

April 25, 2012  
RTI Project No. 0209847.024

# State of Practice for Emerging Waste Conversion Technologies

## Final Project Report

**Prepared for**

U.S. Environmental Protection Agency  
Office of Research and Development  
Research Triangle Park, NC 27709

**Prepared by**

RTI International  
3040 Cornwallis Road  
Research Triangle Park, NC 27709



## Contents

Abbreviations and Acronyms List .....	3
Acknowledgements .....	5
Disclaimer .....	6
Executive Summary .....	7
Section 1: Introduction.....	10
1.1 Conversion Technology Development Stages .....	11
1.2 Challenges for Implementing Conversion Technologies.....	16
1.3 Report Structure .....	18
Section 2: Conversion Technology Categories.....	19
2.1 Pyrolysis.....	19
2.1.1 Types of Pyrolysis.....	19
2.1.2 Issues Associated with Pyrolysis .....	20
2.2 Gasification .....	20
2.2.1 Types of Gasification.....	20
2.2.2 Issues Associated with Gasification .....	21
2.3 Plasma Arc .....	21
2.3.1 Types of Plasma Arc.....	21
2.3.2 Issues Associated With Plasma Arc.....	22
2.4 Anaerobic Digestion .....	22
Section 3: Technology Vendor Case Examples.....	24
3.1 Pyrolysis Technology Vendors.....	25
3.1.1 Agilyx: Tigard, Oregon.....	25
3.1.2 Envion: Derwood, MD (to be relocated to Florida in 2011/2012).....	27
3.1.3 Climax Global Energy: South Carolina .....	29
3.1.4 JBI: Niagara Falls, New York.....	30
3.2 Gasification Technology Vendors .....	31
3.2.1 Enerkem: Westbrook, PQ, Canada .....	31
3.2.2 Plasco: Ottawa, Ontario, Canada .....	35
3.2.4 Ze-gen: Attleboro, MA .....	37

3.2.5	Geoplasma: St. Lucie, Florida.....	38
3.3	Anaerobic Digestion .....	40
3.3.1	County of Yolo Public Works Department: Yolo County, California .....	40
3.3.2	Quasar: Wooster, Ohio .....	41
Section 4: Life Cycle Environmental and Cost Assessment for Technology Categories.....		42
4.1	General Approach for the Environmental and Cost Assessment.....	43
4.1.1	Goals .....	43
4.1.2	Scope and Boundaries .....	43
4.2	Life Cycle Inventory Methodology, Assumptions, and Modules for Waste Conversion Technologies.....	45
4.2.1	Treatment of Material and Energy Recovery .....	46
4.2.2	Items Excluded From the Life Cycle Inventory .....	46
4.2.3	LCI Parameters Tracked and Reported .....	47
4.3	Key Data and Assumptions Used in the LCIs .....	49
4.4	Environmental and Cost Assessment Results .....	53
4.4.1	Gasification Results.....	53
4.4.2	Pyrolysis Results.....	60
4.4.3	Anaerobic Digestion Results .....	65
Section 5: Findings and Recommendations .....		72
5.1	Key Findings.....	72
5.2	Limitations and Recommendations Associated with the Key Findings in This Report.....	74
Resources .....		76
Appendix A: Conversion Technology Data.....		79
Table A-1. Gasification Technology Data.....		80
Table A-2. Pyrolysis Technology Data.....		82
Appendix B: List of Facilities Considered in this Study and Their Status (as of Early 2011) .....		84

## Abbreviations and Acronyms List

AD	Anaerobic Digestion
BOD	Biological Oxygen Demand
BTU	British Thermal Unit
C&D	Construction and Demolition
COD	Chemical Oxygen Demand
CRV	Carbon Recovery Vessel
CT	Conversion Technology
D/F	Dioxins and Furans
DOE	Department of Energy
DST	Decision Support Tool
ECY WA	Department of Ecology Washington
EIA	Environmental Impact Analysis
EIS	Environmental Impact Statement
EOG	Envion Oil Generator
EPA	Environmental Protection Agency
EPIC	Environment and Plastics Industry Council
FGD	Flue Gas Desulfurization
FEMP	Federal Energy Management Program
GHG	Greenhouse Gas
HAP	Hazardous Air Pollutant
HDPE	High Density Polyethylene
HRSG	Heat Recovery Steam Generator
ICE	Internal Combustion Engine
ICI	Industrial Commercial and Institutional
ISO	International Organization for Standardization
KWh	Kilowatt hour
LCA	Life Cycle Analysis
LCI	Life Cycle Inventory
LCIA	Life Cycle Impact Assessment

LDPE	Low Density Polyethylene
LHV	Lower Heating Value
MMBtu	Millions of British Thermal Units (BTUs)
MBtu	Thousands of British Thermal Units (BTUs)
MRF	Materials Recycling Facility
MSW	Municipal Solid Waste
MW	Megawatt
MWh	Megawatt hour
NA	Not Applicable
NGO	Nongovernmental Organization
OARDC	Ohio State University's Agricultural Research and Development Center
PAC	Powered Activated Carbon
PET	Polyethylene Terephthalate
PM	Particulate Matter
P2O	Plastic2Oil
PP	Polypropylene
PS	Polystyrene
PVC	Polyvinyl Chloride
RDF	Refuse Derived Fuel
Syngas	Synthetic gas or Synthesis gas
TCE	Tons of Carbon Equivalent
TNMOC	Total Nonmethane Organic Carbon
TPD	Tons per Day
USDA	U.S. Department of Agriculture
VE	Visible Emissions
VOC	Volatile Organic Compound

## **Acknowledgements**

The research presented here is part of an emerging waste technologies project funded by the U.S. Environmental Protection Agency (EPA) through contract no. with RTI International. The support of EPA's Office of Research and Development project manager, Susan Thorneloe, as well as staff from EPA Region 5 including Susan Mooney and Christopher Newman are acknowledged. The authors would also like to acknowledge the assistance provided by technology vendors that were contacted as part of this project.

## **Disclaimer**

This report includes a summary of existing data and information for emerging waste conversion technologies in North America. The U.S. EPA does not advocate or endorse any particular technology or facility included in this report. The analysis and report were developed from January—September 2011 and the data and information included was collected from informal interviews with technology vendors, independent engineer analyses, vendor product information and presentations, and literature/website reviews. Data and information supplied by technology vendors was not independently verified. The U.S. EPA cannot verify the current viability of the information or data due to the dynamic nature of emerging waste conversion technologies and markets.

## Executive Summary

RTI International (RTI) was contracted by the U.S. Environmental Protection Agency (EPA), Office of Research and Development to conduct research to prepare a “State of Practice” report to support State and local decision-makers on the subject of emerging<sup>1</sup> waste conversion technologies. Focus was placed on the ability of these technologies to manage the currently non-recycled fraction of municipal solid waste (MSW). The specific objectives for this study and report were to develop:

- An overview of each waste conversion technology, identifying the types of feedstock that have or can be used in each process and the air, water, and waste emissions.
- Basic information on energy and mass balance for each technology.
- Basic information on the economics of the technologies to help decision-makers understand the key cost factors and economic feasibility.
- A listing and maps of proposed and operational facilities in the United States and pertinent examples for each technology.
- A summary of key findings and considerations decision-makers should be aware of when evaluating waste conversion technologies.

To address these objectives, a literature search and informal technology vendor interviews were conducted in early 2011. Conversion technology vendors for the general categories of pyrolysis, gasification, and anaerobic digestion (AD) were identified and asked to provide process, environmental, and cost information. Additionally, publicly available data sources were retrieved to complement the data received from each vendor. The technologies researched included **Table ES-1** presents a chart summarizing the main characteristics of the studied conversion technologies.

Based on the data collection and analyses, the study offers the following insights on the science and status of waste conversion technologies in North America:

- **There is a high level of uncertainty about the environmental performance and costs associated with these technologies.** Because most facilities using these technologies are demonstration plants, they are operating in batch-test mode and not as continuous-mode commercial plants. Until there are commercially operating facilities in North America, there will not be good real-world data to characterize the environmental aspects and costs for these technologies. The literature also revealed that most facilities that are commercial-scale are often operating in more of a demonstration mode and do not have waste contracts and/or energy or product contracts in place.

---

<sup>1</sup> This report defines emerging technologies as those where most of the existing examples are in a pre-commercial development stage, but past bench scale (i.e., in the field testing, pilot scale, or semi-commercial stage; see **Figure 1-1. Stages of Waste Conversion Technology Development**). While the application of these technologies to solid waste feedstocks is only emerging in the United States, these technologies have been applied for the management of solid waste in other parts of the world, such as Japan and Europe.

**Table ES-1. Overview of Conversion Technology Characteristics.**

Conversion Technologies	Pyrolysis	Gasification	Anaerobic Digestion
<b>Feedstock</b>	Plastics only	MSW	Food/yard wastes
<b>End Products</b>	Synthetic Oil, Petroleum Wax	Syngas, Electricity, Ethanol	Biogas, Electricity
<b>Conversion efficiency</b>	62–85%	69–82%	60–75%
<b>Capacities</b>	10–30 tons per day	75–330 <sup>2</sup> tons per day	10–100 tons per day (estimated)
<b>Product Energy Value</b>	15,000–19,050 BTU/lb	11,500 <sup>3</sup> -18,800 BTU/lb	6,000–7,000 BTU/lb (estimated)

- Conversion technologies do present another option (in addition to mass burn waste combustion) for managing non-recycled waste.** While conversion technologies present another option for managing non-recycled waste, it will be an estimated 5–10 years before the first-generation demonstration facilities transition to commercial operations based on estimated times for siting, permitting, construction, and contract development. The capability of conversion technologies to meet landfill diversion and/or energy production goals will likely depend heavily on the success of these first-generation facilities.
- It is difficult to compare the cost and performance of pyrolysis, gasification, and AD technologies directly due to differences in feedstock.** In general, pyrolysis technologies utilize only plastics, gasification technologies utilize MSW, and AD utilizes food, yard, and paper waste. Therefore, there are differences in feedstock energy values as well as in beneficial offsets. For pyrolysis, beneficial offsets are primarily based on the conversion of plastics to oil. For gasification, beneficial offsets include energy production and can also include recyclables (e.g., metals, glass, and other inorganics) removed in the up-front sorting process, but this component was not included in the analysis since we assumed post-recycling waste would be the input feedstock and few recyclables would be available. For AD, the benefit offsets are primarily based on the conversion of organic wastes to biogas, which is assumed to be used to produce electrical energy.
- Different technology vendors/facilities have specific variations on the process to enhance conversion efficiency and/or tailor the end product to site-specific markets.** The primary objective of the conversion technologies is to convert waste into useful energy products that can include synthetic or synthesis gas (syngas) or biogas, petroleum, and/or commodity chemicals. Syngas and biogas can be used

<sup>2</sup> Total capacity permitted based on vendor communications. Geoplasma’s St. Lucie, FL plasma gasification plant is permitted up to 686 tons/day, but the vendor could not be reached for confirmation.

<sup>3</sup> LHV of ethanol.

directly in industrial boilers or in an internal combustion engine (ICE) generator set (gen-set) to produce electrical energy. Petroleum and commodity chemicals are typically tailored to specific end-users (e.g., petroleum wax for cosmetics manufacturers). Each end product has different life-cycle offsets that may affect the overall life-cycle environmental impact of the process.

- **Using the currently available data, a high-level<sup>4</sup> life cycle environmental assessment conducted for the conversion technologies indicates that the technologies do offer environmental benefits as compared to landfill disposal.** When energy recovery potential is accounted for, we estimated that gasification (excluding energy production and materials recycling offsets) of MSW saves 6.5–13 million British thermal units (MMBtu) per ton as compared to landfill disposal and pyrolysis of waste plastics saves 2–4 MMBtu per ton as compared to landfill disposal. Likewise, our results show that gasification of MSW saves 0.3–0.6 tons of carbon equivalent (TCE) emissions per ton of MSW treated as compared to landfill disposal. Pyrolysis of waste plastics saves 0.15–0.25 TCE emissions per ton as compared to landfill disposal. AD of organic waste saves 0.11–0.13 TCE per ton compared to landfill disposal. Because many of the technologies focus on feedstocks that are not suitable for typical recycling practices, comparisons were made only to landfill disposal and not to recycling.
- **Pyrolysis for plastics is in a more mature stage of development than other technologies.** Of the current conversion technologies, plastics-to-oil pyrolysis technologies are more mature than MSW-based technologies (typically gasification and AD), in part because of the decreased variability of the incoming feedstock—e.g., three facilities at a commercial stage were identified for plastics-to-oil pyrolysis, while none at a commercial stage were identified for gasification and AD. This study did not look at supplementation with agricultural waste or increased use of agricultural waste for biofuels production.
- **The waste processing costs obtained indicate that the cost/ton for pyrolysis is comparable to MSW options, such as recycling and landfilling, and that the cost/ton for gasification and AD is higher.** The estimated waste processing cost for pyrolysis is approximately \$50/ton of plastics, close to \$90/ton of MSW for gasification, and close to \$115/ton of organics for AD. This cost is generally related to the capital and operating costs required to run the process and dispose of any residuals. U.S. averages for landfill disposal and recycling, for comparison, range from \$30–75/ton of MSW, depending on region.

---

<sup>4</sup> The data used for this assessment were provided by industry vendors and were not independently validated. In addition, the datasets used to characterize the technologies vary in the level of detail and the number of values obtained for particular input parameters, with only one value obtained for certain parameters.

## Section 1: Introduction

New technologies to convert municipal and other waste streams into fuels and chemical commodities, termed conversion technologies, are rapidly developing. Conversion technologies are garnering increasing interest and demand due primarily to alternative energy initiatives. These technologies have the potential to serve multiple functions, such as diverting waste from landfills, reducing dependence on fossil fuels, and lowering the environmental footprint for waste management. Conversion technologies are particularly difficult to define because their market is in development and many of their design and operational features are not openly communicated by vendors.

RTI was contracted by EPA's Office of Research and Development to conduct research to evaluate and develop an easy-to-understand "State of Practice" report for State and local decision-makers on the suite of emerging waste conversion technologies. The technologies information was collected in the December 2010- February 2011 timeframe and include the general categories of pyrolysis, gasification, and AD.

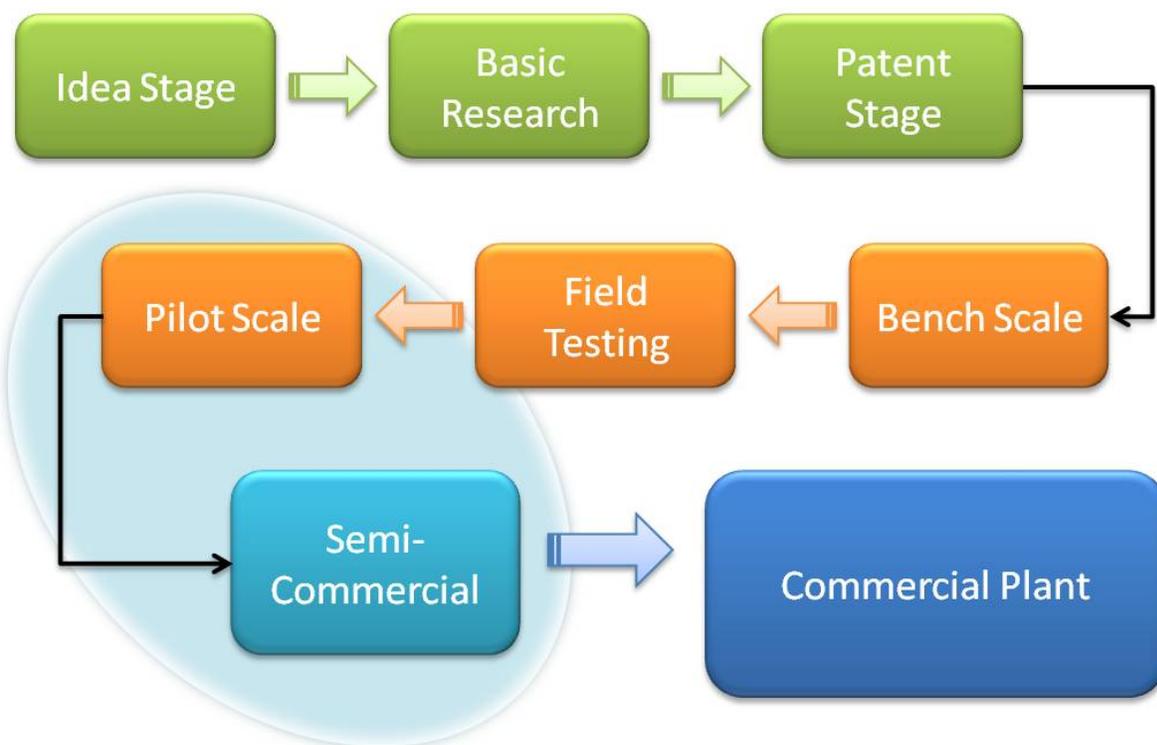
The objectives for this report were to develop:

- An overview of each waste conversion technology, identifying the types of feedstock that have or can be used in each process and the air, water, and waste emissions.
- Basic information on energy and mass balance for each technology.
- Basic information on the economics of the technologies to assist decision-makers in understanding the key cost factors and economic feasibility.
- A list and maps of proposed and operational facilities in North America for each technology.
- A summary of key findings and considerations decision-makers should be aware of when evaluating waste conversion technologies.

To address these objectives, this study evaluated real-world case examples and data and information from the literature. This analysis provides a better understanding of the range of emerging conversion technologies available that use MSW or specific MSW fractions as its feedstock, to identify and profile specific technology vendors. The study was also designed to identify and quantify the potential cost and life cycle environmental burdens/benefits of the technologies as compared to existing landfill disposal. Technology categories are described in detail and potential benefits and impediments are reviewed. Additionally, a life cycle inventory analysis was performed for the general technology categories using data from technology vendors in combination with data obtained from the literature.

## 1.1 Conversion Technology Development Stages

There are a number of ongoing efforts in North America to develop and commercialize waste conversion technologies. The current situation is very dynamic, with new technology proposals, new vendors, mergers and acquisitions, and redesigns or closings occurring almost weekly. It is useful to consider the technology development stages as illustrated in **Figure 1-1** when discussing waste conversion technologies. There are technologies at every stage of the development cycle. At the time the facilities specific data used in this report were collected (early 2011), there were only a few commercial-scale facilities operating.



**Figure 1-1. Stages of Waste Conversion Technology Development.**

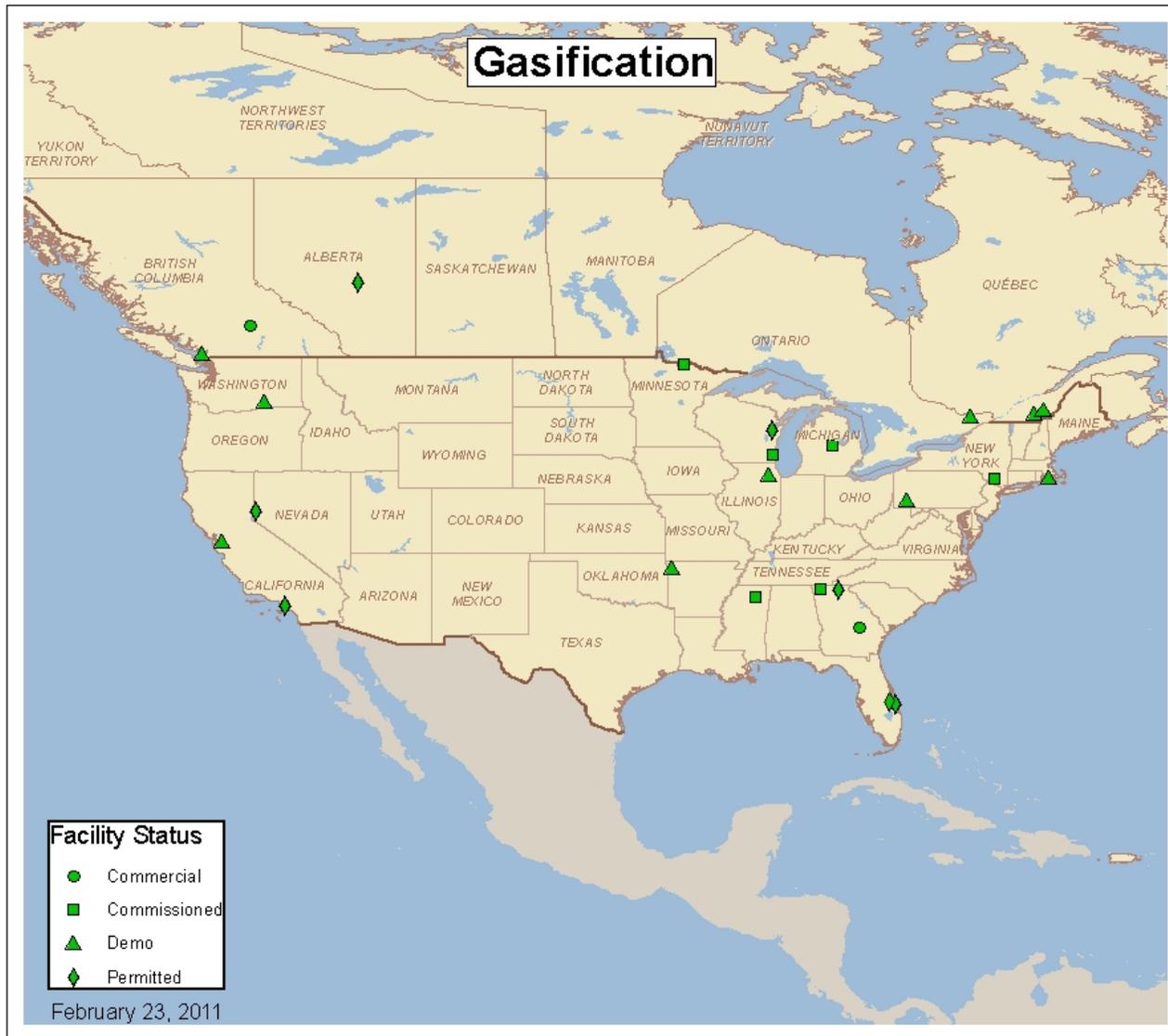
*Note: Most of the facilities investigated in this report are in the stages within the shaded area.*

Most facilities are at a pilot or semi-commercial stage. It was found that even facilities that are commercial-scale are often operating in more of a demonstration mode and most do not have waste contracts and/or energy or product contracts in place.

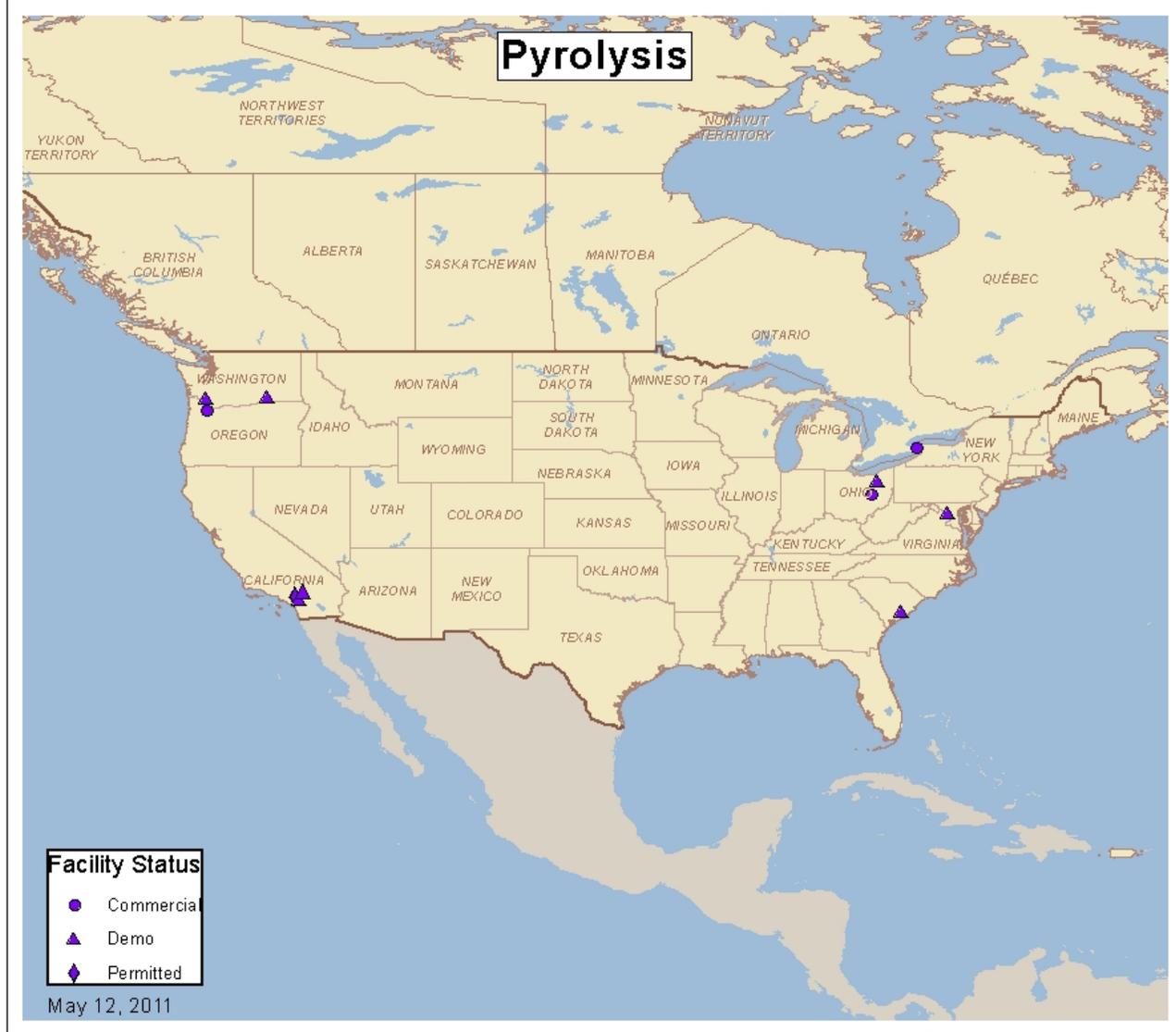
This study focused on technology vendors and facilities that were at the pilot to commercial plant stages. **Figure 1-2** illustrates the locations of existing North American waste conversion facilities by main technology category of AD, concentrated acid hydrolysis, gasification, and pyrolysis. Gasification and pyrolysis are the primary technology categories that can accept MSW (or MSW fractions), whereas AD and concentrated acid hydrolysis primarily accept organics. **Figure 1-3** through **Figure 1-5** illustrate the locations and stages of technology development of gasification, pyrolysis, and AD facilities, respectively. **Appendix B** presents additional information about the facilities presented in the maps and their status.



Figure 1-2. Waste Conversion Facility Types and Locations in North America (data collected in early 2011).



**Figure 1-3. Gasification Facility Locations and Status in North America (data collected in early 2011).**



**Figure 1-4. Pyrolysis Facility Locations and Status in North America (data collected in early 2011).**



Concentrated acid hydrolysis was not included for further consideration in this report. There is only one facility and overall conversion technologies appear to be moving in the direction of AD, gasification, and pyrolysis.

Because there were so few true commercial facilities in operation, it was difficult to present reliable estimates for cost and life cycle environmental aspects. Most of the facilities covered in this report were still in pilot and demonstration stages. As facilities transition to fully operational commercial facilities, one would expect the process inputs/outputs to stabilize and cost and environmental aspects to become more consistent and reliable.

## **1.2 Challenges for Implementing Conversion Technologies**

Three main challenges must be successfully addressed for any of these technologies to be implemented successfully: legislative/regulatory, contract and market development, and social stigma.

One of the most significant barriers that exist for waste conversion facilities is the regulatory aspect. It is difficult to obtain permits from local and state governments. It may be necessary to obtain solid waste handling permits through the appropriate local agency. It is also important for facilities to apply for and acquire air permits in order to regulate criteria pollutants and toxic air pollutants. One such permit would be a Title V Permit, which sanctions construction of permitted emissions units as well as initial operations (FL DEP, 2011). Excess emissions, such as those from startup, shutdown, and malfunction operations are also specified in air permits. Water quality permits are necessary to regulate discharges to surface and ground water. The local or county planning agency likely has requirements for the planned facility that encompass building, grading, water system, shoreline, utility, site plan review, septic system, floodplain development, variance (zoning, shoreline, etc.), and outdoor burning (ECY WA, 2011). The permitting process is overall a complex one, and the facility owners may easily have difficulties that lead to substantial delays in construction. It is not uncommon for companies to apply for and be rejected from these permits several times. Without acquiring permits, these facilities may not begin construction operations.

Before a facility is built, it may be necessary for an Environmental Impact Assessment (EIA) to be prepared. The EIA is a comprehensive evaluation of the positive and negative impacts that the proposed facility may have on the natural environment, as well as social and economic consequences. After the assessment is completed, it is likely that an environmental impact statement (EIS) will need to be written. An EIS is a decision-making tool that is required for proposed projects that may significantly impact the environment. Included in this statement is a discussion of purpose of and need for the project, alternatives, and environmental effects of the proposed project.

