimum of seven blank matrices at a level near or below the MRL. Follow the procedures starting in Section 11 to analyze the samples. Refer to CE-QA011, *Performing Method Detection Limit Studies and Establishing Limits of Detection and Quantification* details of performing the MDL study.

12.2.2. Calculate the average concentration found (x) and the standard deviation of the concentrations for each analyte. Calculate the MDL for each analyte using the correct T value for the number of replicates. MDL's must be verified annually or whenever there is a significant change in the background or instrument response.

- 12.3. For method 6020A, an LLQC sample (a CRA that is carried through the digestion) must be analyzed to verify accuracy at the MRL. The recovery must be 70-130%.



-
- 12.4. Instrument Detection Limits (IDLs) and linear ranges studies are performed quarterly. These will be calculated and made available to the ICP-MS operator. Linear range studies determine the Linear Dynamic Range (LDR) of the each instrument by analysis of a high concentration standard with results with $\pm 10\%$ of the expected value. For non-DoD projects samples may be quantified between the MRL and 90% of the LDR without flagging. The Linear Calibration Range (LCR) is established by the highest calibration standard.
- **Note:** IDLs must be $< \text{LOD}$ for DOD projects. DoD project samples with concentrations above the calibration standard must be diluted to bring results within the quantitation range. The LOQ and cal standard establish the quantitation range. The lab may report a sample result above quantitation range if the lab runs and passes a CCV that is $>$ sample result.
- 12.5. The Initial Calibration Verification (ICV) standard is analyzed immediately after calibration. The results of the ICV must agree within $\pm 10\%$ of the expected value. If the control limits are exceeded, the problem will be identified and the instrument recalibrated.
- 12.6. A Continuing Calibration Verification (CCV) and Continuing Calibration Blank (CCB) are analyzed after calibration then every 10 samples thereafter with a final CCV/CCB closing the final samples of the analytical run.
- 12.6.1. The results of the CCV must agree within $\pm 10\%$ of the expected value.
 - 12.6.2. The CCB measured values must be less than the MRL / LOQ for each element for standard applications. Other project-specific criteria may apply (for DoD QSM projects CCB can have no analytes $>$ the LOD).
 - 12.6.3. If the control limits are exceeded, the problem will be identified and corrective action taken. The instrument recalibrated. The previous 10 samples must be reanalyzed.
- 12.7. The ICSA and ICSAB solutions are analyzed after calibration and before any field samples. The solutions are then reanalyzed every 12 hours. Results of the ICSA are used by the analyst to identify the impact of potential interferences on the quality of the data. Based on these results appropriate action should be taken when interferences are suspected in a field sample including, but not limited to, selecting an alternative isotope for quantification, manual correction of the data, elevating the MRL, selection of an alternative method (e.g. optical ICP, GFAA) or flagging the result as estimated when no other action is possible. Results for the spiked analytes in the ICSAB solution must agree with $\pm 20\%$ of the expected value.

**INTERFERENCE CHECK SAMPLE COMPONENTS AND
CONCENTRATIONS**



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	Solution A	Solution B
	<u>Concentrations (mg/L)</u>	<u>Concentrations (mg/L)</u>
Al	20.0	20.0
Ca	60.0	60.0
Fe	50.0	50.0
Mg	20.0	20.0
Na	50.0	50.0
P	20.0	20.0
K	20.0	20.0
S	20.0	20.0
C	40.0	40.0
Cl	424	424
Mo	0.05	0.05
Ti	0.40	0.40
As	0.0	0.025
Cd	0.0	0.025
Cr	0.0	0.050
Co	0.0	0.050
Cu	0.0	0.050
Mn	0.0	0.050
Ni	0.0	0.050
Se	0.0	0.025
Ag	0.0	0.0125
V	0.0	0.050
Zn	0.0	0.025

NOTE: The concentration of interfering elements in the ICSA and ICSAB solutions are spiked at levels 5 times lower than recommended in Table 1 of Method 6020A. Running the full strength solutions as described in 6020A introduces too much material approximately 0.35 % dissolved solids into the ICP-MS system when trying to conduct low level analysis. Since the ICP-MS instrumentation is able to handle a maximum of 0.2% solids, the 6020A ICSA solution is higher in interfering components than any sample that would run through the instrument. However, the ICS solutions will be analyzed at levels that will provide approximately 0.1% dissolved solids.

- 12.8. Internal standards are used to correct for physical interferences. Masses used as internal standards include; ^{71}Ga , ^{115}In , ^6Li , ^{175}Lu , ^{103}Rh , ^{45}Sc , and ^{89}Y . These internal standards are used in combination to cover the appropriate mass ranges. Internal standard correction is applied to the analytical isotopes via interpolation of the responses from nearest internal standard isotopes (Thermo instruments) or direct correlation of analyte to IS (NexION). This function is performed in real-time by the instruments operating system. Internal standards must be run within 50 AMU of the masses that are analyzed. Internal standard recoveries must fall between 30% and 125% when running method 6020, or 70-125% when running method 6020A Revision 1. If not, then the sample must be reanalyzed after a fivefold or greater dilution has been performed.



- 12.9. A method blank is digested and analyzed with every batch of 20 (or fewer) samples to demonstrate that there are no method interferences. If the method blank shows any hits above the MRL for standard applications, or $> \frac{1}{2}$ the MRL for DoD projects or $> \frac{1}{10}$ the sample result, corrective action must be taken. The MB can only be rerun once. Corrective action includes recalculation, reanalysis, system cleaning, or re-extraction and reanalysis.
- 12.10. Laboratory Control Samples are analyzed at a frequency of 5% or one per batch, whichever is greater. Refer to the current ALS-Kelso DQO spreadsheets for the LCS limits. For method 6020A, the LCS recovery limits are 80-120%. If statistical in-house limits are used, they must fall within the 80-120% range. Project, QAPP, or client-specific control limits may supersede the limits listed, but laboratory limits should be consistent with specified limits in order to establish that the specified limits can be achieved. If the control limits are exceeded, the associated batch of samples will be re-digested and reanalyzed.
- 12.11. A digested duplicate and matrix spike are analyzed at a frequency of 5% or one per batch, whichever is greater. Refer to the current ALS-Kelso DQO spreadsheets for the matrix spike limits. The matrix spike recovery and relative percent difference will be calculated while analysis is in progress. Project, QAPP, or client-specific control limits may supersede the limits listed. If the control limits are exceeded, the samples will be re-digested and reanalyzed, unless matrix interference or sample non-homogeneity is established as cause. In these instances, the data and the report will be flagged accordingly.
- 12.12. A Matrix Spike sample is digested one per batch, or per 20 samples (i.e. 5%). Default spike concentrations are listed in the sample digestion SOPs. Spike concentrations may be adjusted to meet project requirements. The matrix spike recovery will be calculated while the job is in progress. Where specified by project requirements, a matrix spike duplicate may be required. Matrix spike recovery criteria are derived from lab data. For method 6020A, the recovery limits are 75-125%. If statistical in-house limits are used, they must fall within the 75-125% range. In some cases, project-specific QC limits may be required. Unless specified otherwise, for DoD QSM projects the project LCS criteria will be used for evaluation of matrix spikes. If an analyte recovery is outside acceptance limits proceed with the additional quality control tests described in sections 12.13 and 12.14. Based on results of these tests, the physical nature of the sample (e.g. homogeneity), and any specific project requirements, a determination can then be made as to appropriate corrective action (e.g. re-digestion, reporting with a qualifier, alternative methodologies, etc.). If the analyte concentration is $> 4x$ the spike level the spike control limit is no longer applicable and no action is required. For specifics on the preparation and composition of matrix spike solutions refer to the appropriate digestion SOP.
- Note:** For DOD projects a MS/MSD is required with every extraction batch. The %RSD should be $< 20\%$.
- 12.13. Post Digestion Spike Test: When analysis is conducted via 6020 a post digestion spike must be performed for each matrix and each batch of sample. The prepared sample or its dilution is spiked for each element of interest at a concentration sufficiently high to be observed. Typically 20 μ L of 10,000 ppb intermediate stock is added to a 10 mL aliquot of sample. If analyte concentrations are elevated in the sample, spiking at a higher concentration may be required. The post spike should be recovered to within 75-125% of the known value or within the laboratory derived acceptance criteria. When analysis is conducted via 6020A, the post digestion spike test is performed whenever matrix spike or replicate criteria are exceeded. An analyte spike is added to a portion of a prepared sample, or its dilution, and should be recovered to within 80% to 120% of the known value. If this spike fails, then the dilution test



(Sec. 12.14) should be run on this sample. If both the matrix spike and the post digestion spike fail, then matrix effects are confirmed.

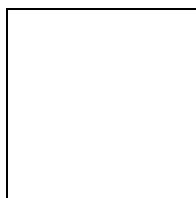
- 12.14. Dilution Test: When analysis is conducted via 6020, a serial dilution test must be performed for each matrix and each batch of sample. For sample concentrations that are sufficiently high (minimally, a factor of greater than 100 times the MDL), the analysis of a fivefold (1+4) dilution must agree within $\pm 10\%$ of the original determination. When analysis is conducted via 6020A, the dilution test is performed whenever matrix spike or replicate criteria and post digestion spike criteria are exceeded. If the dilution test fails then a chemical or physical effect should be suspected. Corrective action can include additional dilution of the sample, the use of alternate methodologies, etc. or the data can be flagged and reported. The exact course of action will be dependent on the nature of the samples and project requirements and should be discussed with the project manager.
- 12.15. Instrument blanks should be evaluated for potential carryover and rinse times need to bring the analyte signal to within the CCB criteria discussed above in section 12.6.2. Results from instrument blanks run after standards or control samples should be used to establish levels at which carryover in samples may occur. Samples exhibiting similar effects of carryover should be reanalyzed.
- 12.16. Refer to the Quality Control section of EPA Methods 6020 and 6020A for additional information describing required QA/QC. Note that the nomenclature of certain QC samples in the method differs from that of the CLP SOW, but the function of those samples is equivalent in both cases.

13. DATA REDUCTION AND REPORTING

13.1. Calculations

Calculate sample results using the data system printouts and digestion information. the digestion and dilution information is entered into the data system. The data system then uses the calculations below to generate a sample result.

Aqueous samples are reported in $\mu\text{g/L}$:



C^* = Concentration of analyte as measured at the instrument in $\mu\text{g/L}$ (in digestate).

Solid samples are reported in mg/Kg :

$$\text{mg/Kg (Sample)} = C^* \times \text{Post Digestion Dilution Factor} \times \frac{\text{Digestion Vol. (ml)}}{\text{Sample wt. (g)}} \times \frac{1 \text{ mg}}{1000 \mu\text{g}} \times \frac{1 \text{ L}}{1000 \text{ ml}} \times \frac{1000 \text{ g}}{1 \text{ Kg}}$$



C* = Concentration of analyte as measured at the instrument in ug/L (in digestate).