After firms receive permits to operate, they must be able to secure contracts with waste facilities in order to have a secure, continuous feedstock. Feedstocks are often one of the most challenging aspects of successfully operating a conversion facility. The quantity of feedstock needs to be relatively constant because the systems are optimized for a specific flow rate. It is

also necessary for quality and volume of feedstock to be taken into account. Brightstar Environmental is an example of one company that encountered issues with feedstock supply. Brightstar was a subsidiary of Energy Developments Limited and located in Australia's New South Wales province in the city of Wollongong. The gasification facility was forced to close in 2004 due to feedstock contractual issues.

Most revenue from these processes comes from the sale of oil and gas products. Therefore, if markets are not developed for recycled products from the pre-sorting process, revenue that otherwise would have been generated is lost. Furthermore, if no market share exists and clients are not found for the oil or gas products, the facilities will be forced to close due to a lack of revenue.

Ash and other residual products from waste conversion technologies can be a regulated hazardous waste or solid waste and will need to be assessed and approved by local or state agencies to determine their potential use (e.g., as aggregate) and appropriate disposal (e.g., conventional versus hazardous waste landfill). Slag produced from a number of technologies is generally characterized by technology vendors as nonleachable and may not pose the same problem, but it may need to be tested for compliance with state and local regulations or standards and will likely need to be approved for reuse applications. If a market is developed for slag and it is approved for reuse, it may be sold. If not, the slag must be landfilled.

The public's negative association with these facilities is another barrier that needs to be overcome. Smell, noise, and visual aesthetics complaints are common from community members after MSW facilities have been installed. The negative stigma has led to some difficulty in locating sites for these plants. Some national nongovernmental organizations (NGOs), such as the Sierra Club, believe facilities that use waste to convert to fuel lead to a disincentive for individuals and communities to reduce their consumption. Global Alliance for Incinerator Alternatives is a conglomeration of over 500 grassroots organizations opposed to incinerators as well as other waste technologies. They argue that the emissions associated with these facilities, including gasification, pyrolysis, and plasma arc fuel climate change, do not address the real issue of overconsumption, and decrease resources that are available for waste reduction and recycling programs. Most easily accessible information that drives public opinion is derived from these NGOs, which leads to a negative perception of these facilities. However, some communities that have installed these facilities in their communities have a much more positive opinion of the facilities.

In order for facilities to be successful, they should reduce smells and noises emanating from operations. To reduce public resistance to these facilities, it would be helpful for companies to provide outreach to the public to educate them about technological advances and other positive aspects of these technologies. Some measures that may help include siting facilities at brownfields (i.e., abandoned or underused industrial and commercial facilities available for reuse), the use of dome designs to hide smokestack visibility, and integrated "utility campuses" that consist of sewage treatment, electricity generation, and water reclamation facilities (Lawrence, 2009).

Some legislative actions are designed to encourage the development of conversion technologies. The federal government provides several grants and loans for feedstock development, biofuels, and biobased product development for technologies such as these conversion facilities. The Biomass Research and Development Initiative is one major source of funding. The initiative is an interagency effort of senior decision-makers from various federal agencies, including the U.S. Department of Agriculture (USDA) and U.S. Department of Energy (DOE) as well as the White House. The USDA awards loans to companies that demonstrate the potential benefits of their conversion technology processes. U.S. DOE provides funding for the conversion of biomass to various fuels, such as those produced through the use of conversion technologies. One company awarded a grant through this program is Enerkem. The company was also awarded an \$80 million loan through the Biorefinery Assistance Program.

Another federal program designed to assist energy efficiency projects is the DOE's Federal Energy Management Program (FEMP). The objectives of FEMP are to lower government costs by advancing energy efficiency and water conservation and increasing renewable resource use. Agencies are guided by FEMP to use private sector financing for energy projects with the use of Utility Energy Service Contracts or DOE's Super Energy Savings Performance Contracts. Other federal assistance programs include EPA's Innovations Work Group, the National Center for Environmental Research, and DOE's Office of Energy Efficiency and Renewable Energy.

State and local governments also provide incentives for the development of alternative energy. For example, Iowa's Department of Natural Resources Land Quality and Assistance Division offers a loan program that "encourages implementation of innovative waste reduction and recycling techniques, develops markets for recyclable materials and products, and encourages the adoption of the best waste management practices" (U.S. EPA, 2011). Other states, such as California, provide extensive research and development opportunities for waste reduction. One such group is the California Energy Commission, which recently announced a \$4.5 million grant to aid the development of an AD plant in Perris, California.

### **1.3 Report Structure**

Section 2 of this report contains general descriptions and definitions for conversion technology categories and subcategories. Section 3 provides profiles of real-world companies that are developing and commercializing technologies and specifics about their respective processes. Section 4 includes the approach, key assumptions, and results for life cycle inventory (LCI) analyses constructed for the main categories of conversion technologies studied based on company-specific data and information and also data and information collected from the literature. Section 5 presents the overall findings and recommendations.

## Section 2: Conversion Technology Categories

In this section, the following categories of thermal and biochemical conversion technologies are described: pyrolysis, gasification, plasma arc, and anaerobic digestion (AD). Thermal conversion processes are characterized by higher temperatures and conversion rates than biochemical processes. These technologies contain a continuum of processes ranging from thermal decomposition in a primarily nonreactive environment (commonly referred to as pyrolysis/cracking processes) to decomposition in a chemically reactive environment (or gasification processes).

The definitions adopted in this report are presented in the following sections. It should be noted that vendor technologies are often difficult to fit under one conversion technology category and sometimes include characteristics common to both gasification and pyrolysis. For example, in a two-stage (pyrolysis-gasification) fixed bed gasification process, some of the oxygen injected into the system is used in reactions that produce heat, so that pyrolysis (endothermic) gasification reactions can initiate, after which the exothermic reactions control and cause the gasification process to be self-sustaining.

### 2.1 Pyrolysis

Pyrolysis is defined as an endothermic process, also referred to as cracking, involving the use of heat to thermally decompose carbon-based material in the absence of air or oxygen (i.e., no burning). Its main product is a gaseous mixture of carbon monoxide (CO) and hydrogen (H<sub>2</sub>) called “syngas” that can be used for steam and electricity generation. Other byproducts of this process are commonly reported, but the list and proportion of each differs depending on reactor design, reaction conditions, and feedstock.

#### 2.1.1 Types of Pyrolysis

Various technology vendors include different variations and names for pyrolysis processes in their technology descriptions, which can be confusing to waste managers. Technologies that are categorized as pyrolysis generally belong to one of the following process categories:

- ***Thermal pyrolysis/cracking***—The feedstock is heated at high temperatures (350–900 degree Celsius) in the absence of a catalyst. Typically, thermal cracking uses mixed plastics from industrial or municipal sources to yield low-octane liquid and gas products. These products require refining to be upgraded to useable fuel products.
- ***Catalytic pyrolysis/cracking***—The feedstock is processed using a catalyst. The presence of a catalyst reduces the required reaction temperature and time (compared to thermal pyrolysis). The catalysts used in this process can include acidic materials (e.g., silica-alumina), zeolites (e.g., HY, HZSM-5, mordenite), or alkaline compounds (e.g., zinc oxide). Research has shown that this method can be used to process a variety of plastic feedstocks, including low density polyethylene (LDPE), high density polyethylene (HDPE), polypropylene (PP), and polystyrene (PS).

- **Hydrocracking** (sometimes referred to as “hydrogenation”)—The feedstock is reacted with hydrogen and a catalyst. The process occurs under moderate temperatures and pressures (e.g., 150–400 °C and 30–100 bar hydrogen). Most research on this method has involved generating gasoline fuels from various waste feedstocks, including MSW plastics, plastics mixed with coal, plastics mixed with refinery oils, and scrap tires.

### 2.1.2 Issues Associated with Pyrolysis

The process of pyrolysis creates residues including char, silica (sand), and bottom ash. Some of these residues may be reused (if approved by an environmental agency) while others must be disposed of in a landfill. The amount of residual waste produced is about 15–20 percent of the overall feedstock used in the process. Litter, odor, traffic, noise, and dust must also be assessed but vary according to the differences in facility technology, size, and feedstock.

## 2.2 Gasification

Gasification is the partial oxygenation of carbon-based feedstock to generate syngas. The process is similar to pyrolysis, except that air or steam is added to promote gasification, forming carbon monoxide, hydrogen, and methane. Gasification is an endothermic process and requires a heat source, such as syngas combustion, char combustion, or steam. The primary product of gasification, syngas, can be converted into heat, power, or chemical products, or used in fuel cells. The current main types of gasification processes for MSW include the following:

### 2.2.1. Types of Gasification

- **High temperature gasification**—High temperature gasification reactors, as described in ARI (2007), can reach up to 1,200 °C and produce an inert byproduct, or slag, that does not need further processing to be stabilized. The syngas and steam (also produced from this process) are used for power generation. Typically, this technology processes a mix of carbonaceous waste including paper, plastics, and other organics with a moisture content of up to 30 percent, which avoids the need for drying. In general, there are no water emissions because conventional water treatment systems are used to convert process discharges to useable process and/or cooling water. Treatment systems include settling and precipitation to capture and remove solids, which are returned to the high-temperature reactor.
- **Low temperature gasification**—Low temperature gasification reactors, as described in ARI (2007) and RTI (2005), operate at temperatures between 600 and 875 °C and produce ash that could be sent to a vitrification process to make it inert and available for other uses. Syngas is the main product from this process and is used for electricity generation. This process can also recover steam energy. Separate estimates of energy from syngas and steam are obtained. This technology is assumed to require a feedstock with a moisture content of 5 percent or less and includes a drying pre-processing. A mix of gases and aerosols are produced from gasification and are sent to be quenched. The resulting liquid is cooled with chilled water and water is recovered and sent to a solids mixing tank. Char, brine, and bio-

oils may also be recovered. Bio-oils are recycled back to the process and char and brine are included as water and solid waste emissions.

- **Plasma gasification**—Plasma gasification converts selected waste streams including paper, plastics, and other organics, hazardous waste, and chemicals to syngas, steam, and slag. In this technology, the gasification reactor uses a plasma torch where a high-voltage current is passed between two electrodes to create a high-intensity arc, which in turn rips electrons from the air and converts the gas into plasma or a field of intense and radiant energy with temperatures of thousands of degrees Celsius. The heated and ionized plasma gas is then used to treat the feedstock. Petroleum coke is assumed to be added to the reactor to support reduction reactions and to stabilize the slag. No drying pre-processing of the feedstock is required and the feedstock is assumed to have up to 30 percent moisture content. Syngas and steam are then used for power generation, included in the estimate of total electricity offsets. The slag, also produced in this process, is quenched prior to use.

### 2.2.2 Issues Associated with Gasification

As with pyrolysis, residues such as slag and ash that are produced in the gasification process may need to be disposed of at a landfill.

Another potential issue that may need to be assessed is the level of pre-sorting necessary. Some pre-processing will be needed for many of these facilities. For some gasification technologies, however, a significant presorting process will be required, including the removal of recyclables, sorting, shredding, and drying. The pre-sorting process is necessary to make the feedstock more homogenous and to increase efficiency of the overall process. The amount of material removed depends on the feedstock composition and the specific process requirements.

## 2.3 Plasma Arc

A plasma arc device is a heating method that uses high temperatures to reduce MSW into elemental byproducts. The plasma itself is a collection of free-moving ions and electrons formed with the use of a large voltage across a gas volume at atmospheric pressure. Electrons in the gas molecules are then able to break apart and move toward the positive side of the voltage. The gas molecules are converted into positively charged ions able to transfer an electric current and produce heat. The same process is intrinsic to the formation of lightning in the atmosphere.

### 2.3.1 Types of Plasma Arc

Two types of plasma arc devices or plasma “torches” exist: the transferred torch and non-transferred torch. The former produces an electric field between an electrode, located at the tip of the torch, and the reactor wall. When the strength of the field is high enough, an electric arc is created. The electric arc is comparable to a spark plug in an automobile. The non-transferred torch creates the electric arc within the torch and sends a process gas through the arc, where it is heated and exits through the torch as a hot gas. Temperatures of 3,870 °C and

higher are generated in the ionized plasma. In the reactor chamber, nonionized gas temperatures can reach 925–1,200 °C. Slag temperatures are approximately 1,650 °C. The high temperatures are able to break apart the molecules to produce simpler molecules including carbon monoxide, hydrogen, and carbon dioxide. An inert glassy slag material is obtained and may be used (pending testing and approval from state and/or local regulatory agencies) for construction purposes as an aggregate.

### 2.3.2 Issues Associated With Plasma Arc

Disadvantages exist for plasma arcs, especially in relation to feedstock size, electricity requirements, and cost issues. Before MSW may be used in this process, the feedstock must be shredded to a size of six inches or less. A large portion of electricity generated is necessary for the operation of the plasma torches. This leads to a net reduction in electricity generation from the facility, which can vary significantly and depends largely on the throughput. The parasitic load, or energy consumed even when the system is not in use, is also significant. Any impacts on energy production are ultimately reflected in the revenues that the plasma arc systems produce. Currently, no plasma arc facilities that process MSW are operating at a commercial scale in the United States.

## 2.4 Anaerobic Digestion

AD is a biochemical conversion process that decomposes organic material in the absence of oxygen (O<sub>2</sub>). Organic waste materials such as manure, agricultural wastes, and biodegradable fractions of industrial, commercial, and MSW (or fractions of MSW) can be used as feedstocks for anaerobic digesters. The main product of AD is a methane-rich biogas, which can be combusted to generate heat and/or electricity, converted to pipeline quality gas, or further refined to create biomethane, a transportation fuel. Byproducts of AD include CO<sub>2</sub> and undigested solids. Depending on the type of feedstock used, the undigested solids may have economic value when refined and used as a fertilizer soil amendment.

### 2.4.1 Types of Anaerobic Digesters

There are several types of anaerobic digesters. Digesters can be classified into “wet” or “dry” systems, depending on the feedstock; into single- or multiple-stage systems, depending on their complexity; and batch or continuous flow systems, depending on the feedstock input method.

- **“Wet” or “dry” systems**—Wet systems generally process feedstocks with a total solid content of less than 15 percent, whereas dry systems process feedstocks with greater than 15 percent total solid content. Wet systems are most appropriate for wastewater AD. Dry systems are preferred for MSW because bacteria have a higher survival rate and less pre-handling is required.
- **Single-stage or multiple-stage systems**—Multi-stage systems are more expensive than single-stage systems to construct and operate; however, they have higher loading rates and greater feedstock flexibility. For MSW, the majority of AD systems are single-stage due to the complexity and cost barriers of multi-stage systems, but the prevalence may change as technology becomes more affordable and standardized.

- ***Batch or continuous flow systems***—Batch systems process waste within a single sealed reactor or holding tank, whereas continuous systems use a series of reactors. Batch systems require less precision and have lower construction costs than continuous systems, but may result in inconsistent biogas production and incomplete degradations. Comparatively, continuous systems increase process efficiency due to increased control over bacterial reactions taking place within the reactors.

AD for MSW is being used effectively in Europe. However, no commercial AD facilities that process MSW are in operation in the United States. A pilot facility, East Bay Municipal Utility District, is currently operating in California. It co-digests food scraps from restaurants within the San Francisco Bay area at its wastewater treatment plant in order to generate biogas that is used to produce electricity. The study has found that converting 100 tons per day (TPD) for five days per week offers enough power for 800–1,400 homes annually.

#### **2.4.2 Issues Associated With Anaerobic Digestion**

Some pre-sorting and pretreatment is necessary to limit the clogging of the pumps and to reduce inert material(s) located in the reactor. Also, it is necessary to remove metals and plastics prior to adding the stream into the AD. Otherwise, the stream may contaminate the process. Generally, the material handling systems include extensive receiving, particle size reduction, and separation processes before the feedstock may be fed into the digester.

## Section 3: Technology Vendor Case Examples

This section describes case examples of specific conversion technology vendors that manage MSW and that were considered for the life-cycle environmental assessment described in **Section 4**. Examples include thermal and biochemical conversion technologies. As described in **Section 2** thermal conversion processes such as pyrolysis and gasification are characterized by higher temperatures and conversion rates than biochemical processes such as AD. The technologies contain a continuum of processes ranging from thermal decomposition in a primarily nonreactive environment (commonly referred to as pyrolysis/cracking processes) to decomposition in a chemically reactive environment (or gasification processes). Even though we will be presenting the conversion technology vendors grouped under the pyrolysis, gasification and AD categories, it should be noted that vendor technologies are often difficult to fit under one category, in particular for thermal processes, and sometimes include characteristics common to both gasification and pyrolysis. For example, in a two-stage (pyrolysis-gasification) fixed bed gasification process, some of the oxygen injected into the system is used in reactions that produce heat, so that pyrolysis (endothermic) reactions can initiate, after which the exothermic reactions control and cause the gasification process to be self-sustaining.

The following are the technology vendors considered. It should be noted that it is not our attempt to compare the performance of the various technology vendors and that the life cycle modeling in **Section 4** will use data ranges defined after considering all the data obtained on these processes.

- **Technology vendors classified as pyrolysis vendors:**
  - Agilyx (OR)
  - Envion (MD)
  - Climax Global Energy (GA)
  - JBI (NY)
- **Technology vendors classified as gasification vendors:**
  - Enerkem (Canada)
  - Plasco (Canada)
  - Ze-gen (MA)
  - Geoplasma (FL)
- **Technology vendors classified as AD vendors:**
  - Yolo County (CA)
  - Quasar (OH)

These vendors were selected because of their relatively advanced stage of technology development and their willingness to participate in the project. **Attachment C** presents the data questionnaires. The data provided by the vendors have not been independently verified and very little information was obtained about the tests and test conditions used to obtain the data.

Gathering this type of information as well as performing an independent verification are part of the recommendations from this analysis.

As part of the data collection process, which took place between December 2010 and January 2011, RTI designed a questionnaire to collect life cycle inventory (LCI) data and sent it to each of the case study vendors. Six of the facilities—Agilyx, Envion, Climax, JBI, Enerkem, and Ze-Gen—participated by e-mailing or phoning in responses. Simultaneously, data and information from additional publicly available data sources for each vendor were identified and compiled. Each section and data table in **Section 3** indicates whether the data was obtained via communication with the facility or through literature, such as engineering reports, outreach materials, websites, and/or environmental assessment reports. For case study facilities that did not participate in the survey process, the data obtained for the study has been collected from literature resources only. Given the emerging nature of these technologies and the likelihood that most data corresponds to testing under controlled batch tests, the uncertainty associated with the data should be considered high.

### 3.1 Pyrolysis Technology Vendors

#### 3.1.1 Agilyx: Tigard, Oregon

Agilyx, formerly known as Plas2Fuel, was founded in 2004 and has an operating demonstration facility in Oregon. Agilyx uses waste plastics of any type (resin identification codes 1–7s) as feedstock and converts it into synthetic crude oil. The plastic waste can be commingled and no pre-sorting or pre-cleaning is needed. The company estimates that approximately 10 tons of plastic may be converted to 60 barrels (or 2,400 gallons) of oil on a daily basis through a pyrolysis process. All of the information and data about Agilyx was obtained via phone survey.

##### *Process Details*

One main purpose of the Agilyx system is to handle any type of plastic feedstock and contamination level, thus reducing time and cost of the process. Agilyx uses custom-designed cartridges to convey feedstock to their processing equipment. Each system is modular and may be located at the collection facility to reduce costs associated with feedstock transportation. These systems may be scaled up or down, based on the amount of feedstock available.

Pre-processing of the plastic waste includes industry-standard grinding and shredding to a density target of 20–21 lbs/ft<sup>3</sup>. The cartridges are filled with plastic feedstock and inserted into a Plastic Reclamation Unit, which is a large processing vessel. A light industrial burner heats air to about 593.3 °C, and the air is circulated around the exterior of the cartridge while the plastics are transformed from a solid to a liquid, and finally a gas. In the gaseous form, the plastics have been broken down into oil-sized molecules. The heating system is closed loop in order to diminish heat loss. The gases are removed from the cartridge into a central condensing system. The gases are cooled in this system and condensed into synthetic crude oil. Waste materials are extracted from the stream, while lightweight gases that do not condense continue downstream. The gases contain about 80 percent methane, propane, and butane species. The gases are then treated by an environmental control device. The crude oil moves into a coalescing and settling process and is eventually moved to an above-ground storage tank

outside the facility for transport to a refinery. Crude oil may be refined into products that conform to ASTM specifications, including ultra-low sulfur diesel. The process is set up to operate on a continuous basis: 24 hours a day, 7 days a week. It is assumed for purposes of this report that operations occur 312 days a year for 24 hours a day.

### ***Performance Information***

Agilyx's performance information includes a process energy ratio, which measures the British thermal units (BTUs) received from the process (output) for each BTU input to the process. According to the company's representatives, the process energy ratio (without including the energy value found in char) is about 5:1. With the energy value of the char included, the ratio is about 6:1. The BTU value of the crude oil produced is about 19,250 BTU/lb. The energy load requirements are purchased from the local utility company. Agilyx has the ability to generate both heat and electricity onsite (i.e., go off-grid), but their costs are lowered by purchasing power. Natural gas is used as a supplemental fuel during startup and emergency situations. Other fuels could be used as well.

### ***Process Emissions***

**Appendix A** presents the emissions data obtained for Agilyx. Water requirements are minimal because it is recycled and filtered for contaminants. Sorbent cartridges, or wastewater treatment filters, are sent to a contractor to be cleaned and then are reused. No other inputs, such as catalysts, are necessary for the process. The primary residual in the process is char, and the company is attempting to find a commercial outlet for the product. About 8 percent of the feedstock generally becomes char, but the values can range from 1–50 percent, depending on the type of plastic used as feedstock.

Air emissions include permitted volatile organic compound (VOC), nitrogen oxide (NO<sub>x</sub>), and carbon monoxide (CO) emissions. Particulate matter (PM) and sulfur dioxide (SO<sub>2</sub>) are considered *de minimus* and are unregulated. Approximately 1,500 short tons per year of carbon dioxide (CO<sub>2</sub>) are emitted from the light industrial burners. Agilyx is permitted to emit 39 short tons per year of nitrogen oxides and 39 short tons per year of VOCs but only discharge around 2.5 short tons of each pollutant. Agilyx is also allowed to emit 99 short tons per year of carbon monoxide, but actually emits about 1.5 short tons. Emissions of hydrogen chloride (HCl), SO<sub>2</sub>, NO<sub>x</sub>, and VOCs are based on a proposed limit, not actual emissions levels.

### ***Cost Information***

Agilyx did not provide any cost information about their processes.

### ***Additional Aspects and Future Outlook***

Agilyx is the only company known to have a refinery off-take agreement within this industry. Currently, Agilyx is shipping crude oil from its demonstration facility in Portland, Oregon, to the U.S. Oil and Refining Co., located in the Pacific Northwest. The impacts of shipping and transportation costs in general were not researched in this study, but they suggest additional burdens that should be considered when evaluating the financial viability of the project.

### 3.1.2 Envion: Derwood, MD (to be relocated to Florida in 2011/2012)

Envion was founded in 2004 and focuses solely on the conversion of waste plastics to oil. Advantages of the process include relatively easy reactor construction and operations as well as the high efficiency and high BTU value of output products. One reactor that demonstrates the company's operations has been running since 2009. In terms of design capacity, a single Envion unit can process up to 10,000 tons of plastic waste annually. The company estimates that each ton of plastic may be converted to three to five barrels of refined petroleum through a pyrolysis process. This technology can be scaled up or down through the addition of reactors. Envion provided energy and emissions data through a combination of personal communications and transmittal of a 2010 independent engineering report, although general process information has been obtained from their website (RW Beck, 2010).

#### *Process Details*

The Envion technology uses chipped plastics as feedstock for the pyrolysis process. An illustration of the process is shown in **Figure 3-1**. The plastics must be chipped to less than 1.5 inches and melted. Approximately 1.22 tons of raw feedstock per hour is able to be processed. About 1.8 tons per hour are processed after water and contaminants are purged. The feedstock is composed of high-density polyethylene (HDPE), polypropylene (PP), low-density polyethylene (LDPE) plastics, and polystyrene (PS), polyethylene terephthalate (PET), and polyvinyl chloride (PVC). PS, HDPE, LDPE, and PP are preferred because they provide the best oil yield. Only restricted amounts of PET containers are used because they lead to much higher values of waste product, mainly sludge. PVC plastics are also used in very small amounts due to the chlorine compounds released in the cracking process. The exact proportions of feedstock types are unknown, but would likely be comparable to the typical MSW plastic composition.

In the pretreatment process, plastics move through a magnetic removal section and into the melting and screening section where they are liquefied at 300 °C. The plastics then go through a screen to filter nonplastic contaminants like glass and nonmagnetic metals. The feedstock and remaining organic contaminants (approximately 6 percent) pass through the screen to the main cracking reactor where plastics become an oil gas vapor. In order to power the cracking process, Envion uses far infrared heaters. Crude oil exits as an "oil gas" through a packed tower in order to remove contaminants. Oil gas is cooled and moved to tanks that separate reactor effluent into three streams: process gas stream, product oil stream, and water stream. Light components in the oil gas stream such as butane, propane, and methane exit the separation tank and are moved to an ICE gen-set to produce electricity for the process. The efficiency of the ICE gen-set depends on the composition of the process gas. The product oil is eventually transferred to primary oil tanks. Waste oil and water contaminants condense to liquid form and are sent to the sludge tank.

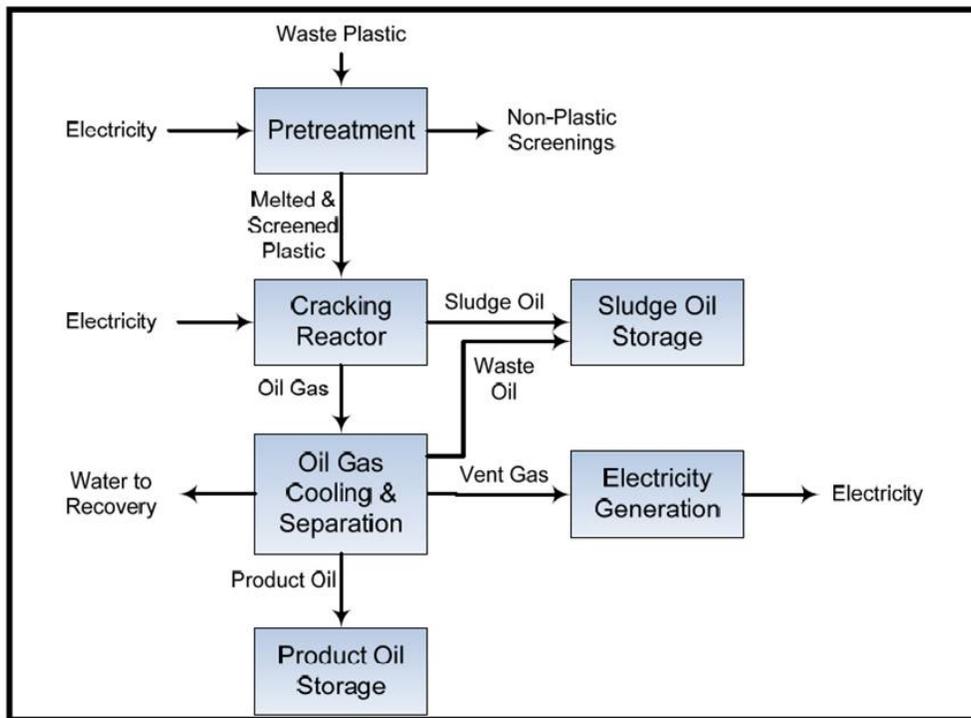
The sludge oil tank remains at an elevated temperature so contents do not solidify. To empty the tank, some product oil is moved to the sludge oil tank to blend the oil so it may be moved to a heated asphalt transfer truck. The gas that does not condense is sent to an ICE generator to produce electricity. A portion of the process electricity may be offset by this gas.

Other inputs for this process include about 750 KW of electricity and up to 0.435 tons of water per ton of raw plastic, depending on the amount of water needed for the cooling tower. Material byproducts include process gas that is currently used to offset 10–25 percent of electricity used in the process. The sludge byproduct accounts for about 15 percent of overall feedstock. Currently, the sludge is stored in barrels since the BTU value of the sludge indicates that it may have market potential as an energy source. Residuals include nonmetal contaminants at a rate of 0.52 TPD, or about 2 percent of the feedstock. About 1.56 TPD of small contaminants are collected, which makes up about 6 percent of the overall feedstock.

**Performance Information**

Approximately 1 short ton of plastic produces 4.22 barrels of oil with a value of about 18,347 BTU through the Envion process. The parasitic load is about 480 KWh/ton of waste after process gas has been combusted to generate electricity. The energy recovery efficiency of the Envion technology can be highly variable depending on the feedstock, but is generally about 62 percent.

**Envion Plastic-to-Oil Technology  
Block Flow Diagram of Plastic-to-Oil Process**



**Figure 3-1. Envion Pyrolysis Process Flow Diagram.**

(Source: [www.envion.com](http://www.envion.com))

**Process Emissions**

**Appendix A** presents the emissions data obtained for Envion. Estimates of emissions were gathered from the vendor and include small amounts of methane, sulfur dioxide, and nitrous

oxide emissions. Mercury emissions are about 0.016 micrograms/ton of waste. Lead emissions are 0.106 mg/L of oil. Envion did not provide any information on water emissions.