NOTE: If results are to be reported on a dry weight basis, determine the dry weight of a separate aliquot of the sample, using the SOP for Total Solids.

- 13.2. Common isobaric interferences are corrected using equations equivalent to those listed in EPA Methods 6020, 6020A, and 200.8. Monitoring of multiple isotopes for a single element provides a mechanism for identifying isobaric interferences. Refer to the Interferences section of EPA methods for additional descriptions of possible interferences and the mechanisms required for adequately compensating for their effects.
- 13.3. Data Review and Reporting
- 13.3.1. The ICP-MS operator reviews the MS data and signs and dates the Data Review Form. A qualified senior staff spectroscopist performs a secondary review of the data and the Data Review Form is signed and dated. The data is then delivered to the report generation area where it is filed in the service request file. Once all of the data for the service request is complete, a CAR is generated.
- 13.3.2. The data is saved on the local hard drive and is also copied to the appropriate directory on the network. The data directories are located at r:\icp\wip\data. The data is kept on the local directory for 1 month. The network files are periodically backed up on disc or network tape.
- 13.3.3. For “non-production” work (such as method development or research/development studies) the analyses are performed under the direction of a senior spectroscopist. All associated data is scrutinized by the senior spectroscopist. Original raw data and associated records are archived in the analytical project file.
- 13.3.4. The final review and approval of all data is performed by qualified spectroscopists.

14. CONTINGENCIES FOR HANDLING OUT-OF-CONTROL OR UNACCEPTABLE DATA

- 14.1. Refer to the SOP for *Nonconformity and Corrective Action* (CE-QA008) for procedures for corrective action. Personnel at all levels and positions in the laboratory are to be alert to identifying problems and nonconformities when errors, deficiencies, or out-of-control situations are detected.
- 14.2. Handling out-of-control or unacceptable data
- 14.2.1. On-the-spot corrective actions that are routinely made by analysts and result in acceptable analyses should be documented as normal operating procedures, and no specific documentation need be made other than notations in laboratory maintenance logbooks, runlogs, for example.
- 14.2.2. Some examples when documentation of a nonconformity is required using a Nonconformity and Corrective Action Report (NCAR):
- Quality control results outside acceptance limits for accuracy and precision
 - Method blanks or continuing calibration blanks (CCBs) with target analytes above acceptable levels



-
- Sample holding time missed due to laboratory error or operations
 - Deviations from SOPs or project requirements
 - Laboratory analysis errors impacting sample or QC results
 - Miscellaneous laboratory errors (spilled sample, incorrect spiking, etc.)
 - Sample preservation or handling discrepancies due to laboratory or operations error

15. METHOD PERFORMANCE

- 15.1. This method was validated through single laboratory studies of accuracy and precision. Refer to the reference method for additional available method performance data.
- 15.2. The method detection limit (MDL), limit of detection (LOD) and limit of quantitation (LOQ) are established using procedures described in CE-QA011, *Performing Method Detection Limit Studies and Establishing Limits of Detection and Quantification*. Method Reporting Limits are established for this method based on MDL studies and as specified in the ALS, Kelso Quality Assurance Manual.

16. POLLUTION PREVENTION AND WASTE MANAGEMENT

- 16.1. It is the laboratory's practice to minimize the amount of solvents, acids, and reagents used to perform this method wherever feasibly possible. Standards are prepared in volumes consistent with methodology and only the amount needed for routine laboratory use is kept on site. The threat to the environment from solvents and/or reagents used in this method can be minimized when recycled or disposed of properly.
- 16.2. The laboratory will comply with all Federal, State, and local regulations governing waste management, particularly the hazardous waste identification rules and land disposal restrictions as specified in the ALS Environmental Health and Safety Manual.
- 16.3. This method uses acid. Waste acid is hazardous to the sewer system and to the environment. All acid waste must be neutralized to a pH of 5-9 prior to disposal down the drain. The neutralization step is considered hazardous waste treatment and must be documented on the treatment by generator record. See the ALS *EH&S Manual* for details.

17. TRAINING

- 17.1. Refer to the SOP ADM-TRAIN, *ALS-Kelso Training Procedure* for standard procedures.
- 17.2. A minimum of two senior level spectroscopists are to be maintained on staff at all times. Senior spectroscopists are defined as individuals with a minimum of ten years combined education and experience in, or related to atomic spectroscopy. Of those ten years, a minimum of two years of ICP-MS experience is required.
- 17.3. All technical staff is encouraged to attend one technical seminar per year. In addition to the technical seminars, senior spectroscopists are required to complete a one week training session offered by the instrument manufacturer.
- 17.4. On-the-job-training occurs daily with the senior spectroscopists providing direction to new operators. The physical operation of the equipment is relatively simple. The data reduction



and troubleshooting requires extensive experience that can only be gained by hands-on operation of the instrument and assisted evaluation of raw data.

17.5. Training outline

17.5.1. Review literature (see references section). Read and understand the SOP. Also review the applicable MSDS for all reagents and standards used. Following these reviews, observe the procedure as performed by an experienced analyst at least three times.

17.5.2. The next training step is to assist in the procedure under the guidance of an experienced analyst. During this period, the analyst is expected to transition from a role of assisting, to performing the procedure with minimal oversight from an experienced analyst.

17.5.3. Perform initial precision and recovery (IPR) study as described above for water or soil samples. Summaries of the IPR are reviewed and signed by the supervisor. Copies may be forwarded to the employee's training file. For applicable tests, IPR studies should be performed in order to be equivalent to NELAC's Initial Demonstration of Capability.

17.6. Training and proficiency is documented in accordance with the SOP ADM-TRAIN.

18. METHOD MODIFICATIONS

18.1. There are no known modifications in this laboratory standard operating procedure from the reference method.

19. REFERENCES

19.1. USEPA, Test Methods for Evaluating Solid Waste, SW-846, 3rd Edition, Update III Method 6020, Revision 0, September 1994.

19.2. USEPA, Test Methods for Evaluating Solid Waste, SW-846, Update IV, Method 6020A, Revision 1, February 2007.

19.3. Agilent and Thermo Elemental Instrument Manuals

20. CHANGES SINCE THE LAST REVISION

20.1. Reformatted SOP to current ALS format.

20.2. Minor changes (correct typos and errors, etc.) throughout SOP.

20.3. Section 1 – revised to eliminate redundant language.

20.4. Section 7.2.7.2 – updated to replace Excell with Agilent

20.5. Section 7.3 – revised to list specific internal standards and concentrations.

20.6. Section 8.1 – updated instrument information.

20.7. Section 11.4.2.3 – revised to correct peak height %.

20.8. Sections 12.10 and 12.12 – revised to refer to DQO tables for QC limits

20.9. Section 16 – revised to include default language.

20.10. Section 17 – revised to include default language and be consistent with 200.8 SOP.

20.11. Table 1 – updated.

20.12. Attachments updated.



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TABLE 1
TARGET ANALYTES, MDLs, and MRLs

METHOD	PREP METHOD	ANALYTE	MATRIX	MDL	MRL mg/kg
6020A	EPA 3050B	Aluminum	Soil	0.6	2
6020A	EPA 3050B	Antimony	Soil	0.02	0.05
6020A	EPA 3050B	Arsenic	Soil	0.2	0.5
6020A	EPA 3050B	Barium	Soil	0.02	0.05
6020A	EPA 3050B	Beryllium	Soil	0.005	0.02
6020A	EPA 3050B	Bismuth	Soil	0.02	0.05
6020A	EPA 3050B	Boron	Soil	0.05	0.5
6020A	EPA 3050B	Cadmium	Soil	0.009	0.02
6020A	EPA 3050B	Chromium	Soil	0.07	0.2
6020A	EPA 3050B	Cobalt	Soil	0.009	0.02
6020A	EPA 3050B	Copper	Soil	0.04	0.1
6020A	EPA 3050B	Lead	Soil	0.02	0.05
6020A	EPA 3050B	Manganese	Soil	0.02	0.05
6020A	EPA 3050B	Molybdenum	Soil	0.02	0.05
6020A	EPA 3050B	Nickel	Soil	0.04	0.2
6020A	EPA 3050B	Selenium	Soil	0.2	1
6020A	EPA 3050B	Silver	Soil	0.005	0.02
6020A	EPA 3050B	Thallium	Soil	0.002	0.02
6020A	EPA 3050B	Tin	Soil	0.02	0.1
6020A	EPA 3050B	Uranium	Soil	0.003	0.02
6020A	EPA 3050B	Vanadium	Soil	0.08	0.2
6020A	EPA 3050B	Zinc	Soil	0.2	0.5



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TABLE 1 – continued

METHOD	PREP METHOD	ANALYTE	MATRIX	MDL	MRL ug/L
6020A	MET-DIG (CLP)	Aluminum	Water	0.2	2
6020A	MET-DIG (CLP)	Antimony	Water	0.01	0.05
6020A	MET-DIG (CLP)	Arsenic	Water	0.05	0.5
6020A	MET-DIG (CLP)	Barium	Water	0.006	0.05
6020A	MET-DIG (CLP)	Beryllium	Water	0.008	0.02
6020A	MET-DIG (CLP)	Bismuth	Water	0.005	0.05
6020A	MET-DIG (CLP)	Boron	Water	0.07	0.5
6020A	MET-DIG (CLP)	Cadmium	Water	0.005	0.02
6020A	MET-DIG (CLP)	Chromium	Water	0.02	0.2
6020A	MET-DIG (CLP)	Cobalt	Water	0.006	0.02
6020A	MET-DIG (CLP)	Copper	Water	0.03	0.1
6020A	MET-DIG (CLP)	Iron	Water	0.3	1
6020A	MET-DIG (CLP)	Lead	Water	0.004	0.02
6020A	MET-DIG (CLP)	Manganese	Water	0.006	0.05
6020A	MET-DIG (CLP)	Molybdenum	Water	0.008	0.05
6020A	MET-DIG (CLP)	Nickel	Water	0.04	0.2
6020A	MET-DIG (CLP)	Selenium	Water	0.4	1
6020A	MET-DIG (CLP)	Silver	Water	0.005	0.02
6020A	MET-DIG (CLP)	Thallium	Water	0.005	0.02
6020A	MET-DIG (CLP)	Tin	Water	0.01	0.05
6020A	MET-DIG (CLP)	Uranium	Water	0.003	0.02
6020A	MET-DIG (CLP)	Vanadium	Water	0.05	0.2
6020A	MET-DIG (CLP)	Zinc	Water	0.09	0.5



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TABLE 1 – continued

METHOD	PREP METHOD	ANALYTE	MATRIX	MDL	MRL mg/kg
6020A	PSEP TISSUE	Aluminum	Tissue	0.2	2
6020A	PSEP TISSUE	Antimony	Tissue	0.002	0.05
6020A	PSEP TISSUE	Arsenic	Tissue	0.02	0.5
6020A	PSEP TISSUE	Barium	Tissue	0.005	0.05
6020A	PSEP TISSUE	Beryllium	Tissue	0.003	0.02
6020A	PSEP TISSUE	Bismuth	Tissue	0.003	0.05
6020A	PSEP TISSUE	Boron	Tissue	0.2	2
6020A	PSEP TISSUE	Cadmium	Tissue	0.002	0.02
6020A	PSEP TISSUE	Chromium	Tissue	0.02	0.2
6020A	PSEP TISSUE	Cobalt	Tissue	0.003	0.02
6020A	PSEP TISSUE	Copper	Tissue	0.02	0.1
6020A	PSEP TISSUE	Iron	Tissue	0.2	1
6020A	PSEP TISSUE	Lead	Tissue	0.0005	0.02
6020A	PSEP TISSUE	Manganese	Tissue	0.008	0.05
6020A	PSEP TISSUE	Molybdenum	Tissue	0.008	0.05
6020A	PSEP TISSUE	Nickel	Tissue	0.02	0.2
6020A	PSEP TISSUE	Selenium	Tissue	0.2	1
6020A	PSEP TISSUE	Silver	Tissue	0.006	0.02
6020A	PSEP TISSUE	Thallium	Tissue	0.0009	0.02
6020A	PSEP TISSUE	Tin	Tissue	0.003	0.05
6020A	PSEP TISSUE	Uranium	Tissue	0.0008	0.02
6020A	PSEP TISSUE	Vanadium	Tissue	0.007	0.2
6020A	PSEP TISSUE	Zinc	Tissue	0.06	0.5



Table 2
Target Element Masses

Analyte	ISOTOPES ANALYZED	ISOTOPE REPORTED
Aluminum	27	27
Antimony	121,123	123
Arsenic	75	75
Barium	135,137,138	137
Beryllium	9	9
Cadmium	111,112,114	111
Chromium	52,53	52
Cobalt	59	59
Copper	63,65	65
Lead	206,207,208	208
Manganese	55	55
Molybdenum	95,97,98	98
Nickel	60,61,62	60
Selenium	77,78,82	82
Silver	107,109	107
Thallium	203,205	205
Uranium	238	238
Vanadium	51	51
Zinc	66,67,68	66



ATTACHMENT A
Example Standard Sheets**SOLUTION: ICP-MS, 200.8 INTERMEDIATE STOCK****MATRIX: 2% HNO₃**

		ALIQUOT OF	CONCENTRATION
ELEMENT		1000 ppm Std./1000ml	(µg/L)
HNO ₃		50.0 ml.	5%
Al		1.0 ml.	1000
Sb		1.0 ml.	1000
As		1.0 ml.	1000
Ba		1.0 ml.	1000
Be		1.0 ml.	1000
Cd		1.0 ml.	1000
Cr		1.0 ml.	1000
Co		1.0 ml.	1000
Cu		1.0 ml.	1000
Fe		1.0 ml.	1000
Pb		1.0 ml.	1000
Mn		1.0 ml.	1000
Mo		1.0 ml.	1000
Ni		1.0 ml.	1000
Se		1.0 ml.	1000
Tl		1.0 ml.	1000
V		1.0 ml.	1000
U		1.0 ml.	1000
Zn		1.0 ml.	1000



SOLUTION: ICP-MS, 200.8 SILVER INTERMEDIATE STOCK
MATRIX: 5% HNO₃

		ALiquot of	CONCENTRATION
ELEMENT		1000 ppm Std./1000ml	(µg/L)
HNO ₃		50.0	5%
Ag		1.0	1000

SOLUTION: ICP-MS 25ppb Calibration Standard and CCV
MATRIX: As Required

	ALiquot PER	CONCENTRATION
SOURCE	100 ml.	(µg/L)
HNO ₃ (Ultrex)	As Required	As Required
INTERMEDIATE STOCK	2.5	25.0
SILVER INTERMEDIATE STOCK	2.5	25.0



ATTACHMENT B
Isobaric Interference Corrections

Interference Equations:

Equation Name: Default

?SW82 = I82 * 0.7
?%SE77 = ?SE82 * 0.8484163
?%ARCL77 = I77 - ?%SE77
?%ARCL75 = ?%ARCL77 * 3.0650407
?AS75 = I75 - ?%ARCL75
?%CR53 = I52 * 0.1133652
?%CLO53 = I53 - ?%CR53
?%CLO51 = ?%CLO53 * 3.0650407
?V51 = I51 - ?%CLO51
?PB208 = I208 + I207 + I206

ALS Standard Operating Procedure

DOCUMENT TITLE:	DETERMINATION OF METALS AND TRACE ELEMENTS BY INDUCTIVELY COUPLED PLASMA ATOMIC EMISSION SPECTROMETRY (ICP)
REFERENCED METHOD:	EPA 200.7/6010C
SOP ID:	MET-ICP
REVISION NUMBER:	25
EFFECTIVE DATE:	01/01/2015



STANDARD OPERATING PROCEDURE

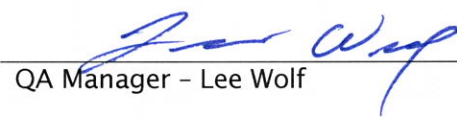
SOP No.: MET-ICP
Revision: 25
Effective: 01/01/2015
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DETERMINATION OF METALS AND TRACE ELEMENTS BY INDUCTIVELY COUPLED PLASMA ATOMIC EMISSION SPECTROMETRY (ICP)

ALS-KELSO

SOP ID:	MET-ICP	Rev. Number:	25	Effective Date:	01/01/2015
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Approved By:  Date: 12/8/14
Department Supervisor/Technical Director - Jeff Coronado

Approved By:  Date: 12/8/14
QA Manager - Lee Wolf

Approved By:  Date: 12/8/14
Laboratory Director - Jeff Grindstaff

Issue Date: _____ Doc Control ID#: _____ Issued To: _____

ANNUAL REVIEW

SIGNATURES BELOW INDICATE NO PROCEDURAL CHANGES HAVE BEEN MADE TO THE SOP SINCE THE APPROVAL DATE ABOVE. THIS SOP IS VALID FOR TWELVE ADDITIONAL MONTHS FROM DATE OF THE LAST SIGNATURE UNLESS INACTIVATED OR REPLACED BY SUBSEQUENT REVISIONS.

Signature _____

Title _____

Date _____

Signature _____

Title _____

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Date _____



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DETERMINATION OF METALS AND TRACE ELEMENTS BY INDUCTIVELY COUPLED PLASMA ATOMIC EMISSION SPECTROMETRY (ICP)

1. SCOPE AND APPLICATION

- 1.1. This procedure describes the steps taken for the analysis of soil, sludge surface water and drinking water digestates using EPA methods 6010C, 200.7, and CLP ILM04.0 for a variety of elements. The digested samples and QC standards are all diluted in a similar acid matrix. A procedure is also given for calculation of hardness by Standard Methods 2340B.
- 1.2. The Method Reporting Limits (MRLs) for common elements are listed in Table 1. Equivalent nomenclature for MRL includes Estimated Quantitation Limit (EQL). Therefore, $MRL=EQL$. The reported MRL may be adjusted if required for specific project requirements, however, the capability of achieving other reported MRLs must be demonstrated. The Method Detection Limits (MDLs) that have been achieved are listed in Table 1. The MDL and MRL may change as annual studies are performed.
- 1.3. In cases where there is a project-specific quality assurance plan (QAPP), the project manager identifies and communicates the QAPP-specific requirements to the laboratory. In general, project specific QAPP's supersede method specified requirements. An example of this are projects falling under DoD ELAP or project which require older versions of EPA methods (i.e. 6010B). QC requirements defined in the SOP *Department of Defense Projects – Laboratory Practices and Project Management (ADM-DOD)* may supersede the requirements defined in this SOP.

2. METHOD SUMMARY

- 2.1. A representative aliquot of sample is prepared as described in the applicable digestion SOP. The digestate is analyzed for the elements of interest using ICP spectrometry. The instrument measures characteristic emission spectra by optical spectrometry. The intensity of emission lines are monitored.
- 2.2. Final results are calculated using the digestion information and the results from the ICP analysis. Data is reported using standard ALS procedures and formats, or following project specific reporting specifications.
- 2.3. Deviations from the reference method(s): This SOP contains no deviations from the reference methods.

3. DEFINITIONS

- 3.1. **Batch** - A batch of samples is a group of environmental samples that are prepared and/or analyzed together as a unit with the same process and personnel using the same lot(s) of reagents. It is the basic unit for analytical quality control.



-
- 3.1.1. Preparation Batch - A preparation batch is composed of one to twenty field samples, all of the same matrix, and with a maximum time between the start of processing of the first and last samples in the batch to be 24 hours.
 - 3.1.2. Analysis Batch - Samples are analyzed in a set referred to as an analysis sequence. The sequence begins with instrument calibration (initial or continuing verification) followed by sample extracts interspersed with calibration standards (CCBs, CCVs, etc.) The sequence ends when the set of samples has been injected or when qualitative and/or quantitative QC criteria indicate an out-of-control situation.
 - 3.2. **Sample**
 - 3.2.1. Field Sample - An environmental sample collected and delivered to the laboratory for analysis; a.k.a., client's sample.
 - 3.2.2. Laboratory Sample - A representative portion, aliquot, or subsample of a field sample upon which laboratory analyses are made and results generated.
 - 3.3. **Quality System Matrix** - The *matrix* of an environmental sample is distinguished by its physical and/or chemical state and by the program for which the results are intended. The following sections describe the matrix distinctions. These matrices shall be used for purpose of batch and quality control requirements.
 - 3.3.1. Aqueous - Any groundwater sample, surface water sample, effluent sample, and TCLP or other extract. Specifically excluded are samples of the drinking water matrix and the saline/estuarine water matrix.
 - 3.3.2. Drinking water - Any aqueous sample that has been designated a potable or potential potable water source.
 - 3.3.3. Saline/Estuarine water - Any aqueous sample from an ocean or estuary or other salt-water source.
 - 3.3.4. Non-aqueous Liquid - Any organic liquid with <15% settleable solids.
 - 3.3.5. Animal tissue - Any tissue sample of an animal, invertebrate, marine organism, or other origin; such as fish tissue/organs, shellfish, worms, or animal material.
 - 3.3.6. Solids - Any solid sample such as soil, sediment, sludge, and other materials with >15% settleable solids.
 - 3.3.7. Chemical waste - Any sample of a product or by-product of an industrial process that results in a matrix not described in one of the matrices in Sections 3.3.1 through 3.3.6. These can be such matrices as non-aqueous liquids, solvents, oil, etc.
 - 3.3.8. Miscellaneous matrices - Samples of any composition not listed in 3.3.1 – 3.3.7. These can be such matrices as plant material, paper/paperboard, wood, autoluff, mechanical parts, filters, wipes, etc. Such samples shall be batched/grouped according to their specific matrix.