#### ***Cost Information***

The cost per design capacity is estimated to be \$7.6 million per unit or \$280,700/TPD. In terms of process cost per ton, estimates range from \$17 to \$60, assuming 80 percent of electricity use in the production process is from the grid. Costs would be lower if the process relied solely on their own power generation.

#### ***Additional Aspects and Future Outlook***

The sludge currently considered a waste byproduct has an energy value. If a market niche is found, the sludge could be sold as an energy product. Selling the sludge as an energy product would also mean that disposal costs for the sludge would be reduced.

In terms of the final oil products, Envion says that they have flexibility to meet customers' needs. The Envion Oil Generator (EOG) can create kerosene, jet fuel, diesel, or gasoline fuels. The facility may also be used to create more plastic products or other petroleum-derived manufactured goods.

### **3.1.3 Climax Global Energy: South Carolina**

Climax Global Energy is a company that exclusively uses plastics as their feedstock in order to produce high-quality synthetic oil and wax. Climax currently operates a demonstration facility and claims to be able to accept any type of plastic. Their source material comes from municipalities and private companies within a 50-mile radius. No pre-cleaning or pre-sorting processes are necessary (although shredding is required); feedstocks are fed directly into a pyrolysis chamber. In order to power this process, microwave energy or diesel generators may be used. Vitrified solid residuals are one byproduct of this process. Approximately 5–10 percent of the original mass of the feedstock is nontoxic ash that must be landfilled. Climax Global Energy provided LCI data through submittal of the data questionnaire and follow-up communications.

#### ***Process Details***

Climax Global technology uses mixed, post-consumer plastics as feedstock for their pyrolysis process. The plastics must be chipped and shredded prior to being processed. Approximately 20 tons of raw feedstock per day is processed. Moisture content of the feedstock ranges from 0 to 5 percent. One ton of waste plastic yields 5 barrels of synthetic oil.

The feedstock is converted using average bulk reactor temperatures of 400 °C. Inputs to the process include a minimal amount of inert nitrogen and 1–3 gallons of water per minute. Three to 4 tons of light gas (e.g., methane, propane) are produced as byproducts. One to 3 tons of solid carbonaceous residue and any inert materials from feedstock stream, such as rocks, dirt, and glass, are removed as a part of the process.

#### ***Performance Information***

Climax Global Energy claims an energy recovery efficiency of approximately 75 percent. The commodity wax has approximately 6 million BTUs per barrel. The internal parasitic power

requirement is expected to be about 18,000 KW per day. No external fuel use is required in order for the facility to begin operations.

### ***Process Emissions***

**Appendix A** presents the emissions data obtained for Climax. The facility emits PM, CO<sub>2</sub> and hydrocarbons, SO<sub>2</sub>, N<sub>2</sub>O, VOCs, NO<sub>x</sub>, and CO. Byproducts of the process include inorganic residue and ash. Additionally, less than 1 gallon of water effluent per hour is produced during the process.

### ***Cost Information***

The cost per design capacity is estimated to be \$250,000/TPD, including materials, handling, and other plant costs.

### ***Additional Aspects and Future Outlook***

Similar to the other pyrolysis operations profiled, Climax claims it is able to create many different products out of its plastic feedstock. For example, commodity wax is one product that has a variety of uses such as cosmetics, adhesives, and coatings. The company can also produce oils that can be refined into ultra-low sulfur diesel and high-grade synthetic lubricants such as automobile motor fuels.

#### **3.1.4 JBI: Niagara Falls, New York**

JBI uses a proprietary pyrolysis process, Plastic2Oil (P2O), to convert mixed, nonrecyclable plastic waste to gasoline, diesel, and light-fraction gases. JBI receives feedstock from a variety of sources, including commercial and industrial partners, and is currently seeking a permit to use MSW-based feedstock. JBI has been operating at a commercial status in Niagara Falls, New York, since 2010 and anticipates one jointly-operated site in Canada and several in Florida. JBI provided LCI data through submittal of the data questionnaire and follow-up communications.

The P2O processor is highly automated and runs continuously, as long as feedstock is loaded into the hopper. Approximately 1,800 pounds of feedstock can be converted per hour. The process currently converts up to 20 tons of plastics per day. However, 30-ton-per-day units are in development. The footprint for the processing equipment is less than 1,000 square feet.

### ***Process Details***

Feedstock is first shredded or pre-melted and conveyed to the reactor via a hopper and conveyor system. The reactor cracks the plastics into short, gaseous hydrocarbons. The heavy fraction gases are condensed and stored in fuel tanks and the light fraction gases are compressed and set aside to be used to internally power the P2O process or sold separately.

Inputs include natural gas for start-up and proprietary catalysts, and water and electricity during P2O processing. P2O is permitted to generate electricity onsite using process gases as fuel. Since the process can convert approximately 8 percent of the plastic feedstock into these light-fraction process gases, the grid electricity requirement averages around 67 KWh.

For every ton of plastic processed, approximately 5 pounds of nonhazardous solids, 136 pounds of char, and spent catalysts are produced in addition to the gasoline, diesel, syngas, and residual light-fraction gases. Residues are removed automatically.

### ***Performance Information***

The Plastic2Oil process has a recovery efficiency rate of approximately 92 percent. Each ton of plastic produces 1,734 pounds of gasoline and diesel. Additional byproducts include residuals, which have been found to have a heating value of 10,600 BTU, and syngas. These products and byproducts may then be blended with other fuels and additives, depending on the market and/or needs of the purchaser.

The company reports a high level of vertical integration due to co-locating with plastic waste sources, scaling the equipment to meet the feedstock supply, and using a highly automated process. Additionally, the process partially relies on the off-gases generated internally, reducing the operating costs and offsetting electricity grid mix emissions.

### ***Process Emissions***

**Appendix A** presents the emissions data obtained for JBI. Primary air emissions from the P2O process include particulate matter, carbon dioxide, nitrogen oxides, hydrocarbons, and VOCs. However, JBI is not required to monitor emissions or install emissions control technologies. In terms of GHG emissions, converting 1 ton of plastic using JBI's P2O process yields approximately 0.29 pounds of carbon equivalents. The process also reports 2.41 pounds of NOx emitted for every ton of waste plastics. JBI reports that the atmospheric emissions are less than those of a natural gas furnace.

Water is used for gas cooling and wastewater from this step is reused, but no water effluent is generated.

### ***Cost Information***

The estimate for cost per design capacity is \$587,000 for the entire machine. Operational costs to cold start and power the processing equipment average about \$7 per hour. Plastics are generally provided to JBI at no cost.

### ***Additional Aspects and Future Outlook***

In addition to receiving permits to begin commercial operations in New York, JBI recently announced a joint venture with OxyVinyl Canada to produce oil onsite using the waste plastics generated by OxyVinyl. JBI is currently focusing on creating additional partnerships with organizations that have existing permits and high-volume waste plastic streams to maximize consistent feedstock volume while minimizing the permitting processes.

## **3.2 Gasification Technology Vendors**

### **3.2.1 Enerkem: Westbrook, PQ, Canada**

Enerkem uses a gasification process to convert waste materials to syngas as an intermediate product. Sources of feedstock include MSW, refuse-derived fuel (RDF) from sorted MSW,

woody wastes from construction and demolition, used telephone poles, and other wastes from industrial, commercial, and institutional (ICI) processes. Electricity, ethanol, and other green chemicals are options for final products.

The company currently has two operational facilities including a pilot-scale demonstration plant in Sherbrooke, PQ, Canada; and an operating commercial-scale demonstration plant in Westbrook, PQ, Canada. Enerkem also has begun construction on two additional facilities: one in Pontotoc, MS, and one in Edmonton, AB, Canada, that are anticipated to begin full operations in 2011 and 2012, respectively. Information regarding the status of the operations can be found on the City of Edmonton's and the company's websites. All information about the anticipated Pontotoc, MS, plant was obtained from the Environmental Assessment (U.S. DOE, 2010). Information about the Canadian facilities was obtained from a combination of personal communications and literature search.

The commercial-scale demonstration facility has been in operation since 2009 and in its demonstration stage, has managed approximately 39 tons per day of feedstock on a dry basis. Commercial-scale demonstration signifies that the facility is in the next-to-final stage of the technology development cycle and is a commercial-scale facility running smaller "batches" of waste to refine the process. The planned commercial facilities will have a capacity of approximately 330 dry tons per day.

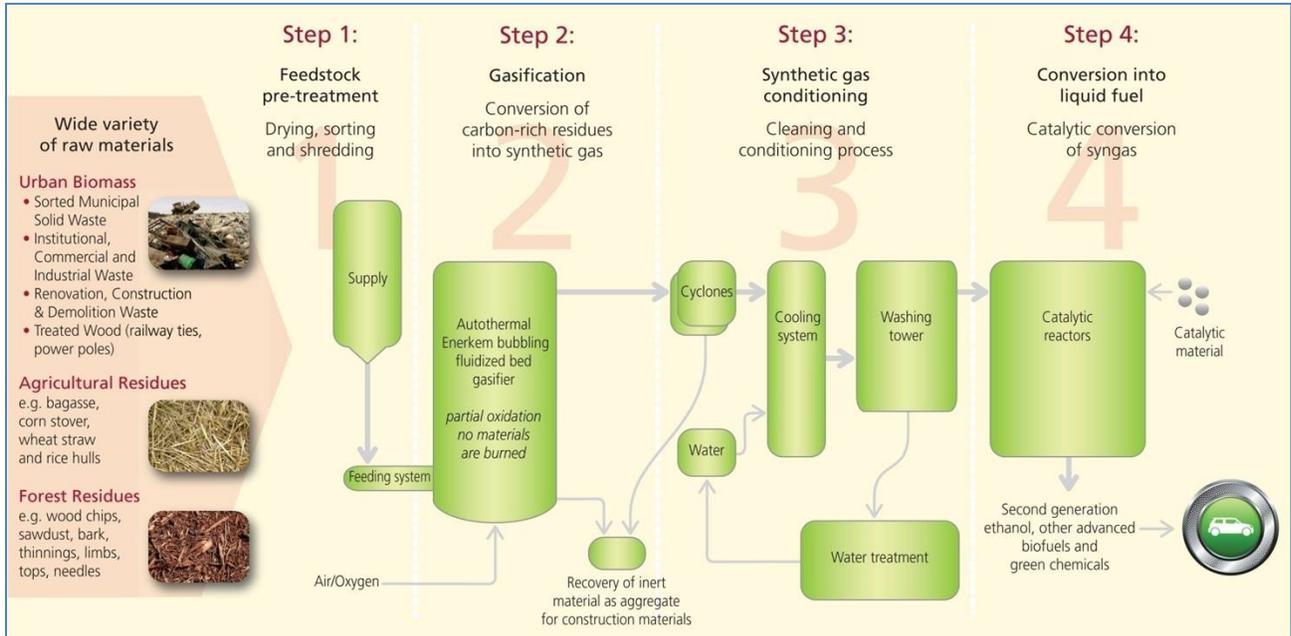
### ***Process Details***

The Enerkem gasification process is illustrated in **Figure 3-2**. The first steps in the process are to dry, sort, and shred the waste. Three types of feedstock are used: (a) refuse-derived fuel (RDF) that has been sorted from MSW, (b) construction and demolition (C&D) waste, and (c) institutional, commercial, and small industry (ICI) waste. The pre-sorting of RDF waste includes sorting and biological treatment followed by processing to a "fluff." The facility can also accept more traditional pelletized RDF. C&D wood is shredded and ICI is sorted and also shredded. All pre-processing occurs at the facility. The inorganic matter content of each type of feedstock is generally 15 percent of total weight for RDF and ICI, while C&D wood is less than 5 percent.

The shredded "fluff" from MSW, C&D, and ICI waste is fed into a bubbling fluidized gasifier. The waste is converted into syngas. Inert residues are removed and can be used as aggregate for construction (if approved). Next, the syngas goes through a series of steps that clean and condition the syngas. These systems include cyclones, a cooling system, water treatment, and a washing tower. Wastewater is a main byproduct of this portion of the process, but is reused. The heating value of syngas is between 6 and 12 megajoules per standard cubic meter, depending on the gasification process. Electricity can be produced with the use of syngas in an internal combustion engine (ICE) generator-set. Enerkem is currently installing an ICE gen-set that is scheduled to be operational in 2011. Alternatively, the syngas can enter catalytic reactors where it is converted into liquid fuel including second generation ethanol, advanced biofuels, and/or green chemicals. Conversion to ethanol requires oxygen and steam inputs for this step of the process. The exact process configuration and end product(s) will be tailored to the markets and contractual arrangements.

### Performance Information

Performance information includes the efficiency of the process in terms of converting BTUs of waste input into BTUs of syngas and/or ethanol output, as well as the reliability of the technology in commercial operating conditions. Enerkem characterizes the conversion



**Figure 3-2. Enerkem Gasification Process Flow Diagram.**  
(Source: <http://enerkem.com/en/our-solution/technology/process.html>)

efficiency of their gasification process as the ratio of the lower heating value (LHV) of the syngas to the LHV of the input feed and states that it is higher than 72 percent. In addition, high- or low-grade heat recovery is an option that Enerkem states can provide 5–10 percent of additional conversion efficiency. The internal parasitic power requirement to operate the gasification process is approximately 600 KWh per dry ton when electricity is the end product and 490 KWh per dry ton when ethanol is the end product. In addition, natural gas is required (15.72 lbs per ton of MSW) for facility start-up, but is not used as a co-fuel for normal process operation.

Since Enerkem is operating as a demonstration facility, information about the reliability of the process at commercial operating conditions is not available at this time.

### Process Emissions

**Appendix A** presents the emissions data obtained for Enerkem. In general, the data quality for emissions estimates is low since the facility is still in the demonstration stage. As the facility transitions to a fully operational commercial stage, one would expect the process inputs/outputs to stabilize and emissions to be more consistent for measurement.

Primary air emissions from the Enerkem process include CO<sub>2</sub> and NO<sub>x</sub> as well as traces of methane, HCl, hydrocarbons, SO<sub>2</sub>, and CO. Mercury, cadmium, lead, ammonia, dioxin, and

furan emissions are all below Canadian and U.S. (and EU) regulatory limits. Ammonia is also an emission that must be controlled. It has to be scrubbed out and thus removed from the circulating scrubbing water. The recovered ammonia ( $\text{NH}_3$ ) can be sold or reintroduced in the gasifier, where it is converted into nitrogen ( $\text{N}_2$ ) and hydrogen ( $\text{H}_2$ ). A steady state level of  $\text{NH}_3$  is thus achieved and the syngas maintains a concentration below the regulations.

In terms of GHG emissions, Enerkem estimates that approximately 40 percent of the carbon in the feed is turned into  $\text{CO}_2$ , but approximately 75 percent of the produced  $\text{CO}_2$  is recovered and reused. The ratio of biogenic to fossil carbon in  $\text{CO}_2$  depends on the ratio of biogenic to fossil material in the RDF feed stream. Personal communications with Enerkem indicate that the biogenic to fossil carbon fraction is typically 3 to 4 to 1 for the RDF since it contains about 20 percent plastics and 60 to 70 percent biomass.

Water is used for gas cooling and wastewater from this step is reused. The process itself is a net water producer. Enerkem estimates that it purges 1 ton of process water per ton of feed (dry basis). The facility cleans this water and returns about 80 percent of the purged water to the process. The remaining excess water generated is evaporated in a cooling tower or discharged as wastewater. Enerkem data provide a range of 544 to 1,270 pounds of water generated per ton of waste processed, depending on the moisture content removed in the drying/dehydrating step.

Residual wastes produced by the process include primarily gasifier char and spent or residual catalysts from the catalytic synthesis stage. No estimate for char production was provided, but the char would require disposal. If the process is tailored to produce alcohol fuels as the main product, then residual catalysts would be produced and also require disposal.

### ***Cost Information***

Estimates for capital and operating costs were collected through publicly available sources as well as through a questionnaire and follow-up telephone calls with Enerkem staff. Similar to emissions, reliable cost estimates are difficult to present since the facility is still in the demonstration stage. As the facility transitions to a fully operational commercial facility, one would expect the process inputs/outputs to stabilize and costs to be more consistent and reliable.

Estimates for cost per design capacity for the Enerkem Pontotoc, MS, facility is \$424,242 per dry ton. For their 330 dry ton per day facility, the total capital cost would be approximately \$140 million. Additionally, an external source presentation indicates Enerkem receives feedstock at no cost or at a gain of approximately \$45 per ton of waste for the Quebec facility.

### ***Additional Aspects and Future Outlook***

Electricity, ethanol, and other green chemicals are options for final products for the planned facilities. The exact process configuration and operation specifics will be tailored to the markets and contractual arrangements.

### 3.2.2 Plasco: Ottawa, Ontario, Canada

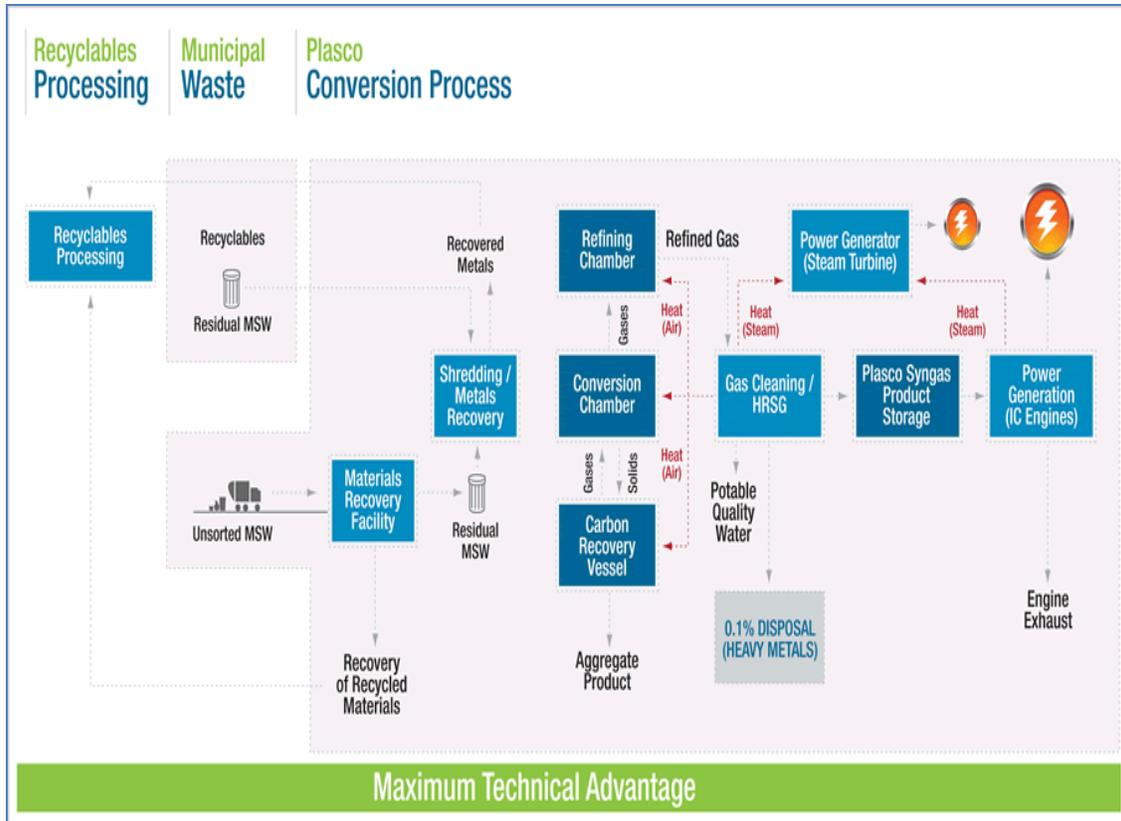
Plasco Energy Group operates a commercial-scale demonstration facility working closely with the city of Ottawa. The partnership began in April 2006 and the facility was constructed at the site of the operating Trail Road Landfill. Currently, the facility is permitted to process 93 ton per day of solid waste and is able to generate 4 MW of electricity. Plasco Energy Group provided RTI with an independent comparative analysis of Plasco and other waste-to-energy (WTE) facilities as well as with a process brochure (Pembina, 2009; Plasco, 2011). Additionally, general process information and semi-annual emissions reports were obtained from the company's website (Plasco, 2010).

#### *Process Details*

Plasco Energy Group's Ottawa Trail Road Facility is a WTE facility that utilizes non-recycled MSW. MSW is first shredded and then goes into the conversion chamber, which converts waste into crude syngas with the use of recycled heat. Solid residue is removed to another chamber called the Carbon Recovery Vessel (CRV), where solids are melted with a plasma torch. Plasma heat stabilizes solids and transfers any volatile compounds and fixed carbon into crude syngas, which then flows back to the conversion chamber.

The crude syngas moves to the refinement chamber and plasma torches are utilized to clean and refine the gas. This refined gas is termed PlascoSyngas. At this point, PlascoSyngas moves to the Gas Quality Control Suite, which removes heavy metals and PM found in the MSW. It also neutralizes acid gases. PlascoSyngas is now able to be used in ICE gen-sets to generate electricity. A percentage of the process water must be disposed of through a licensed carrier. However, Plasco will be a net producer of water because the excess moisture in the waste is removed at high temperatures. The water is then filtered and cleaned to sewer water standards.

The process flow diagram for Plasco is **Figure 3-3**.



**Figure 3-3. Plasco Gasification Process Flow Diagram.**  
 (Source: [www.plascoenergygroup.com](http://www.plascoenergygroup.com))

***Performance Information***

No data were obtained for the energy conversion efficiency of the process. However, Plasco reports that 98 percent of the waste processed is converted to marketable products. Additionally, a 2009 study comparing Plasco to standard waste-to-energy processes indicates that each ton of waste produces between 2,000 and 3,000 cubic meters of syngas with an energy content of 3 to 5 megajoules per cubic meter, depending on the feedstock content. Higher energy content in feedstock yields higher energy content and higher volumes of syngas. If accurate, syngas would yield approximately 3,165 to 7,912 BTUs per pound of waste. This estimate is significantly less than the other gasification processes studied here. However, the study estimates that when used to generate electricity, the Plasco process produces more energy than incineration for energy recovery and landfill gas to energy.

***Process Emissions***

**Appendix A** presents the emissions data obtained for Plasco. Slag is one residual from the Plasco process. Slag is transformed to pellets, which are inert vitrified, or glass, residues. Converter ash is a byproduct that is also produced when the carbon recovery vessel (CRV) is not running. The ash is then landfilled. Baghouse ash is sent offsite as hazardous waste.

### ***Cost Information***

No cost information for the Plasco technology was provided; however, the website indicates that approximately \$270 million in capital has been raised and invested in Plasco since 2005. Additionally, an external source presentation<sup>5</sup> indicates that capital costs are approximately US\$86/ton of waste.

### ***Additional Aspects and Future Outlook***

Since the Plasco facility is still in a demonstration phase, details of the facility's operations may not necessarily be representative of the actual levels of efficiency and waste outputs that will occur under a commercial facility. Although the demonstration facility may not perform as well as the planned commercial-scale one, a technical review conducted in 2009 displayed results in favor of Plasco's operations. An independent research organization conducted an analysis of the commercial version of the current demonstration facility in comparison with incineration, AD, and landfill gas with gas capture facilities located around the world (Pembina, 2009). The life cycle analysis results showed that air emissions were lower or about the same for Plasco when compared to other systems, with the exception of heavy metals and PM. Plasco had a heightened ability to generate a greater energy value per waste unit. The company was also capable of generating more marketable products from a given waste stream, and was also able to remove more sulfur, heavy metals, and PM before combustion than the other companies. The results of the study lead to a favorable conclusion of Plasco's planned commercial-scale facility in terms of environmental effects and efficiency levels.

### **3.2.4 Ze-gen: Attleboro, MA**

Ze-gen was founded in 2004. The company was expected to complete construction and begin operations in 2012 of the Attleboro Clean Energy Project, located within the Attleboro Corporate Campus in Massachusetts. However, declines in natural gas prices and difficulties obtaining permits led to the cancellation of the project. The city council banned any gasification plant on account of potential "toxic dust." The facility was going to be co-located with an industrial wastewater treatment facility. The design capacity was expected to be between 75 and 150 tons per day. The energy products were expected to be steam and synthesis gas (syngas) with one-quarter the energy density of natural gas and expected to replace natural gas. The company also has a demonstration facility, mainly for research and development, located in New Bedford, MA, that opened in 2007.

### ***Process Details***

Ze-gen will construct a liquid metal gasification facility that utilizes post-recycled, processed waste material. The facility will accept the following feedstocks: creosote treated railroad ties, nonrecyclable plastics, and clean wood waste. Pre-processing of the feedstock will be necessary and will occur through a contracted processor off-site. After pre-processing is complete, the moisture content of the feedstock will be less than 20 percent and the inorganic matter content will be less than 5 percent. Other inputs are required in order to achieve air emissions control, such as sodium hydroxide, calcium hydroxide, aqueous ammonia, and activated carbon.

---

<sup>5</sup> <http://www.seas.columbia.edu/earth/wtert/meet2010/Proceedings/presentations/CASTALDI.pdf>.

Synthesis gas (syngas) will be created through a thermo-chemical process with the use of liquid copper. The temperature of the gasifier will be about 1,204 °C. The process of gasification will divide organic and inorganic components. The organic components will be reformulated to produce syngas, while inorganic components will be removed. The syngas will be used in a boiler that will produce steam and power a generator to yield electricity.

### ***Performance Information***

The Attleboro Clean Energy Project is expected to have an energy recovery efficiency of approximately 48 percent. The internal parasitic power requirement is expected to be less than one MW. The regional electricity grid mix displaced by delivered electricity is 9 percent coal, 38 percent natural gas, 25 percent oil, and 14 percent hydroelectric power and renewable. In order for the facility to begin operations, supplemental fuel use will be necessary at a rate of approximately 1,500 MMBtu of natural gas per startup.

### ***Process Emissions***

**Appendix A** presents the emissions data obtained for Ze-gen. Some process emissions that will be regulated by the Massachusetts Department of Environmental Protection include PM, CO<sub>2</sub>, CH<sub>4</sub>, HCl, NO<sub>x</sub>, VOCs, CO, NH<sub>3</sub>, mercury (Hg), cadmium (Cd), and lead (Pb). The Commonwealth of Massachusetts does not treat carbon emissions as neutral, unlike most other states. In their report, Ze-gen computes carbon contributions in three ways: avoided emissions, total carbon + biogenic, and carbon without including biogenic emissions. Ze-gen provided a range of emissions, and for this report the upper bounds of emissions levels were used. Wastewater will be another byproduct of the gasification process, and will occur at a rate of about 45 gallons per minute. Residuals will also be present from those inorganic components that have been removed from liquid metal. The components will be made into vitreous glass-like slag. About 1.5 tons of slag is expected to be generated per day.

### ***Cost Information***

No cost information for the Ze-gen technology was provided or found through literature and Web searches.

### ***Additional Aspects and Future Outlook***

Currently, Ze-gen is testing the viability of using various feedstocks, including its ability to use marine debris plastic floating along the surface of the ocean. If successful, the company could remove some of the waste that is detrimental to the overall ecosystem health of the ocean while converting waste to usable fuel.

### **3.2.5 Geoplasma: St. Lucie, Florida**

Jacoby Development, Inc. formed Geoplasma, LLC in 2003 in order to work on research and development for conversion technologies. Geoplasma is a planned facility that has received its final air permit from the Florida Department of Environmental Protection. The facility is set to produce 22 MW of power with the use of 600 tons of waste on a daily basis. Geoplasma, St. Lucie will be constructed at the St. Lucie County Solid Waste Facility. All information about the anticipated St. Lucie plant was obtained from the Environmental Assessment (U.S. DOE, 2010).

### ***Process Details***

The facility will use Class I waste, which in Florida includes solid waste that is not hazardous and waste not banned from disposal in a lined landfill. It will also process construction and demolition (C&D) waste, tires, and yard waste. Geoplasma will reduce monetary and time costs associated with transport of waste to the facility because they will be co-located with the waste facility. The feedstocks will be received in the existing receiving and baling recycling building. Supplementary storage will be constructed similar to the existing one. A conveyer system will transport waste fuel to the initial processing location to reduce the size of the material. The moisture content of the feedstock value is assumed to be 30 percent. In order to minimize fugitive emissions and odors, air for the gasifier will be pulled from the waste processing area and conveyer system.

The waste will also be mixed with coke and limestone. Coke will be necessary to mix with MSW and tire fuel to have a porous bed at the bottom of the gasifier. Limestone will be used in the flue gas desulfurization (FGD). The mixed feedstock will be fed into the plasma heat gasifier. The organic constituents will undergo a conversion process into a syngas, which will then be combusted in a multi-stage thermal oxidizer, and then a heat recovery steam generator (HRSG) to produce high-pressure and high-temperature steam. The steam will power a steam turbine electrical generator that will supply electricity to the grid. Exhaust gas from the HRSG will be filtered through an emissions control system before it is discharged to reduce harmful pollutants.

### ***Performance Information***

No information on the energy performance of Geoplasma's anticipated facility was supplied by the vendor; however, the Florida Department of Environmental Protection cites that the facility is anticipated to produce approximately 22 MW of power from approximately 600 tons per day of waste.

### ***Process Emissions***

Geoplasma is considered a major source of hazardous air pollutant (HAP) emissions and is in accordance with Title V a major source category. No water emissions data were available. Since the facility is not yet functioning, the potential to emit value was used instead of actual emissions levels. The facility was also assumed to be operating 312 days a year on a 24-hour basis. Emissions that have limits include NO<sub>x</sub>, CO, SO<sub>2</sub>, VOC, HCl, PM, Lead, Hg, Cd, dioxins and furans (D/F), visible emissions (VE), and NH<sub>3</sub>. Limestone is used in air pollution control equipment to minimize SO<sub>2</sub> emissions. Another input is powered activated carbon (PAC) delivery, which will be used to manage Hg, trace metals, and complex organic compounds.

Byproducts of the plasma gasification process include vitrified inorganic residue. The bottom of the gasifier will also discharge some residue metals into water. Sand-like aggregate and metal nodules will be produced from this mixture at a rate of 13,200 lb/hr. The two byproducts are planned to be separated, stored, and loaded into trucks to be sold offsite. Spent PAC will be accumulated in the system baghouse and moved to a storage silo at a rate of 900 lb/hour. In order to reduce PM emissions, the PAC will be transferred through an enclosed conveyer to the

silos. Gypsum is another process byproduct, and is expected to be produced by the FGD system at a rate of 900 lb/hour.

The Geoplasma data collected was not analyzed during this analysis for several reasons. Most importantly, the Geoplasma process data were the only data we were able to collect for the plasma arc process. Additionally, we were not able to obtain all of the process information needed for the LCI.

#### ***Cost Information***

No cost information was provided by the company and was not available at the time of this report.

#### ***Additional Aspects and Future Outlook***

According to the public's comments on the draft permit, support for the facility is widespread. One potential issue that may need to be addressed in the future is that excess emissions are allowed during startup, shutdown, or malfunction. The Blue Ridge Environmental Defense League specifically cited that this flexibility in emissions levels is unacceptable. If there are issues during Geoplasma's operations that lead to significantly higher emissions, it is possible that this issue may come up again.

### **3.3 Anaerobic Digestion**

#### **3.3.1 County of Yolo Public Works Department: Yolo County, California**

This project is not a conventional AD project in that the digester unit was constructed as a cell on top of an existing landfill cell as it will be described in detail under the **Process Details** section. The California Department of Resources Recycling and Recovery (CalRecycle) commissioned the design and construction of this pilot-scale anaerobic digester. The system is co-located at Yolo County Central Landfill and has been in operation since 2007. This system takes advantage of the facility's landfill gas-to-energy infrastructure to increase the energy recovery efficiency of the digester and also allows for the recovery of the residual material to be used as compost. Due to the innovations in the project, this AD approach has the potential to be more cost-effective at a larger scale than many other AD systems (CalRecycle, 2010) when compared with landfill disposal and other waste management techniques.

#### ***Process Details***

The anaerobic digester was constructed on top of an existing landfill cell. The digester cell was lined and then layered with 1,894 tons of green waste, 34 tons of wood chips, 130 tons of aged horse manure, and 25 pounds of limestone and capped with a liner cover. Pipes that are distributed between the waste layers transfer the gas to the energy facility located onsite. Leachate and water are re-circulated to promote anaerobic degradation.

#### ***Performance***

The process has demonstrated the ability to produce 1,680 cubic feet of methane per dry ton of waste. Due to these results, it was suggested that further pilot projects be initialized in order to

understand the technological barriers such as high moisture waste and odor issues associated with food waste. It is also recommended that a better quantification of emissions associated with composting be assessed. Overall, the study found that California would greatly benefit from a wider implementation of waste diversion to produce methane gas and electricity.

### **3.3.2 Quasar: Wooster, Ohio**

Quasar joined Ohio State University's Agricultural Research and Development Center (OARDC) to build an AD system called the ecoFARMsystem550 (F550) system in the BioHio Research Park located in Wooster, Ohio. The F550 system uses regional food and crop waste as well as grass and manure from the university's farm operations in order to produce renewable energy and other byproducts.

#### ***Process Details***

A receiving hopper is filled with biomass within the plant building in order to control odors. Live bottom hopper augers move the waste toward the middle of the hopper. Biomass is discharged and passes through a grinder to the process lines. Fresh biomass mixes with heated recycled biomass and moves into the biomass equalization tank. The liquid waste passes through a strainer in order to remove any leftover solid materials. Liquid biomass is then added to the equalization tank, which may store biomass for up to six days. A pipe connects the head space in the equalization tank with the digester tank and dual purpose tank in order to sustain equalized pressure. The space also provides an area for displaced gas from filling or discharging or temperature expansion.

After digestion is complete, the biomass is pasteurized in order to remove pathogens. Energy recovery is completed through a biomass to biomass heat exchanger as it is pumped to a holding tank. Energy recovered is transferred and temperature elevated through a heat exchanger. Suspended solids in the digested material are mechanically separated in a dewatering process. The resulting cake biomass is about 25 percent dry solids.

#### ***Performance***

The system capacity is 550,000 gallons, and biomass may be stored for about 3 days, while the average digestion time is about 28 days. The system can handle 19,382 wet tons of waste annually, and produces 5,256 MWh of electricity. The digester is currently operational and is able to offset half of OARDC's electricity demand.

## Section 4: Life Cycle Environmental and Cost Assessment for Technology Categories

For the environmental and cost assessment we wanted to use a systematic and standard methodology and to evaluate the life cycle environmental aspects and potential impacts of the diversion alternatives so that we could compare the environmental performance of the systems. We adopted the methodologies used to develop life cycle inventories (LCIs) as part of a life cycle assessment (LCA). LCA is a technique for assessing the environmental aspects and potential impacts of a system from raw materials acquisition through production, use, and disposal. According to the internationally accepted ISO 14040 standard, conducting an LCA includes compiling an inventory (called an LCI – life cycle inventory) of relevant inputs and outputs of a system, evaluating the potential environmental and health impacts of those inputs and outputs (called an LCIA – life cycle impact assessment), and interpreting the results in relation to the objectives of the study. In this study, we developed high-level<sup>6</sup> LCIs aiming to identify and evaluate the general environmental performance and cost of the conversion technologies and to compare them to a reference waste management option (landfill).

### First Law of Thermodynamics

In conducting a life cycle inventory for conversion technologies, it is useful to review the laws of thermodynamics. These laws succinctly define the possibilities and the necessary limitations on conversion processes. The First Law of Thermodynamics states that energy can neither be created nor destroyed. Because energy must always be conserved, it can only be converted from one form to another and “energy out” must balance the “energy in.”

Therefore, the resource produced during conversion must equal the net addition of the energy extracted from the waste supply, any feedstock energy used, and the losses in the form of heat or waste product. This corresponds with the Second Law of Thermodynamics, which states that although energy cannot be created or destroyed, some of the usable energy will be converted into unusable energy during conversion (e.g., uncaptured heat, unusable byproducts). Also, energy must always flow from high temperatures to low temperatures. Taken together, the laws indicate that 100 percent conversion efficiency is not possible.

Using a life cycle perspective encourages planners and decision-makers to consider the environmental aspects of the entire waste management system. These include activities that occur outside of the traditional framework of activities, from the point-of-waste collection to final disposal. For example, anyone evaluating options for recycling should consider the net environmental benefits (or additional burdens), including any potential displacement of raw materials or energy. Similarly, when energy is recovered through waste combustion, conversion technologies, or landfill gas-to-energy, the production of fuels and the generation of electricity from the utility sector is displaced.

In this respect, LCA (and LCI) can be a valuable tool to ensure that a given technology creates actual environmental improvements rather than just transfers environmental burdens from one

---

<sup>6</sup> The data used for this assessment were provided by industry vendors and were not independently validated. In addition, the datasets used to characterize the technologies vary in the level of detail and the number of values obtained for particular input parameters, with only one value obtained for certain parameters.

life cycle stage to another or from one environmental media to another. This analysis is also useful for screening systems to identify the key drivers behind their environmental performance.

## 4.1 General Approach for the Environmental and Cost Assessment

An LCI methodology was used to guide the environmental and cost assessment. The focus of this study was only on those materials currently destined for the landfill and not those currently being recycled. Our general approach was to develop inventories of energy, emissions, and cost for the pyrolysis, gasification, and anaerobic digestion (AD) systems and to utilize the Municipal Solid Waste Decision Support Tool (MSW DST), a tool developed under a cooperative agreement between RTI and the U.S. Environmental Protection Agency (U.S. EPA), to capture the other life cycle components (e.g., materials pre-processing [separation], landfill disposal, energy production, transportation, and materials production activities). The data and models in the MSW DST have been developed for the U.S. EPA and have been extensively peer reviewed for quality assurance.

The environmental and cost inventories for gasification, pyrolysis, and AD do not represent any one specific technology or vendor. Rather, data collected for selected technology vendors as profiled in **Section 3** were supplemented with data collected from the literature and lower–upper bound ranges were developed for the three technology classes (i.e., gasification, pyrolysis, and AD).

### 4.1.1 Goals

The overall goal of the analysis is to estimate the impacts that mixed waste and plastic waste conversion technologies have on the environment and public health. In general, the analysis will seek to answer questions in two categories:

- What are the life cycle environmental burdens/benefits of conversion technologies?
- How do the life cycle environmental burdens/benefits of conversion technologies compare to the baseline practice of landfill disposal for non-recycled material?

The goal of the LCA is not necessarily to make definitive conclusions about conversion technologies or the environmental preference of conversion technologies compared to the existing landfill base case. Rather, the goal is to better understand the potential environmental burdens and benefits that may result from the commercialization of conversion technologies, the tradeoffs of employing conversion technologies as alternatives to existing MSW management practices, and the variables that influence the potential environmental impacts of conversion technologies.

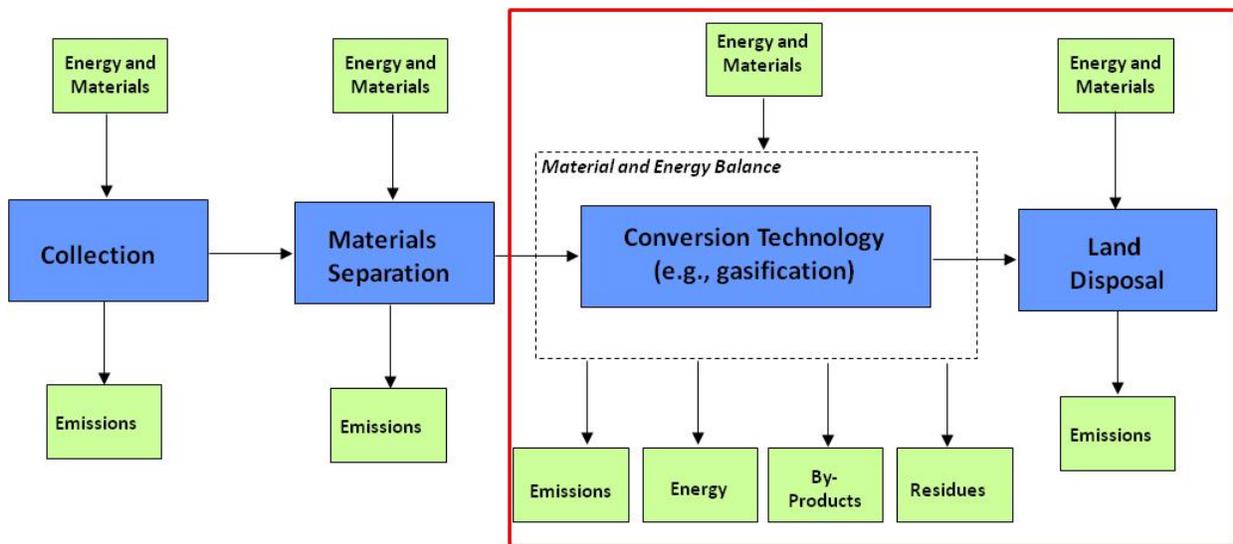
### 4.1.2 Scope and Boundaries

Since gasification, pyrolysis, and AD have different functional units, we did not directly compare the three systems. The function of the gasification technology system is to transform the mixed waste fraction of non-recycled waste (i.e., residual waste after recycling and composting) into energy and useful products. The functional unit is then a mass unit (e.g., a ton of MSW) of mixed waste. The pyrolysis technology system manages plastic waste. Therefore, the functional

unit is a mass unit of plastics waste (e.g., a ton of plastics). AD accepts organics, mostly food waste, so the functional unit is then a mass unit of organic waste (e.g., a ton of organics).

**Figure 4-1** illustrates the system boundaries defined for a conversion technology (CT) in this assessment. In the figure, the boundaries include not only the conversion technology and other MSW management operations, but also the processes that supply inputs to those operations, such as fuels, electricity, and materials production. Likewise, any useful energy or materials produced from the conversion technology system are included in the study boundaries as offsets. An offset is the displacement of energy or materials produced from primary (virgin) resources that result from using secondary (recycled) energy or materials.

RTI used a gate-to-grave approach for this assessment and assumed that all waste is fed to the conversion technologies after collection and separation for recycling or composting. Some technologies will require additional screening of the feedstock prior to the conversion process and this is included as part of the conversion technology subsystem. Therefore, the boundaries of this assessment were defined by the red box in **Figure 4-1**.



**Figure 4-1. General Life Cycle Boundaries for a Conversion Technology System.**

Once the specific conversion technology designs were identified based on the technical evaluation of technology vendors, detailed process descriptions and process flow diagrams were prepared to identify mass flows, energy consumption, environmental releases, and other significant waste production and resource utilization parameters. An important aspect of this step was identifying the key aspects (for example, facility construction and operation parameters) of each process that needed to be considered and ensuring that all conversion technology systems were defined in a consistent manner. For example, if one conversion technology system included the production of materials used for pollution control, then all conversion technology systems should include this aspect.

In comparing conversion technologies to existing landfill disposal practices, we needed to have consistent data for each burden (for example, dioxin/furan emissions) across all unit processes in the waste management system. Therefore, if data for any given burden was not consistently available across all processes included in the system, then the burden was not included in the comparative results of conversion technologies to existing management practices. However, we did consider all burdens in this report when describing specific conversion technologies (**Section 3**). In general, the main categories of inputs and outputs that are reported for each conversion technology system are consistent with those reported in the MSW DST. These include annual estimates for energy consumption, air emissions, water pollutants, and solid waste. In deciding upon which LCI burdens to present in this report, we chose energy and greenhouse gas (GHG) emissions since the input data used to estimate these results were consistently available for the various processes included in the LCIs.

## 4.2 Life Cycle Inventory Methodology, Assumptions, and Modules for Waste Conversion Technologies

As part of the life cycle inventory (LCI) approach adopted for the environmental and cost assessment, we collected data to quantify the relevant inputs and outputs of the technology systems. We collected, reviewed, and compiled data based on the conversion technology system boundaries (**Figure 4-1**). We worked with the internal and external contacts to identify available data for each of the conversion technologies. Data were collected from the following sources:

- Technology vendors.
- Publicly available literature.
- Federal reports.
- State and municipal governments.
- Industry reports.
- Trade associations.
- Waste collection, processing, and disposal facility records and reports.

The scope and boundaries for each major conversion technology category are based on the technology class definitions and vendor-specific process flow diagrams presented in **Sections 2 and 3** of this report as well as other information collected from the literature. Each process flow diagram shows the major process steps that occur in processing and converting waste input. In addition, the diagrams show the main material and energy inputs and outputs for each conversion technology.

As shown by the process flow diagrams, the processes for which data are presented are not cradle-to-grave, but rather gate-to-gate. This is because the conversion technologies by themselves are just one process step within the system. Only after all of the pieces of life cycle inventory data from each process step within the system boundaries are assembled can the inventory module for each conversion technology be completed. These inventory modules rely on the material and energy data provided by the vendors and/or obtained from the literature as a starting point and then add the inventory information for upstream and downstream steps.

In general, the construction of the LCI module for each conversion technology is depicted as follows:

***LC input/output burdens – Offsets = Net LCI Coefficients***

For example, gasification may use natural gas as a supplemental fuel. The amount of natural gas consumed for a given tonnage of waste processed is calculated in the material and energy model. This amount is multiplied by the environmental burdens associated with producing the natural gas and added to the inventory for the technology. Similarly, the gasification process generates some residual waste and char that is landfilled. The environmental burden associated with the transportation and landfill disposal of these residuals was added to the inventory for the technology.

Material and energy offsets are netted out of the LCI. In the case of pyrolysis, the main products are waxes and liquid fuels, each having a number of possible end uses. For this study, we assumed that it would be used as a replacement for fuel oil. The quantity of commodity oil that is produced by the process (as given by the material and energy model) is converted to an equivalent function amount of fuel oil. That amount of fuel oil offset is then multiplied by the inventory burdens associated with fuel oil production, and these burdens are netted out of the inventory for the technology.

**4.2.1 Treatment of Material and Energy Recovery**

Only energy recovery was included within the conversion technologies' boundaries. Some gasification vendors report material recovery from pre-processing of the mixed waste arriving at their facilities. However, because we assumed non-recycled waste would be the input feedstock, there would likely be only small amounts of available recyclables. In addition, the current test/demonstration nature of the facilities means they do not yet have contracts in place and/or data for materials recovery and available markets. For these reasons, as well as maintaining consistent treatment across the technologies, materials recycling related benefits were not included in the assessment.

For energy-related offsets, we assumed that electrical energy produced from landfill gas-to-energy and conversion technology systems displaces electrical energy produced from fossil sources. The exact mix of fossil fuels displaced is based on the U.S. average grid mix. Electrical energy is produced mainly from the gasification and AD technologies.

For the pyrolysis/cracking technologies, commodity oils/waxes are the main product. We assumed that the commodity oils/waxes displace petroleum-based crude oil.

**4.2.2 Items Excluded From the Life Cycle Inventory**

A number of items have been excluded from the LCI because they are typically found to be negligible in terms of the inventory totals. These items are described below.

The energy and environmental burdens associated with the manufacture of capital equipment is not included in the life cycle profiles. This includes equipment to manufacture buildings,

motor vehicles, and industrial machinery. The life cycle burdens associated with such capital equipment generally, for a ton of materials, becomes negligible when averaged over the millions of tons of product that the capital equipment manufactures compared to the burdens associated with the processing steps.

The fuels and power consumed to heat, cool, and light manufacturing establishments are omitted from the calculations. For most industries, space conditioning energy is quite low compared to process energy. Energy consumed for space conditioning is usually less than 1 percent of the total energy consumption for the manufacturing process. The energy associated with research and development, sales, and administrative personnel or related activities have not been included in this analysis.

For each system evaluated, small amounts of miscellaneous materials are associated with the processes that are not included in the LCI results. Generally these materials make up less than 1 percent of the mass of raw materials for the system. For example, the use of biocides and other conditioning chemicals for cooling water are not documented and included in the inventory results, except to the extent that these materials contributed to waterborne emissions from the facilities.

The Geoplasma data collected was not analyzed during this analysis for several reasons. Most importantly, the Geoplasma process data was the only data we were able to collect for the plasma arc process. Additionally, we were not able to establish contact with the vendor during the investigation and were not able to obtain all of the process information needed for the LCA.

#### **4.2.3 LCI Parameters Tracked and Reported**

The main categories of LCI inputs and outputs that were tracked and reported as part of this study include annual estimates for the following:

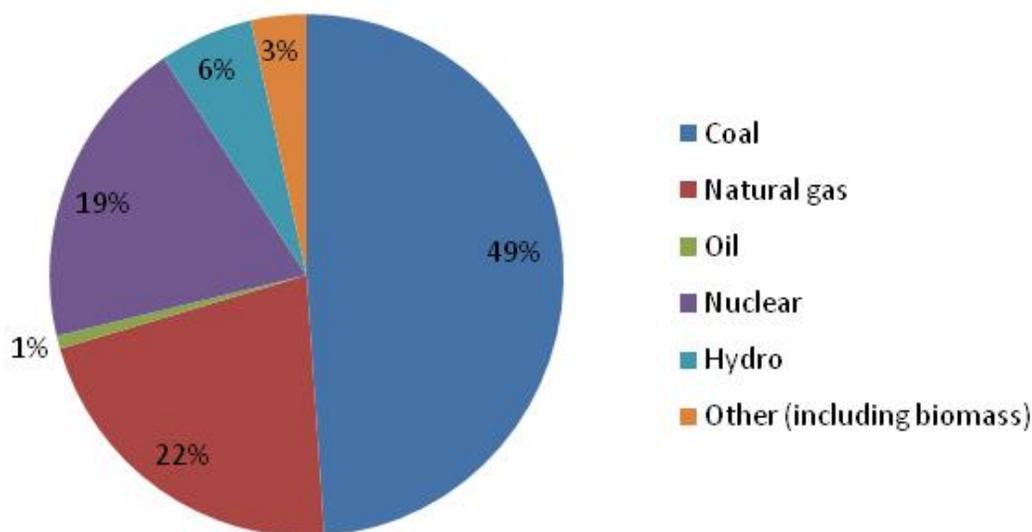
- Energy consumption and production.
- Criteria air emissions
- Greenhouse gas emissions.
- Waterborne pollutants.
- Residual solid wastes.

Descriptions of what comprises each of these main categories are provided in the following sections.

##### ***Energy Consumption***

Annual energy consumed is aggregated across process and transportation steps in the life cycle of each conversion technology module. All fuel and electrical energy units are converted to British thermal unit (BTU) values. Electricity production assumes the average U.S. conversion efficiency of fuel to electricity and accounts for transmission and distribution losses in the power lines. Therefore, the KWh value is the aggregated amount of electricity used by the system, as delivered to the various facilities in the life cycle. The BTU value accounts for the average mix of fuels (for example, coal, natural gas, hydroelectricity, nuclear) used by utilities to produce electricity in the United States.

Where energy is produced by a process and displaces the production of electricity or a fuel by a utility or the petroleum sector, respectively, such as the combustion of MSW with energy recovery, a credit is given to the extent that it displaces power generation by the utility sector or production of the fuel. For this study, we used the U.S. average electrical energy grid mix to calculate the life cycle inventory burdens associated with electrical energy consumption, as well as the credits associated with electrical energy offsets. **Figure 4-2** presents the fuel mix in the U.S. average electrical energy grid (U.S. EIA, 2009).



**Figure 4-2. U.S. Average Electrical Energy Grid Mix of Fuels (U.S. EIA, 2009).**

### *Air Emissions*

Air emissions can result from two primary sources in the life cycle: process-related activities or fuel-related activities. Process emissions are those that are emitted during a processing step, but not as a result of fuel combustion. For example, calcination of limestone to produce lime emits CO<sub>2</sub>. The quantity of CO<sub>2</sub> emitted from this process would be listed under process air emissions. Fuel-related emissions are those emissions that result from the combustion of fuels. For example, the combustion of wood byproducts in a paper mill produces a fuel-related solid waste, ash. The emissions reported on the data tables in the product summaries are the quantities that reach the environment (air, water, and land) after pollution control measures have been taken.

Atmospheric emissions include substances released to the air that are regulated or classified as pollutants. Emissions are reported as pounds of pollutant per annual tonnage of waste managed. Atmospheric emissions also include CO<sub>2</sub> releases, which are calculated from fuel combustion data or process chemistry. CO<sub>2</sub> emissions are not regulated; however, we are

reporting them in this study because of the growing concern about global warming. CO<sub>2</sub> emissions are labeled as being from either fossil or nonfossil fuels.

CO<sub>2</sub> released from the combustion of fossil carbon sources (for example, coal, natural gas, or petroleum) or released during the reaction of chemicals derived from these materials is classified as fossil CO<sub>2</sub>. CO<sub>2</sub> released from mineral sources (for example, the calcining of limestone to lime), is also classified as fossil CO<sub>2</sub>. CO<sub>2</sub> from sources other than fossil carbon sources (that is, from biomass) is classified as nonfossil carbon dioxide. Nonfossil CO<sub>2</sub> includes CO<sub>2</sub> released from the combustion of plant or animal material or released during the reaction of chemicals derived from these materials. The labeling of the CO<sub>2</sub> releases as either fossil or nonfossil is done to aid in the interpretation of the life cycle inventory data. The source of CO<sub>2</sub> releases is an important issue in the context of the natural carbon cycle and global warming.

#### ***Waterborne Pollutants***

Waterborne wastes are produced from both process activities and fuel-production activities. These are reported as pounds of pollutant per tonnage of waste managed. Similar to air emissions, the waterborne pollutants include substances released to surface water and groundwater that are regulated or classified as pollutants. The values reported are the average quantity of pollutants still present in the wastewater stream after wastewater treatment and represent discharges into receiving waters.

Air or waterborne emissions that are not regulated or reported to regulatory agencies are not reported in the inventory results presented in the material summaries. Reliable data for any such emissions would be difficult to obtain, except for a site-specific study where additional testing was authorized. Conversely, some air and waterborne emissions data that are regulated and reported may not have been included in the inventory results. The data used represent the best available from existing sources.

#### ***Solid Waste***

Similar to air and water emissions, solid wastes are produced from process and fuel production activities and are reported as pounds of pollutant per tonnage of waste managed. Process solid wastes include mineral processing wastes (such as red mud from alumina manufacturing); wastewater treatment sludge; solids collected in air pollution control devices; trim or waste materials from manufacturing operations that are not recycled; and packaging materials from material suppliers.

Fuel-related solid wastes are fuel production and combustion residues, such as the ash generated by burning coal or wood.

### **4.3 Key Data and Assumptions Used in the LCIs**

**Table 4-1** presents key LCI assumptions for the different conversion processes.

**Table 4-1. Key Assumptions Used in the LCIs.**

<b>Parameter</b>	<b>Assumption</b>
<b>General</b>	
Waste Input	Gasification: 1 ton of mixed non-recycled waste
	Pyrolysis: 1 ton of plastics
	Anaerobic digestion (AD): 1 ton of organics
Waste Composition	Gasification: average U.S. post-recovery composition from U.S. EPA (2008)
	Pyrolysis: 100% plastics
	AD: food waste, yard waste, paper
<b>Transportation Distances</b>	
Conversion facility to ash landfill	30 miles one way
Gasification facility to landfill	30 miles one way
AD facility to landfill	30 miles one way
<b>Gasification</b>	
Basic Design	Accepts mixed waste; includes recyclables recovery; syngas as the main product
Waste Input Heating Value	12 MMBtu/ton (based on waste composition)
Assumed Offset for Energy Recovery	Solid waste to electricity: U.S. average electricity grid mix of fuels
<b>Pyrolysis</b>	
Basic Design	Only accepts plastics; does not include recyclables recovery; oil/wax as the main product
Waste Input Heating Value	28 MMBtu/ton (plastics only)
Assumed Offset for Energy Recovery	Fuel oil
<b>AD</b>	
Basic Design	Only accepts food, yard, and paper wastes; assumed biogas as the main product
Waste Input Gas Yield	3,281 ft <sup>3</sup> /ton
Assumed Offset for Energy Recovery	Solid waste to electricity: U.S. average electricity grid mix of fuels
<b>Landfill</b>	
Basic Design	Conventional, Subtitle D Type
Time Period for Calculating Emissions	100 years
Landfill Gas Collection Efficiency	75%
Landfill Gas Management	Energy Recovery
Assumed Offset	U.S. average electricity grid mix of fuels

At least three gasification and three pyrolysis technology vendors were contacted to obtain process data. However, not all the vendors provided data and, in some cases, the data provided or obtained was not complete. Therefore, RTI reviewed the literature for additional data and selected ranges for relevant input parameters to reflect the uncertainty and variability in the data obtained. There are currently no stand-alone AD facilities and thus literature data was used for AD process data. **Tables 4-2 through 4-4** present the data ranges used for modeling and **Appendix A** presents the original data used to select those ranges.