-
- 3.4. Matrix Spike/Duplicate Matrix Spike (MS/DMS) Analysis - In the matrix spike analysis, predetermined quantities of target analytes are added to a sample matrix prior to sample preparation and analysis. The purpose of the matrix spike is to evaluate the effects of the sample matrix on the method used for the analysis. Duplicate samples are spiked, and analyzed as a MS/DMS pair. Percent recoveries are calculated for each of the analytes detected. The relative percent difference (RPD) between the duplicate spikes (or samples) is calculated and used to assess analytical precision. The concentration of the spike should be at the midpoint of the calibration range or at levels specified by a project analysis plan.
- 3.5. Laboratory Duplicates (DUP) - Duplicates are additional replicates of samples that are subjected to the same preparation and analytical scheme as the original sample. The relative percent difference (RPD) between the sample and its duplicate is calculated and used to assess analytical precision.
- 3.6. Method Blank (MB) - The method blank is an artificial sample composed of analyte-free water or solid matrix and is designed to monitor the introduction of artifacts into the analytical process. The method blank is carried through the entire analytical procedure.
- 3.7. Laboratory Control Samples (LCS) - The LCS is an aliquot of analyte free water or analyte free solid to which known amounts target analytes are added. The LCS is prepared and analyzed in exactly the same manner as the samples. The percent recovery is compared to established limits and assists in determining whether the batch is in control.
- 3.8. Laboratory fortified Blank (LFB) - A laboratory blank that has been fortified with target analyte at the method reporting limit and used to determine if the laboratory can detect contaminants at the method reporting limit.
- 3.9. Independent Verification Standard (ICV) - A mid-level standard injected into the instrument after the calibration curve and prepared from a different source than the initial calibration standards. This is used to verify the validity of the initial calibration standards
- 3.10. Continuing Calibration Verification Standard (CCV) - A standard analyzed at specified intervals and used to verify the ongoing validity of the instrument calibration.
- 3.11. Instrument Blank (CCB) - The instrument blank (also called continuing calibration blank) is a volume of blank reagent of composition identical to the digestates. The purpose of the CCB is to determine the levels of contamination associated with the instrumental analysis.

4. INTERFERENCES

- 4.1. Interferences from contaminated reagents must be eliminated. The purity of acids must be established by the laboratory as being high enough to eliminate the introduction of contamination above the MRL (or above ½ the RL for DoD work).
- 4.2. Background emission and stray light can be compensated by background correction.
- 4.3. Spectral overlaps resulting in interelement contributions can be corrected for by using interelement correction factors. Interelement correction factors are established for each instrument and are maintained by the analyst at the workstation.

5. SAFETY



- 5.1. Chemicals, reagents and standards must be handled as described in the ALS safety policies, approved methods and in MSDSs where available. Refer to the ALS Environmental, Health and Safety Manual and the appropriate MSDS prior to beginning this method.
- 5.2. Hydrochloric, Nitric and Hydrofluoric Acids are used in this method. These acids are extremely corrosive and care must be taken while handling them. A face shield should be used while pouring acids. Safety glasses, lab coat and gloves should be worn while working with the solutions.
- 5.3. High Voltage - The power unit supplies high voltage to the RF generator which is used to form the plasma. The unit should never be opened. Exposure to high voltage can cause injury or death.
- 5.4. UV Light -The plasma when lit is a very intense light, and must not be viewed with the naked eye. Protective lenses are in place on the instrument. Glasses with special protective lenses are available.

6. SAMPLE COLLECTION, CONTAINERS, PRESERVATION AND STORAGE

- 6.1. Samples are prepared using methods 3005A, 3010A, 3050, or CLPILM04.0 (ALS SOPs MET-3005A, MET-3010A, MET-3050, and MET-DIG). Samples are received in the ICP lab as completed digestates. Samples are stored in 50 mL plastic centrifuge tubes, 100 mL digestion vessels or in 100 mL volumetric flasks.
- 6.2. Water samples analyzed by EPA method 200.7 are preserved after arrival at the laboratory. These samples are held for a minimum of 24 hours and the pH verified to be <2 prior to digestion.
- 6.3. Soil samples are diluted prior to instrumental analysis by a factor of 2. This allows the method to meet the required 1 g of sample to 200 mL dilution during digestion.
- 6.4. Following analysis, digestates are stored until two weeks after all results have been reviewed and then brought to $3 < \text{pH} < 10$ and disposed of through the sewer system.

7. STANDARDS, REAGENTS, AND CONSUMABLE MATERIALS

7.1. Standards Preparation

- 7.1.1. Stock standard solutions may be purchased from a number of vendors. All reference standards, where possible, must be traceable to SI units or NIST certified reference materials. The preparation for all laboratory prepared reagents and solutions must be documented in a laboratory logbook. Refer to the SOP *Reagent/Standards Login and Tracking (ADM-RTL)* for the complete procedure and documentation requirements. Manufacturer's expiration dates are used to determine the viability of standards.

7.1.2. Calibration Standards

Calibration standards are prepared from commercially purchased single element 1000 ppm or 10,000 ppm stock standards as well as pre-mixed multi element stock standards. All standards are aliquoted using Class A volumetric pipettes, or calibrated



fixed and adjustable volume autopipettors. All dilutions are made in Class A volumetric glassware.

The standard mixes for each ICP system vary based on the requirements of each instrument. The composition of the ICAP 6500 standards are outlined in Table 2.

7.1.3. Continuing Calibration Verification (CCV) Standards

CCV standards are analyzed at the midpoint of the calibration. These standards are produced by making a two-fold dilution of each calibration standard. The CCV standards are then run in sequence during the analytical run.

7.1.4. Initial Calibration Verification (ICV) Standards

The ICV working standards are produced by direct dilution of two certified mixed stock solutions (QCP-CICV1 and QCP-CICV3 purchased from Inorganic Ventures or another qualified vendor and various single element stock solutions from sources different than the calibration standards. The composition of these standards is outlined in Table 3.

7.1.5. Interference Check Solutions (ICSA & ICSAB)

The ICSA and ICSAB working standards are produced by direct dilution of certified mixed stock solutions (CLPP-ICS-A and CLPP-ICS-B or equivalent.) Antimony is also added to the ICSAB solution from a 1000 ppm single element stock standard. The composition of these standards is outlined in Table 4.

7.1.6. CRI/Low Level Calibration Verification

The CRI, Low Level Initial Calibration Verification (LLICV), and Low Level Continuing Calibration Verification (LLCCV) are produced by diluting 1000 or 10000ppm single stock standards into a 100X intermediate standard and then diluted 1/100 to obtain the MRL level. Note: The level used is that of the normal MRL used for both instruments.

7.1.7. The solutions and materials used for the LCS and matrix spikes are described in the applicable digestion SOP.

7.1.8. Standard Log

The analyte, source, initial volume, final volume, final concentration and expiration date are recorded in a standard logbook kept in the ICP lab. The operator who prepares the standard must date and initial the entry in the standards logbook. The operator also places his initials and the date prepared on the standard container. In addition to working standards used in calibration, all other standards used in the analytical run such as ICVs, MRL standards, and other project or client specific standards shall be documented in the standard logbook.

7.2. High Purity Argon.

7.3. Capillary, rinse and peristaltic pump tubing.



- 7.4. 17 x 100mm polypropylene test tubes.

8. APPARATUS AND EQUIPMENT

- 8.1. Inductively Coupled Plasma Atomic Emission Spectrometer
 - 8.1.1. Thermo Scientific ICAP 6500 (AES-03).
 - 8.1.2. Thermo Scientific ICAP 6500 (AES-04).
- 8.2. Concentric nebulizers.
- 8.3. Microflow nebulizer for ICAP 6500.
- 8.4. Torches and injector tips for each ICP.
- 8.5. Cyclonic spray chambers for each instrument.
- 8.6. Water coolers for each ICP.
- 8.7. Argon Humidifiers for the ICAP 6500.
- 8.8. ESI SC4 DX Autosampler with Fast System for ICAP 6500.
- 8.9. Peristaltic Pumps for each Spectrometer.
- 8.10. RF Generators for each ICP (internal on the IRIS and ICAP 6500).

9. PREVENTIVE MAINTENANCE

- 9.1. All maintenance activities are recorded in a maintenance logbook kept for each instrument. Pertinent information (serial numbers, instrument I.D., etc.) must be in the logbook. This includes the routine maintenance described in section 9. The entry in the log must include: date of event, the initials of who performed the work, and a reference to analytical control.
- 9.2. Torch, nebulizer, and spray chambers are cleaned as required. All instrument filters are vacuumed monthly. Dirty ICP torches and mixing chambers are soaked in aqua regia overnight, rinsed and placed in a clean dry area. The conical nebulizer is back flushed with acid or DI water as needed. The microflow nebulizer is not back flushed. Use the obstruction removal kit.

10. RESPONSIBILITIES

- 10.1. It is the responsibility of the analyst to perform the analysis according to this SOP and to complete all documentation required for data review. Analysis and interpretation of the results are performed by personnel in the laboratory who have demonstrated the ability to generate acceptable results utilizing this SOP. This demonstration is in accordance with the training program of the laboratory. Final review and sign-off of the data is performed by the department supervisor/manager or designee.
- 10.2. It is the responsibility of the department supervisor/manager to document analyst training. Training and proficiency is documented in accordance with the SOP *ADM-TRANDOC*.

11. PROCEDURE



11.1. Operating Parameters

11.1.1. For each Thermo Scientific ICAP 6500, the operating parameters are defined in the Method file. Default operating parameters are given in Tools/Options/New Method Parameters. However, each unique set of operating parameters is saved as a new file and the analyst must select and use the correct Method file for the application. Refer to the method files on the workstation for a listing of parameters for each file. The interelement correction factors to be used are established for the ICAP 6500 and are saved on the workstation also. Since these parameters change with method and correction factor updates, and due to the large amount of hardcopy printout for listing these parameters, it is not practical to include the parameters in this SOP.

11.2. Calibration/Standardization

11.2.1. ICAP 6500

11.2.1.1. Plasma is ignited and instrument is allowed to warm up for at least 30 minutes.

11.2.1.2. An internal standard is used for routine analyses on this instrument. Yttrium and Indium are used as internal standards. The internal standard solution is introduced into the analyzed solutions (standards, blanks, QC, samples, etc.) at 0.8 ug/mL for Y, and 1.6 ug/mL for In.

11.2.1.3. Run a peak check standard and adjust peaks as needed.

11.2.1.4. Standardize by running a Blank and a High Standard for each element in the analytical method. Analyst will initial and date the first page of the standardization.

11.2.2. Standardization is completed by analyzing an ICV for each analyte to be determined. For method 200.7 the result must be within $\pm 5\%$ of the true value. For method 6010B/C the result must be within $\pm 10\%$ of the true value. If the ICV fails when running method 6010C, either the calibration standards or the ICV must be prepared fresh and the instrument re-standardized. If the ICV fails when running methods 200.7 and 6010B only re-standardization is necessary.