**Table 4-2. Gasification Process Data.**

Parameters		Units	Value			
<b>Process Inputs and Outputs</b>						
Inputs	Power consumption/parasitic load	KWh/dry ton	200	-	490	
	Other inputs (e.g., water, oxygen, etc.)	Oxygen	lb/dry ton			1,446
		Catalysts and chemicals	lb/dry ton			107
		Diesel for preprocessing	gal/dry ton			0.05
		Caustic for gas cleaning and cooling	lb/dry ton			10
		Activated Carbon for gas cleaning and cooling	gal/dry ton			0.2
		Feldspar for gas cleaning and cooling	gal/dry ton			0.1
	Water	gal/dry ton	540	-	1,622	
	Supplemental fuel use	Natural Gas	lb/dry ton	16	-	87
	Outputs	Energy product (e.g., syngas, ethanol, hydrogen, electricity, steam)	Electricity	KWh/dry ton	925	-
Material byproducts		Residual gas	lb/dry ton			428
		Sulfur	lb/dry ton	2.6	-	2.7
		Salt	lb/dry ton	9	-	13
		Slag	lb/dry ton	24	-	424
Residuals (e.g., ash, char, slag, etc.)		Char	lb/dry ton			297
		Slag	lb/dry ton			75
		Gasifier solid residues	lb/dry ton	25	-	120
		Spent catalysts and chemicals	lb/dry ton			3
		Inorganic sludge	lb/dry ton			45
		Nonhazardous solid waste	lb/dry ton			13
<b>Air Emissions Data</b>						
PM			lb/dry ton	0.01	-	0.35
PM10			lb/dry ton			0.001
Biogenic Carbon Dioxide (CO2bio)			lb/dry ton	467		
Fossil Carbon Dioxide (CO2fossil)			lb/dry ton	345	-	1,048
Methane (CH4)			lb/dry ton	2.E-04	-	2
HCl			lb/dry ton	0	-	0.03
Sulfur Dioxide (SO2)		lb/dry ton	0	-	0.4	
Sulfur Oxide		lb/dry ton			5.E-05	
Nitrous Oxide (N2O)		lb/dry ton	0.001	-	0.40	
NOx expressed as NO2		lb/dry ton	0.2	-	1	
Carbon Monoxide (CO)		lb/dry ton	0.1	-	1	

Parameters		Units	Value		
	Mercury (Hg)	lb/dry ton			6.E-07
	Cadmium (Cd)	lb/dry ton			8.E-06
	Lead	lb/dry ton			1.E-05
	VOC	lb/dry ton	1	-	0.04
	HAP	lb/dry ton			0.1
	Acetaldehyde	lb/dry ton			0.1
	Total non-methane organic carbon (TNMOC)	lb/dry ton	0	-	0.2
	Dioxins and Furans	lb/dry ton			0
	<b>Water Emissions Data</b>				
Water Effluent	gal/dry ton	600	-	1,400	
<b>Cost Data</b>					
Cost per design capacity		\$/dtpd	499,109		

**Table 4-3. Pyrolysis Process Data.**

Parameters		Units	Value			
<b>Process Inputs and Outputs</b>						
Inputs	Power consumption/parasitic load	KWh/dry ton	0.3	-	480	
	Other inputs (e.g., water, oxygen, etc.)	Water gal/dry ton	30	-	216	
	Supplemental fuel use	Natural Gas MMBtu/dry ton			0.03	
Outputs	Energy product (e.g., syngas, ethanol, hydrogen, electricity, steam)	Syngas	MMBtu/dry ton		0.2	
		Synthetic crude oil	lb/dry ton	37	-	39
		Light fraction (liquid)	lb/dry ton	300	-	400
		Gas fraction	lb/ dry ton	200	-	500
		Gasoline	lb/ dry ton			23
		Diesel	lb/dry ton			1,711
	Residuals (e.g., ash, char, slag, etc.)	Char	lb/dry ton	136	-	160
		Solid residues	lb/dry ton			160
		Inorganic sludge	lb/dry ton			300
		Nonhazardous solid waste	lb/dry ton			5
	Water losses	gal/dry ton			25	
	<b>Air Emissions Data</b>					
	PM		lb/dry ton	0.04	-	15
	Fossil Carbon Dioxide (CO2Fossil)		lb/dry ton	500	-	962
	Methane (CH4)		lb/dry ton	26	-	65
	HCl		lb/dry ton			3.E-04
	Hydrocarbons		lb/dry ton	0.01	-	8
Nitrous Oxide (N2O)		lb/dry ton			2	
NOx expressed as NO2		lb/dry ton	0.3	-	91	
Carbon Monoxide (CO)		lb/dry ton		-	9	
Lead		lb/dry ton	2.E-04	-	0.02	
VOC		lb/dry ton	3.E-04	-	2	
<b>Cost Data</b>						
Cost per design capacity		\$/dtpd	29,350	-	280,699	

**Table 4-4. AD Process Data.**

Parameters		Units	Value			Data Sources
<b>Process Characteristics</b>						
Power consumption/parasitic load		% energy produced	22	-	30	RTI (2005) and ARI (2007)
Total Solids		%			70	RTI (2005)
Volatile Solids		%			70	RTI (2005)
Biodegradable Volatile Solids		%			75	RTI (2005)
Conversion Efficiency waste to methane		%	60	-	75	RTI (2005) and ARI (2007)
Conversion Efficiency methane to electricity		%	33	-	39	RTI (2005) and ARI (2007)
<b>Air Emissions Data</b>						
PM		lb/dry ton			0.12	RTI (2005)
HCl		lb/dry ton			0.02	RTI (2005)
Nitrogen Oxides		lb/dry ton			0.61	RTI (2005)
Sulfur Oxides		lb/dry ton			0.03	RTI (2005)
Carbon Monoxide (CO)		lb/dry ton			1.15	RTI (2005)
Carbon Dioxide (biomass)		lb/dry ton			137	RTI (2005)
<b>Cost Data</b>						
Cost per design capacity		\$/dtpd			82	ARI (2007)

## 4.4 Environmental and Cost Assessment Results

Results for energy consumption and greenhouse gas (GHG) emissions as carbon equivalents, as well as cost results, are presented and discussed in this section of the report.

Results are presented as net totals, burdens minus benefits. Therefore, negative energy results mean that more energy is recovered than that needed to run the processes; negative GHG emissions mean that there are more emissions savings as a result of energy and fuels production using the waste material relative to using virgin material; and negative cost results mean that the revenues are higher than the costs.

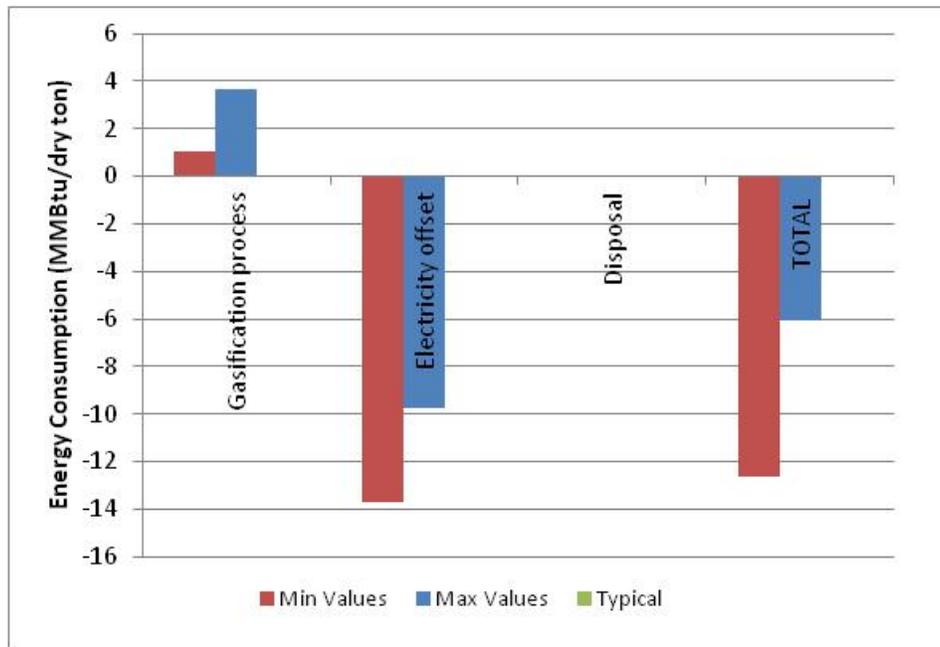
### 4.4.1 Gasification Results

The cost and LCI results for energy and GHG emissions for gasification of MSW are presented in this section. Since gasification technologies typically accept MSW that must be pre-processed, recyclables are recovered and residual unwanted wastes must be disposed. Thus the cost and LCI results include burdens associated with the pre-processing of MSW, transportation of recyclables and residuals, and the disposal of residuals. The benefits include not only the electricity recovered, but also the materials recovery and recycling.

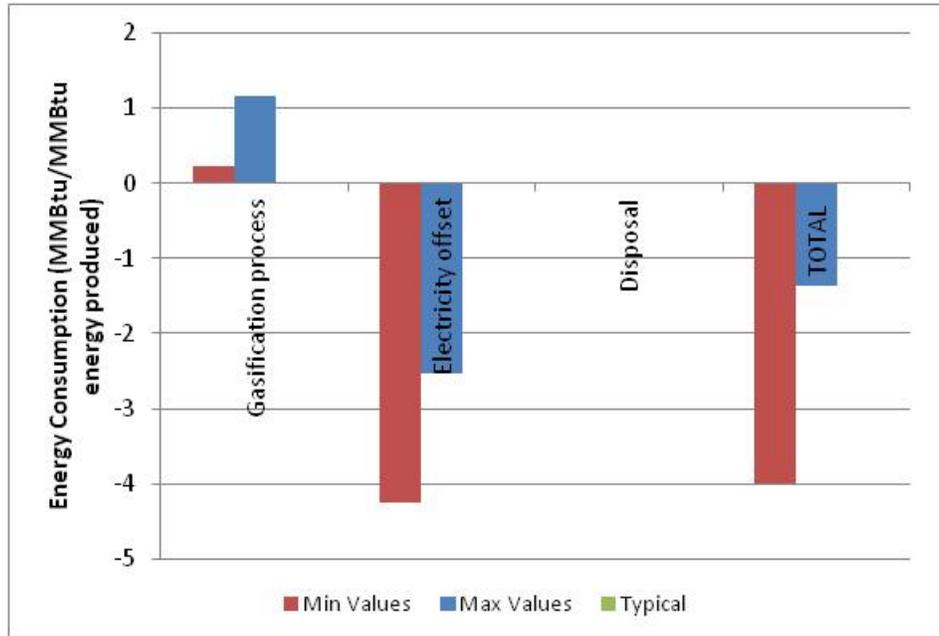
#### *Energy*

For gasification, energy is consumed to pre-process the incoming MSW, power the gasifier and ancillary systems, transport residuals, and dispose of residuals in a landfill. Energy in the form of syngas is the main output from the gasification process. Typically this syngas is combusted onsite in an internal combustion engine (ICE) generator set (gen-set) to produce electricity. This is the process modeled in the LCI. The syngas can be directly used or converted to liquid fuel, but these options were not modeled because they are less common.

The net energy consumption results for gasification are shown in **Figure 4-3** on a per-ton basis and in **Figure 4-4** per MMBtu of energy produced. As shown in the figures, the energy (in the form of electrical energy) produced from the gasification process generate significant energy offsets. The gasification process itself is a net electricity producer (i.e., the energy produced exceeds the energy consumed) with some variation (according to the data obtained from the different vendors and the literature) in the amount of energy produced in the range of over 4 MMBtu per ton or over 2 MMBtu per MMBtu of energy produced.



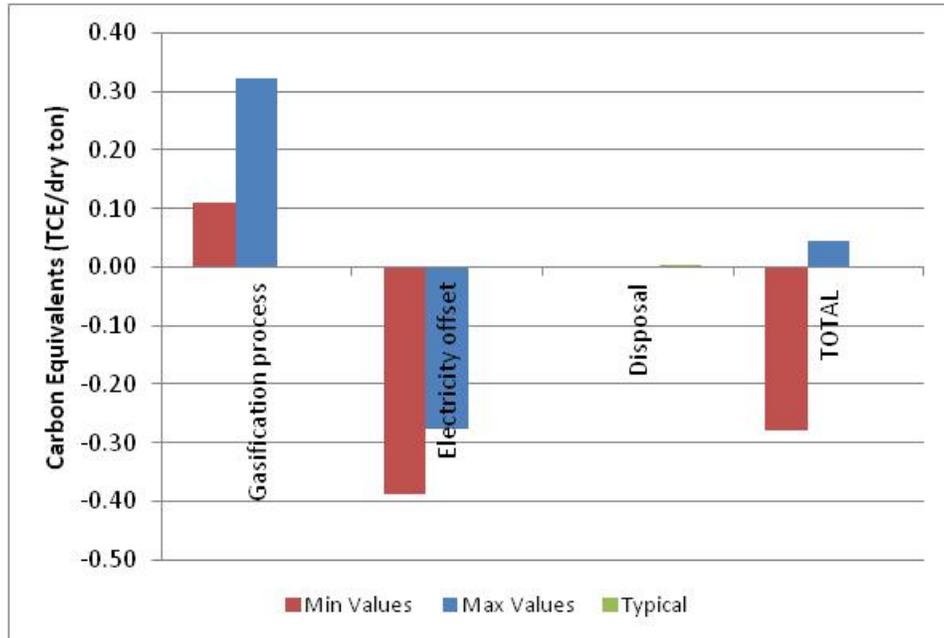
**Figure 4-3. Net Energy Consumption Per Ton for Gasification of MSW.**



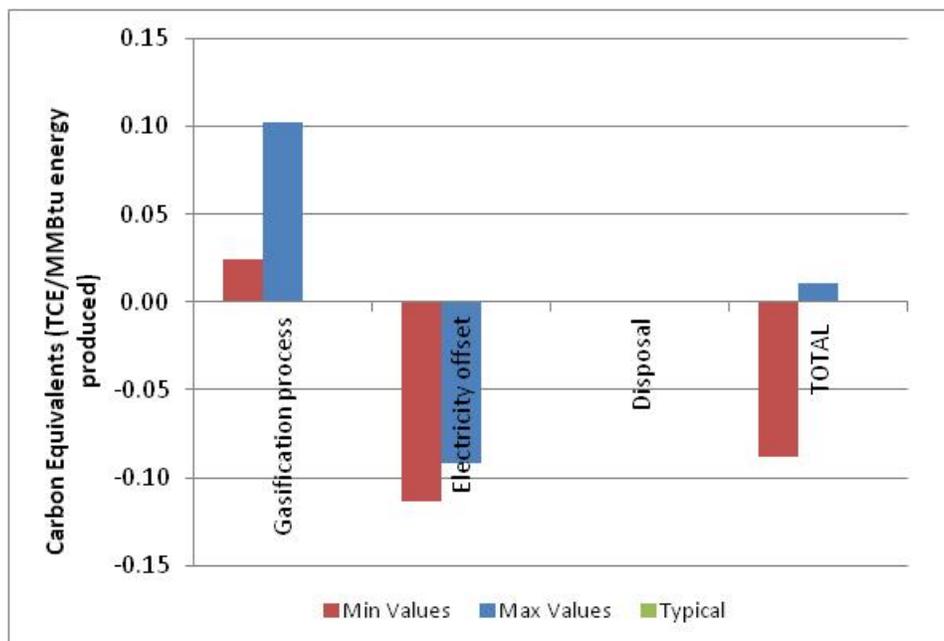
**Figure 4-4. Net Energy Consumption Per MMBtu for Gasification of MSW.**

***GHG Emissions***

Figures 4-5 and 4-6 show the gasification process producing a net GHG emissions savings at the lower end of emissions generation from the process, which results from the displacement of conventional electricity production (assuming displacement of fossil fuels in the U.S. average grid mix of fuels for electricity production). The emissions data obtained for the gasification piece of the LCI exhibits a wide range of variation from a net savings of approximately 0.28 TCE/dry ton (~0.09 TCE/MMBtu energy produced) to a burden of 0.05 TCE/dry ton (~0.01 TCE/MMBtu energy produced) as illustrated by the minimum and the maximum bars.



**Figure 4-5. Net Carbon Equivalents Per Ton for Gasification of MSW.**

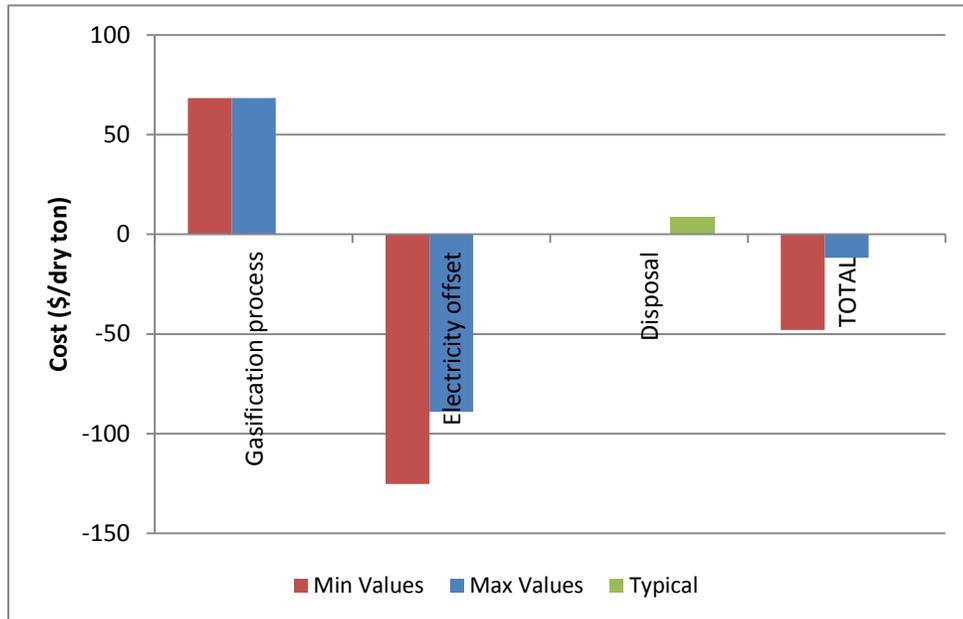


**Figure 4-6. Net Carbon Equivalents Per MMBtu for Gasification of MSW.**

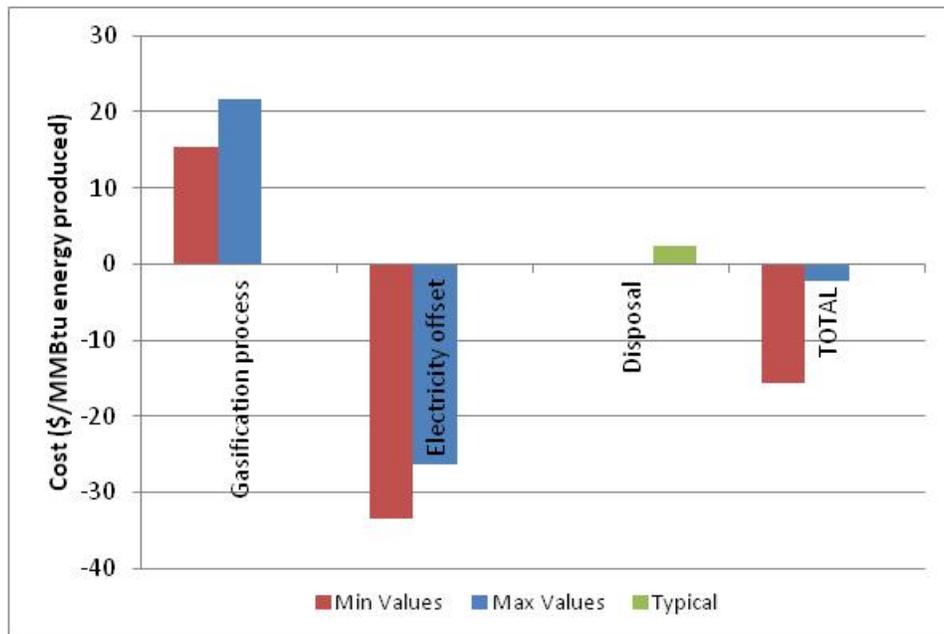
**Cost**

Cost data were only available for one of the gasification technology vendors in **Section 3**. **Figures 4-7 and 4-8** show the cost (or revenue) by process as well as the total net cost of approximately (\$48) to (\$12) per ton of MSW or (\$16) to (\$2) per MMBtu of energy produced. This signifies that the revenues received from the sale of electricity are greater than the costs to

process the MSW via the gasification technology. In **Figures 4-7 and 4-8**, the process cost/revenue is per vendor-supplied values and the remaining residuals disposal costs are per RTI's MSW DST.



**Figure 4-7. Net Cost Per Ton for Gasification of MSW.**



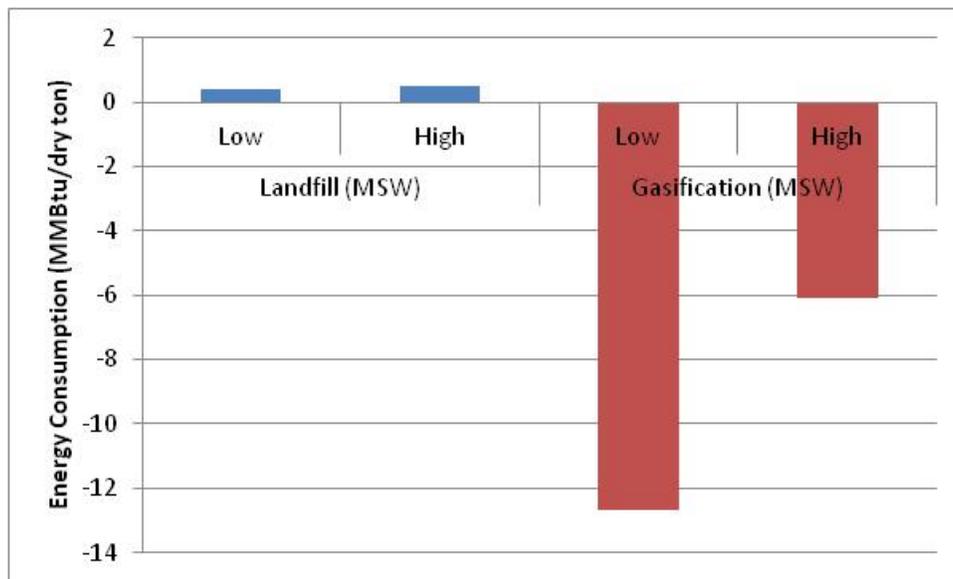
**Figure 4-8. Net Cost per MMBtu for Gasification of MSW.**

**Comparison to Landfill Base Case**

The results for gasification of MSW were compared to results for a landfill base case for MSW. A low–high range was developed for the landfill base case using a landfill with gas collection

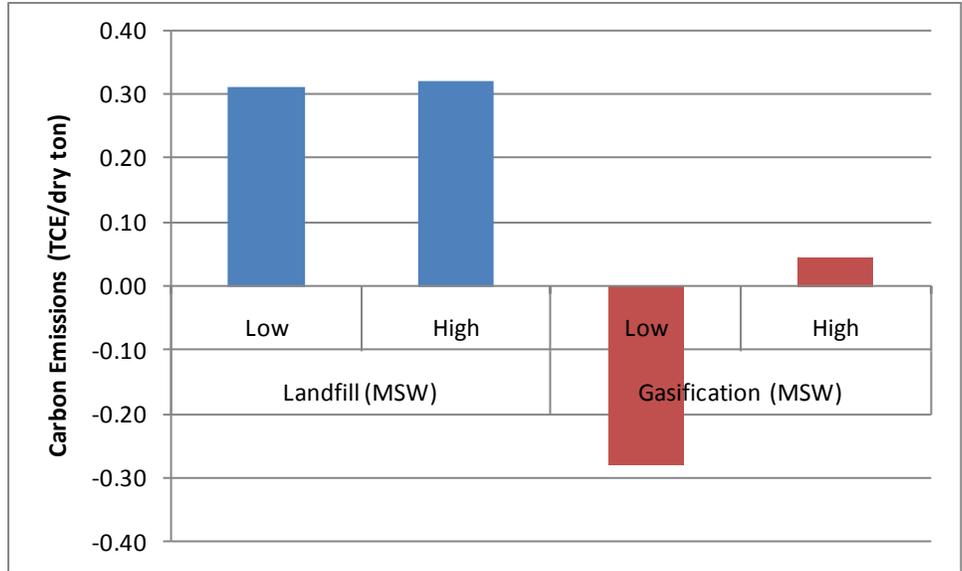
and flaring for the “low” end of the range and a landfill with gas collection and energy recovery for the “high” end of the range. The landfill base case was modeled using RTI’s MSW DST and is representative of a U.S. average.

**Figure 4-9** shows the results for net energy consumption (i.e., energy consumed minus energy produced). According to this figure, the net energy saved using the gasification technology versus landfill disposal is approximately 6.5–13 MMBtu per dry ton of MSW. These savings are mostly associated with the energy produced by the gasification facility. For example, when compared to a landfill with energy recovery (i.e., the low landfill energy consumption bar in **Figure 4-9**), these savings indicate that the gasification facility is much more efficient at producing energy than the landfill facility.



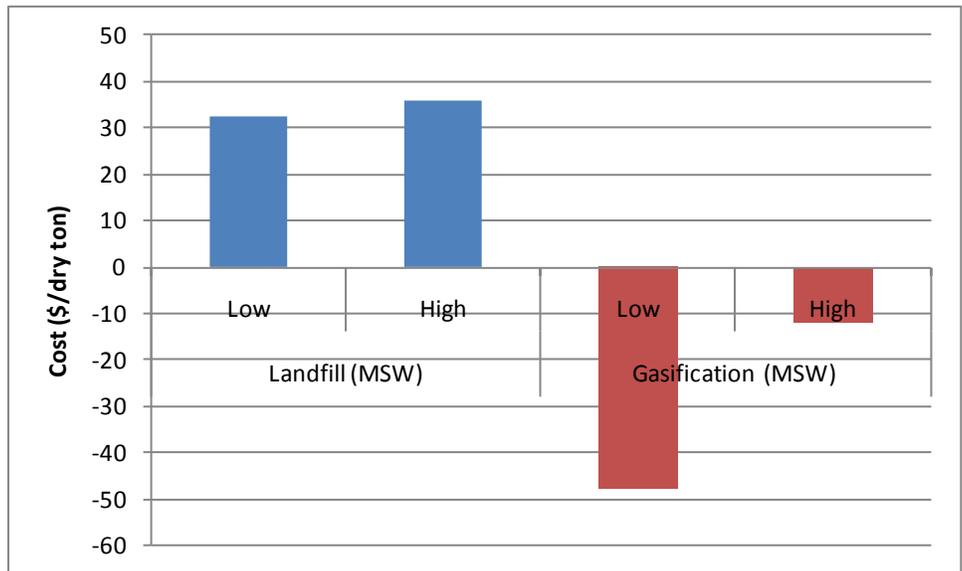
**Figure 4-9. Net Energy Consumption for Landfill Base Case and Gasification of MSW.**

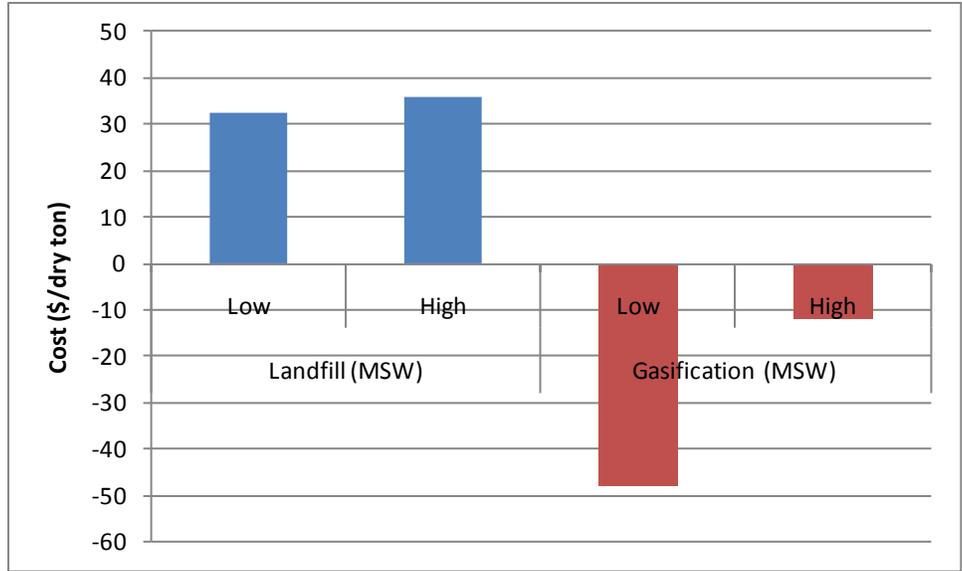
**Figure 4-10** shows the results for net carbon emissions (i.e., carbon emissions minus savings). According to this figure, the gasification technology results in a net reduction of approximately 0.3–0.6 TCE per dry ton of MSW processed when compared to landfills. This reduction is mostly associated with the energy produced by the gasification facility. For example, **Figure 4-9** indicates that the gasification facility is more efficient at energy production than the landfill with energy recovery, so the emissions savings associated with energy production using waste versus virgin materials are also greater for gasification facilities.



**Figure 4-10. Net Carbon Equivalents for Landfill Base Case and Gasification of MSW.**

**Figure 4-11** shows the results for net cost (i.e., cost minus revenues). According to this figure, the gasification technology results in a net reduction of approximately \$40–85 per dry ton of MSW processed when compared to landfills. Consistent with the energy and GHG emissions results, this reduction is mostly associated with the energy produced by the gasification facility. For example, by having larger energy savings, as illustrated in **Figure 4-9**, the gasification facility will also get more revenues from energy sales than the landfill with energy recovery.





**Figure 4-11. Net Cost for Landfill Base Case and Gasification of MSW.**

#### 4.4.2 Pyrolysis Results

The cost and LCI results for energy and GHG emissions for pyrolysis of plastics are presented in this section. Pyrolysis results include transportation and disposal of residuals. Thus the cost and LCI results include the burdens associated with the pyrolysis facility as well as with transportation and disposal of residuals. The benefits are those associated with fuels recovery.

##### *Energy*

For pyrolysis, energy is consumed to power the process and ancillary systems and transport and dispose of residuals in a landfill. Energy in the form of petroleum products (e.g., diesel and petroleum wax) is the main output from the pyrolysis process. Typically this product is transported off-site for use.

The LCI results for energy consumption for pyrolysis are shown in **Figure 4-12** on a per-ton basis and in **Figure 4-13** per MMBtu of energy produced. According to these figures, the petroleum product output generates large energy offsets. The pyrolysis process can be considered an energy producer (i.e., the energy produced exceeds the energy consumed), with some variation in the amount of energy produced, according to the data obtained from the vendors and the literature.

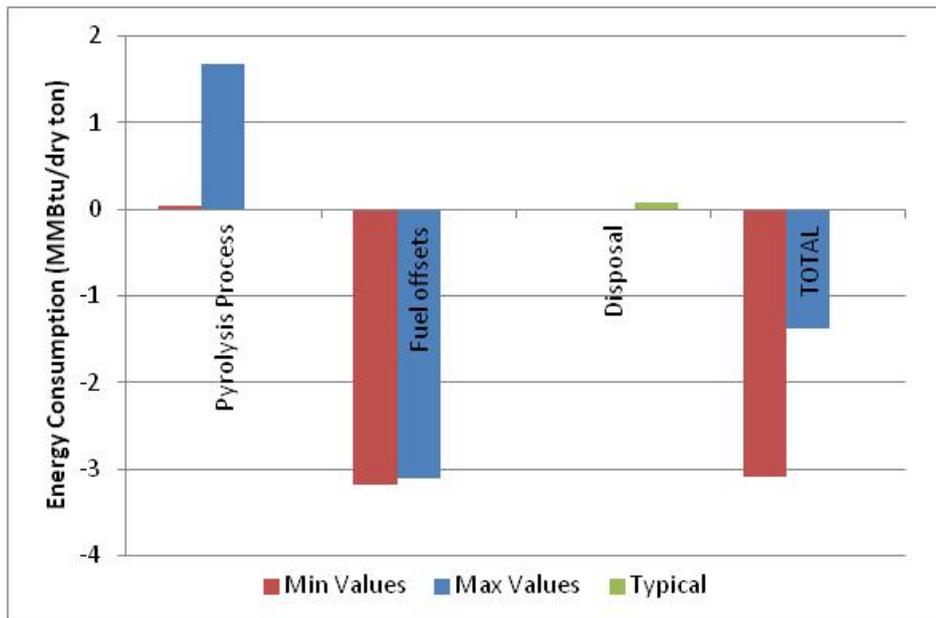


Figure 4-12. Net Energy Consumption Per Ton for Pyrolysis of Plastics.

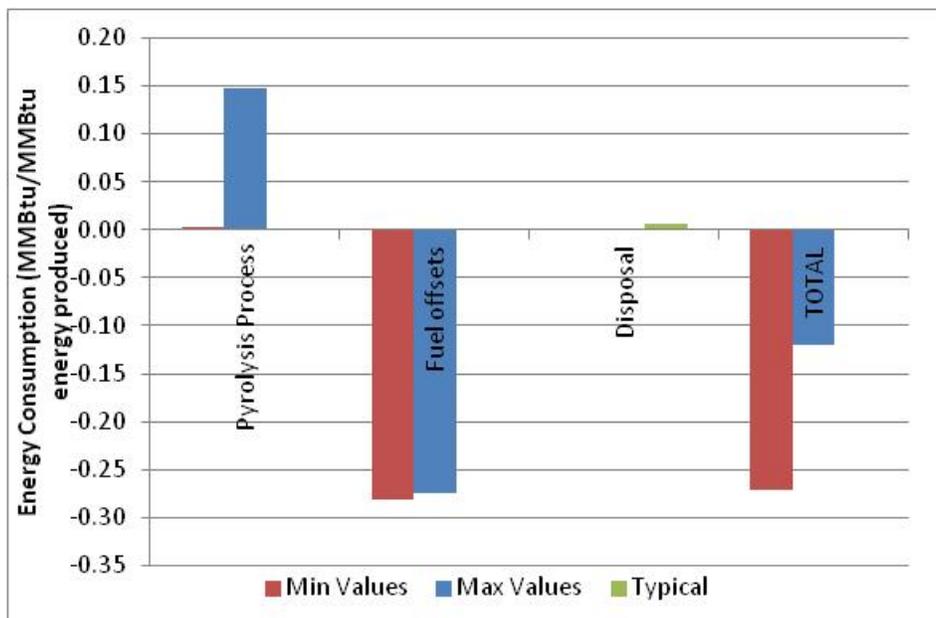


Figure 4-13. Net Energy Consumption Per MMBtu for Pyrolysis of Plastics.

### GHG Emissions

Consistent with the energy results, **Figures 4-14 and 4-15** show that pyrolysis of plastics results in GHG emissions savings, which are mostly due to emissions savings from the replacement of conventional energy (petroleum) products. The emissions data obtained for pyrolysis exhibits a wide range of variation, as illustrated by the minimum and the maximum bars.

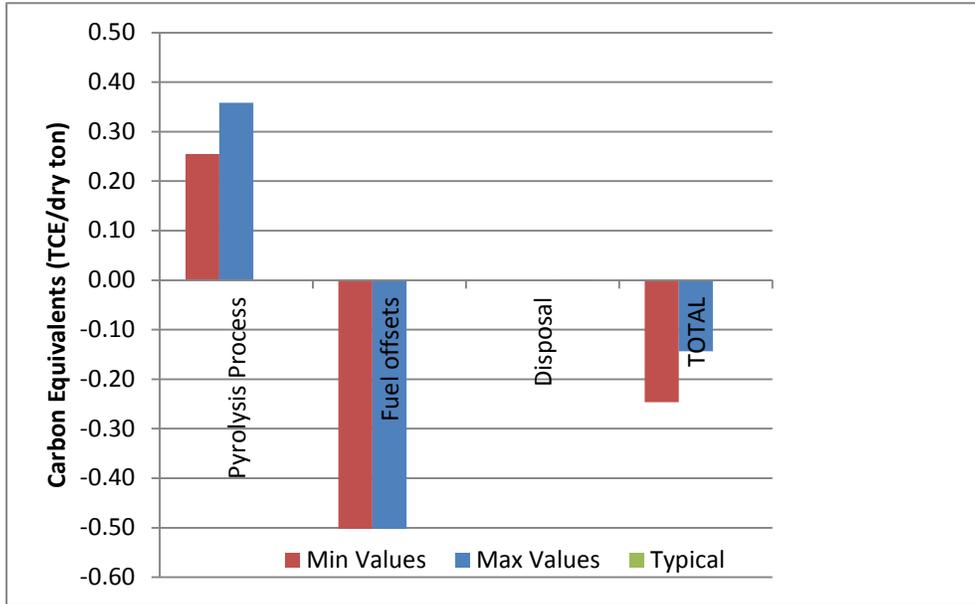


Figure 4-14. Net Carbon Equivalents Per Ton for Pyrolysis of Plastics.

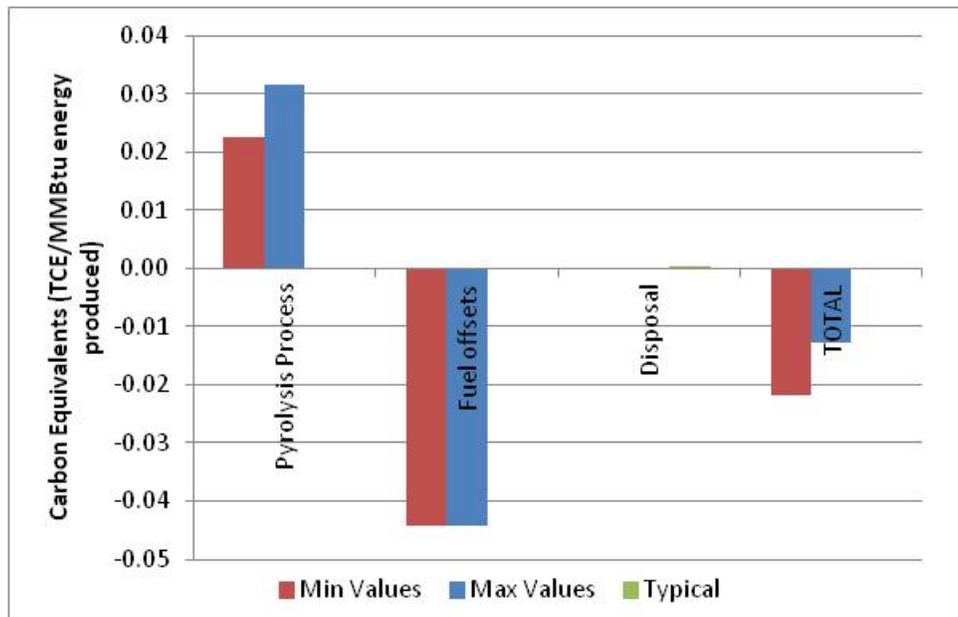
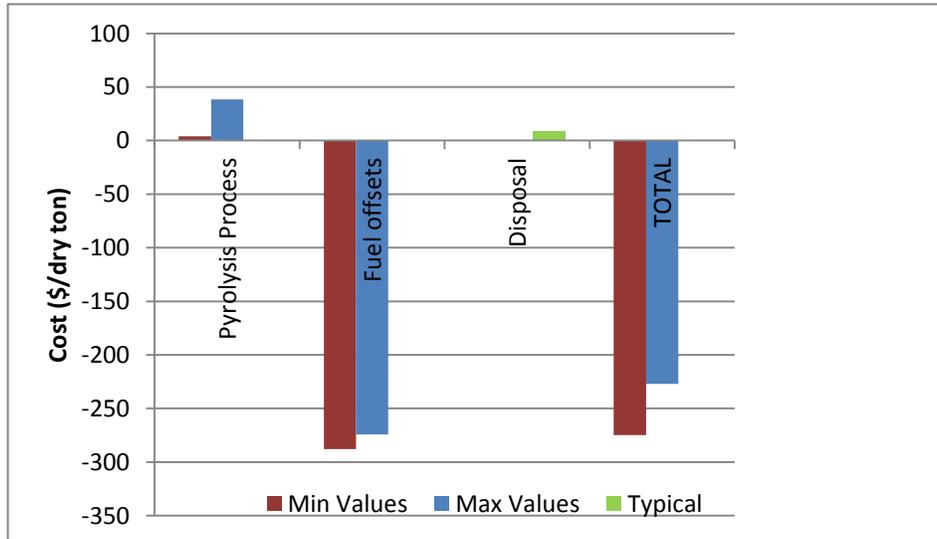


Figure 4-15. Net Carbon Equivalents Per MMBtu for Pyrolysis of Plastics.

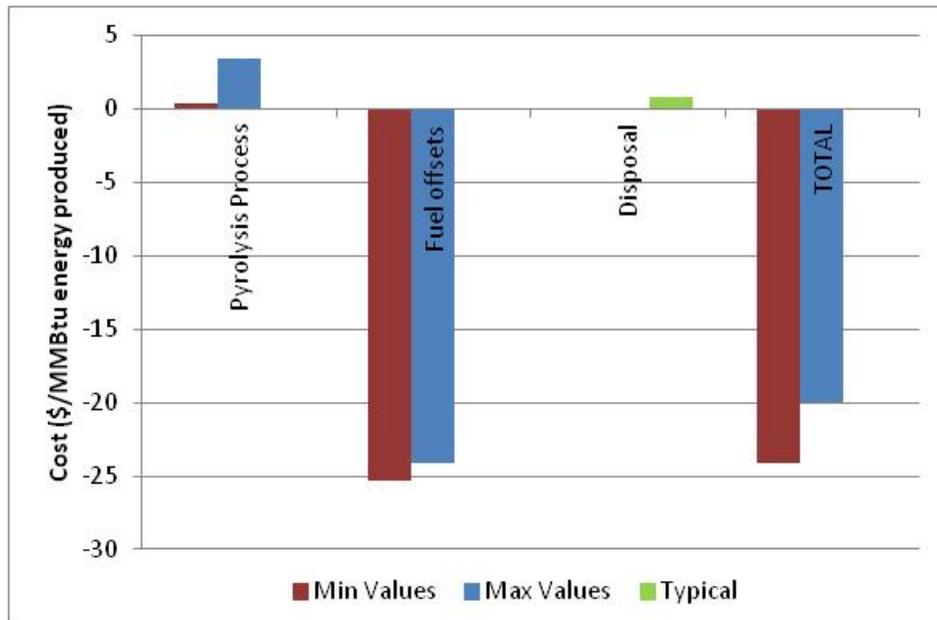
**Cost**

The net cost (expenses minus revenues) per ton for pyrolysis of plastics is shown in **Figures 4-16 and 4-17**. As shown in these figures, the net cost range is negative, signifying a net revenue stream that results from the market value of the petroleum product being greater than the cost to process the plastics into petroleum via the pyrolysis process.

The conversion efficiency (e.g., number of barrels of oil per ton of plastics) and contracted market price for the recovered petroleum product are highly significant to the net cost. Facilities will likely align their specific technology to obtain the specific petroleum product (e.g., diesel and petroleum wax) that yields the highest market price.



**Figure 4-16. Net Cost Per Ton for Pyrolysis of Plastics.**



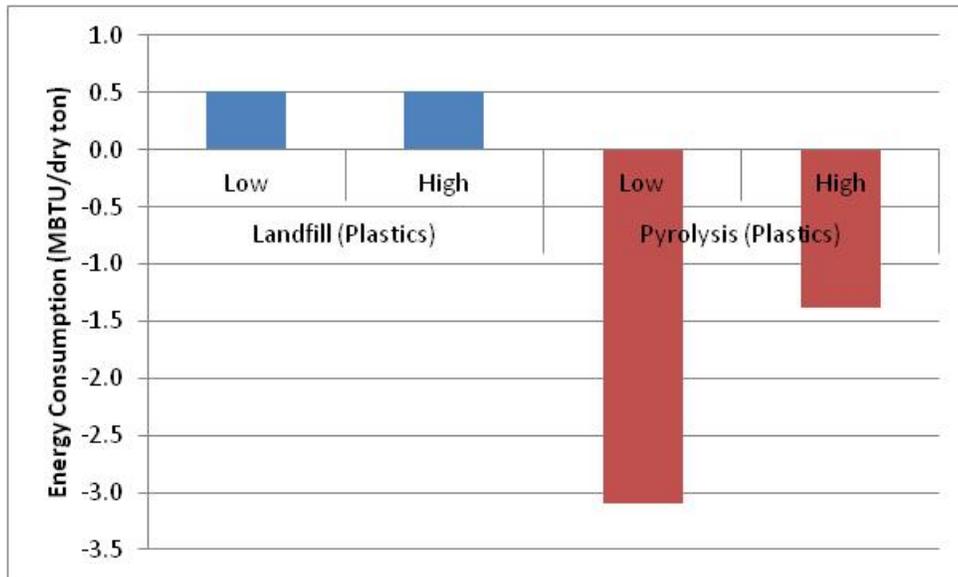
**Figure 4-17. Net Cost Per MMBtu for Pyrolysis of Plastics.**

**Comparison to Landfill Base Case**

In this section, the results for pyrolysis of plastics are compared to results for a landfill base case for plastics. A low–high range was developed for the landfill base case using a landfill with gas collection and flaring for the “low” end of the range and a landfill with gas collection and energy recovery for the “high” end of the range. However, since plastics waste isn’t expected

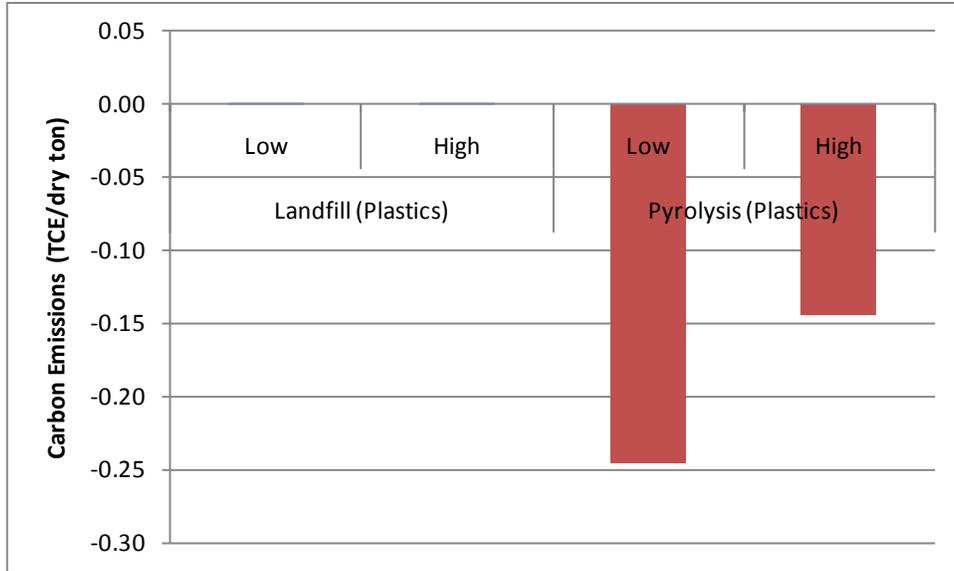
to produce any gas, this distinction is not relevant and only done to be consistent with the gasification results. Again, the landfill base case was modeled using RTI’s MSW DST and is representative of a U.S. average.

**Figure 4-18** shows the results for net energy consumption (i.e., energy consumed minus energy produced). According to this figure, the net energy saved using the pyrolysis technology versus landfill disposal is approximately 2–4 MMBtu per dry ton of plastics. These savings are mostly associated with the fuels produced by the pyrolysis facility and the fact that there is no energy recovery potential (i.e., there is no methane generation) from landfill disposal of plastics.



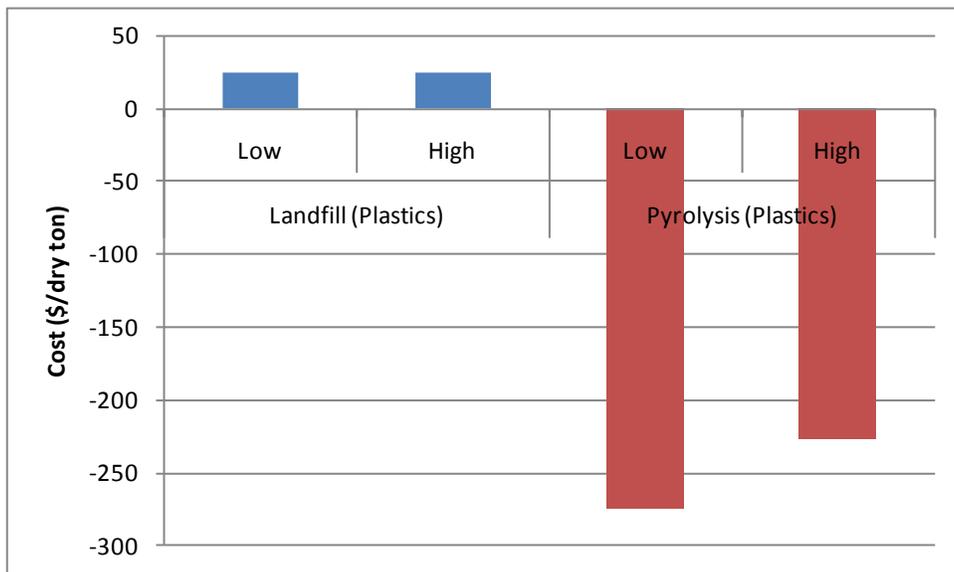
**Figure 4-18. Net Energy Consumption for Landfill Base Case and Pyrolysis of Plastics.**

**Figure 4-19** shows the results for net carbon emissions (i.e., carbon emissions minus savings). According to this figure, the pyrolysis technology results in a net reduction of approximately 0.15–0.25 TCE per dry ton of plastics processed when compared to landfills. This reduction is mostly associated with the fuels produced by the gasification facility and the fact that no energy is generated from landfill disposal of plastics. In particular, emissions savings are estimated when fuels production from waste is compared with fuels production from virgin materials.



**Figure 4-19. Net Carbon Equivalent for Landfill Base Case and Pyrolysis of Plastics.**

**Figure 4-20** shows the results for net cost (i.e., costs minus revenues). According to this figure, the pyrolysis technology results in a net reduction of approximately \$250–300 per dry ton of plastics processed when compared to landfills. Consistent with the energy and GHG emissions results, this reduction is mostly associated with the fuels produced by the pyrolysis facility. For example, the pyrolysis facility will get revenues from fuels sale.



**Figure 4-20. Net Cost for Landfill Base Case and Pyrolysis of Plastics.**

#### 4.4.3 Anaerobic Digestion Results

The cost and LCI results for energy and GHG emissions for AD of organics (namely, food and yard wastes) are presented in this section. AD results include transportation and disposal of

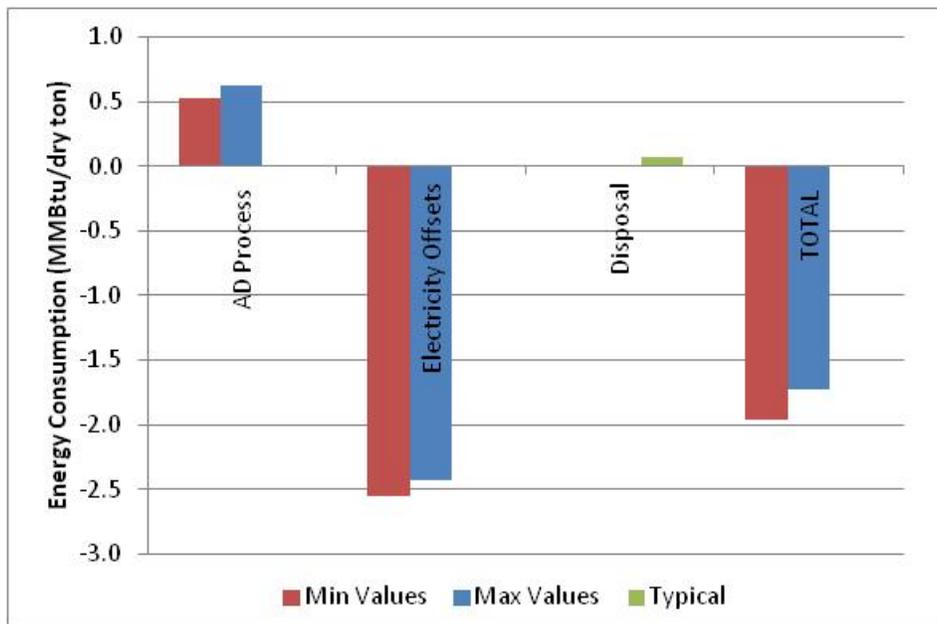
residuals. Thus the cost and LCI results include the burdens associated with the AD facility as well as with transportation and disposal of residuals. The benefits are those associated with electricity recovery.

Results are presented as net totals, burdens minus benefits. Therefore, negative energy results mean that there is more energy recovered than that needed to run the processes, negative GHG emissions mean that there are more emissions savings as a result of energy and fuels production using the waste material than using virgin material, and negative cost results mean that the revenues are higher than the costs.

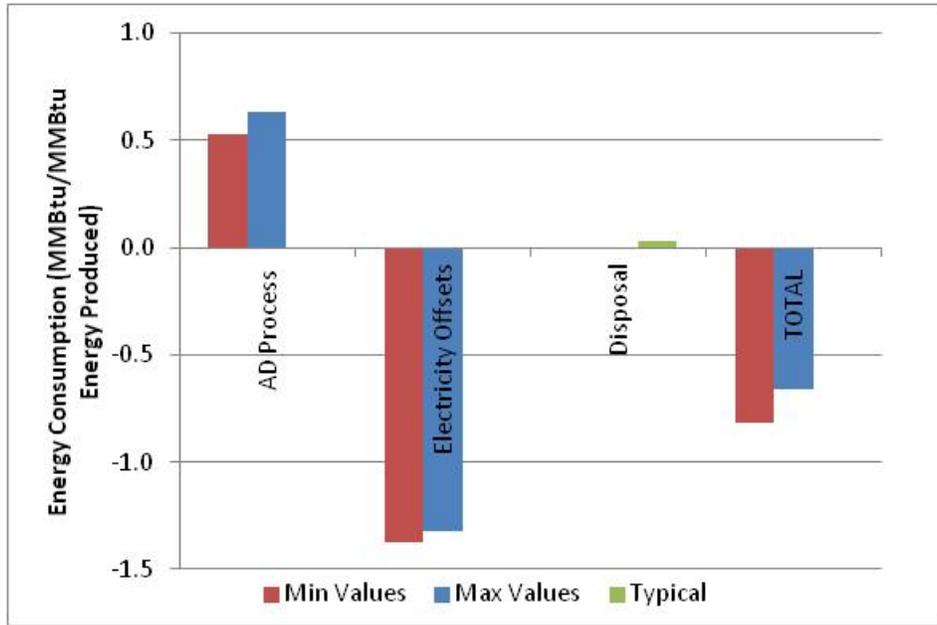
**Energy**

For AD, energy is consumed in feedstock pre-processing, digestate post processing, ancillary systems, and transport and dispose of residuals in a landfill. Energy in the form of methane-rich biogas, which can be combusted to generate heat and/or electricity or further refined to create biomethane, a transportation fuel, is the main output from the AD process. For this analysis we assumed the biomethane will be used to generate electricity.

The LCI results for energy consumption for AD are shown in **Figures 4-21 and 4-22**. According to **Figure 4-21**, the electricity output generates energy offsets. The AD process can be considered an energy producer (i.e., the energy produced exceeds the energy consumed), with some variation in the amount of energy produced, according to the data obtained from the literature.



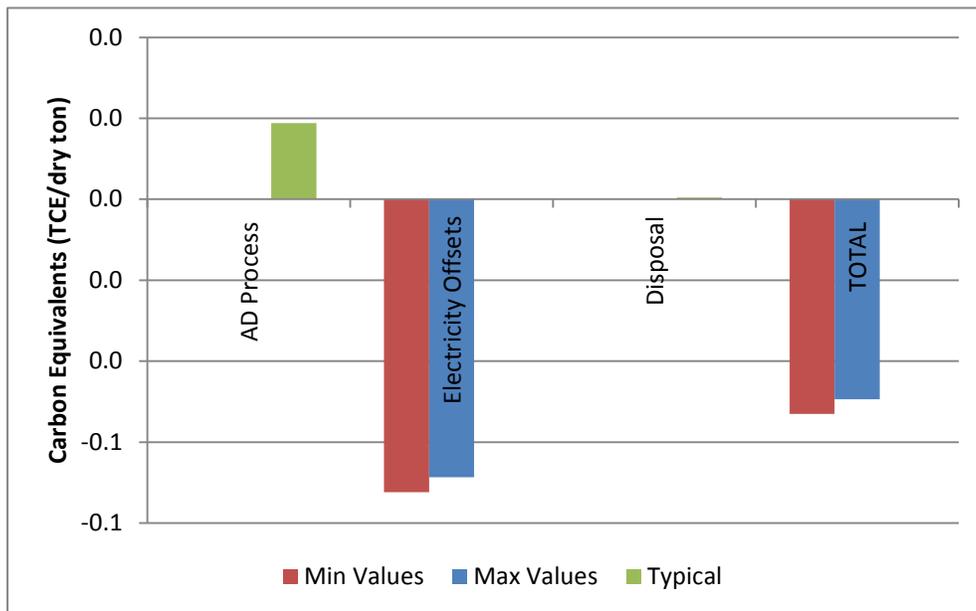
**Figure 4-21. Net Energy Consumption Per Ton for AD of Organics.**



**Figure 4-22. Net Energy Consumption Per MMBtu for AD of Organics.**

***GHG Emissions***

Consistent with the energy results, **Figures 4-23 and 4-24** show that the AD of organics results in GHG emissions savings, which results from the displacement of conventional electricity production (assuming displacement of fossil fuels in the U.S. average grid mix of fuels for electricity production).



**Figure 4-23. Net Carbon Equivalents Per Ton for AD of Organics.**

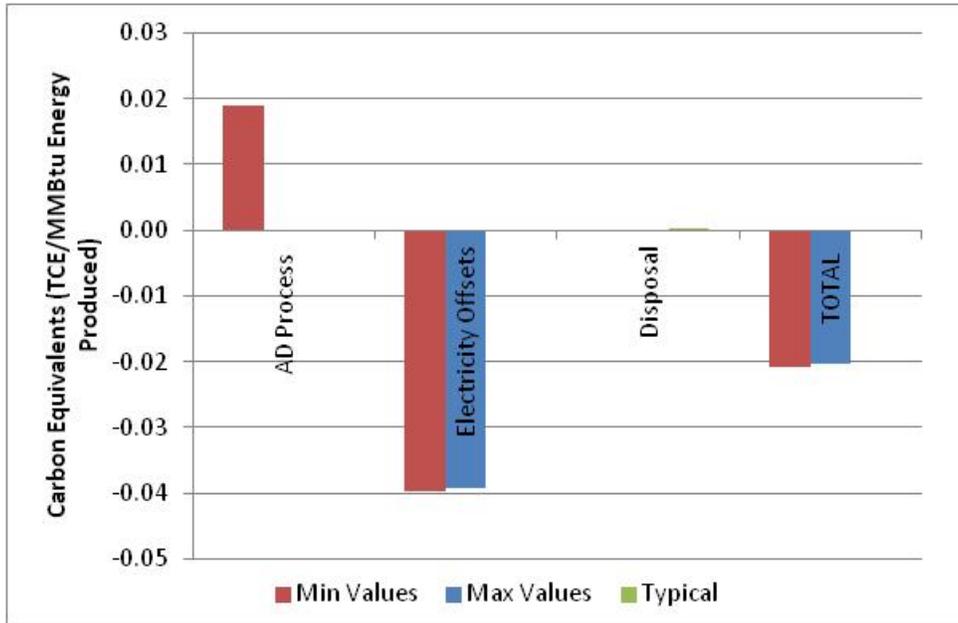


Figure 4-24. Net Carbon Equivalents Per MMBtu for AD of Organics.