11.2.3. Method 6010C also requires a LLICV be analyzed at the MRL level. The result must be within $\pm 30\%$ of the true value. The LLICV need not be made up with stock standards different than those of the calibration standards.

11.3. Analytical Run

11.3.1. Following standardization and ICV analysis, the remainder of the run is determined by what analytical method is being performed. These are listed below.

11.3.1.1. CLP ILM04.0: ICB, CCV, CCB, CRI, ICSA, ICSAB, CCV, CCB, routine samples. The CRI, ICSA, and ICSAB will be analyzed every 20 samples.



They will be labeled with an F indicating Final. Each set will be numbered in increasing order, i.e. ICSAF1, ICSAF2.

11.3.1.2.Methods 200.7 and 6010B/C: ICB, LLICV, CCV, CCB, CRI, ICSA, ICSAB, routine samples.

11.3.2. Evaluate the initial QC using the following criteria:

11.3.2.1.For methods 200.7 and 6010B/C, the following criteria apply:

- The ICB and CCB results are evaluated using method specified requirements. The following guidelines should also be used to determine acceptability:
- For 200.7, the result should be less than 3 times the standard deviation of the mean background signal.
- For method 6010B, the result should be less than the Method Detection Limit (MDL). In cases where the associated sample results are being reported to the Method Reporting Limit (MRL) the result may be greater than the MDL if the result does not adversely impact data quality.
- For method 6010C, the result should be less than the Lower Limit of Quantitation (LOQ).
- Where project specifications allow, the result may be over the MDL if the result does not adversely impact data quality.
- The CCV immediately following standardization must verify within $\pm 10\%$ of the true values with a relative standard deviation of $<5\%$ from 2 replicate integrations for methods 6010B/C. For 200.7, the first CCV must verify within $\pm 5\%$ with a RSD of $<3\%$ from 4 replicates. Calculate %RSD as follows:

$$\%RSD = \frac{StdDev_{CCV}}{Average_{CCV}} \times 100$$

where: StdDevccv = Standard deviation of the replicate integrations
Averageccv = Average of the replicate CCV integrations

- The LLICV or CRI is a low level standard with concentrations at the RL. For DoD projects, the LLICV standard concentrations will be equal to the project RLs. For method 6010C the CRI results should be within 30% of the true value. For 200.7 and 6010B the LLICV/CRI results should be greater than the MDL and less than 2X the MRL. For method 6010C, the LLICV results should be $\pm 30\%$ of the true value.



- The ICSA is run to check the validity of the Inter-element Correction Factors (IECs).

Note: DoD QSM requires this to be run at the beginning of each analytical run.

- The ICSAB must be within 20% of the expected value for the CLPP-ICS-B elements and Sb.

11.3.2.2. The ICV, LLICV, ICB, CCV, CCB, CRI, and ICSAB must meet the criteria listed. Reanalyze any elements that fail.

11.3.2.3. For CLP, refer to SOW ILM04.0 for acceptance criteria.

11.3.3. Continuing Calibration Verification

11.3.3.1. CCVs are analyzed after every 10 samples and at the end of the analytical run. They must verify within $\pm 10\%$ of the expected value with a RSD of $< 10\%$.

11.3.3.2. CCBs are analyzed after every 10 samples and at the end of the analytical run. CCBs are evaluated as in section 11.3.2.1.

11.3.3.3. Method 6010C requires a LLCCV be analyzed at the end of each analysis batch. The LLCCV is at the MRL level and must verify within $\pm 30\%$ of the true value. Reanalyze any elements to be reported at low levels that are bracketed by the LLCCV if the standard fails.

11.3.4. If the CCV or CCB solutions fail, reanalyze any elements to be reported.

12. QA/QC REQUIREMENTS

12.1. Initial Precision and Recovery Validation

The accuracy and precision of the procedure must be validated before analysis of samples begins, or whenever significant changes to the procedures have been made. To do this, four LCS aliquots are prepared and analyzed. The average percent recovery for each analyte must meet LCS criteria and the RSD $< 30\%$.

12.2. Method Detection Limits

12.2.1. A Method Detection Limit (MDL) study must be undertaken before analysis of samples can begin. To establish detection limits that are precise and accurate, the analyst must perform the following procedure. Spike a minimum of seven blank replicates at a level near or below the MRL. Follow the procedures in Section 11 to analyze the samples. Refer to the SOP CE-QA011, *Performing Method Detection Limit Studies and Establishing Limits of Detection and Quantification*.

12.2.2. Calculate the average concentration found (\bar{x}) and the standard deviation of the concentrations for each analyte. Calculate the MDL for each analyte using the correct



T value for the number of replicates. MDLs must be performed whenever there is a significant change in the background or instrument response.

12.2.3. A Limit of Detection (LOD) check must be performed after establishing the MDL and at least annually (quarterly if DoD) afterward. A blank is spiked with analytes at 2-4X the MDL and carried through the preparation and analytical procedure. The LOD is verified when the signal/noise ratio is > 3 for all analytes.

12.3. Limit of Quantitation Check(LOQ)/Lower Limit of Quantitation Check(LLQC)

For Method 6010C and drinking waters by method 200.7 a Lower Limit of Quantitation Check (LOQ/LLOQ) sample must be analyzed after establishing the MRL and at least annually (quarterly if DoD) afterward to demonstrate the desired detection capability. The LOQ/LLOQ sample is spiked at 1-2X the MRL and must be carried through the entire preparation and analytical procedure. Limits of quantitation are verified when all analytes are detected within 30% of their true value.

12.4. Linear Dynamic Range

The upper limit of the LDR must be established for each wavelength utilized. It must be determined from a linear calibration prepared in the normal manner using the established analytical operating procedure for the instrument. The LDR should be determined by analyzing at least three succeeding higher standard concentrations of the analyte until the observed analyte concentration is no more than 10% above or below the stated concentration of the standard. Determined LDRs must be documented and kept on file. The LDR which may be used for the analysis of samples should be judged by the analyst from the resulting data. Sample analyte concentrations that are greater than 90% of the determined upper LDR limit must be diluted and reanalyzed. The LDRs are verified semi-annually or whenever, in the judgment of the analyst, a change in analytical performance caused by either a change in instrument hardware or operating conditions would dictate they be redetermined.

12.5. Instrument Detection Limit

On a quarterly basis, the instrument detection limits for all analytes are determined as per procedures outlined in ILM04.0 (Section E, paragraph 10, 12 resp.). IDLs are determined using blanks and this data is kept on file.

12.6. Interelement Correction Factors

Semi-annually, instrument interferences are calculated as per ILM04.0 (Section E, paragraph 11) and Method 6010B/C. During the course of routine work, other interferences may be found. They are verified by the operator during the analytical run and data is manually corrected. Copies of this data are kept on file. Data can be manually corrected or automatically corrected using iTEVA software.

12.7. Internal Standard

Internal standard values are tracked by the instrument software. Values should remain within 60-125% of the value found in the calibration blank. If a sample is found to have an internal standard outside this value, the sample will be diluted to bring the internal standard into range.



12.8. Ongoing QC Samples required are described in the ALS-Kelso Quality Assurance Manual and in the SOP for *Sample Batches*. Additional QC Samples may be required in project specific quality assurance plans (QAPP). For example projects managed under the DoD ELAP must follow requirements defined in the DoD *Quality Systems Manual for Environmental Laboratories*. General QA requirements for DoD QSM are defined in the laboratory SOP, Department of Defense Projects – *Laboratory Practices and Project Management (ADM-DOD)*. General QC Samples are:

12.8.1. Each sample preparation batch must have a method blank associated with it. The method blank result should be $< \text{MRL}$. If the method blank is found to be contaminated, it may be reported if the concentration in the associated samples is at least 20 times the amount found in the method blank for methods 200.7 and 6010B, otherwise redigest the batch. For Method 6010C, the method blank may be reported if the concentration in the associated samples is at least 10 times the amount found in the method blank. A contaminated method blank (MB) may also be reported if all of the associated samples are non-detect (ND).

Note: DoD QSM requires contamination in the MB be $< 1/2$ the RL or $< 1/10$ any sample amount.

12.8.2. A Laboratory Control Sample (LCS) is digested one per batch, or per 20 samples. For soil samples, the recovery must fall within the ranges specified for the reference material. For CLP, use the prescribed limits for the SOW in use. If the LCS fails the acceptance criteria, redigest the batch of samples. For specifics on the preparation and composition of LCS samples refer to the appropriate digestion SOP.

12.8.3. A Duplicate sample is digested one per batch, or per 20 samples (i.e. 5%) for 6010B/C analysis, or per 10 samples (i.e. 10%) for 200.7 analyses. If the RPD is outside acceptance limits, either redigest the sample batch or flag the data appropriately, depending on the physical nature of the samples (e.g. non-homogenous).

12.8.4. A Laboratory fortified Blank (LFB) at the MRL is digested and analyzed with every batch of drinking water samples (method 200.7). The default acceptance criteria of 50-150% are to be used until sufficient data points are acquired to calculate in-house control limits.

12.8.5. A Matrix Spike sample is digested one per batch, or per 20 samples (i.e. 5%) for 6010B/C analysis, or per 10 samples (i.e. 10%) for 200.7 analyses. Where specified by project requirements, a matrix spike duplicate may be required. If the recovery is outside acceptance limits, either redigest the sample batch or flag the data appropriately, depending on the physical nature of the samples (e.g. non-homogenous). If the sample concentration is $> 4x$ the spike level, no action is required and data is flagged accordingly. For specifics on the preparation and composition of matrix spike solutions refer to the appropriate digestion SOP.

12.8.6. Acceptance criteria

12.8.6.1. Current ALS control limits and acceptance criteria for ongoing QC analyses are listed in the current ALS-Kelso DQO tables. Criteria are subject to change as statistical data are generated. The default method criteria may be used if



statistically generated criteria are broader or insufficient points are available for accurate statistical limits.

12.8.6.2. For all QC analyses, project-specific or program-specific (e.g. DOD) acceptance criteria may supersede ALS criteria. For analyses under the CLP SOW use the prescribed limits for the SOW in use.

12.8.7. Matrix Interference

12.8.7.1. When an analyst suspects that there may be any matrix interferences present, a post digestion spike may be performed. The recovery should be $\pm 20\%$.

12.8.7.2. If the post spike fails, a 1:5 serial dilution test shall be performed. The dilution should be within $\pm 10\%$ of the original result.

12.8.7.3. A 1:5 serial dilution shall be performed for all Tier III or IV deliverables.

Note: DoD QSM recovery acceptance limits are 75-125%.

12.8.7.4. Post spikes for 6010C shall be performed for Tier III and Tier IV.

12.9. Additional QC measures include control charting and compiling of QC data for generation of control limits.

12.10. CLP analyses are performed as per the QA/QC guidelines in the most current CLP SOW.

13. DATA REDUCTION, REVIEW, AND REPORTING

13.1. Calculate sample results using the data system printouts and digestion information. The digestion and dilution information is entered into the data system. The data system then uses the calculations below to generate a sample result. The wavelengths used to quantify each metal are summarized in Table 5 for the IRIS and Table 6 for the ICAP6500.

Aqueous samples are reported in $\mu\text{g/L}$:

$$\mu\text{g/L}(\text{Sample}) = C^* \times \text{Digestion Dilution Factor} \times \text{Post Digestion Dilution Factor} \times 1000 \mu\text{g} / \text{mg}$$

Solid samples are reported in mg/Kg :

$$\text{mg/Kg}(\text{Sample}) = C^* \times \text{Post Digestion Dilution Factor} \times \frac{\text{Digestion Vol. (ml)}}{\text{Sample wt. (g)}} \times \frac{1\text{L}}{1000\text{ml}} \times \frac{1000\text{g}}{1\text{Kg}}$$

C^* = Concentration of analyte as measured at the instrument in mg/L .

13.2. If total hardness is to be reported, use Calcium and Magnesium results to calculate as follows. For reporting calcium hardness, use only the calcium portion of the equation.

$$\text{Hardness, mg equivalent CaCO}_3/\text{L} = 2.497 [\text{Ca, mg} / \text{L}] + 4.118 [\text{Mg, mg} / \text{L}]$$



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- 13.3. A daily run log of all samples analyzed is maintained. All CLP data should be printed and stored after operator has checked for evenness of burns. A copy of this document will go with each package of Tier III or higher data run that day.
- 13.4. Data Review and Reporting
- 13.4.1. It is the analyst's responsibility to review analytical data to ensure that all quality control requirements have been met for each analytical run. Results for QC analyses are calculated and recorded as specified in section 12. The data is then placed in a work order file until complete. When the work order is complete, a report is generated. A final review is performed and the data is delivered to the project management department.

14. CONTINGENCIES FOR HANDLING OUT-OF-CONTROL OR UNACCEPTABLE DATA

- 14.1. Refer to the SOP for *Nonconformance and Corrective Action* (CE-QA008) for procedures for corrective action. Personnel at all levels and positions in the laboratory are to be alert to identifying problems and nonconformities when errors, deficiencies, or out-of-control situations are detected.
- 14.2. Handling out-of-control or unacceptable data
- 14.2.1. On-the-spot corrective actions that are routinely made by analysts and result in acceptable analyses should be documented as normal operating procedures, and no specific documentation need be made other than notations in laboratory maintenance logbooks, runlogs, for example.
- 14.2.2. Some examples when documentation of a nonconformity is required using a Nonconformity and Corrective Action Report (NCAR):
- Quality control results outside acceptance limits for accuracy and precision
 - Method blanks or continuing calibration blanks (CCBs) with target analytes above acceptable levels
 - Sample holding time missed due to laboratory error or operations
 - Deviations from SOPs or project requirements
 - Laboratory analysis errors impacting sample or QC results
 - Miscellaneous laboratory errors (spilled sample, incorrect spiking, etc.)
 - Sample preservation or handling discrepancies due to laboratory or operations error

15. METHOD PERFORMANCE

- 15.1. This method was validated through single laboratory studies of accuracy and precision. Refer to the reference method for additional available method performance data.
- 15.2. The method detection limit (MDL) is established using the procedure described in the SOP CE-QA011, *Performing Method Detection Limit Studies and Establishing Limits of Detection and Quantification*. Method Reporting Limits are established for this method based on MDL studies and as specified in the ALS Quality Assurance Manual.



16. POLLUTION PREVENTION AND WASTE MANAGEMENT

- 16.1. It is the laboratory's practice to minimize the amount of solvents, acids, and reagents used to perform this method wherever feasibly possible. Standards are prepared in volumes consistent with methodology and only the amount needed for routine laboratory use is kept on site. The threat to the environment from solvents and/or reagents used in this method can be minimized when recycled or disposed of properly.
- 16.2. The laboratory will comply with all Federal, State, and local regulations governing waste management, particularly the hazardous waste identification rules and land disposal restrictions as specified in the ALS Environmental Health and Safety Manual.
- 16.3. This method uses acid. Waste acid is hazardous to the sewer system and to the environment. All acid waste must be neutralized to a pH of 3-10 prior to disposal down the drain. The neutralization step is considered hazardous waste treatment and must be documented on the treatment by generator record. See the ALS EH&S Manual for details.

17. TRAINING

17.1. Training outline

- 17.1.1. Review literature (see references section). Read and understand the SOP. Also review the applicable MSDS for all reagents and standards used. Following these reviews, observe the procedure as performed by an experienced analyst at least three times.
- 17.1.2. Assist in the procedure under the guidance of an experienced analyst for approximately two weeks. During this period, the analyst is expected to transition from a role of assisting, to performing the procedure with minimal oversight from an experienced analyst.
- 17.1.3. Perform initial precision and recovery (IPR) study as described above for water samples. Summaries of the IPR are reviewed and signed by the supervisor. Copies may be forwarded to the employee's training file. For applicable tests, IPR studies should be performed in order to be equivalent to NELAP Initial Demonstration of Capability.

- 17.2. Training is documented following the *ALS Kelso, Training Procedure* (ADM-TRAIN) and the Corporate *Training Policy* (CE-QA003).

NOTE: When the analyst training is documented by the supervisor on internal training documentation forms, the supervisor is acknowledging that the analyst has read and understands this SOP and that adequate training has been given to the analyst to competently perform the analysis independently.

18. METHOD MODIFICATIONS

- 18.1. There are no known modifications in this laboratory standard operating procedure from the reference method.

19. REFERENCES



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- 19.1. USEPA, Contract Laboratory Program, SOW #ILM04.0
 - 19.2. Thermo Jarrell Ash ICAP61 Manual
 - 19.3. USEPA, Test Methods for Evaluating Solid Waste, SW-846, 3rd Edition, Update III, Method 6010B, Revision 2, December 1996.
 - 19.4. USEPA, Test Methods for Evaluating Solid Waste, SW-846, 3rd Edition, Update III, Method 6010C, Revision 3, February 2007.
 - 19.5. USEPA, Methods for Determination of Metals in Environmental Samples, Supplement I, EPA/600/R-94/111, Method 200.7, Revision 4.4, May 1994.
 - 19.6. *Hardness by Calculation, Method 2340B*, Standard Methods for the Examination of Water and Wastewater, 20th ed., 1998.

20. CHANGES SINCE THE LAST REVISION

- 20.1. Updated to current ALS format.
- 20.2. Revised internal document references from CAS to ALS
- 20.3. Minor typographical and format corrections.
- 20.4. Section 3– updated several definitions to standard definitions for SOPs.
- 20.5. Section 7.1.4 – corrected standard composition to reflect current practice.
- 20.6. Section 9.2 – revised to reflect current practice.
- 20.7. Section 11.3.2.1 – LL ICV criteria revised to reflect current practice.
- 20.8. Section 12.2.3 – LOD spike level corrected.
- 20.9. Section 12.4 – revised to reflect current practice (LDR semi-annual).
- 20.10. Sections 12.8.2. – 12.8.5 revised to remove outdated/redundant QC criteria and added new section 12.8.6.
- 20.11. Section 14 – updated to standard language.
- 20.12. Section 17 – updated to standard language.
- 20.13. Tables reference errors corrected and tables updated.



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TABLE 1

Target Elements, Method Reporting Limits, and Method Detection Limits

METHOD	PREP METHOD	ANALYTE	MATRIX	MDL	MRL
0.7	EPA 3050B	Aluminum	Soil	0.5	2
200.7	EPA 3050B	Antimony	Soil	2	4
200.7	EPA 3050B	Arsenic	Soil	2	4
200.7	EPA 3050B	Barium	Soil	0.3	0.8
200.7	EPA 3050B	Beryllium	Soil	0.08	0.2
200.7	EPA 3050B	Bismuth	Soil	3	8
200.7	EPA 3050B	Boron	Soil	0.7	4
200.7	EPA 3050B	Cadmium	Soil	0.09	0.2
200.7	EPA 3050B	Calcium	Soil	1	4
200.7	EPA 3050B	Chromium	Soil	0.3	0.8
200.7	EPA 3050B	Cobalt	Soil	0.2	0.4
200.7	EPA 3050B	Copper	Soil	0.4	0.8
200.7	EPA 3050B	Iron	Soil	2	4
200.7	EPA 3050B	Lead	Soil	0.7	2
200.7	EPA 3050B	Lithium	Soil	0.6	4
200.7	EPA 3050B	Magnesium	Soil	0.2	2
200.7	EPA 3050B	Manganese	Soil	0.04	0.2
200.7	EPA 3050B	Molybdenum	Soil	0.2	0.8
200.7	EPA 3050B	Nickel	Soil	0.2	0.8
200.7	EPA 3050B	Phosphorus	Soil	3	8
200.7	EPA 3050B	Potassium	Soil	10	40
200.7	EPA 3050B	Selenium	Soil	2	4
200.7	EPA 3050B	Silver	Soil	0.3	0.8
200.7	EPA 3050B	Sodium	Soil	5	40
200.7	EPA 3050B	Strontium	Soil	0.05	0.2
200.7	EPA 3050B	Sulfur	Soil	4	8
200.7	EPA 3050B	Thallium	Soil	0.8	2
200.7	EPA 3050B	Tin	Soil	0.6	4
200.7	EPA 3050B	Titanium	Soil	0.2	0.4
200.7	EPA 3050B	Vanadium	Soil	0.3	0.8
200.7	EPA 3050B	Zinc	Soil	0.2	1



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METHOD	PREP METHOD	ANALYTE	MATRIX	MDL	MRL
200.7	MET-DIG (CLP)	Aluminum	Water	4	10
200.7	MET-DIG (CLP)	Antimony	Water	6	20
200.7	MET-DIG (CLP)	Arsenic	Water	5	10
200.7	MET-DIG (CLP)	Barium	Water	0.6	4
200.7	MET-DIG (CLP)	Beryllium	Water	0.5	1
200.7	MET-DIG (CLP)	Bismuth	Water	6	40
200.7	MET-DIG (CLP)	Boron	Water	4	20
200.7	MET-DIG (CLP)	Cadmium	Water	0.5	1
200.7	MET-DIG (CLP)	Calcium	Water	0.9	20
200.7	MET-DIG (CLP)	Chromium	Water	0.9	4
200.7	MET-DIG (CLP)	Cobalt	Water	1	2
200.7	MET-DIG (CLP)	Copper	Water	2	4
200.7	MET-DIG (CLP)	Iron	Water	3	20
200.7	MET-DIG (CLP)	Lead	Water	5	10
200.7	MET-DIG (CLP)	Lithium	Water	4	20
200.7	MET-DIG (CLP)	Magnesium	Water	0.3	5
200.7	MET-DIG (CLP)	Manganese	Water	0.3	1
200.7	MET-DIG (CLP)	Molybdenum	Water	0.9	4
200.7	MET-DIG (CLP)	Nickel	Water	0.6	4
200.7	MET-DIG (CLP)	Phosphorus	Water	6	40
200.7	MET-DIG (CLP)	Potassium	Water	60	200
200.7	MET-DIG (CLP)	Selenium	Water	9	20
200.7	MET-DIG (CLP)	Silicon	Water	20	200
200.7	MET-DIG (CLP)	Silver	Water	2	4
200.7	MET-DIG (CLP)	Sodium	Water	20	200
200.7	MET-DIG (CLP)	Strontium	Water	0.2	1
200.7	MET-DIG (CLP)	Sulfur	Water	20	40
200.7	MET-DIG (CLP)	Thallium	Water	4	10
200.7	MET-DIG (CLP)	Tin	Water	3	20
200.7	MET-DIG (CLP)	Titanium	Water	0.8	2
200.7	MET-DIG (CLP)	Vanadium	Water	1	4
200.7	MET-DIG (CLP)	Zinc	Water	0.6	4