**Cost**

The net (expenses–revenue) cost per ton for the AD of organics is shown in **Figures 4-25 and 4-26**. As shown in these figures, the net cost range is positive, signifying a net cost stream that results from the capital and operating costs of AD being greater than the revenues brought by electricity sale.

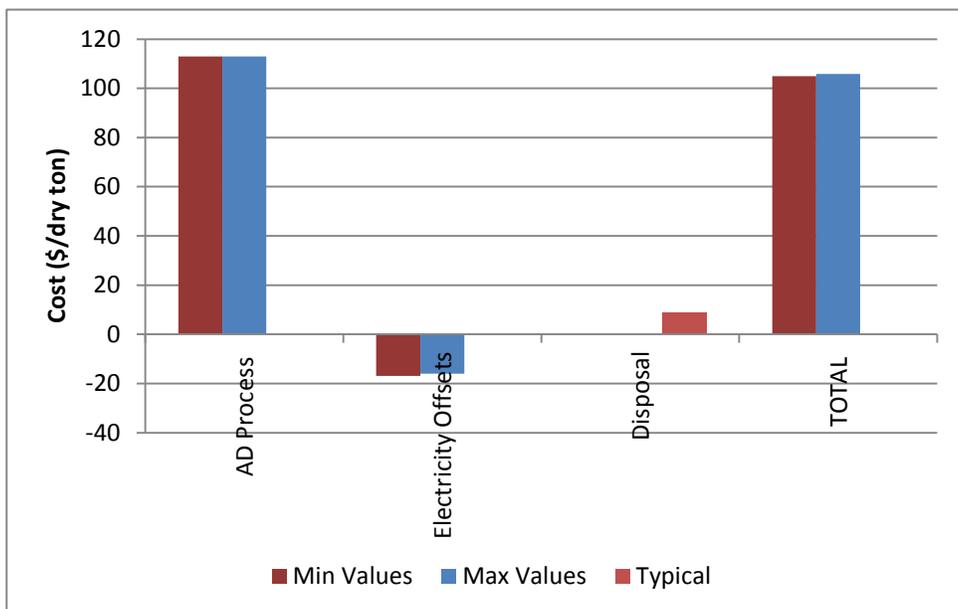
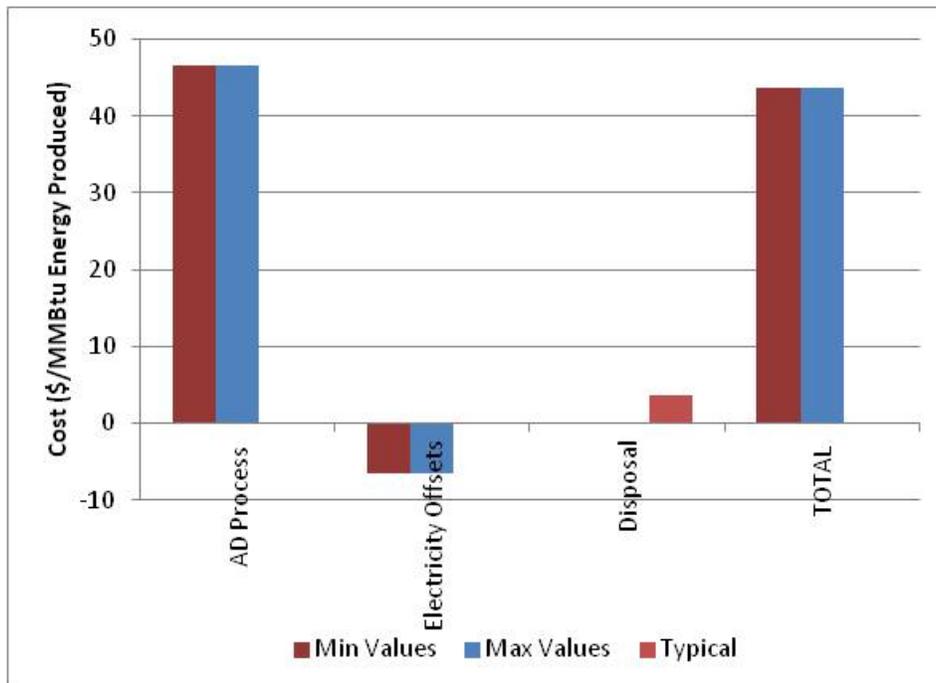


Figure 4-25. Net Cost Per Ton for AD of Organics.



**Figure 4-26. Net Cost Per MMBtu for AD of Organics.**

#### *Comparison to Landfill Base Case*

In this section, the results for the AD of organics are compared to results for a landfill base case for organics. A low–high range was developed for the landfill base case using a landfill with gas collection and flaring for the low end of the range and a landfill with gas collection and energy recovery for the high end of the range. The landfill base case was modeled using RTI’s MSW DST and is representative of a U.S. average.

**Figure 4-27** shows the results for net energy consumption (i.e., energy consumed minus energy produced). According to this figure, the net energy saved using the AD technology versus landfill disposal is approximately 0.6–2.5 MMBtu per dry ton of organics. These savings are mostly associated with the energy produced by the AD facility. For example, when compared to a landfill with energy recovery (i.e., the low landfill energy consumption bar in **Figure 4-27**), these savings indicate that the AD facility is much more efficient at producing energy than the landfill facility.

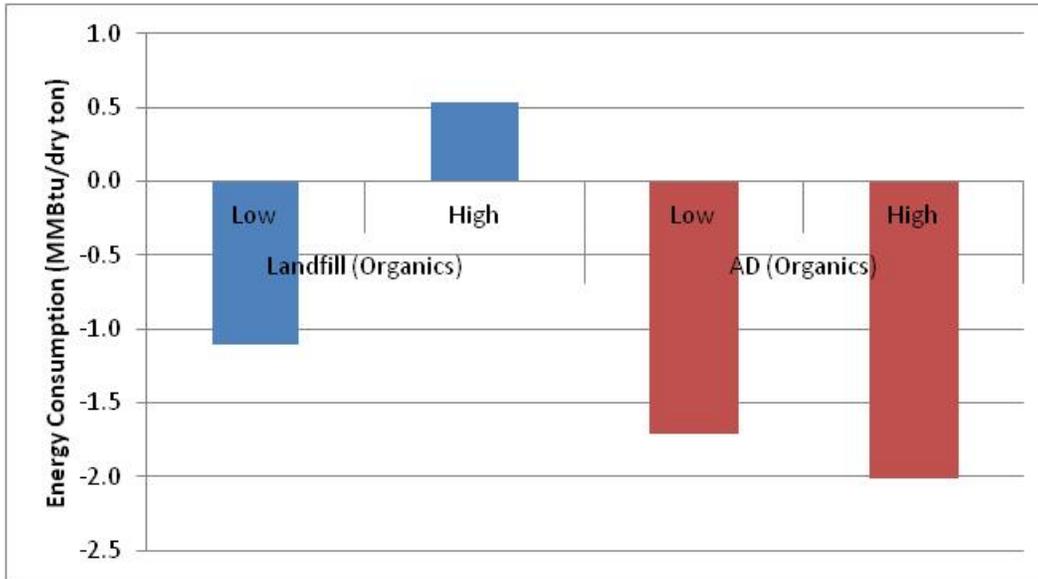


Figure 4-27. Net Energy Consumption for Landfill Base Case and AD of Organics.

Figure 4-28 shows the results for net carbon emissions (i.e., carbon emissions minus savings). According to this figure, the AD technology results in a net reduction of approximately 0.11–0.13 TCE per dry ton of organics managed via AD as compared to landfill disposal. This reduction is mostly associated with the energy produced by the AD facility. For example, Figure 4-27 indicates that the AD facility is more efficient at energy production than the landfill with energy recovery, so the emissions savings associated with energy production using waste versus virgin materials are also greater for AD facilities.

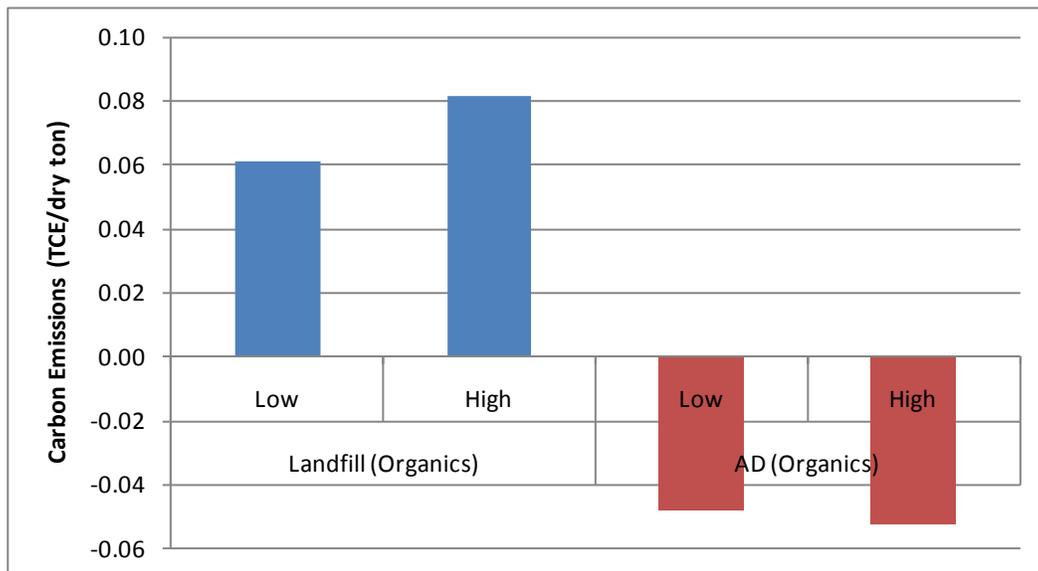
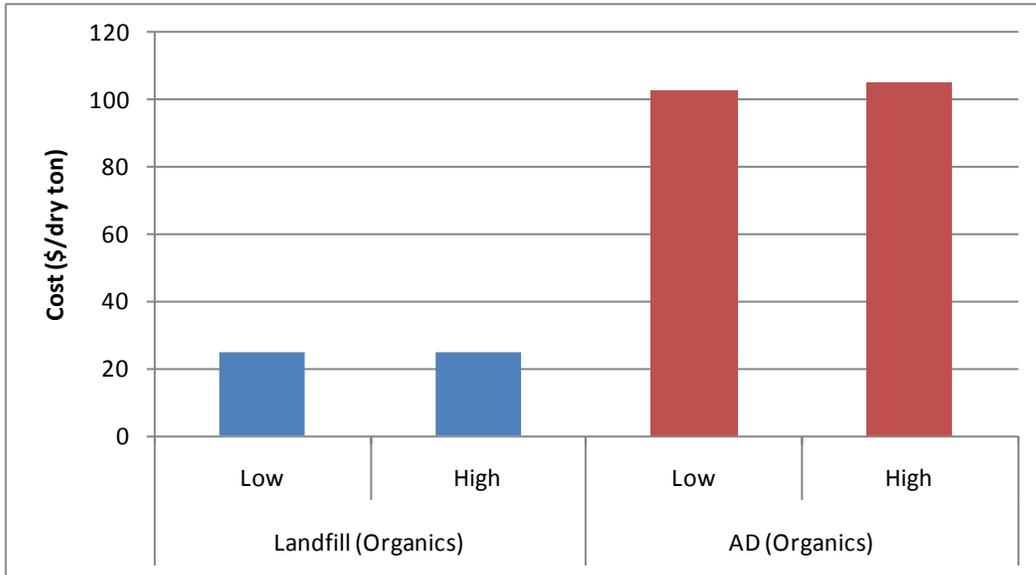


Figure 4-28. Net Carbon Equivalents for Landfill Base Case and AD of Organics.

Figure 4-29 shows the results for net costs (i.e., costs minus revenues). According to this figure, the AD technology results in a net reduction of approximately \$75 per dry ton of organics

processed when compared to landfills. Consistent with the energy and GHG emissions results, this reduction is mostly associated with the energy produced by the AD facility. For example, by having larger energy savings, as illustrated in **Figure 4-27**, the AD facility will also get more revenues from energy sales than the landfill with energy recovery.



**Figure 4-29. Net Cost for Landfill Base Case and AD of Organics.**

## Section 5: Findings and Recommendations

Based on the data collection and analyses performed in this study, the following sections are intended to provide insights on the science and status of waste conversion technologies in North America.

### 5.1 Key Findings

Waste conversion technologies present alternative options to landfill disposal for managing non-recycled MSW; however, there are currently very few commercially operating facilities in North America. At the time of this study, we estimated there were 9 pyrolysis, 7 gasification, and 10 AD demonstration and commercially operating facilities in North America<sup>7</sup> that process municipal wastes. In general, we found that plastics-to-oil pyrolysis facilities were more at the commercially operating stage in the United States, while bulk MSW (typically gasification) and organics (typically AD) technologies were still in the demonstration phase at the time of this report. Plastics-to-oil pyrolysis technologies were reported to receive plastics from both materials recycling facilities and, more so, from industrial partnerships. Due to the size and flexibility of plastics-to-oil facilities, industrial and other small-scale partnerships that divert nonrecycled plastics from landfills directly to these facilities may represent an efficient and economic opportunity. Other findings include:

- **There is a high level of uncertainty associated with the environmental and cost data associated with these technologies.** Because most conversion facilities are demonstration plants, they are operating in batch-test mode and not as a continuous-mode commercial plant. Until there are commercially operating facilities in North America, there will not be good real-world data to characterize the environmental aspects and costs for these technologies. It was found that even facilities that are commercial-scale are often operating in more of a demonstration mode and do not have waste contracts and/or energy or product contracts in place.
- **Conversion technologies do present another option (in addition to mass burn waste combustion) for managing non-recycled waste.** While conversion technologies present another option for managing post-recovered waste, it will be 5-10 years before the first-generation demonstration facilities transition to commercial operations. The capability of conversion technologies to meet landfill diversion and/or energy production goals will depend heavily on the success of these first-generation facilities.
- **It is difficult to compare the cost and performance of pyrolysis, gasification, and AD technologies directly due to differences in feedstock.** In general, pyrolysis technologies utilize only plastics, gasification technologies utilize MSW, and AD utilizes food, yard, and paper waste. Therefore, there are differences in feedstock energy value as well differences in beneficial offsets. For pyrolysis, beneficial offsets are primarily based on the conversion of plastics to oil. For gasification, beneficial offsets include energy production and can also include the recyclables (e.g., remove metals, glass, and other

---

<sup>7</sup> This includes demonstration and commercial scale facilities only. Proposed and planned facilities were not included.

inorganics) in the up-front sorting process but this component was not included in this analysis since we assumed non-recycled waste would be the input feedstock and little recyclables available. For AD, the benefit offsets are primarily based on the conversion of organic wastes to biogas which is assumed to be used to produce electrical energy.

- **Using the currently available data, a high-level life cycle environmental assessment conducted for the conversion technologies indicates that the technologies do offer environmental benefits as compared to landfill disposal.** Specifically, we estimated that gasification (excluding energy production and materials recycling offsets) of MSW saves between 6.5—13 MMBtu per ton as compared to landfill disposal. Pyrolysis of waste plastics saves between 2—4 MMBtu per ton as compared to landfill disposal. Likewise, our results show that gasification of MSW saves between 0.3—0.6 TCE emissions per ton of MSW treated as compared to landfill disposal. Pyrolysis of waste plastics saves between 0.15 and 0.25 TCE emissions per ton as compared to landfill disposal. AD of organics waste saves between 0.11—0.13 TCE per ton compared to landfill disposal. We used a combination of vendor provided and publicly available data for these estimates as described in previous sections of this report.
- **Different technology vendors/facilities have specific variations on the process to enhance conversion efficiency and/or tailor the end product to site-specific markets. The primary objective of the conversion technologies is to convert waste into useful energy products that can include syngas or biogas, petroleum, and/or commodity chemicals.** Syngas and biogas can be used directly in industrial boilers or in an ICE gen-set to produce electrical energy. Petroleum and commodity chemicals are typically tailored to specific end-users (e.g., petroleum wax for cosmetics manufacturers). Each end product has different life-cycle offsets which may affect the overall environmental impact of the process.
- **Pyrolysis for plastics is in a more mature stage of development than other technologies.** Of the current conversion technologies, plastics-to-oil pyrolysis technologies seem to be more mature than MSW-based technologies (typically gasification and AD), in part because of the decreased variability of the incoming feedstock.
- **Estimates provided by technology vendors indicate cost/ton is comparable to other MSW options, such as recycling and landfilling.** Vendors estimate that the cost to process the waste is approximately \$50/ton for pyrolysis and gasification technologies, and approximately \$100/ton for AD. This cost is generally related to the capital and operating costs required to run the process and the market price for products. U.S. averages for landfill disposal and recycling, for comparison, range from \$30-75/ton depending on region.

Given the developmental stage and the current capacities of technologies, our preliminary estimates suggest that conversion technologies would offset significantly less than 1 percent of annual North American oil consumption. For example, the average size of a plastics-to-oil facility is in the range of 10–30 tons per day. If there were 100 plastics-to-oil facilities in North America by 2015, conversion production could offset approximately 6,000–18,000 barrels of oil

per day, assuming 1 ton of plastic is equivalent to 6 barrels of oil. Total consumption of oil in North America is forecast to be 21.57 million barrels per day in 2015.<sup>8</sup> Therefore, according to these estimates, 100 commercial-scale plants would supply, at most, a tenth of 1 percent of U.S. oil consumption. While MSW-based conversion facilities are anticipated to convert 7–10 times more waste to energy, estimates still indicate significantly less than 1 percent of annual North American oil consumption.

There are a number of proposed pyrolysis, gasification, and AD facilities in North America, but the exact number is unknown (up-to-date information on proposed developments was generally difficult to ascertain). Anecdotal evidence suggests that project viability may in part be affected by difficulties encountered scaling up facilities from demonstration to commercial scale (especially MSW-based plants), financial backing/economic conditions, and the highly variable permitting classifications per local, state, and provincial level.

Ultimately, the findings from this research show that the different categories of waste conversion technologies are designed to handle two very different types of waste feedstock. Pyrolysis technologies are generally designed to use only waste plastics, whereas gasification is generally designed to use bulk MSW and AD is generally designed to use only certain organic fractions of MSW. In this regard, all technologies have their benefits (and burdens) and decisions about their adoption will likely be done on a site- or region-specific basis and depend on characteristics such as waste composition, contracts for assuring steady waste feedstock supply, market prices for electricity and fuels, availability of markets for products, and distance to those markets.

The economic sustainability of conversion facilities will depend on the markets for energy and commodity products. Each facility will likely tailor its process to match site-specific market conditions and contractual arrangements. For example, according to common behavior if the price of crude oil continues to increase, technologies that convert plastics and MSW to synthetic petroleum and/or liquid transportation fuels will be able to generate more revenue from the sale of products and become more cost competitive.

## **5.2 Limitations and Recommendations Associated with the Key Findings in This Report**

Real-world cost and environmental information for North America is difficult to obtain, due primarily to the current stage of development of conversion technologies in the United States. As more commercial-scale facilities are built and operating, it would be beneficial to reassess the cost and life cycle environmental aspects of conversion technologies as compared to competing waste management alternatives. Also, with the appropriate caveats, data from facilities outside North America, e.g., in Europe and Japan, could be considered to fill in gaps in the North America dataset or for comparison purposes with results obtained using North America data.

---

<sup>8</sup> [http://www.researchandmarkets.com/research/96a49e/united\\_states\\_oil\\_and\\_gas\\_report\\_q1\\_2011](http://www.researchandmarkets.com/research/96a49e/united_states_oil_and_gas_report_q1_2011).

Additional research that could be done in the near term to advance the understanding of conversion technologies might include examining sensitivities and “break-even” points relative to cost and environmental aspects for key parameters such as:

- Feedstock composition (e.g., high vs. low BTU value feedstock)
- Plant energy conversion efficiency
- Recovery of materials for recycling (for MSW technologies)
- Beneficial offsets for different end product alternatives
- Distance to market for liquid fuels
- Market prices for energy products
- Market prices for recyclable and other byproduct streams.

The costs considered in this study were scarce (i.e., one data point for gasification, three data points for pyrolysis, and one data point for AD) and based on facilities that are not operating at a commercial stage. Therefore, there is inherent uncertainty in these data. For example, at the waste processing cost estimated for pyrolysis of plastics (approximately \$50/ton of plastics) full-scale commercial plants should be fairly common in areas with high-cost landfill disposal.

For AD, we assumed the facilities would be stand-alone and not exceed wastewater treatment digester capacity. This assumption was made to simplify the cost and life cycle environmental assessments. However, it is expected that AD economics would be favorable for utilizing unused wastewater treatment capacity. For example, according to CalRecycle (2009), the estimates for the state of California indicate that if 75 percent of the capacity comes from existing wastewater treatment facilities and 25 percent from stand-alone facilities, the total annual costs are almost six times higher than when all the capacity comes from existing facilities,

Information about the financing mechanisms proposed or in place for the facilities identified in this report was not collected. Information such as existing or planned government subsidies and private sector off-take agreements, in addition to the net cost estimates provided in this study, would give a better picture of the financial viability of these technologies.

## Resources

- Agilyx, 2011. The Agilyx Process. Available at: <http://www.agilyx.com/our-technology/>
- ARI (Alternative Resources, Inc.), 2007. Los Angeles County Conversion Technology Evaluation Report- Phase II. Prepared for The County of Los Angeles, Department of Public Works. October.
- Beck, R.W., 2010. Independent Engineer's Report, Envion Plastic-to-Oil Technology. Prepared for Envion. June.
- CalRecycle (California Department of Resources Recycling and Recovery), 2010. Landfill-Based Anaerobic Digester-Compost Pilot Project at Yolo County Central Landfill. California Natural Resources Agency. April.
- CalRecycle. 2009. Life Cycle Assessment and Economic Analysis of Organic Waste Management and Greenhouse Gas Reduction Options. Available at <http://www.calrecycle.ca.gov/climate/Events/LifeCycle/2009/default.htm> . As obtained on March 8, 2011.
- Chornet, Esteban (2011). Telephone Conversation regarding Enerkem processes and environmental performance. January 2011.
- Enerkem, 2011. The Enerkem Process. Available at: <http://enerkem.com/en/technology-platform/process.html>
- Environment and Plastics Industry Council (EPIC) Canadian Plastics Industry Association, 2004. The Gasification of Residual Plastics Derived from Municipal Recycling Facilities. Available at: [http://www.plastics.ca/articles\\_merge/gasification\\_pf.php](http://www.plastics.ca/articles_merge/gasification_pf.php).
- Envion, 2011. Envion Process. Available at: <http://envion.com/technology/etechnology.php>
- FL DEP (Florida Department of Environmental Protection, Office of Permitting and Compliance), 2011. Emission Sources: NSR/PSD Construction Permits - St. Lucie Plasma Gasification Facility, St. Lucie County, Florida. Available at: <http://www.dep.state.fl.us/Air/emission/construction/geoplasma.htm>.
- Griffith, John (2011). Telephone Conversation regarding Climax Global Energy processes and environmental performance. January 2011.
- Gradman, Gideon (2011). Telephone Conversation regarding Ze-gen processes and environmental performance. January 2011.

- Goco, Pio (2010). Telephone Conversation regarding Envion processes and environmental performance. December 2010.
- ICF International, 2009. A Life Cycle Assessment of Plasma Conversion at Trail Road Landfill in Ottawa and Landfilling of Waste at the Halton Regional Landfill, Phase 1 Final Report. Prepared for Ontario Ministry of Environment. March.
- JBI Inc, 2011. Patent Pending Process. Available at: <http://www.plastic2oil.com/site/patent-pending>
- Lawrence, Mackinnon, 2009. Innovation, Cost Parity Offset by Public Anxiety. Biofuels Digest May 29, 2009.
- Plasco Energy Group, 2010. Semi-Annual Report. January 22 – July 31, 2010. 141 RT 3388 Rev A. Available at: [http://www.zerowasteottawa.com/docs/141-RT-3388\\_RevA\\_Plasco%20Energy%20Group%20Demonstration%20Project%20Jan%202010%20-%20July%202010%20Semi%20Annual%20Progress%20Report%20Plasco%20Trail%20Road%20FINAL.pdf](http://www.zerowasteottawa.com/docs/141-RT-3388_RevA_Plasco%20Energy%20Group%20Demonstration%20Project%20Jan%202010%20-%20July%202010%20Semi%20Annual%20Progress%20Report%20Plasco%20Trail%20Road%20FINAL.pdf).
- Plasco Energy Group, 2010. The Plasco Process. Available at: <http://www.plascoenergygroup.com/our-technology/the-plasco-process/>
- Plasco Energy Group, 2011. What Becomes of 1 Ton of Waste? February.
- The Pembina Institute, 2009. Life-cycle Environmental Performance of Waste-to-Energy Technologies: A Comparative Analysis for Plasco Energy. March. Prepared by: Matt McCulloch, P.Eng; Jeremy Moorhouse, E.I.T; Rich Wong, E.I.T.
- RTI International, 2005. Anaerobic Digestion (AD) Model Documentation Prepared for URS Corporation for Use in “The County of Los Angeles Department of Public Works and The Los Angeles County of Solid Waste Management Committee Conversion Technology Evaluation Report.”
- RTI International, 2009. Life Cycle Inventory (LCI) of MASDAR’s Potential Solid Waste Management Alternatives. Prepared for CH2 MHill. July 2009. U.S. DOE (Department of Energy), 2010. Construction and Operation of a Heterogeneous Feed Biorefinery Enerkem Corporation Pontotoc, Mississippi. Report No.: EA-1790: Final Environmental Assessment (2010/09) Available at: <http://energy.gov/nepa/downloads/ea-1790-final-environmental-assessment>.
- U.S. EIA (Energy Information Administration), 2009. Electric Power Monthly with data for August 2009. Report No.: DOE/EIA-0226 (2009/11) Available at: [http://www.eia.doe.gov/cneaf/electricity/epm/epm\\_sum.html](http://www.eia.doe.gov/cneaf/electricity/epm/epm_sum.html).

U.S. EPA (Environmental Protection Agency), 2008. Municipal Solid Waste Generation, Recycling, and Disposal in the United States: Facts and Figures for 2008. Available at: <http://www.epa.gov/waste/nonhaz/municipal/pubs/msw2008rpt.pdf>.

U.S. EPA (Environmental Protection Agency), 2011. Funding for Solid Waste Management and Recycling. Available at: <http://www.epa.gov/region7/waste/solidwaste/funding.htm>

Ulum, Chris (2011). Telephone Conversation regarding Agilyx processes and environmental performance. January 2011.

ECY WA (Washington Department of Ecology), 2011. SEPA Guide for Project Applicants. Available at: <http://www.ecy.wa.gov/programs/sea/sepa/apguide/apguide1.htm>.