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TABLE 1 – continued

METHOD	PREP METHOD	ANALYTE	MATRIX	MDL	MRL
6010C	EPA 3050B	Aluminum	Soil	0.5	2
6010C	EPA 3050B	Antimony	Soil	2	4
6010C	EPA 3050B	Arsenic	Soil	2	4
6010C	EPA 3050B	Barium	Soil	0.3	0.8
6010C	EPA 3050B	Beryllium	Soil	0.08	0.2
6010C	EPA 3050B	Bismuth	Soil	3	8
6010C	EPA 3050B	Boron	Soil	0.7	4
6010C	EPA 3050B	Cadmium	Soil	0.09	0.2
6010C	EPA 3050B	Calcium	Soil	1	4
6010C	EPA 3050B	Chromium	Soil	0.3	0.8
6010C	EPA 3050B	Cobalt	Soil	0.2	0.4
6010C	EPA 3050B	Copper	Soil	0.4	0.8
6010C	EPA 3050B	Iron	Soil	2	4
6010C	EPA 3050B	Lead	Soil	0.7	2
6010C	EPA 3050B	Lithium	Soil	0.6	4
6010C	EPA 3050B	Magnesium	Soil	0.2	2
6010C	EPA 3050B	Manganese	Soil	0.04	0.2
6010C	EPA 3050B	Molybdenum	Soil	0.2	0.8
6010C	EPA 3050B	Nickel	Soil	0.2	0.8
6010C	EPA 3050B	Phosphorus	Soil	3	8
6010C	EPA 3050B	Potassium	Soil	10	40
6010C	EPA 3050B	Selenium	Soil	2	4
6010C	EPA 3050B	Silver	Soil	0.3	0.8
6010C	EPA 3050B	Sodium	Soil	5	40
6010C	EPA 3050B	Strontium	Soil	0.05	0.2
6010C	EPA 3050B	Sulfur	Soil	4	8
6010C	EPA 3050B	Thallium	Soil	0.8	2
6010C	EPA 3050B	Tin	Soil	0.6	4
6010C	EPA 3050B	Titanium	Soil	0.2	0.4
6010C	EPA 3050B	Vanadium	Soil	0.3	0.8
6010C	EPA 3050B	Zinc	Soil	0.2	1
6010C/AVS-SEM	EPA 821/R-91-100	Antimony	Soil	0.0008	0.003
6010C/AVS-SEM	EPA 821/R-91-100	Arsenic	Soil	0.002	0.005
6010C/AVS-SEM	EPA 821/R-91-100	Cadmium	Soil	0.00007	0.0002
6010C/AVS-SEM	EPA 821/R-91-100	Chromium	Soil	0.0004	0.002
6010C/AVS-SEM	EPA 821/R-91-100	Copper	Soil	0.0005	0.002
6010C/AVS-SEM	EPA 821/R-91-100	Lead	Soil	0.0005	0.001
6010C/AVS-SEM	EPA 821/R-91-100	Nickel	Soil	0.0003	0.001
6010C/AVS-SEM	EPA 821/R-91-100	Silver	Soil	0.0003	0.001
6010C/AVS-SEM	EPA 821/R-91-100	Zinc	Soil	0.0003	0.003



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TABLE 1 – continued

METHOD	PREP METHOD	ANALYTE	MATRIX	MDL	MRL
6010C	MET-DIG (CLP)	Aluminum	Water	4	10
6010C	MET-DIG (CLP)	Antimony	Water	6	20
6010C	MET-DIG (CLP)	Arsenic	Water	5	10
6010C	MET-DIG (CLP)	Barium	Water	0.6	4
6010C	MET-DIG (CLP)	Beryllium	Water	0.5	1
6010C	MET-DIG (CLP)	Bismuth	Water	6	40
6010C	MET-DIG (CLP)	Boron	Water	4	20
6010C	MET-DIG (CLP)	Cadmium	Water	0.5	1
6010C	MET-DIG (CLP)	Calcium	Water	0.9	20
6010C	MET-DIG (CLP)	Chromium	Water	0.9	4
6010C	MET-DIG (CLP)	Cobalt	Water	1	2
6010C	MET-DIG (CLP)	Copper	Water	2	4
6010C	MET-DIG (CLP)	Iron	Water	3	20
6010C	MET-DIG (CLP)	Lead	Water	5	10
6010C	MET-DIG (CLP)	Lithium	Water	4	20
6010C	MET-DIG (CLP)	Magnesium	Water	0.3	5
6010C	MET-DIG (CLP)	Manganese	Water	0.3	1
6010C	MET-DIG (CLP)	Molybdenum	Water	0.9	4
6010C	MET-DIG (CLP)	Nickel	Water	0.6	4
6010C	MET-DIG (CLP)	Phosphorus	Water	6	40
6010C	MET-DIG (CLP)	Potassium	Water	60	200
6010C	MET-DIG (CLP)	Selenium	Water	9	20
6010C	MET-DIG (CLP)	Silicon	Water	20	200
6010C	MET-DIG (CLP)	Silver	Water	2	4
6010C	MET-DIG (CLP)	Sodium	Water	20	200
6010C	MET-DIG (CLP)	Strontium	Water	0.2	1
6010C	MET-DIG (CLP)	Sulfur	Water	20	40
6010C	MET-DIG (CLP)	Thallium	Water	4	10
6010C	MET-DIG (CLP)	Tin	Water	3	20
6010C	MET-DIG (CLP)	Titanium	Water	0.8	2
6010C	MET-DIG (CLP)	Vanadium	Water	1	4
6010C	MET-DIG (CLP)	Zinc	Water	0.6	4



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METHOD	PREP METHOD	ANALYTE	MATRIX	MDL	MRL
6010C	PSEP TISSUE	Aluminum	Tissue	0.3	1
6010C	PSEP TISSUE	Antimony	Tissue	0.5	2
6010C	PSEP TISSUE	Arsenic	Tissue	0.5	1
6010C	PSEP TISSUE	Barium	Tissue	0.07	0.4
6010C	PSEP TISSUE	Beryllium	Tissue	0.05	0.1
6010C	PSEP TISSUE	Boron	Tissue	0.8	2
6010C	PSEP TISSUE	Cadmium	Tissue	0.04	0.1
6010C	PSEP TISSUE	Calcium	Tissue	2	4
6010C	PSEP TISSUE	Chromium	Tissue	0.08	0.4
6010C	PSEP TISSUE	Cobalt	Tissue	0.07	0.2
6010C	PSEP TISSUE	Copper	Tissue	0.2	0.4
6010C	PSEP TISSUE	Iron	Tissue	1	2
6010C	PSEP TISSUE	Lead	Tissue	0.3	1
6010C	PSEP TISSUE	Lithium	Tissue	0.3	2
6010C	PSEP TISSUE	Magnesium	Tissue	0.6	2
6010C	PSEP TISSUE	Manganese	Tissue	0.03	0.1
6010C	PSEP TISSUE	Molybdenum	Tissue	0.2	0.4
6010C	PSEP TISSUE	Nickel	Tissue	0.2	0.4
6010C	PSEP TISSUE	Phosphorus	Tissue	2	4
6010C	PSEP TISSUE	Potassium	Tissue	9	20
6010C	PSEP TISSUE	Selenium	Tissue	0.9	2
6010C	PSEP TISSUE	Silicon	Tissue	4	20
6010C	PSEP TISSUE	Silver	Tissue	0.2	0.4
6010C	PSEP TISSUE	Sodium	Tissue	2	20
6010C	PSEP TISSUE	Strontium	Tissue	0.04	0.1
6010C	PSEP TISSUE	Thallium	Tissue	0.4	1
6010C	PSEP TISSUE	Tin	Tissue	0.3	2
6010C	PSEP TISSUE	Titanium	Tissue	0.08	0.2
6010C	PSEP TISSUE	Vanadium	Tissue	0.2	0.4
6010C	PSEP TISSUE	Zinc	Tissue	0.2	0.4



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TABLE 1 - continued

METHOD	PREP METHOD	ANALYTE	MATRIX	MDL	MRL
6010C	1311/3010A	Antimony	TCLP	0.03	0.1
6010C	1311/3010A	Arsenic	TCLP	0.025	0.05
6010C	1311/3010A	Barium	TCLP	0.5	1
6010C	1311/3010A	Beryllium	TCLP	0.001	0.005
6010C	1311/3010A	Cadmium	TCLP	0.001	0.05
6010C	1311/3010A	Chromium	TCLP	0.01	0.05
6010C	1311/3010A	Cobalt	TCLP	0.0035	0.01
6010C	1311/3010A	Copper	TCLP	0.01	0.1
6010C	1311/3010A	Lead	TCLP	0.02	0.05
6010C	1311/3010A	Manganese	TCLP	0.0025	0.005
6010C	1311/3010A	Nickel	TCLP	0.0035	0.1
6010C	1311/3010A	Selenium	TCLP	0.025	0.1
6010C	1311/3010A	Silver	TCLP	0.004	0.05
6010C	1311/3010A	Thallium	TCLP	0.1	0.25
6010C	1311/3010A	Zinc	TCLP	0.1	1



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TABLE 2					
Standard A for ICAP 6500 ICP-OES					
		Source		Final	Final
Analyte	Source	Concentration	Aliquot	Volume	Concentration
		(ppm)	(mL)	(mL)	(ppm)
Antimony	(1)	100	5	1000	0.5
Beryllium	(1)	100	5	1000	0.5
Boron	(1)	100	5	1000	0.5
Cadmium	(1)	100	5	1000	0.5
Calcium	Ca stock	1000	0.5	1000	1.0*
Chromium	(1)	100	5	1000	0.5
Cobalt	(1)	100	5	1000	0.5
Copper	(1)	100	5	1000	0.5
Iron	(1)	100	5	1000	0.5
Lead	(1)	100	5	1000	0.5
Magnesium	(1)	100	5	1000	0.5
Manganese	(1)	100	5	1000	0.5
Molybdenum	(1)	100	5	1000	0.5
Nickel	(1)	100	5	1000	0.5
Selenium	(1)	100	5	1000	0.5
Silver	(1)	100	5	1000	0.5
Tin	Elemental Stock	1000	0.5	1000	0.5
Thallium	(1)	100	5	1000	0.5
Titanium	(1)	100	5	1000	0.5
Vanadium	(1)	100	5	1000	0.5
Zinc	(1)	100	5	1000	0.5
Hydrochloric Acid	-	-	50	1000	5%
Nitric Acid	-	-	10	1000	1%
(1) Mixed Standard, QCS-26					
* 0.5mL 1000ppm Ca added to 5mL QCS-26(100ppm Ca), 1000mL Final Volume					



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TABLE 3
ICP ICV Standards

ICV1 Solution

Analyte	Source	Source Concentration (ppm)	Aliquot (mL)	Final Volume (mL)	Final Concentration (ppm)
Aluminum	QCP-CICV-1	1000	2.5	500	5.0
Antimony	QCP-CICV-1	1000	1.25	500	2.5
Arsenic	QCP-CICV-3	500	2.5	500	2.5
Barium	QCP-CICV-1	1000	2.5	500	5.0
Beryllium	QCP-CICV-1	25	2.5	500	0.125
Cadmium	QCP-CICV-3	250	2.5	500	1.25
Calcium	QCP-CICV-1	2500	2.5	500	12.5
Chromium	QCP-CICV-1	100	2.5	500	0.5
Cobalt	QCP-CICV-1	250	2.5	500	1.25
Copper	QCP-CICV-1	125	2.5	500	0.625
Iron	QCP-CICV-1	500	2.5	500	2.5
Lead	QCP-CICV-3	500	2.5	500	2.5
Magnesium	QCP-CICV-1	2500	2.5	500	12.5
Manganese	QCP-CICV-1	250	2.5	500	1.25
Molybdenum	Elemental Stock	1000	1.0	500	2.0
Nickel	QCP-CICV-1	250	2.5	500	1.25
Potassium	QCP-CICV-1	2500	2.5	500	12.5
Selenium	QCP-CICV-3	500	2.5	500	2.5
Silver	QCP-CICV-1	125	2.5	500	0.625
Sodium	QCP-CICV-1	2500	2.5	500	12.5
Thallium	QCP-CICV-3	500	2.5	500	2.5
Titanium	Elemental Stock	1000	1.0	500	2.0
Vanadium	QCP-CICV-1	250	2.5	500	1.25
Zinc	QCP-CICV-1	250	2.5	500	1.25
Hydrochloric Acid	-	-	25	500	5%
Nitric Acid	-	-	5	500	1%



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TABLE 4
ICP Interference Check Solutions

ICSA Solution

Analyte	Source	Source Concentration (ppm)	Aliquot (mL)	Final Volume (mL)	Final Concentration (ppm)
Aluminum	CLPP-ICS-A	5000	50	500	500
Calcium	CLPP-ICS-A	5000	50	500	500
Iron	CLPP-ICS-A	2000	50	500	200
Magnesium	CLPP-ICS-A	5000	50	500	500
Hydrochloric Acid	-	-	25	500	5%
Nitric Acid	-	-	5	500	1%

ICSAB Solution

Analyte	Source	Source Concentration (ppm)	Aliquot (mL)	Final Volume (mL)	Final Concentration (ppm)
Aluminum	CLPP-ICS-A	5000	50	500	500
Antimony	Elemental Stock	1000	0.5	500	1
Barium	CLPP-ICS-B	50	5	500	0.5
Beryllium	CLPP-ICS-B	50	5	500	0.5
Cadmium	CLPP-ICS-B	100	5	500	1
Calcium	CLPP-ICS-A	5000	50	500	500
Chromium	CLPP-ICS-B	50	5	500	0.5
Cobalt	CLPP-ICS-B	50	5	500	0.5
Copper	CLPP-ICS-B	50	5	500	0.5
Iron	CLPP-ICS-A	2000	50	500	200
Lead	CLPP-ICS-B	100	5	500	1
Magnesium	CLPP-ICS-A	5000	50	500	500
Manganese	CLPP-ICS-B	50	5	500	0.5
Nickel	CLPP-ICS-B	100	5	500	1
Silver	CLPP-ICS-B	100	5	500	1
Vanadium	CLPP-ICS-B	50	5	500	0.5
Zinc	CLPP-ICS-B	100	5	500	1
HCl	-	-	25	500	0.05
HNO ₃	-	-	5	500	0.01



TABLE 5
IRIS Analytical Wavelengths

<u>Analyte</u>	<u>Wavelength</u>	
Aluminum	237.3	
Antimony	206.8	
Arsenic	189.0	
Barium	233.5	
Beryllium	313.0	
Boron	249.7	
Cadmium	226.5	
Calcium	317.9	
Calcium	211.2	High Line
Chromium	267.7	
Cobalt	228.6	
Copper	324.7	
Iron	259.9	
Iron	271.4	High Line
Lead	220.3	
Lithium	670.7	
Magnesium	279.5	
Magnesium	202.5	High Line
Manganese	257.6	
Manganese	293.9	High Line
Molybdenum	202.0	
Nickel	231.6	
Phosphorus	214.9	
Potassium	766.4	
Selenium	196.0	
Silicon	251.6	
Silver	328.0	
Sodium	589.5	
Strontium	407.7	
Thallium	190.8	
Tin	189.9	
Titanium	323.4	
Vanadium	310.2	
Zinc	206.2	



TABLE 6
ICAP 6500 Analytical Wavelengths

<u>Analyte</u>	<u>Wavelength</u>	
Aluminum	167.0	Low Line
Aluminum	394.4	
Antimony	206.8	
Antimony	217.5	Alternate
Arsenic	189.0	
Barium	455.4	
Beryllium	234.8	
Boron	249.6	
Cadmium	226.5	
Cadmium	214.4	Alternate
Calcium	315.8	
Calcium	393.3	Low Line
Chromium	267.7	
Cobalt	230.7	
Cobalt	228.6	Alternate
Copper	327.3	
Copper	224.7	Alternate
Iron	259.9	
Lead	220.3	
Lithium	670.7	
Magnesium	279.0	High Line
Magnesium	279.5	Low Line
Magnesium	285.2	
Manganese	257.6	
Manganese	260.5	High Line
Molybdenum	202.0	
Nickel	221.6	
Nickel	231.6	Alternate
Phosphorus	214.9	
Phosphorus	178.2	Alternate
Potassium	766.4	
Selenium	196.0	
Silicon	251.6	
Silver	328.0	
Sodium	588.9	Alternate
Sodium	589.5	



TABLE 6
ICAP 6500 Analytical Wavelengths, continued

<u>Analyte</u>	<u>Wavelength</u>	
Strontium	407.7	
Thallium	190.8	
Tin	189.9	
Titanium	336.1	
Vanadium	292.4	
Zinc	206.2	
Zinc	213.8	Alternate

Quality Assurance Project Plan Addendum
Pines Area of Investigation

Section: Attachment B
Revision: 3
Date: May 2015

Attachment B

Gamma Spectroscopy Library

Title: AECOM1

Nuclide Name	Nuclide Type	Half Life	Key Line?	No Wtmean?	Energy (keV)	%Abn
K-40		1.25E+09Y	*		1460.82	10.66
CO-60		1925.28D			1173.23	99.85
			*		1332.49	99.98
BA-137M		2.55M	*		661.66	89.90
CS-137		30.08Y	*		661.66	85.10
TL-208		3.05M			277.37	6.60
			*		583.19	85.00
					860.56	12.50
PB-210		22.20Y	*		46.54	4.25
BI-211		2.13M	*		351.06	12.92
PB-211		36.10M	*		404.85	3.78
					427.09	1.76
					832.01	3.52
BI-212		60.55M	*		727.33	6.67
					785.37	1.10
					1620.50	1.47
PB-212		10.64H	*		238.63	43.60
					300.09	3.30
BI-214		19.90M	*		609.32	45.49
					768.36	4.89
					1120.29	14.92
					1238.12	5.83
					1764.49	15.30
PB-214		26.60M			242.00	7.25
					295.22	18.42
			*		351.93	35.60
RN-219		3.96S			271.23	10.80
			*		401.81	6.60
FR-223		21.80M			50.09	34.00
					79.65	8.70
			*		234.75	3.00
RA-223		11.43D			144.24	3.27
					154.21	5.70
			*		269.46	13.90
					338.28	2.84
TH-227		18.72D			50.13	11.40
					235.96	17.50
			*		256.23	9.50
					329.85	4.00
AC-228		6.25H			338.32	11.27
					463.00	4.40
					794.95	4.25
					835.71	1.61
			*		911.20	25.80
					968.97	15.80
PA-231		3.28E+04Y	*		283.69	1.70
					300.07	2.42
					302.65	2.20

			330.06	1.40
TH-231	25.52H		81.23	0.90
		*	84.21	6.60
			89.95	1.00
			163.10	0.15
PA-234M	1.16M		766.42	0.32
		*	1001.03	0.84
TH-234	24.10D	*	63.29	3.70
			92.59	4.23
U-235	7.04E+08Y	*	143.76	10.96
			163.33	5.08
			185.72	57.20
			205.31	5.01
AM-241	432.60Y	*	59.54	35.90

Quality Assurance Project Plan Addendum

Pines Area of Investigation

Section: Attachment C
Revision: 3
Date: May 2015

Attachment C

Radioisotope Calculation Spreadsheet

EXAMPLE TABLE
GAMMA SPECTROSCOPY RESULTS AFTER 28-DAY INGROWTH PERIOD

PINES AREA OF INVESTIGATION
SUPPLEMENTAL SOIL CHARACTERIZATION

	Measured				Calculated	Measured								Calculated	Calculated	Calculated	Measured		Calculated	Measured
Sample I.D.	Be-7	Am-241	K-40	Ba-137m	Cs-137 (a)	Co-60	Tl-208	Pb-210	Bi-212	Pb-212	Bi-214	Pb-214	Ac-228	Ra-228 (b)	Ra-226 (c)	Total Ra (d)	Pa-234m	Th-234	U-238 (e)	U-235
GS001	1.0	1.0	1.0	1.0	1.1	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	2.0	1.0	1.0	1.0	1.0
					0.0									0.0	0.0	0.0				
					0.0									0.0	0.0	0.0				
					0.0									0.0	0.0	0.0				
					0.0									0.0	0.0	0.0				
					0.0									0.0	0.0	0.0				
					0.0									0.0	0.0	0.0				
					0.0									0.0	0.0	0.0				
					0.0									0.0	0.0	0.0				
					0.0									0.0	0.0	0.0				
					0.0									0.0	0.0	0.0				
					0.0									0.0	0.0	0.0				

- Notes:
- Equations:

(a) Cs-137 = 0.946 x Ba-137m

(b) Ra-228 = Ac-228

(c) Ra-226 = (Pb-214 + Bi-214) / 2

(d) Total Ra = Ra-228 + Ra-226

(e) U-238 = (Pa-234m + Th-234) / 2