Ze-gen, 2011. Our Technology: Strategy. Available at: [http://www.ze-gen.com/#tech\\_strategy](http://www.ze-gen.com/#tech_strategy)

## **Appendix A: Conversion Technology Data**

**Table A-1. Gasification Technology Data**

Parameters	Units	Information/data from Vendor			Vendor's information/data available in the Literature					
		High Temperature Gasification		High Temperature Gasification		High Temperature Gasification With Plasma Arc Syngas Cleaning				
		Energem (Pontotoc)		Ze-gen		Plasco*				
General Information										
Location	NA	Pontotoc, MS (PROPOSED)		Narragansett Bay, Massachusetts		Ottawa, Ontario, Canada (ICF, 2009)				
Years of Operation	yr	expected to open late 2011		expected to open 2012						
Process Information										
Cost per design capacity	\$/dtpd	499,109								
Design capacity	dtpd			75-150		147				
Type of Feedstock (% compositions, if available)	NA	post-MRF-sorted MSW, industrial waste, construction and demolition waste, treated wood, bagasse, corn stover, wheat straw, rice hulls, wood chips, sawdust, bark, thinning, limbs, needles		95% wood based material, consisting of railroad crossties (90%), clean wood waste (5%), non-recyclable source-separated plastics (5%)		Paper and paperboard (24.3%), Plastics (16.2%), Metals (7.2%), Glass (6.1%), Rubber & Leather (3.3%), Textiles (5.9%), Wood (7.4%), Food Scraps (18%), Yard Trimmings (7.3%), Miscellaneous Inorganic Waste (2.2%), Other (2%)				
Feedstock preprocessing requirements	NA	Yes- MSW would be sorted, shredded and dried. Shredding and sorting done at waste manager site.		Yes- MSW would be sorted, shredded and dried. Shredding and sorting by contracted processor.		Yes-The first stage in the waste conversion process is the reclamation of metals from the waste stream for recycling. The remaining material is shredded to facilitate consistent flow into the conversion chamber.				
Inorganic matter content of feedstock	<%			<5						
Moisture content of feedstock	<%	15		<20		NR				
Efficiency of the electricity generating unit (ICE)	%			85						
Energy recovery efficiency	%			48						
Process Inputs and Outputs										
Inputs	Tonnage of feedstock (actual capacity)	dtpd	3.E+02		1.E+02	[1]	1.E+02	[2]		
	Power consumption / parasitic load	KWh/dry ton	5.E+02		2.E+02					
	Other inputs (e.g., water, oxygen, etc.)	Oxygen	lb/dry ton	1.E+03						
		Diesel for preprocessing	gal/dry ton					5.E-02		
		Caustic for gas cleaning and cooling	lb/dry ton					1.E+01		
		Activated Carbon for gas cleaning and cooling	gal/dry ton					2.E-01		
		Feldspar for gas cleaning and cooling	gal/dry ton					1.E-01		
	Supplemental fuel use	Water	gal/dry ton	2.E+03		5.E+02				
		Natural Gas	lb/dry ton	2.E+01		2.E+03	[5]	9.E+01		
	Outputs**	Energy product (e.g., syngas, ethanol, hydrogen, electricity, steam)	Electricity	KWh/dry ton				9.E+02	-	1.E+03
Syngas			MMBtu/dry ton		9.E+01	[3]				
Steam			MMBtu/dry ton		1.E+01					
Material By-Products		Ethanol	lb/dry ton	560-615						
		Residual gas	lb/dry ton	4.E+02				2.6E+00	-	2.7E+00
		Sulphur	lb/dry ton					9.E+00	-	1.E+01
		Salt	lb/dry ton					2.E+01	-	4.E+02
Residuals (e.g., ash, char, slag, etc.)		Slag	lb/dry ton							
		Char	lb/dry ton	3.E+02						
		Slag	lb/dry ton			8.E+01				
		Gasifier solid residues	lb/dry ton	1.E+02		3.E+01				
		Spent catalysts and chemicals	lb/dry ton	3.E+00						
		Inorganic sludge	lb/dry ton	4.E+01						
		Non-hazardous solid waste	lb/dry ton	1.E+01						
Water losses		Potable water	gal/dry ton				5.E+02	-	3.E+03	
	gal/dry ton	1.E+03			5.E+02					
Air Emissions Data										
PM	lb/dry ton		4.E-01		1.E-02		4.2.E-02	-	4.4.E-02	
PM10	lb/dry ton						7.E-04			
Biogenic Carbon Dioxide (CO2bio)	lb/dry ton						5.E+02			
Fossil Carbon Dioxide (CO2fossil)	lb/dry ton		4.E+02		3.E+02		1.E+03			
Methane (CH4)	lb/dry ton		2.E+00				2.E-04			
HCl	lb/dry ton				8.E-03	[4]	2.E-02	-	3.E-02	
Hydrocarbons	lb/dry ton									
Sulphur dioxide (SO2)	lb/dry ton		2.E-01		4.E-01		1.E-01	-	2.E-01	
Sulphur oxide	lb/dry ton						5.E-05			

Parameters	Units	Information/data from Vendor				Vendor's information/data available in the Literature			
		High Temperature Gasification		High Temperature Gasification		High Temperature Gasification With Plasma Arc Syngas Cleaning			
		Enerkem (Pontotoc)		Ze-gen		Plasco*			
Nitrous Oxide (N2O)	lb/dry ton	4.E-01				7.E-04			
NOx expressed as NO2	lb/dry ton	1.E+00		2.E-01		2.E-01	-	2.E-01	
Carbon monoxide (CO)	lb/dry ton	1.E+00		1.E-01		4.E-01	-	4.E-01	
Mercury (Hg)	lb/dry ton			3.E-06	[4]	6.E-07			
Cadmium (Cd)	lb/dry ton			5.E-07	[4]	8.E-06			
Lead	lb/dry ton			7.E-06	[4]	1.E-05			
VOC	lb/dry ton	9.E-01		4.E-02					
HAP	lb/dry ton	1.E-01							
NH3 slip	lb/dry ton								
Dioxins and furans	lb/dry ton					0.E+00			
H2S	lb/dry ton								
NH3	lb/dry ton								
AS	lb/dry ton								
AL	lb/dry ton								
B	lb/dry ton								
BA	lb/dry ton								
BE	lb/dry ton								
CR	lb/dry ton								
CU	lb/dry ton								
FE	lb/dry ton								
MN	lb/dry ton								
NI	lb/dry ton								
SB	lb/dry ton								
SE	lb/dry ton								
SN	lb/dry ton								
ZN	lb/dry ton								
Acetaldehyde	lb/dry ton	6.E-02							
TNMOC	lb/dry ton					2.E-01	-	2.E-01	
<b>Water Emissions Data</b>									
Water Effluent	gal/dry ton	600-1400							
BOD	lb/dry ton								
COD	lb/dry ton								
PB	lb/dry ton								
H2S	lb/dry ton								
AS	lb/dry ton								
AL	lb/dry ton								
B	lb/dry ton								
BA	lb/dry ton								
BE	lb/dry ton								
CD	lb/dry ton								
CR	lb/dry ton								
CU	lb/dry ton								
FE	lb/dry ton								
HG	lb/dry ton								
MN	lb/dry ton								
NI	lb/dry ton								
SB	lb/dry ton								
SE	lb/dry ton								
SN	lb/dry ton								
ZN	lb/dry ton								

**Table A-2. Pyrolysis Technology Data**

Parameters	Units	Information/Data From Vendors				Information/Data From Literature				
		Envion	Agilyx	Climax	JBI	H. Smart	Veiba	BP		
<b>General Information</b>										
Location	NA	Derwood, MD	Tigard, OR	Fairfax, SC	Niagara Falls, NY		Bottrop, Germany			
Years of Operation	yr	1 reactor, 1.5 years			Over 1 year					
<b>Process Information</b>										
Cost per design capacity	\$/dry tons	280,699		250,000	29,350					
Design capacity	dtpd	29		20	20	53	641			
Type of Feedstock (% compositions, if available)	NA	100% plastic		100% plastic	2,4,5 for highest yield		polyolefins			
Feedstock preprocessing requirements	NA	feedstock is chipped to 1.5 inches or smaller	industry standard grinding/shredding		shred or pre-melt					
Inorganic matter content of feedstock	<%	100	100	100	<5%		4.5			
Moisture content of feedstock	<%	2		0-5	<10%					
Efficiency of the electricity generating unit (ICE)	%				N/A					
Energy recovery efficiency	%	30-80%, with 62-70% being bell curve high probability	82-85% see comment	75	92%			80		
<b>Process Inputs and Outputs</b>										
Inputs	Tonnage of feedstock (actual capacity)	dtpd	3.E+01	1.E+01		20				
	Power consumption / parasitic load	KWh/dry ton	5.E+02			3.E-01		2.E+02		
	Other inputs (e.g., water, oxygen, etc.)	Oxygen	lb/dry ton				N/A		3.E-02	
		Catalysts and chemicals	lb/dry ton				trade secret	8.E-01		
		CaO	lb/dry ton						1.E-04	
		Ammonia	lb/dry ton				0.E+00		6.E-01	
		sand	lb/dry ton				0.E+00		4.E-06	
		Hydrogen	lb/dry ton						2	
		E-Gas	lb/dry ton						22	
	Supplemental fuel use	HCl	lb/dry ton						2	
		Water	gal/dry ton	1.E+02		2.E+02	3.E+01		4.E+02	
		Natural Gas	MBTU/dry ton				3.E+01		4.E+00	
	Residuals (e.g., ash, char, slag, etc.)	Steam	MBTU/dry ton						7.E-02	
		Syngas	Syngas	MMBTu/dry ton				2.E-01		
			synthetic crude oil	MMBTu/dry ton	3.7.E+01	3.9.E+01	3.E+01			2.E+02
light fraction (liquid)			lb/dry ton			300-400			3.E-02	
gas fraction			lb/dry ton	200-500					7.E-05	
diesel			lb/dry ton				1.7.E+03	1.5.E+03		
Residuals (e.g., ash, char, slag, etc.)		Gasoline	lb/dry ton				2.3.E+01	2.0.E+01		
		CaCl2	lb/dry ton						2.E+00	
		Char	lb/dry ton		2.E+02		1.E+02			
		Solid residues	lb/dry ton	2.E+02						
		Spent catalysts and chemicals	lb/dry ton				trade secret	6.E+01		
		Inorganic sludge	lb/dry ton	3.E+02			n/a			
		Spent SCR catalyst	lb/dry ton					2.E-01		
Residuals (e.g., ash, char, slag, etc.)		Non-hazardous solid waste	lb/dry ton				5	1.E-02	2.E+01	
		Residue to incineration	lb/dry ton						2.E+00	
	waxy filter to incineration	lb/dry ton						3.E-05		
Water losses	gal/dry ton				3.E+01	4.E+02				
<b>Air Emissions Data</b>										
PM	lb/dry ton	negligible		2.E+01	4.E-02					
PM10	lb/dry ton									
Fossil Carbon Dioxide (CO2fossil)	lb/dry ton		1.E+03	5.E+02	0.E+00					
CO2	lb/dry ton	7.38-18.45			0.E+00	9.E+02				
Methane (CH4)	lb/dry ton	26-65			0.E+00					
HCl	lb/dry ton				3.E-04					

Parameters	Units	Information/Data From Vendors				Information/Data From Literature		
		Envion	Agilyx	Climax	JB I	H. Smart	Veba	BP
Hydrocarbons	lb/dry ton			8.E+00	1.E-02		4.E+00	
Sulphur dioxide (SO2)	lb/dry ton				0.E+00			1.E-06
Sulphur oxide	lb/dry ton			TBD (minimal)	0.E+00			
Nitrous Oxide (N2O)	lb/dry ton			TBD (minimal)	2.E+00			
NOx expressed as NO2	lb/dry ton	36.2-90.5	2.E+00	TBD (minimal)	3.E-01	2.E-01		2.E-07
Carbon monoxide (CO)	lb/dry ton	3.58-8.95	1.E+00	TBD (minimal)	0.E+00	6.E-01		
Mercury (Hg)	lb/dry ton				0.E+00			
Cadmium (Cd)	lb/dry ton							
Lead	lb/dry ton	2.E-04			2.E-02			
VOC	lb/dry ton	Negligible	2.E+00	TBD (minimal)	3.E-04	2.E-01		
HAP	lb/dry ton							
NH3 slip	lb/dry ton				0.E+00			
Dioxins and furans	lb/dry ton				0.E+00			
H2S	lb/dry ton				0.E+00			
NH3	lb/dry ton				0.E+00		1.E-02	
AS	lb/dry ton				0.E+00			
AL	lb/dry ton				0.E+00			
B	lb/dry ton				0.E+00			
BA	lb/dry ton				0.E+00			
BE	lb/dry ton				0.E+00			
CR	lb/dry ton				0.E+00			
CU	lb/dry ton				0.E+00			
FE	lb/dry ton				0.E+00			
MN	lb/dry ton				0.E+00			
NI	lb/dry ton				0.E+00			
SB	lb/dry ton				0.E+00			
SE	lb/dry ton				0.E+00			
SN	lb/dry ton				0.E+00			
ZN	lb/dry ton				0.E+00			
Acetaldehyde	lb/dry ton				0.E+00			
TNMOC	lb/dry ton				0.E+00			
<b>Water Emissions Data</b>								
Water Effluent	gal/dry ton							
BOD	lb/dry ton							
COD	lb/dry ton							
PB	lb/dry ton							
H2S	lb/dry ton							
AS	lb/dry ton							
AL	lb/dry ton							
B	lb/dry ton							
BA	lb/dry ton							
BE	lb/dry ton							
CD	lb/dry ton							
CR	lb/dry ton							
CU	lb/dry ton							
FE	lb/dry ton							
HG	lb/dry ton							
MN	lb/dry ton							
NI	lb/dry ton							
SB	lb/dry ton							
SE	lb/dry ton							
SN	lb/dry ton							
ZN	lb/dry ton							

## Appendix B: List of Facilities Considered in this Study and Their Status (as of Early 2011)

Process Type	Vendor	Status	Feedstock	Location	Main Product	Source
Anaerobic digestion	City of Riverside	Commercial	grease from restaurants	City of Riverside	Biogas	<a href="http://www.riversideca.gov/sewer/project-grease.asp">http://www.riversideca.gov/sewer/project-grease.asp</a>
Anaerobic digestion	Quasar Energy Group	Commercial	MSW, crop waste, grass, and manure	Wooster, OH	Electricity	<a href="http://www.schmackbioenergy.com/pages/wooster.html">http://www.schmackbioenergy.com/pages/wooster.html</a>
Anaerobic digestion	Quasar Energy Group	Commissioned	MSW	Columbus, OH	Electricity	<a href="http://www.schmackbioenergy.com/pages/columbus.html">http://www.schmackbioenergy.com/pages/columbus.html</a>
Anaerobic digestion	Central Marin Sanitation	Commissioned	MSW	City of San Rafael –	Biogas	<a href="http://www.cityofsanrafael.org/Assets/Methane+Gas+Study.pdf">http://www.cityofsanrafael.org/Assets/Methane+Gas+Study.pdf</a>
Anaerobic digestion	Humboldt Co. Waste Authority	Commissioned	grease from restaurants	Humboldt Co. Waste Authority	Biogas	<a href="http://www.hwma.net/HRFWDFS.pdf">http://www.hwma.net/HRFWDFS.pdf</a>
Anaerobic digestion	Terrabon	Demo	MSW, sewage sludge, forest/ag residues	Bryan, TX	Gasoline	<a href="http://www.terrabon.com/mixalco_semiworksplant.php">http://www.terrabon.com/mixalco_semiworksplant.php</a>
Anaerobic digestion	B & D Geerts	Demo	Food waste, green material, and mixed solid waste	Yolo County, CA	Biogas	<a href="http://www.epa.gov/region9/organics/symposium/2010/Pors9-15-10.pdf">http://www.epa.gov/region9/organics/symposium/2010/Pors9-15-10.pdf</a>

Process Type	Vendor	Status	Feedstock	Location	Main Product	Source
Anaerobic digestion	East Bay Municipal Utility District	Demo	MSW	East Bay Municipal Utility District	Biogas	<a href="http://www.epa.gov/region9/organics/symposium/2010/Pors9-15-10.pdf">http://www.epa.gov/region9/organics/symposium/2010/Pors9-15-10.pdf</a>
Anaerobic digestion	Sacramento Co. Regional WWTP	Demo	MSW	Elk Grove, CA	Biogas	<a href="http://www.biomassmagazine.com/articles/3376/building-on-its--biomass-base/">http://www.biomassmagazine.com/articles/3376/building-on-its--biomass-base/</a>
Anaerobic digestion	Arrow Ecological	Permitted	MSW	Perris, CA	Electricity	<a href="http://dpw.lacounty.gov/prg/pressroom/printview.aspx?ID=370&amp;newstype=PRESS">http://dpw.lacounty.gov/prg/pressroom/printview.aspx?ID=370&amp;newstype=PRESS</a>
concentrated acid hydrolysis	Blue Fire Ethanol	Permitted	Wood chips, forest residuals, urban wood waste	Fulton, Mississippi	Ethanol	<a href="http://www.eere.energy.gov/golden/PDFs/ReadingRoom/NEPA/1%20BlueFire%20DOE%20Final%20EA%206-4-10.pdf">http://www.eere.energy.gov/golden/PDFs/ReadingRoom/NEPA/1%20BlueFire%20DOE%20Final%20EA%206-4-10.pdf</a>
Gasification	Nexterra	Commercial	Wood residues	Heffley Creek, BC	Syngas	<a href="http://www.nexterra.ca/PDF/Project_Profile_Tolko_20100118.pdf">http://www.nexterra.ca/PDF/Project_Profile_Tolko_20100118.pdf</a>
Gasification	RangeFuels	Commercial	Non-food biomass, such as woody biomass and grasses	Soperton, GA	Syngas	<a href="http://www.rangefuels.com/range-fuels-produces-cellulosic-methanol-from-first-commercial-cellulosic-biofuels-plant.html">http://www.rangefuels.com/range-fuels-produces-cellulosic-methanol-from-first-commercial-cellulosic-biofuels-plant.html</a>
Gasification	Cirque Energy LLC	Commissioned	Wood chips	Midland MI; Dow Corning	Syngas	<a href="http://www.dowcorning.com/content/news/midland_biomass_plant_dow_corning.aspx?bhcp=1">http://www.dowcorning.com/content/news/midland_biomass_plant_dow_corning.aspx?bhcp=1</a>
Gasification	Alter NRG	Commissioned	MSW	Milwaukee, WI	Syngas	<a href="http://www.wisbusiness.com/index.iml?Article=209527">http://www.wisbusiness.com/index.iml?Article=209527</a>

Process Type	Vendor	Status	Feedstock	Location	Main Product	Source
Gasification	Taylor Biomass	Commissioned	paper, fiber, food residuals, leather, some textiles and wood products from MSW	Montgomery, NY	Syngas	<a href="http://www.taylorbiomassenergy.com/taylorbiomass04_mont_mn.html">http://www.taylorbiomassenergy.com/taylorbiomass04_mont_mn.html</a>
Gasification	Primenergy	Commissioned	Carpet Residues	Dalton, Georgia	Syngas	<a href="http://www.shawfloors.com/pv_obj_cache/pv_obj_id_D731E1B2C5B45ECE98332FBBC1FB427478F32A00/filename/EnviroBrochure2009.pdf">http://www.shawfloors.com/pv_obj_cache/pv_obj_id_D731E1B2C5B45ECE98332FBBC1FB427478F32A00/filename/EnviroBrochure2009.pdf</a>
Gasification	Westing-house/ Coronal	Commissioned	MSW, woody waste and WWT biosolids	International Falls, Minnesota	Syngas	<a href="http://alternrg.com/press_release_94443">http://alternrg.com/press_release_94443</a>
Gasification	Westing-house/ Coskata	Demo	Non-food based feedstocks, forest/ag waste and construction waste	Madison, Pennsylvania	Cellulosic Ethanol	<a href="http://www.westinghouse-plasma.com/technology/demonstration-facility">http://www.westinghouse-plasma.com/technology/demonstration-facility</a>
Gasification	Energkem	Demo	MSW, wood chips, treated wood, sludge, petcoke, spent plastics, wheat straw	Sherbrooke, Quebec, Canada	syngas, methanol, acetates, second generation ethanol	<a href="http://www.energkem.com/en/our-locations/overview.html">http://www.energkem.com/en/our-locations/overview.html</a>
Gasification	Westing-house/Coskata	Demo	Building waste, forest waste	Warrenville, IL		<a href="http://www.coskata.com/facilities/?source=C3C8A85B-7736-4D64-87F8-9E4FBD45D1B0">http://www.coskata.com/facilities/?source=C3C8A85B-7736-4D64-87F8-9E4FBD45D1B0</a>
Gasification	InEnTech, LLC	Demo	MSW	Richland, WA	Syngas	<a href="http://www.inentec.com/pem-facilities.html">http://www.inentec.com/pem-facilities.html</a>

Process Type	Vendor	Status	Feedstock	Location	Main Product	Source
Gasification	Plasco Energy	Demo	MSW	Ottawa, Canada		<a href="http://www.bioenergyproducers.org/documents/ucr_emissions_report.pdf">http://www.bioenergyproducers.org/documents/ucr_emissions_report.pdf</a>
Gasification	Ze-gen	Demo	MSW- wood wastes, non-recyclable plastics, carpet, and glycol (anti-freeze)	New Bedford, MA	Syngas	<a href="http://attleboroproject.com/ga.html">http://attleboroproject.com/ga.html</a>
Gasification	Enerkem	Demo	MSW/used electricity and telephone poles	Westbury, Quebec, Canada	Syngas	<a href="http://www.rdno.ca/services/swr/docs/swmpr/waste_to_energy.pdf">http://www.rdno.ca/services/swr/docs/swmpr/waste_to_energy.pdf</a> <a href="http://www.bioenergyproducers.org/documents/ucr_emissions_report.pdf">http://www.bioenergyproducers.org/documents/ucr_emissions_report.pdf</a> <a href="http://www.gemcanadawaste.com/52901.html?*session*id*key*=session*id*val*">http://www.gemcanadawaste.com/52901.html?*session*id*key*=session*id*val*</a>
Gasification	Fulcrum, InEnTech, LLC	Demo	Post-recycled MSW	Pleasanton, California	Ethanol	<a href="http://fulcrum-bioenergy.com/documents/IEReport-032610-FINAL.pdf">http://fulcrum-bioenergy.com/documents/IEReport-032610-FINAL.pdf</a>
Gasification	Ineos	Demo	Pre-processed MSW	Fayetteville, Arkansas	Ethanol	<a href="http://www.dep.state.fl.us/Air/emission/bioenergy/indian_river/INEOS_technical_evaluation.pdf">http://www.dep.state.fl.us/Air/emission/bioenergy/indian_river/INEOS_technical_evaluation.pdf</a>
Gasification	Nexterra	Demo	Sawmill residues	Vancouver, BC	Syngas	<a href="http://www.unbc.ca/releases/2010/11_25biomass_ignition.html">http://www.unbc.ca/releases/2010/11_25biomass_ignition.html</a>
Gasification	InEnTech /WM	Permitted	MSW	Columbia Ridge, OR	Syngas	<a href="http://www.inentec.com/images/stories/documents/PressReleases/releases--s4%20or%20announcement-march-2010_final.pdf">http://www.inentec.com/images/stories/documents/PressReleases/releases--s4%20or%20announcement-march-2010_final.pdf</a>
Gasification	Entech Renewable Energy	Permitted	MSW	Huntington Beach, CA	Syngas	<a href="http://www.socalconversion.org/news.html">http://www.socalconversion.org/news.html</a>
Gasification	Ineos	Permitted	Yard, wood, agricultural and vegetative wastes	Vero Beach, FL	Ethanol	<a href="http://www.ineosbio.com/76-Press_releases-13.htm">http://www.ineosbio.com/76-Press_releases-13.htm</a>

Process Type	Vendor	Status	Feedstock	Location	Main Product	Source
Gasification	Fulcrum, InEnTech, LLC	Permitted	MSW	Reno, Nevada	Ethanol	<a href="http://fulcrum-bioenergy.com/biofuel-plants.html">http://fulcrum-bioenergy.com/biofuel-plants.html</a>
Gasification	Enerkem	Permitted	MSW	Edmonton, Canada	Ethanol (as well as Syngas, Methanol, Acetates)	<a href="http://www.enerkem.com/en/our-locations/plants/edmonton-alberta.html">http://www.enerkem.com/en/our-locations/plants/edmonton-alberta.html</a>
Gasification	Enerkem	Permitted	MSW	Pontotoc, MS	Ethanol (as well as Syngas, Methanol, Acetates)	<a href="http://www.enerkem.com/en/our-locations/plants/pontotoc-mississippi.html">http://www.enerkem.com/en/our-locations/plants/pontotoc-mississippi.html</a>
Pyrolysis	Agilyx	Commercial	Plastics	Tigard, OR	Synthetic crude oil	<a href="http://www.sustainablebusinessoregon.com/articles/2010/06/plas2fuel_opens_showcase_facility_changes_name_to_agilyx.html">http://www.sustainablebusinessoregon.com/articles/2010/06/plas2fuel_opens_showcase_facility_changes_name_to_agilyx.html</a>
Pyrolysis	Intrinergy Coshocton, LLC	Commercial	Blends of crumb rubber, shredded carpet fluff, wood chips, and biomass	Coshocton, OH	Synthetic Oil	<a href="http://www.rdno.ca/services/swr/docs/swmpr/waste_to_energy.pdf">http://www.rdno.ca/services/swr/docs/swmpr/waste_to_energy.pdf</a>
Pyrolysis	JB1	Commercial	Plastics	Niagara Falls, NY	Synthetic Oil	-
Pyrolysis	Envion	Demo	PET, HDPE, LDPE/LLDPE, PP, PE, PS and PVC (less than 10%)	Derwood, MD	Synthetic Oil	<a href="http://inhabitat.com/new-envion-facility-turns-plastic-waste-into-10barrel-fuel/">http://inhabitat.com/new-envion-facility-turns-plastic-waste-into-10barrel-fuel/</a>
Pyrolysis	Climax Global Energy	Demo	Plastics	Fairfax, SC	Synthetic Oil	<a href="http://blog.cleantech.com/sector-insights/waste/on-stage-in-new-york-climax-global-energy/">http://blog.cleantech.com/sector-insights/waste/on-stage-in-new-york-climax-global-energy/</a>

Process Type	Vendor	Status	Feedstock	Location	Main Product	Source
Pyrolysis	International Environmental Solutions	Demo	MSW	Romoland, California	Syngas	<a href="http://www.rdno.ca/services/swr/docs/swmpr/waste_to_energy.pdf">http://www.rdno.ca/services/swr/docs/swmpr/waste_to_energy.pdf</a> <a href="http://www.bioenergyproducers.org/documents/ucr_emissions_report.pdf">http://www.bioenergyproducers.org/documents/ucr_emissions_report.pdf</a>
Pyrolysis	Vadxx	Demo	Plastics, synthetic fibers, used industrial solvents, waste oils	Akron, OH	Synthetic oil, natural gas	<a href="http://www.wksu.org/news/story/26888">http://www.wksu.org/news/story/26888</a>
Pyrolysis	Agriplas	Demo	Agricultural film, mixed nursery and jug material, food containers, and other low- or zero-value plastics	Kelso, WA	Synthetic Oil	<a href="http://www.green-energy-news.com/nwslnks/clips309/mar09019.html">http://www.green-energy-news.com/nwslnks/clips309/mar09019.html</a>
Pyrolysis	Green Power Inc	Demo	Plastics	Pasco, WA	Synthetic oil	<a href="http://www.cleanenergyprojects.com/Summary.html">http://www.cleanenergyprojects.com/Summary.html</a>
Pyrolysis	International Environmental Solutions	Permitted	MSW	Riverside, CA	Syngas	<a href="http://dpw.lacounty.gov/prg/pressroom/printview.aspx?ID=370&amp;newstype=PRESS">http://dpw.lacounty.gov/prg/pressroom/printview.aspx?ID=370&amp;newstype=PRESS</a>
Pyrolysis	Oneida Tribe	Permitted	MSW	Green Bay, WI	Syngas	<a href="http://www.greenbaypressgazette.com/article/20101102/GPG0101/11020584/Oneida-Seven-Generation-gasification-project-begins">http://www.greenbaypressgazette.com/article/20101102/GPG0101/11020584/Oneida-Seven-Generation-gasification-project-begins</a>

## **Appendix C: Data Questionnaires**

## Emerging Technologies Questionnaire

Please fill out the following questionnaire to the extent possible. Please feel free to add additional information or skip areas which may not be known. Any information which can contribute to a life cycle analysis of the process is most appreciated.

The information requested here will be used for life cycle analysis purposes and will not be provided in its entirety in the final, publicly available report without review and consent of the contributing organization (questionnaire submitter). Furthermore, the contributing organization will not be named without review and consent of the questionnaire submitter.

Please feel free to contact Jennifer Richkus at [jrichkus@rti.org](mailto:jrichkus@rti.org), or 202-974-7831, with any questions or concerns.

General Data	Value	Units
Location		
Years of Operation		
Cost per design capacity		\$/TPD
Design capacity		TPD

General Data	Specific Characteristic	Value	Units
Type of Feedstock (% compositions, if available)			%
			%
			%
Energy product (e.g., syngas, ethanol, electricity, steam)			e.g., TPD , ton/ton, million LPD
If producing electricity, mode of production (i.e., turbine, ICE gen set)			
Regional Electricity Grid Mix (for offsetting purposes, if applicable)			%
			%

Process Data	Value	Units
Preprocessing of feedstock? (e.g., Yes/No, Explain)		
Tonnage of feedstock		e.g., TPD , ton/ton
Inorganic matter content of feedstock		%
Moisture content of feedstock		%
BTU value of liquid/gas fuel produced		e.g., btu/gal, btu/ft3
Efficiency of the electricity generating unit (if applicable)		%
Water requirements of process (e.g., losses due to evaporation)		million LPD
Power consumption / parasitic load		kW PD
Temperature of reactor and/or thermal steps		C or F
Supplemental fuel use (e.g., Natural gas/diesel)		e.g., KWH, BTUs
Energy recovery efficiency		%

Process Data	Specific Characteristic	Value	Units
Other Inputs (e.g., water, oxygen, catalysts, etc.)			e.g., TPD , ton/ton, million LPD
Material By-Products (e.g., water, hydrogen, CO <sub>2</sub> )			e.g., TPD , ton/ton, million LPD
Residuals (e.g., ash, char, slag, residual catalysts, etc.)			e.g., TPD , ton/ton, million LPD

Air Emissions Data	Value	Units (e.g. TPY, g/ton, g TEQ/ton)
PM		
Carbon Dioxide (CO <sub>2</sub> )		
Biogenic Carbon Dioxide (CO <sub>2</sub> bio)		
Fossil Carbon Dioxide (CO <sub>2</sub> fossil)		
Methane (CH <sub>4</sub> )		
HCl		
Hydrocarbons		
Sulfur dioxide (SO <sub>2</sub> )		
Nitrous Oxide (N <sub>2</sub> O)		
NO <sub>x</sub>		

Air Emissions Data	Value		Units (e.g. TPY, g/ton, g TEQ/ton)
VOCs			
Carbon monoxide (CO)			
Mercury (Hg)			
Cadmium (Cd)			
Lead			
Dioxins and furans			
Process Data	Specific Characteristic	Value	Units
Additional Compounds Measured (e.g., Metals, Ammonia, Hydrogen Sulfide)			

Water Emissions Data	Value		Units (e.g. TPY, g/ton, ton/ton)
Water Effluent			
BOD			
COD			
Process Data	Specific Characteristic	Value	Units
Additional Compounds Measured (e.g., Metals, Ammonia, Hydrogen Sulfide)			